

1 **Apatite-glaucony association in the Ediacaran Doushantuo Formation, South China**
2 **and implications for marine redox conditions**

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15 **Abstract**

16 Major phosphogenic events took place in the late Paleoproterozoic and the Ediacaran,
17 both of which followed an episode of atmospheric oxygenation. These apparent
18 coincidences have led to suggestions that the Precambrian phosphorus cycle was partly
19 regulated by oceanic redox conditions, although direct evidence is lacking. In this paper,
20 we report the discovery of glaucony in drill cores from the phosphorite-rich Ediacaran
21 Doushantuo Formation at Weng'an, South China, which provides new insights into redox
22 conditions and geochemical cycles during the deposition and early diagenesis of the lower
23 Doushantuo Formation. The presence of flaser and lenticular bedding in the sandstones
24 suggests deposition in a tidally-influenced shallow water environment. Apatite is present
25 as millimeter-scale flasers in sandstones, but more commonly as intraclastic grains.
26 Glaucony typically occurs as interstitial aggregates cementing and replacing detrital
27 minerals and phosphatic grains, with an upsection decrease in abundance. The Doushantuo
28 glauconies have consistently high K₂O (7.6 – 9.6 wt%). In particular, phosphorite-hosted

29 glaucony has generally higher total iron (TFe_2O_3) than sandstone-hosted glaucony,
30 consistent with Al^{3+} - Fe^{3+} substitution in the octahedral site, as supported by Raman peaks
31 slightly shifted towards lower wavenumbers for higher Fe content. A decrease in detrital
32 input probably resulted in limited Al availability during early diagenesis, which in turn led
33 to the waning of glaucony upsection. The phosphate grains do not show any negative Ce
34 anomalies, consistent with precipitation of authigenic apatite in suboxic environments with
35 insignificant preferential remobilisation of Ce during deposition and early diagenesis. REE
36 + Y features show variable diagenetic REE enrichment in phosphorites, but identical
37 concave-down patterns. The wide occurrence of glaucony cement further documents the
38 presence of a generally suboxic early diagenetic environment on the seafloor, which was
39 in active exchange with bottom seawater. Slight fluctuations of redox states in the
40 porewater and bottom water would have promoted the cycling of Fe, enrichment of
41 phosphate and REE, and extensive glaucony authigenesis. Therefore, the glaucony-bearing
42 phosphorites in the Doushantuo Formation are a direct result of Fe-redox driven diagenesis.
43 Apatite-glaucony association likely forms in suboxic environments and is considered as a
44 useful mineralogical proxy for paleoredox conditions on the Precambrian ocean floors.

45 Keywords: Mineralogy; diagenesis; rare earth element; phosphogenesis; glauconitization

46

47 **1. Introduction**

48 The distribution of major phosphate deposits in the Precambrian sedimentary record
49 is restricted to the Paleoproterozoic era and the Ediacaran period, both of which witnessed
50 similar types of global biogeochemical changes (Papineau, 2010). These fundamental
51 changes have generally been related to oxygenation of the Earth's atmosphere and oceans
52 (Och and Shields-Zhou, 2012; Pufahl and Hiatt, 2012; Lyons et al., 2014). Recent studies
53 have shown that Precambrian phosphorites mainly occur in peritidal environments, in
54 contrast to Phanerozoic phosphorites that accumulated across the entire shelf (Nelson et
55 al., 2010). These observations have led to the suggestion that Precambrian phosphogenesis
56 is closely related to redox variations of paleo-seawater that regulated the cycling of
57 phosphorus through the adsorption and reductive desorption of phosphate on Fe-

58 oxyhydroxides (Nelson et al., 2010; Pufahl and Hiatt, 2012; Hiatt et al., 2015; She et al.,
59 2014; Cui et al., 2016; Zhang et al., 2019).

60 The Doushantuo Formation (ca. 635-551 Ma) in South China is an important site for
61 constraining the Ediacaran ocean redox conditions and biological evolution (Xiao et
62 al., 1998; Bailey et al., 2007; McFadden et al., 2008; Li et al., 2010; Shen et al., 2010; Sahoo
63 et al., 2012). Redox sensitive trace element (RSTE) and Fe speciation data suggest that
64 Ediacaran seawater along the South China Craton margin featured redox stratification, with
65 an oxic surface ocean that was underlain by a thin ferruginous layer, a metastable wedge
66 of euxinic water resting on the open shelf, and a deep ferruginous ocean (e.g., Li et al.,
67 2010; Wang et al., 2012; Muscente et al., 2015; Fan et al., 2018). More recently, studies
68 have shown that Ediacaran marine redox chemistry was highly heterogeneous, even at the
69 kilometer-scale (e.g., Li et al., 2015). While Fe speciation and RSTEs data can provide
70 relatively accurate information on marine bottom-water redox conditions suitable for
71 siliclastic rocks (Tribovillard et al., 2006; Lyons and Severmann, 2006), most of these
72 proxies have not been applied to phosphorites. The Doushantuo Formation at the shallow
73 water Weng'an section is largely composed of carbonate and phosphorite and the redox
74 conditions during phosphogenesis remain unclear.

75 Fe-bearing authigenic minerals have the potential to record the chemistry and redox
76 conditions of the pore water and probably of the bottom water. Glauconite [(K, Na, Ca)
77 (Fe, Al, Mg, Mn)₂ (Si, Al)₄ O₁₀ (OH)₂] is a K- and Fe-rich phyllosilicate mineral of
78 dioctahedral mica group with 2:1 + interlayer ion structures, while the term glaucony is
79 used to represent a series of green clay minerals with a wide range of
80 chemical/mineralogical compositions including glauconitic mica, glauconitic smectite, and
81 ferric illite (Odin and Létolle, 1980; Odin and Matter, 1981; Meunier and El Albani, 2007;
82 Banerjee et al., 2015, 2016b). Glaucony is known to occur in marine sedimentary rocks
83 ranging in age from Precambrian to Holocene, where it commonly occurs in green clay
84 aggregates usually 60 -1000 μm in diameter (Odin and Matter, 1981; Dasgupta et al., 1990;
85 Chafetz and Reid, 2000; Lee et al., 2002; El Albani et al., 2005; Meunier and El Albani,
86 2007; Amorosi et al., 2007; Banerjee et al., 2015; 2016b). Although glauconitic minerals
87 are generally considered to be an effective indicator of low sedimentation rate in marine

88 environments and is commonly associated with transgression (Odin and Matter, 1981;
89 Amorosi, 1997; Harris and Whiting, 2000; Giresse and Wiewio, 2001; Chatteraj et al.,
90 2009; Banerjee et al., 2008; 2015; 2016b), they have also been reported from regressive
91 sediments with a normal sedimentation rate (Chafetz and Reid, 2000). In the modern ocean,
92 glaucony forms on the continental shelves at a depth range between 50 and 500 m and is
93 most abundant between the upper slope and the outer shelf at depths around 200-300 m
94 (Odin and Matter, 1981). Likewise, Phanerozoic glaucony is interpreted to have formed in
95 middle to outer shelf depositional conditions (Banerjee et al., 2016b). Precambrian
96 glaucony, however, is thought to have occurred in shallow marine depositional conditions,
97 mostly within the shoreface region (Banerjee et al., 2016b), and is rarely preserved
98 probably due to its alteration or metamorphic recrystallization into other phyllosilicates.

99 Apatite-glaucony associations have been observed in modern offshore environments
100 associated with significant upwelling, which also cause phosphatization, such as off the
101 coast of Peru, Chile, California, South Africa, Namibia, Spain, northwest Africa, east
102 Australia, and west India (Burnett, 1980; Mullins and Rasch, 1985; Odin and Lamboy,
103 1988; Rao et al., 1993; Wigley and Compton, 2007; Banerjee et al., 2016a; Föllmi, 2016)
104 and in Phanerozoic successions as old as the early Cambrian (Braisler, 1980). In this paper,
105 we report on the discovery of apatite-glaucony association in the lower Doushantuo
106 Formation at Weng'an, South China. Drill core samples from three boreholes near
107 Weng'an are studied and complemented by analyses of the sedimentology, mineralogy,
108 and geochemistry of glaucony and apatite to provide new insights into the redox conditions
109 during the deposition and early diagenesis of the Doushantuo Formation.

110 **2. Geological setting**

111 The Doushantuo Formation is a carbonate-black shale sequence that is widespread in
112 South China (Jiang et al., 2011). It is economically important because of the discovery of
113 major phosphorite deposits that account for 51% of the total phosphorus reserves in China
114 according to an early estimate (Fig.1a; Ye et al., 1989). According to an earlier
115 paleogeographic reconstruction, the Doushantuo Formation in the Yangtze platform was
116 deposited on a rimmed carbonate shelf with Weng'an located on the shelf margin (Fig.1a;
117 Jiang et al., 2011). In the Weng'an and nearby Kaiyang areas, however, the common

118 occurrence of terrigenous clasts, cross stratification and wave ripples in the Doushantuo
119 Formation is more consistent with deposition in a more proximal, mixed clastic-chemical
120 costal environment (Chen et al., 2015; Zhang et al., 2019). The current study is focused on
121 the Doushantuo Formation in drillholes close to the northern margin of the Baiyan anticline
122 (Fig. 1b). In this area, the Neoproterozoic Qingshuijiang Formation occurs along the axis
123 of the anticlines while the Cambrian System is exposed on the limbs, with the Ediacaran
124 Doushantuo and Dengying Formations sandwiched in between.

125 The Doushantuo Formation was first typified in the Yangtze Gorges area where it
126 overlies Cryogenian diamictite of the Nantuo Formation and underlies the upper Ediacaran
127 Dengying Formation (Xiao et al., 2014). Lithostratigraphically, the Doushantuo Formation
128 is subdivided into four members, including the cap dolostone unit (Member I) at the base,
129 followed upsection by interbedded dolostone and black shale (Member II), carbonate
130 (Member III), and black shale (Member IV) at the top (e.g., Zhou et al., 2007). At Weng'an,
131 the stratigraphy of the Doushantuo Formation is broadly correlated with its counterpart in
132 the Yangtze Gorges area according to the underlying Cryogenian diamictite and the
133 overlying dolostones of the Dengying Formation, but a one-to-one correlation to the four
134 members is not straightforward. Instead, the Doushantuo Formation is subdivided into five
135 lithological units (Fig. 1c). It consists of, in ascending order, a basal cap dolostone unit
136 (Unit 1), a thin-bedded, peloidal phosphorite unit (Unit 2, the 'lower phosphorite'), a
137 massive dolostone unit (Unit 3) with a prominent karstification surface on top, an
138 intraclastic phosphorite unit (Unit 4, the 'upper phosphorite'), which is further divided into
139 the lower Unit 4A (black facies) and the upper Unit 4B (grey facies), and a phosphatic
140 dolostone unit (Unit 5) probably containing an additional exposure surface (Xiao et al.,
141 2014).

142 The depositional age of Doushantuo Formation in the Yangtze Gorges area has been
143 constrained as 635-551 Ma (Condon et al., 2005). For the Weng'an counterpart, previous
144 studies have yielded Pb-Pb isochron ages of 572 ± 36 Ma for Unit 4A (Chen et al., 2009),
145 599 ± 4 Ma for Unit 4B (Barfod et al., 2002) and 576 ± 14 Ma (Chen et al., 2004). Recently,
146 a SIMS zircon U-Pb age of 609 ± 5 Ma was reported from a tuff layer which lies
147 immediately above the upper phosphorite unit (correlated with unit 4 at Weng'an) in

148 Yichang, 600 km to the northeast of Weng'an (Zhou et al., 2017). The new age data
149 suggests that the major phosphorite beds were likely deposited between 635 and 609 Ma.

150 In this study, stratigraphic sections of three boreholes (Zk1202, Zk208, Zk115; GPS
151 coordinates: 27°4'29.2"N / 107°24'19.8"E, 27°04'56.7"N / 107°24'48.9"E, 27°04'45.4"N /
152 107°24'24.5"E) in the Weng'an area are studied (Fig. 1b). In these drill cores, the lithology
153 of Units 1 to 5 vary (Fig. 1c), and their thicknesses are 1.4–3.8 m, 18.5–24.3 m, 0.6–4.4 m,
154 0.7–5.0 m and ~3.83 m, respectively. Moreover, the basal unit is dominated by glaucony-
155 bearing siliciclastic sediments rather than dolostone. Such a basal siliciclastic unit is
156 considered synchronous with widespread cap carbonate deposition (Zhang et al., 2019).
157 The glaucony-bearing sandstones are generally characterized by sandstone layers
158 intercalated with flasers of mudrock or phosphorite (Fig. 2a), suggesting deposition in a
159 tidal-influenced environment. In Zk1202 and Zk208, glaucony is also present in the lower
160 phosphorite unit (Unit 2), although it shows a generally decreasing trend in abundance (Fig.
161 1c). The base of the Doushantuo Formation features glauconitic sandstones, consistent with
162 deposition in coastal environments as a result of marine transgression following the
163 Marinoan glaciation (Zhang et al., 2019). Upsection, the transition to bedded phosphorite
164 and dolostone in Units 2-5 points to a decrease of terrigenous input and an increase of
165 authigenic phosphate and carbonate, consistent with a transgression into a deeper
166 environment. We focus on the glaucony-bearing Units 1 and 2, which are termed as the
167 lower Doushantuo Formation in this study.

168 **3. Material and methods**

169 **3.1. Samples**

170 Ninety-nine samples were collected from the three drill cores (Zk1202, Zk208 and
171 Zk115) in Weng'an (Fig. 1b). The samples were cut into chips and only the fresh parts are
172 used for thin section and mineralogical/geochemical analyses. Prepared thin sections and
173 rock slabs were further polished with 500 nm Al₂O₃ and cleaned with deionized water prior
174 to analyses with light microscopy, scanning electron microscopy, Raman micro-
175 spectroscopy, energy dispersive X-ray spectrometry and laser ablation inductively coupled
176 plasma mass spectrometry. For x-ray diffractometry and electron probe microanalysis, a

177 selection of samples was cleaned, dried and powdered (~200 mesh) in an agate mortar that
178 was carefully washed and dried before the grinding. Glaucony-rich grains were separated
179 from the samples using a combination of magnetic, electrostatic and handpicking
180 techniques. Care was taken to avoid cross-contamination in each step.

181 **3.2. Optical microscopy**

182 Petrographic studies were carried out on polished thin sections and rock slabs using a
183 Zeiss Axio Scope.A1 microscope attached to an Axio MRe5 camera, at School of Earth
184 Sciences, China University of Geosciences (Wuhan). Mineralogy and microfabrics were
185 documented with 5×, 10×, 20×, 50× and 100× objectives under the transmitted light and
186 reflected light modes, with targets of interests mapped for further *in situ* microanalysis.

187 **3.3. Scanning electron microscopy and energy dispersive X-ray spectrometry**

188 Scanning electron microscopy (SEM) was carried out on polished thin sections using
189 a Zeiss Supra 55 field- emission SEM (FE-SEM) under 20 kV accelerating voltage, at the
190 State Key Laboratory of Biogeology and Environmental Geology, China University of
191 Geosciences (Beijing). A backscattered electron (BSE) detector was used to characterize
192 compositional difference. Samples were coated with ~10-nm-thick platinum for electric
193 conduction prior to analysis.

194 Quantitative element analysis of micron-sized spots was conducted using an Oxford
195 energy dispersive X-ray spectrometer (EDS) attached to the Zeiss Supra 55 FE-SEM,
196 operated at an acceleration voltage of 20 kV with a working distance of 15 mm, specimen
197 current of 200 nA, and a beam diameter of ~1 μm. Minerals as well as synthetic phases
198 (MINM25-53) were used as standards. Duplicate analyses of individual points show
199 analytical error less than 2%.

200 **3.4. Raman micro-spectroscopy**

201 Raman micro-spectroscopy was conducted at the State Key Laboratory of Biogeology
202 and Environmental Geology, China University of Geosciences (Wuhan) using a WITec α
203 300 Confocal Raman system coupled with a Peltier cooled EMCCD detector. Laser

204 excitation at 532 nm with output power between 3 to 10 mW was used. Spectra were
205 obtained using a $100\times$ (N.A. = 0.9) magnification objectives with a 50 μm diameter optic
206 fiber. A grating of 600 g/mm was used, yielding a spectral resolution of $\sim 4\text{ cm}^{-1}$.
207 Considering the weak Raman scattered signal of glaucony, presumably due to the fine
208 crystallinity, we set the integration time for each spectrum from 30 to 80 s, and the number
209 of accumulations from 2 to 5, in order to achieve a sufficiently high signal-to-noise ratio.
210 The Raman spectra were collected at confocal depths of at least one μm below the thin
211 section surface. The WITec Project Five 5.1 Plus software was used to process for all
212 Raman spectra. The spectra were processed with a background subtraction polynomial fit,
213 typically of the order of 4-6.

214 **3.5. Electron probe microanalysis**

215 Glaucony separates were mounted in epoxy resin, polished and coated with a thin
216 carbon layer with a thickness of ca. 200 \AA . Major element analyses were performed with a
217 Cameca SX100 electron microprobe equipped with five spectrometers and “PeakSight”
218 operation software at the Institute of Mineralogy, Leibniz University Hannover, Germany.
219 The major elements (including total Fe expressed as TFe_2O_3) were measured using
220 calibration standards of synthetic oxides (Fe_2O_3 , Al_2O_3 , Cr_2O_3 , Mn_3O_4 , MgO and TiO_2),
221 natural wollastonite (for Si and Ca), orthoclase (for K), and jadeite (for Na). The
222 measurements were conducted with an accelerating voltage of 15 kV, a beam current of 15
223 nA and a defocused beam (diameter 5 μm). The quantification of all major elements was
224 based on $\text{K}\alpha$ intensities, and raw data were corrected using the standard PAP procedure
225 ([Pouchou and Pichoir, 1991](#)).

226 **3.6. X-ray diffractometry**

227 The mineralogical composition of a glaucony-enriched separate (ZK1202-04) was
228 determined by the X-ray diffraction (XRD) technique using a Dandong TD-3500 advance
229 diffractometer equipped with nickel filtered $\text{CuK}\alpha$ radiation (35 kV, 25 mA) at State Key
230 Laboratory of Biogeology and Environmental Geology, China University of Geosciences
231 (Wuhan). The air-dried, powdered sample was scanned from 3° to 65° with a step size of
232 $0.02^\circ 2\theta$ and a scan speed of $0.25^\circ/\text{min}$.

233 3.7. Laser ablation inductively coupled plasma mass spectrometry

234 Major and trace element concentrations in apatite grains were measured by laser
235 ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) at Wuhan Sample
236 Solution Analytical Technology Co., Ltd. Prior to analysis, target spots on apatite grains
237 devoid of veins and secondary cement were selected and mapped in polished slabs (with a
238 last 0.5 μm alumina polish step) under the reflected light. In total, 68 spots were analyzed
239 *in situ* on 17 polished slabs. Laser sampling was performed using a GeolasPro laser ablation
240 system that consists of a MicroLas optical system and a COMPexPro 102 ArF excimer
241 laser (wavelength 193 nm). A spot size of 32 μm was used for all analyses, with laser
242 repetition rate of 5 Hz and energy density of $\sim 5.5 \text{ j/cm}^2$. Ion-signal intensities were acquired
243 with an Agilent 7700e ICP-MS. Helium was used as a carrier gas and Argon was used as
244 the make-up gas and mixed with the carrier gas via a T-connector before entering the ICP.
245 Each analysis incorporated a background acquisition of approximately 20-30 s (gas blank)
246 followed by 50 s of data acquisition from the sample. Element contents were calibrated
247 against multiple-reference materials (BCR-2G, BIR-1G and BHVO-2G) without applying
248 internal standardization (Liu et al., 2008). An Excel-based software ICPMSDataCal was
249 used to perform off-line selection and integration of background and analyzed signals,
250 time-drift correction and quantitative calibration for major and trace element analyses (Liu
251 et al., 2010). The analytical precision (RSD) is better than 10% for major and trace elements,
252 whereas the accuracy is better than 10% except for Pr, Gd, Ho, and Tm (better than 15 %).

253 4. Results

254 4.1. Petrography and occurrences of phosphate and glaucony

255 A compilation of phosphorite and glaucony occurrences in the Doushantuo Formation
256 shows that phosphorite deposits are mainly distributed in coastal environments and
257 shallower parts of the shelf, whereas glaucony is restricted to coastal environments (Table
258 1, Fig. 1a). Apatite-glaucony association is only reported from central Guizhou, including
259 Weng'an and Kaiyang.

260 At Weng'an, the lower Doushantuo Formation mainly consists of glaucony-bearing
261 sandstone in the lower part and granular phosphorite in the upper part (Fig. 1c, Fig. 2). The
262 former may contain a dolostone interbed. The lower Doushantuo rocks contain variable
263 amounts of terrigenous grains, e.g., detrital quartz, feldspar and muscovite, and intraclastic
264 grains (Fig. 2c-n). In the sandstones, dark brown, millimeter- to submillimeter- scale
265 phosphatic flasers were observed and can be locally abundant (Fig. 2a, c-d). They exhibit
266 an anastomosing microstructure and are composed of organic-rich, cryptocrystalline
267 apatite mixed with micrometer-sized detrital particles and sulfides (Fig. 2d). The detrital
268 minerals commonly occur as angular to subangular grains that are 100 to 300 μm in
269 diameter. The intraclasts are dominated by phosphate (francolite) and glauconitized
270 phosphate (Gl-Ap) grains, which account for 20 to 30% of the total clasts in glauconitic
271 sandstones and up to 100% in granular phosphorites. They occur as elongate to elliptical,
272 angular to sub-rounded grains ranging from 200 to 750 μm in diameter (Fig. 2k-n). The
273 grains contain variable amounts of glaucony and apatite, giving rise to their greenish to
274 brownish colors (Fig. 2d-e, k-n). Other than that, in Gl-Ap grains, glaucony is mainly
275 present as interstitial cement in the sandstones and phosphorites (Fig. 2d-g, i-k). Two types
276 of glaucony have been distinguished according to their host lithology.

277 *Type 1 Sandstone-hosted glaucony.* It mainly occurs as interstitial authigenic cement
278 in sandstones (Fig. 2d-g, i-k; 3a-b). Glaucony accounts for up to 30% by volume. Partial
279 replacement of detrital quartz (Fig. 2f; 3a), K-feldspar (Fig. 2h; 3b) and muscovite (Fig.
280 2i-j) by glaucony was observed, suggesting that these detrital grains acted as the original
281 substrate for the growth of glaucony. In the BSE images, gray, porous aggregates of
282 glaucony are distinguishable from dark gray, smooth quartz grains (Fig. 3a-b). EDS
283 analyses confirm that these porous aggregates are glaucony that contains K, Si, Al, Mg and
284 Fe (Fig. 3g). They also document the presence of K-feldspar and that the interstitial cement
285 is dominated by glaucony that contains Si, K, Al, Mg and small amounts of Fe (Fig. 3g).

286 *Type 2 Phosphorite-hosted glaucony.* This type occurs as bluish to greenish patches in
287 Ap-Gl grains (Fig. 2l-n; 3e) or as interstitial cement (Fig. 2k-l; 3c, e). The grains typically
288 consist of a mixture of glaucony, apatite and sulfide (Fig. 2l-n), occasionally with barite
289 (Fig. 3f). They have distinct physical boundaries with the surrounding phosphatic grains.

290 In the glaucony-hosting phosphorite, glaucony accounts for about 15% of the samples by
291 volume. High magnification secondary electron images (Fig. 3d) further show a distinct
292 lamellar and platy habit, typical of phyllosilicate minerals, including glaucony. EDS spot
293 analyses confirm the presence of K, Si, Al, Mg and Fe in type 2 glaucony (Fig. 3g). EDS
294 elemental maps show the coexistence of Si, Al, K and Fe in the glaucony, roughly
295 complementary to the distribution of P (Fig. 3h). These observations demonstrate the close
296 association of apatite and glaucony at the micrometer scale.

297 Sulfides, mostly pyrite, are also common in the samples. They occur as anhedral to
298 euhedral grains with variable sizes of 10 – 200 μm (Fig. 2d-e, g, k-n; 3f). In particular,
299 micrometer-sized sulfide crystals are ubiquitously present with high abundance in the
300 phosphate-glaucony grains and glaucony cement (Fig. 3l-n), making it impossible to obtain
301 pure glaucony separate with standard separation techniques. Sulfides are relatively more
302 abundant in the phosphorites than in the sandstones.

303 **4.2. Mineralogical characterization by XRD**

304 The X-ray diffraction spectrum of the glauconitic sandstone shows that the bulk
305 sample is composed of glauconite, illite, quartz and dolomite (Fig. 4). The sample exhibits
306 characteristic peaks of glauconite characterized by basal reflection (001) at 10.20 \AA , and
307 two reflections (020) and (005) at 4.51 \AA and 1.97 \AA , respectively (Fig. 4). In addition, the
308 presence of basal reflections (11 $\bar{2}$) at ca. 3.70 \AA and (112) at 3.07 \AA further distinguishes
309 glaucony from illite. Instead, the reflections ($\bar{1}$ 13) at 2.89 \AA , the (032) at 2.79 \AA , the ($\bar{2}$ 01)
310 at 2.57 \AA , and the (060) at 1.50 \AA d-spacing confirm the presence of illite. This suggest the
311 presence of illite as a mixture with glaucony, which is difficult to physically separate. The
312 sample also exhibits quartz peaks of intense reflection (101) at 3.35 \AA and evident
313 reflections (100) at 4.25 \AA , (003) at 1.81 \AA and (211) at 1.54 \AA . The presence of dolomite
314 is also shown by the strong reflection (113) at 2.19 \AA , and weak reflections (021), (202)
315 and (116) at d-spacings between 2.06 and 1.78 \AA (Fig. 4).

316 4.3. Raman spectra of glaucony and associated minerals

317 The Raman spectra of glaucony (Fig. 5) show characteristic bands in the range 100 –
318 800 cm^{-1} , and less well-defined peaks and broad bands in the range of $\text{H}_2\text{O}/\text{OH}$ vibrational
319 modes (3000 – 3700 cm^{-1}). Strong bands at 263 cm^{-1} and 701 cm^{-1} are observed for
320 glaucony associated with phosphorite (type 2), and bands at 267 cm^{-1} and 705 cm^{-1}
321 observed for glaucony associated with sandstone (type 1). Moreover, two weak, broad
322 bands at 189 cm^{-1} and 443 cm^{-1} are present in all glaucony-bearing samples. The
323 coexistence of the apatite peak (964 cm^{-1}) and glaucony peaks in the apatite target attests
324 to the close spatial association of these two phases. The presence of a less well-defined
325 peak near 3615 cm^{-1} in type 1 (sandstone-hosted) glaucony documents the presence of
326 hydroxyl group. In type 2 glaucony, however, only a few broad, weak bands occur in the
327 3200 – 3700 cm^{-1} region, which likely indicates the presence of structural H_2O in the
328 mineral (Wang et al., 2015). In type 2 (phosphorite-hosted) glaucony and apatite grains,
329 weak bands for organic matter are also observed at ca. 1355 cm^{-1} (disordered carbon) and
330 1592 cm^{-1} (graphitic carbon).

331 It has been shown that strong bands in the range of 600 – 750 cm^{-1} characterize the
332 Si–O–Si vibrations of phyllosilicates, whereas the exact peak positions are related to
333 properties of cations (Mg, Fe, Ca, Al, etc.) that occupy adjacent octahedral sites (Wang et
334 al., 2015). The systematic downshifts of main Raman peaks (267 to 263 cm^{-1} , 705 to 701
335 cm^{-1} , Fig. 5) from type 1 glaucony to type 2 glaucony probably reflect compositional
336 variability caused by substitution of cations (e.g., Fe^{3+} for Al^{3+}) at the octahedral site. It is
337 also noted that no discernable peak at 540 – 550 cm^{-1} , which is commonly present in Raman
338 spectra of Phanerozoic glauconite (Ospitali et al., 2008; Wang et al., 2015), is observed for
339 the Doushantuo glaucony.

340 4.4. Major element composition and structural formula of glaucony

341 Major element composition of the two types for glaucony (sandstone-hosted, ZK1202-
342 04, 05) and (phosphorite-hosted, ZK1202-10) determined by EPMA and EDS analyses are
343 given in Tables 2 and 3 (55 analyses in total). The measured total Fe is expressed as TFe_2O_3

344 for the calculation of structural formula (Banerjee et al., 2015; Tang et al., 2017). The
345 calculated structural composition of two types of glaucony in the Doushantuo Formation
346 are also presented in Tables 2 and 3. All elemental determinations were made on an anion
347 equivalent basis to the structural formulae per $O_{10}(OH)_2$.

348 The major element compositions of the Doushantuo glaucony are comparable to those
349 of Phanerozoic glauconies (Fig. 6a). However, the K_2O content in the Doushantuo
350 glaucony are higher, falling in a narrow range between 7.6 to 9.6 wt%, which is
351 independent of the type of glaucony. The TFe_2O_3 content of the glaucony types varies
352 considerably in the range of 1.1 to 8.1 wt% with the lowest value in type 1 (sandstone-
353 hosted) glaucony (avg. 4.0 wt%) and the highest in type 2 (phosphorite-hosted) glaucony
354 (avg. 6.8 wt%). However, there is no correlation between K_2O and TFe_2O_3 , because the
355 two types of glaucony have distinct TFe_2O_3 contents (Fig. 6a). The K_2O vs. TFe_2O_3
356 relationship of the studied glaucony is similar to those reported from other Precambrian
357 sedimentary successions (Fig. 6a; Tang et al, 2017; Banerjee et al., 2008, 2015; Dasgupta
358 et al., 1990; Deb and Fukuoka, 1998; Drits et al., 2010; Ivanovskaya et al., 2006; Guimaraes
359 et al., 2000; Sarkar et al., 2014). The Al_2O_3 content of the glaucony varies from 14.0 to
360 29.5 wt% and is higher in type 1 glaucony (avg. 23.3 wt%) than in type 2 (avg. 18.1 wt%).
361 The good negative correlation between Al^{3+} (octahedral) and Fe^{3+} (octahedral) ($r^2 = 0.88$,
362 Fig. 6b) indicates predominant $Al^{3+} - Fe^{3+}$ substitution in the octahedral site (Odin and
363 Matter, 1981; Dasgupta et al., 1990; Banerjee et al., 2008, 2012a, 2012b). The MgO content
364 increases from 2.7 – 5.4 wt% for type 1 to 5.6 – 9.0 wt% for type 2 (Tables 2-3). The SiO_2
365 content shows small variations among the two types of glaucony, with average SiO_2
366 increases from (avg. 56.7 wt%) for type 1 to (avg. 57.4 wt%) for type 2. Na_2O , CaO , MnO ,
367 Cr_2O_3 and P_2O_5 concentrations of the Doushantuo glaucony are negligible.

368 According to the Association Internationale Pour l'Étude des Argiles (AIPEA),
369 glauconite is defined as a dioctahedral mica with tetrahedral Al^{3+} (or Fe^{3+}) usually greater
370 than 0.2 atoms per formula unit and octahedral R^{3+} greater than 1.2 atoms (Bailey, 1980).
371 In the octahedral site, Al^{3+} is the major cation, varying between 0.85 to 1.63 atoms per
372 formula unit (avg. 1.32), whereas the total Fe^{3+} is comparatively low, varying between 0.03
373 and 0.23 (average 0.13) atoms per formula unit. The octahedral R^{3+} of the Doushantuo

374 glaucony varies from 1.08 and 1.66 atoms, with an average of 1.45 atoms per formula unit,
375 Mg^{2+} has an average of 0.57 atoms per formula unit (Tables 2-3), higher than most of the
376 reported values (Jarrar et al., 2000; Banerjee et al., 2015; Tang et al., 2017). The average
377 Si^{4+} and Al^{3+} contents in the tetrahedral sites are 3.76 and 0.33 atoms per formula unit,
378 respectively, indicating low Al^{3+} - Si^{4+} substitution at the tetrahedral sites in the Doushantuo
379 glaucony. Meunier and El Albani (2007) proposed a cross plot of $M^+ / 4Si$ (M^+ =
380 interlayered cations) vs. (Fe octahedral) / Sum Oct. (sum of octahedral charge) to
381 characterize the compositional ranges of minerals forming the solid solution series of
382 glauconite, Fe-illite and Fe-Al smectite. Most of the analyzed Doushantuo glaucony falls
383 in the field of glaucony and Fe-illite (Fig. 6c).

384 4.5. Rare earth element composition of phosphorite

385 REE+Y concentrations of 68 spots analyzed on 17 granular phosphorite slabs (3-4
386 spots for each slab) are presented in Table 4. The average composition of each sample is
387 also calculated after data screening using the criteria of $Al_2O_3 < 0.5\%$ and $P_2O_5 > 30\%$. *In*
388 *situ* analysis with LA-ICP-MS allows site-specific measurements on REE-rich apatite
389 granules and specifically avoids other REE-bearing phases such as carbonate and silica
390 minerals. Patterns normalized to Post-Archean Australian Shale (PAAS; McLennan, 1989)
391 are presented for the average composition of each sample (Fig. 7a). Because relative
392 enrichment of La is common in many samples, the conventional calculation ($Ce/Ce^* = 2 \times$
393 $Ce_{(SN)} / [La_{(SN)} + Pr_{(SN)}]$, where $_{(SN)}$ refers to normalization of concentrations against PAAS)
394 would result in apparent negative or false Ce anomalies (Bau and Dulski, 1996; Lawrence
395 et al., 2006). Therefore, we calculate Ce anomalies using the equation $Ce/Ce^* = Ce_{(SN)} /$
396 $[Pr_{(SN)}^2 / Nd_{(SN)}]$ as suggested by Lawrence et al. (2006). Other parameters, including Y/Ho,
397 $Eu/Eu^* [= Eu_{(SN)} / (0.66Sm_{(SN)} + 0.33Tb_{(SN)})]$ and $Y/Y^* [= 2Y_{(SN)} / (Dy_{(SN)} + Ho_{(SN)})]$ are
398 also given (Table 4).

399 The Doushantuo phosphorites at Weng'an have variable REE concentration ranging
400 from 198 to 846 ppm with an average of 388 ppm ($n = 17$), and can be categorized into
401 two groups (I, II). PAAS-normalized patterns for all samples, however, are highly
402 consistent, as characterized by a hat-shape pattern with elevated MREE, relatively depleted

403 LREE and HREE, and relative enrichment of Y (Fig. 7a). The Y/Ho ratios range from
404 36.37 to 41.67, with an average of 38.64. The Eu/Eu* values range from 0.83 to 1.21,
405 without an evident Eu anomaly. All samples lack a negative Ce anomaly (Fig. 7a), with
406 Ce/Ce* values varying between 0.98 and 1.12 (Table 4). There is a clear decrease in Σ REE
407 (total REE concentration) upsection from > 500 ppm for group I (samples 10-14) to < 350
408 ppm for group II (samples 15-19 and 23-29), whereas no clear stratigraphic trend for
409 Ce/Ce* is observed although slight variations is present (Fig. 7b).

410 **5. Discussion**

411 **5.1. Sedimentary and diagenetic evolution**

412 Based on the above petrographic observations, a reconstruction can be attempted of
413 the sedimentary and diagenetic evolution of the glaucony-bearing sandstone and
414 phosphorites in the lower Doushantuo Formation. The following stages are hereby
415 proposed:

416 1) Formation of authigenic (pristine) apatite. These are represented by the phosphatic
417 flasers in the glauconitized sandstones (Fig. 2a, c-d). They show anastomosing fabrics,
418 resembling microbial mats deposited *in situ* on the seafloor. The common presence of
419 phosphatic intraclasts in the sandstone implies the wide presence of these phosphatic
420 flasers, which must have been largely reworked later and deposited along with terrigenous
421 detritus. In the Doushantuo Formation at Kaiyang, 50 km to the west of Weng'an,
422 laminated phosphatic marl has also been observed and interpreted as a pristine phosphorite
423 facies where apatite precipitated authigenically in sediments (Zhang et al., 2019).

424 2) Deposition of terrigenous clasts and phosphorite intraclasts. During the initial stage
425 of the lower Doushantuo Formation, relatively high rates of terrigenous input lead to the
426 deposition of sandstones, with subordinate reworked phosphatic intraclasts. With the
427 transgression, terrigenous input decreases whereas phosphatic intraclasts gradually
428 dominate the grains, resulting in the formation of granular phosphorites.

429 3) Authigenesis and early cementation and replacement. Penecontemporaneous
430 cementation of the clasts by authigenic glaucony took place near the sediment/water

431 interface where both ferric and ferrous iron were available. An earlier generation of sulfides
432 likely resulted from microbial sulfate reduction around organic debris in the microbial mat.
433 Meanwhile, detrital quartz, feldspar, muscovite grains and phosphate intraclasts were
434 partly replaced by glaucony.

435 4) Diagenetic transformation and replacement. These include recrystallization and
436 overgrowth of sulfides, transformation of glaucony to other clay minerals (e.g., illite), and
437 later replacement of phosphatic grains and glaucony by dolomite.

438 **5.2. Origin and chemical evolution of the Doushantuo glaucony**

439 [Odin and Matter \(1981\)](#) categorized glaucony into four types based on the K₂O
440 content: (1) the nascent (2 – 4 wt% K₂O), (2) the slightly evolved (4 – 6 wt% K₂O), (3) the
441 evolved (6 – 8 wt% K₂O), and (4) the highly evolved (> 8 wt% K₂O). The Doushantuo
442 glaucony is characterized by high K₂O (mostly > 8 wt%), consistent with the highly
443 evolved type. This is supported by the poorly developed (11 $\bar{2}$) reflection and the absence
444 of (112) reflection in the XRD spectrum which are similar to evolved to highly evolved
445 glaucony containing ca. 10% expandable layers ([Odin and Matter, 1981](#)). The lamellar and
446 platy habit observed by SEM (Fig. 3d) further corroborates the ‘highly evolved nature’
447 ([Odin and Matter 1981](#); [Wigley and Compton 2007](#)).

448 Three hypotheses have been proposed to explain the chemical evolution of authigenic
449 glaucony (Fig. 6a): 1) The ‘layer lattice theory’ ([Burst, 1958](#); [Hower, 1961](#)), which invokes
450 the simultaneous incorporation of K and Fe into degraded layer silicates for the formation
451 of glaucony; 2) The ‘verdissement theory’ ([Odin and Matter, 1981](#)), which considers initial
452 precipitation of glauconitic smectite within micropores of the substrate followed by
453 subsequent maturation by the incorporation of K into the glaucony structure without the
454 increase of TFe₂O₃; and 3) The ‘pseudomorphic replacement theory’ ([Dasgupta et al., 1990](#);
455 [Banerjee et al., 2008](#); [Banerjee et al., 2016b](#)), which emphasizes the supply of K and silica
456 (Si) into porewater through dissolution of K-feldspars and quartz in substrates that
457 facilitates glaucony authigenesis. The Doushantuo glaucony shows systematically high
458 K₂O and variable TFe₂O₃, consistent with the trend predicted by the ‘pseudomorphic
459 replacement theory’, like most Precambrian glaucony (Fig. 6a). This is supported by

460 petrographic observations that show partial replacement of K-feldspar, quartz and
461 muscovite grains by glaucony (Fig. 2d, f-h; 3a-b).

462 It should be noted, however, the Doushantuo glaucony primarily occurs as interstitial
463 cement, and thus more likely resulted from *in situ* authigenesis in the pore space.
464 Dissolution of feldspar and mica during diagenesis was a major source K source for
465 glauconitization in the Doushantuo Formation, whereas the dissolution of quartz grains
466 likely supplied Si in the pore water environment. The high alkalinity of the Precambrian
467 seawater and shallow porewater probably favored these processes (Banerjee et al., 2015).
468 The SiO₂ content (avg. 57.6 wt%) in the Doushantuo glaucony is also comparable to that
469 of Precambrian glaucony pellets (Ivanovskaya et al., 2006; Banerjee et al., 2008, 2015;
470 Tang et al., 2017; Guimaraes al., 2000) but is higher than that of Phanerozoic evolved
471 glaucony pellets (45.4 to 52.6 wt%, Odin and Matter, 1981), consistent with high Si
472 concentrations in Precambrian oceans.

473 It is noted that type 2 (phosphorite-hosted) glaucony has systematically higher TFe₂O₃
474 content than type 1 (sandstone-hosted) (Fig. 6a), which likely results from the substitution
475 of Al³⁺ by Fe³⁺ at the octahedral site as suggested in Fig. 6b. These kinds of cation
476 substitution are common in minerals, leading to changes in crystal structure and small shifts
477 of Raman peaks (e.g., Wang et al., 2001; Kuebler et al., 2006; Rividi et al., 2010).
478 Specifically, following increase of the Fe content in solid solution micas, the Raman peaks
479 show a systematic position shift towards lower wavenumber, which is attributed to
480 substitution of Al³⁺ by Fe³⁺ in the octahedral site that affects the (Si₂O₅)²⁻ vibrational modes
481 (Wang et al., 2015). The systematic downshifts of main Raman peaks from type 1 to type
482 2 glauconies (Fig. 5) thus confirm that Al³⁺ is substituted by Fe³⁺ along with the progressive
483 incorporation of Fe during the growth and maturation of the Doushantuo glaucony. The
484 upsection decrease in glaucony abundance and its final disappearance is interpreted as a
485 result of a decrease in detrital input and Al availability since the formation of glauconite is
486 ultimately dependent on the available Al ion amounts in the system (Meunier and Albani,
487 2007).

488 Potential Fe sources for glaucony include iron-bearing freshwater (Odin and Matter,
489 1981; Kozłowska and Maliszewska, 2015), hydrothermal fluids, Fe-rich detrital minerals

490 (e.g., iron oxides and biotite), or Fe-rich seawater. The upsection increase of TFe_2O_3 in
491 glaucony from type 1 (sandstone-hosted) to type 2 (phosphorite-hosted) excludes Fe-rich
492 detrital minerals as a main Fe source, because type 1 is associated with abundant
493 terrigenous clasts whereas type 2 mainly occur with apatite. Considering the widespread
494 ferruginous conditions in deep water environment during the Ediacaran (e.g., [Li et al., 2010;](#)
495 [2015](#)), seawater was likely a main source of Fe^{2+} for the formation of the Doushantuo
496 glaucony.

497 **5.3. Redox conditions for the lower Doushantuo Formation**

498 To assess the redox conditions during phosphogenesis, we use REE + Y patterns as
499 well as Ce anomalies recorded in the Doushantuo phosphorites as proxies. The REE + Y
500 in phosphorites might be affected by (i) the geochemical processes in seawater and pore
501 water, (ii) terrigenous detrital contamination, and (iii) later metamorphism and weathering.
502 Considering the fact that our samples are all from fresh drill cores and targets for *in situ*
503 LA-ICP-MS analyses are carefully selected, surface weathering is unlikely to have
504 significantly affected the REE signatures of the Doushantuo phosphorites. This is
505 supported by the generally high $\sum\text{REE}$, because REE will tend to be lost during weathering
506 processes ([Sheilds & Stille, 2001](#)). High-temperature metamorphism is also unlikely to
507 have affected the REE + Y patterns of the phosphatic grains because they commonly retain
508 their original cryptocrystalline nature ([She et al., 2013](#)) and their peak metamorphic
509 temperature was lower than 300 °C ([She et al., 2016](#)). Although low-temperature alteration
510 is common in the Doushantuo Formation ([Zhang et al., 2019](#)), it mainly affects the
511 interstitial phases rather than the phosphatic grains because the latter consist of densely
512 packed apatite nanocrystals ([She et al., 2013](#)) and hence have very low permeability. The
513 low Al_2O_3 content and the absence of a positive correlation between Al_2O_3 and $\sum\text{REE}$ (Fig.
514 8a) indicate that detrital contributions to the REE + Y patterns of our samples are
515 insignificant. This is also supported by the relatively high Y/Ho ratios and poor correlation
516 between Th and Y/Ho ratio (Fig. 8b), because terrigenous detrital components typically
517 have low Y/Ho ratios (~ 28 ; [Bau, 1996](#)) and high Th concentrations. In fact, shale bands
518 interbedded in the Weng'an phosphorites have much lower $\sum\text{REE}$ (81-146 ppm) and show

519 a distinct PAAS-like REE pattern (Chen et al., 2003). These exclude the possibility that
520 the REE + Y features of Doushantuo phosphorites have been significantly affected by
521 terrigenous detritus.

522 It has been shown that REE signatures of granular phosphorite primarily records
523 variations in the local conditions of diagenesis, while diagenetic redox conditions may
524 broadly reflect local bottom-water redox conditions (e.g., Shields and Stille, 2001). All the
525 Doushantuo samples show nearly identical REE patterns characterized by a slight MREE
526 enrichment with depletions of LREE and HREE (Fig. 7a). These concave-down patterns
527 are consistent with bulk rock REE data of equivalent outcrop samples of the Doushantuo
528 Formation at Weng'an, although the latter have much lower $\sum\text{REE}$ (94 ppm in average, n
529 = 48; Chen et al., 2003; Shields et al., 2004). The good correlation between $\sum\text{REE}$ and
530 $\text{Pr}_{(\text{SN})}/\text{Yb}_{(\text{SN})}$ (Fig. 8c) suggests that diagenetic REE uptake tends to increase both REE
531 abundance and LREE/HREE fractionation. Group I phosphorites, which are more closely
532 related to glaucony (Fig. 7b), must have experienced much stronger diagenetic REE uptake
533 as a result of more intense Fe redox cycling in a slightly more oxidising condition. The
534 slight shifts of $\text{La}_{(\text{SN})}/\text{Sm}_{(\text{SN})}$ and $\text{La}_{(\text{SN})}/\text{Yb}_{(\text{SN})}$ ratios from seawater composition (Fig. 8d)
535 suggest that REE systematics was probably influenced, to a minor extent, by both crystal
536 surface adsorption and bulk crystal substitution during early diagenesis (Reynard et al.,
537 1999).

538 Negative Ce anomalies in authigenic sediments are commonly considered as an
539 indicator of formation in an oxic environment, because Ce(III) is readily oxidized into less
540 soluble Ce(IV) in modern oxic ocean systems and then scavenged by suspended particles
541 that sink through the water column and accumulate in sediments (Sholkovitz et al., 1994;
542 Bau and Dulski, 1996). The absence of a negative Ce anomaly suggests that the formation
543 of the Doushantuo phosphorites may have occurred largely in suboxic or anoxic
544 environments without significant preferential scavenging of Ce from the ambient water.
545 Alternatively, the phosphorites originally could have had negative Ce anomalies but these
546 signatures were disguised during early diagenetic alteration in anoxic conditions. However,
547 diagenetic alteration of Ce anomaly signals is expected to result in various Ce features,
548 including highly positive Ce anomalies (Shields and Stille, 2001). Except for three samples

549 that have slightly positive Ce / Ce* values (1.06-1.12), most Doushantuo phosphorites
550 show highly consistent, near unity Ce / Ce* even at the micrometer scale, as shown by the
551 *in situ* REE data in Table 4. This suggest insignificant preferential Ce remobilization
552 relative to the neighbouring REE during diagenesis, although an original small negative Ce
553 anomaly cannot be completely excluded.

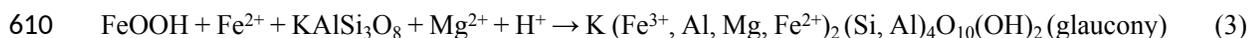
554 Since both Fe²⁺ and Fe³⁺ are simultaneously required for glauconitization, glaucony
555 preferably precipitates in suboxic environments around the Fe redoxcline where Fe²⁺ is
556 partially oxidized to Fe³⁺ (Meunier and El Albani, 2007; Tang et al., 2017). In K- and Fe-
557 rich, fully reducing environments (e.g., ferruginous conditions), formation of berthierine is
558 favored, while oxic conditions may promote goethite precipitation (Velde, 1992).
559 Therefore, the wide occurrence of glaucony in the lower Doushantuo Formation supports
560 a generally suboxic early diagenetic environment, which was in active exchange with more
561 oxygenated bottom seawater. In addition, slight variations of redox states in the porewater
562 and bottom water would have promoted extensive glaucony authigenesis (Tang et al.,
563 2017). These variations would not be captured by the phosphorite Ce signature as long as
564 the environmental redox potential does not exceed the threshold for Ce (III) oxidation.
565 Banerjee et al (2016b) have shown that slightly reducing conditions favour the origin of
566 high-Fe glaucony, whereas glaucony formed in slightly oxidising conditions is depleted in
567 Fe. Therefore, the higher Fe content in type 2 (phosphorite-hosted) glaucony than in type
568 1 (sandstone-hosted) glaucony appears to reflect less oxidising conditions for type 2
569 glaucony.

570 Collectively, mineralogical context and REE geochemistry suggest that the lower
571 Doushantuo Formation likely formed on a suboxic seafloor with slight fluctuations in redox
572 states of pore and bottom water. In contrast, the upper Doushantuo phosphorites at
573 Weng'an contain much lower REE and exhibit a distinct negative cerium anomaly, which
574 implies predominantly oxic conditions in local seawater as well as during both
575 phosphogenesis and early diagenetic REE scavenging (Chen et al., 2003; Shields et al.,
576 2004). Therefore, available phosphorite REE data document a transition from a generally
577 suboxic environment to an oxic environment through the Doushantuo Formation.

578 5.4. Fe-redox driven phosphatisation and glauconitization

579 Shaffer (1986) first noted the presence of a phosphorus concentration gradient near
580 the redoxcline in modern sediments in the Black Sea, and proposed the ‘Fe-redox pump’
581 model to explain the concentration of phosphorus in the pore water. Subsequent studies on
582 modern marine and lacustrine sediments have demonstrated that Fe-redox pumping is a
583 main driver for the cycling of phosphorus (Froelich et al., 1988; Heggie et al., 1990; Wheat
584 et al., 1996; Algeo and Ingall, 2007; Jilbert and Slomp, 2013; Cosmidis et al., 2014). Fe-
585 redox pumping is a cyclic mechanism that enriches pore water phosphate levels by first
586 transporting phosphate to the seafloor through adsorption onto Fe-(oxyhydr)oxides, which,
587 upon burial, are dissolved below the suboxic-anoxic boundary and liberate adsorbed
588 phosphate to pore water. The escape of phosphate is prevented by re-adsorption of
589 phosphate onto Fe-(oxyhydr)oxides just above this redox interface (Heggie et al., 1990;
590 Nelson et al., 2010). In the Precambrian, it is expected that anoxic intermediate and bottom
591 water would have prevented Fe-redox pumping from operating in deeper settings (Pafahl
592 and Hiatt, 2012). In shallow water environments, however, Fe-redox mediated cycling of
593 phosphorus is often invoked as a key mechanism during phosphogenesis (e.g., Crosby et
594 al., 2014; She et al., 2014; Cui et al., 2016) although direct evidence is lacking.

595 The occurrence of glaucony in the Doushantuo Formation and its association with
596 phosphate at both macroscopic and microscopic scales indicate phosphatisation and
597 glauconitization near the anoxic-oxic boundary in the sediments. As discussed earlier, the
598 oxygen concentration in the pore water was probably not high enough to cause preferential
599 Ce remobilisation. However, the extensive presence of glaucony demonstrate that redox
600 cycling of Fe was active during early diagenesis. As Ce (IV) has a redox potential close to
601 Mn (IV), which is higher than Fe (III) (Algeo & Li, 2020), glaucony authigenesis in the
602 lower Doushantuo Formation likely took place around the ferric iron reduction zone, and
603 below the manganese reduction zone. Fe-redox pumping thus led to the concentration of
604 phosphorus and REE, as supported by the coincident decreases in Σ REE and glaucony
605 abundance (Fig. 7b). Possible reactions related to the formation of apatite-glaucony
606 associations are expressed in the following equations (equations 2 and 3 not balanced
607 because of ion substitutions):



611 It is also noted that sulfide is ubiquitously present in the Doushantuo phosphatic
612 granules (Fig. 2; [She et al., 2013](#); [She et al., 2016](#); [Cui et al., 2016](#)), which suggests that
613 the precursors of granules are enriched in both P and Fe. These precursors likely consist of
614 apatite, glaucyony and possibly also vivianite (a ferrous phosphate). Diagenetic
615 transformation of glaucyony to less Fe-rich clays such as illite might have liberated Fe^{2+} ,
616 which, when combined with HS^- produced by microbial sulfate reduction, led to the
617 formation of diagenetic sulfides.

618 Authigenic glaucyony precipitation in Phanerozoic and modern oceans takes place
619 mostly in middle shelf to upper slope environments (200-300 m depth) with low
620 depositional rate or sediment starvation ([Odin and Matter, 1981](#); [Banerjee et al., 2016b](#)).
621 In contrast, Precambrian glaucyony is more common in stratigraphic successions deposited
622 in shallower-water environments with variable depositional rates ([Banerjee et al., 2016b](#)),
623 as also observed in the Doushantuo Formation. Typical Precambrian shallow-water
624 glaucyony include those in sandstones of the 1.0-Ga Bhalukona Formation ([Banerjee et al.,](#)
625 [2015](#)), central India and those associated with stromatolites in the 1.44-Ga Tieling
626 Formation, North China ([Tang et al., 2017](#)). This distinction between Phanerozoic and
627 Precambrian sediments also applies to marine phosphorites, which are restricted to shallow
628 water environments in the Precambrian but formed across the entire shelf in the
629 Phanerozoic ([Nelson et al., 2010](#); [Pufahl and Hiatt, 2012](#); [She et al., 2014](#); [Cui et al., 2016](#)).
630 In South China, Ediacaran phosphorite deposits are mostly distributed in coastal areas to
631 shallow shelf environments, including the inner and outer shelves near or above the
632 chemocline (Fig. 1a), whereas no significant phosphorite bed has been reported from the
633 intra-shelf and slope sections, which were deposited below the chemocline (Table 1; [Jiang](#)
634 [et al., 2011](#)). This further demonstrates that phosphorus enrichment regulated by Fe-redox
635 cycling was a prerequisite for massive accumulation of phosphate and that the apatite-
636 glaucyony association in Doushantuo is a direct result of Fe-redox driven authigenesis.

637 **5.5 A model for phosphogenesis and glaucogenesis**

638 Based on previous paleogeographic reconstructions, we propose a model to explain
639 the formation and spatial distribution of phosphorite and glaucony during the early
640 Doushantuo period (Fig. 9). Major phosphorite deposits are all formed in shallow water
641 environments above or near the redoxcline because in these areas Fe-redox pumping of
642 phosphorus could operate either in the water column or within the sediments. Glaucony
643 formation mainly occurs on the seafloor just above the redoxcline where Fe^{2+} was more
644 readily available because of the presence of deep ferruginous water below the redoxcline.
645 Degradation of organic matter through ferric iron reduction and sulfate reduction leads to
646 the release of Fe^{2+} and HS^- , as well as FeOOH-bound and organic-bound phosphorus to the
647 pore water. Replacement of K-feldspar and other detrital phases begins with the
648 incorporation of Fe^{2+} to form a precursor K-Al-Si-Fe phase. In such reducing conditions,
649 pyrite could have formed around the organic matter.

650 With the increase of redox potential in the bottom and pore water, the ferrous iron was
651 partially oxidized and Fe^{3+} was incorporated into the precursor K-Al-Si-Fe phase at the
652 expense of Al. Meanwhile, concentration of phosphate and precipitation of apatite were
653 probably promoted by sulfide-oxidizing bacteria which are known to be able to metabolize
654 polyphosphates and release orthophosphate into pore waters (Schulz & Schulz, 2005). With
655 frequent, transient redox fluctuations, repeated cycles of these processes led to progressive
656 addition of Fe^{2+} and Fe^{3+} that eventually resulted in the formation of glaucony. At Weng'an,
657 Fe-redox cycling resulted in the concentration of phosphorus and Fe^{2+} just below the
658 water/sediment interface, which in turn led to the precipitation of calcium phosphate.
659 Subsequent reworking and winnowing produced intraclasts of phosphate. Early diagenetic
660 formation of glaucony is favored by fluctuations of the redoxcline, which facilitated the
661 incorporation of both ferric and ferrous Fe (Fig. 9).

662 The occurrence of the Doushantuo apatite-glaucony association supports previous
663 recognition of the tempo-spatial distribution of phosphorites (Nelson et al., 2010; Pufahl
664 and Hiatt, 2012) and glaucony (Banerjee et al., 2015) that characterizes distinct
665 environments of phosphogenesis and glaucogenesis in the Precambrian and Phanerozoic.
666 It should be noted that although glaucony is abundant in Phanerozoic sedimentary

667 successions, it is relatively rare in the Precambrian (Banerjee et al., 2016b), probably due
668 to post-depositional alteration to chlorite (Bansal et al., 2020) or surface weathering.
669 Further investigations of the mineralogical relationships between phosphorite and
670 glaucony (or its alteration products) in Precambrian successions hold great potential to
671 further understanding of the paleoredox conditions of past seawater.

672 **6. Conclusions**

673 The lower Doushantuo Formation at Weng'an begins with terrigenous clastic
674 deposition in shallow water peritidal environments, followed by a transgression that led to
675 a decrease of detrital input and increase of intraclasts. Pristine phosphorite formed as
676 phosphatic flasers in glauconitic sandstones and were subsequently reworked and
677 redeposited, whereas glaucony occur as authigenic cement between detrital minerals and
678 phosphate grains and as partial replacement of K-feldspar, quartz, muscovite and phosphate
679 grains.

680 The Doushantuo glaucony is characterized by high K_2O but low and variable TFe_2O_3
681 contents, similar to other Precambrian glaucony. The compositional variability and Raman
682 spectral features support the formation of glaucony by progressive incorporation of Fe into
683 the mineral lattice, including Fe^{3+} substitution for Al^{3+} . Early diagenetic glaucony
684 cementation occurred in suboxic pore water just below the sediment-water interface where
685 both ferric and ferrous Fe were available and incorporation of iron into glaucony was
686 probably promoted near the redoxcline where deep ferruginous water provided Fe^{2+} .

687 The phosphate grains exhibit consistent concave-down REE + Y patterns and do not
688 show a negative Ce anomaly, which suggests the formation of pristine phosphorites in
689 suboxic environments without significant preferential remobilisation of Ce during early
690 diagenesis. The tempo-spatial distribution of glaucony and phosphorite in the lower
691 Doushantuo Formation demonstrates the presence of a shallow redoxcline during the early
692 Doushantuo Period. Phosphorus and Fe^{2+} were concentrated in anoxic bottom water and
693 pore water by repeated cycles of oxidation and reduction of Fe-bearing phases and organic
694 matter. Apatite-glaucony association is an indication of suboxic early diagenetic
695 environments and can be a reliable mineralogical proxy, which, when present, to decipher
696 shallow-water redox conditions in the Precambrian.

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- 960

Algabri et al_Fig.1

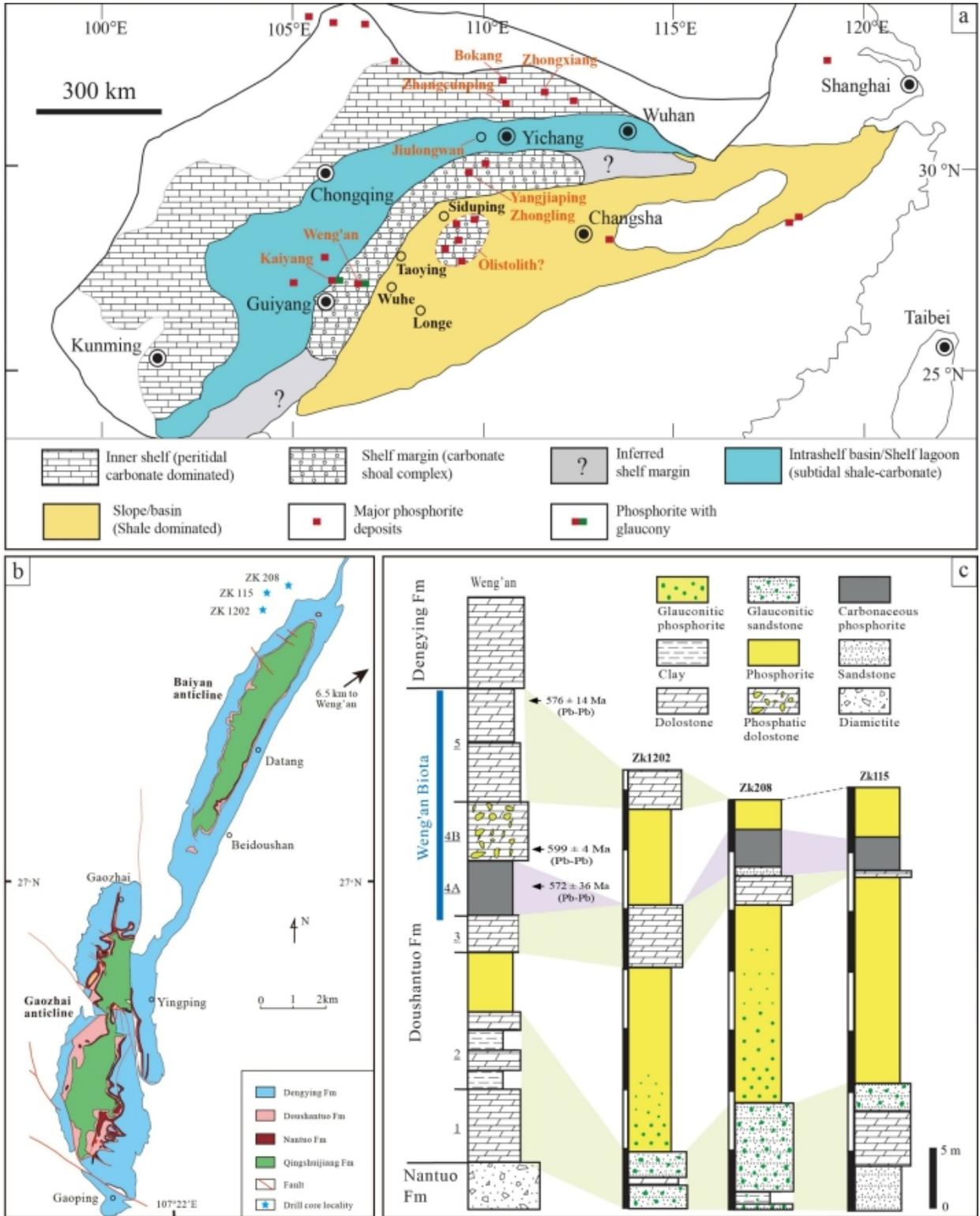


Fig. 1. Geological map and stratigraphic columns. a) A generalized paleogeographic map of the Yangtze Craton during the early Ediacaran Period (modified after Jiang et al., 2011), showing major phosphate deposits in the Doushantuo Formation (Ye et al., 1989) and a possible olistolith (Vernhet et al., 2007). b) A simplified geological map showing the Precambrian geology of the Baiyan - Gaozhai area with locations of studied drill holes. c) Stratigraphic columns of the studied drill cores (Zk1202, Zk208 and Zk115) of the Doushantuo Formation highlighting glaucony-bearing horizons. Note the upward decreasing trend of glaucony abundance. Age data are from Barford et al. (2002), Chen et al. (2004) and Chen et al. (2009).

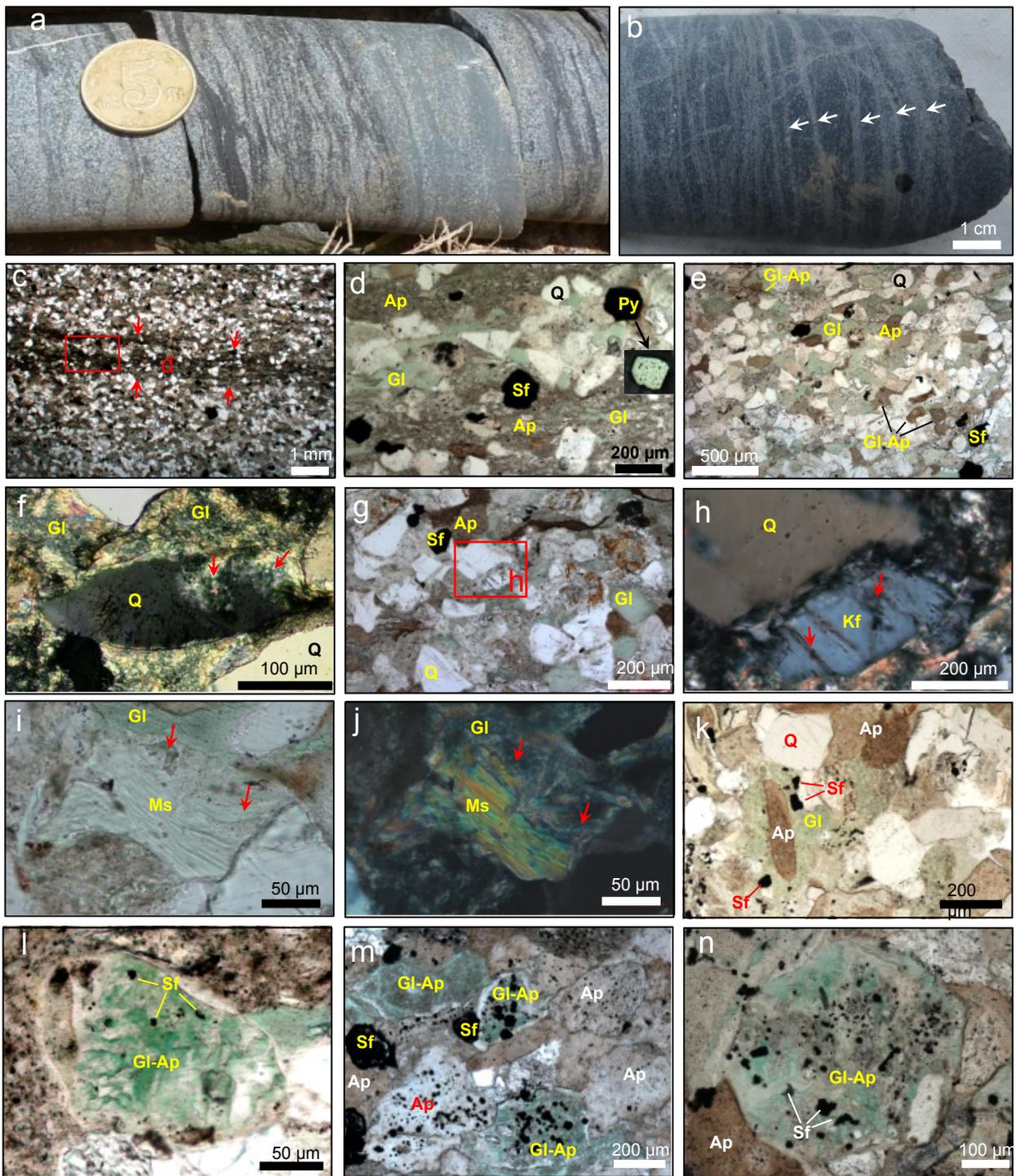


Fig. 2. Drill core photos (a-b) and photomicrographs (c-n) of glauconitic sandstones and phosphorites in the lower Doushantuo Formation. a) Glauconitic sandstone showing lenticular and flaser bedding. b) Granular phosphorite containing carbonate bands (arrows). c-j) Glauconitic sandstones consisting of detrital minerals, apatite grains and glaucony cement (type 1). c) Phosphatic flasers (arrows) in glauconitic sandstone. d) Magnified view of the flasers in (c) showing anastomosing structures with detrital grains embedded in a network of black-brownish apatite mass (inset, reflected light image of the pyrite grain). e) Glaucony cementing and replacing detrital quartz and phosphatic grains. f-g) Glaucony cementing partly replacing detrital quartz (f, arrows) and K-feldspar (g). h) Close-up of the K-feldspar partly replaced by glaucony (arrows) in (g). i-j) Glauconitized muscovite grain (arrows). k-n) Phosphorite containing glaucony (type 2) and sulfides. k) Glaucony cementing apatite grains. l-n) Glauconitized apatite grains containing abundant sulfides. Mineral abbreviations: Gl, glaucony; Ap, apatite; Gl-Ap, glauconitized apatite grains; Q, quartz; Sf, sulfide; Ms, muscovite; Kf, K-feldspar. Coin for scale in (a) is 20.5 mm in diameter. c-e, g, i, k-n), plane polarized light mode; f, h, j), crossed polarizers mode.

Algabri et al_Fig. 3

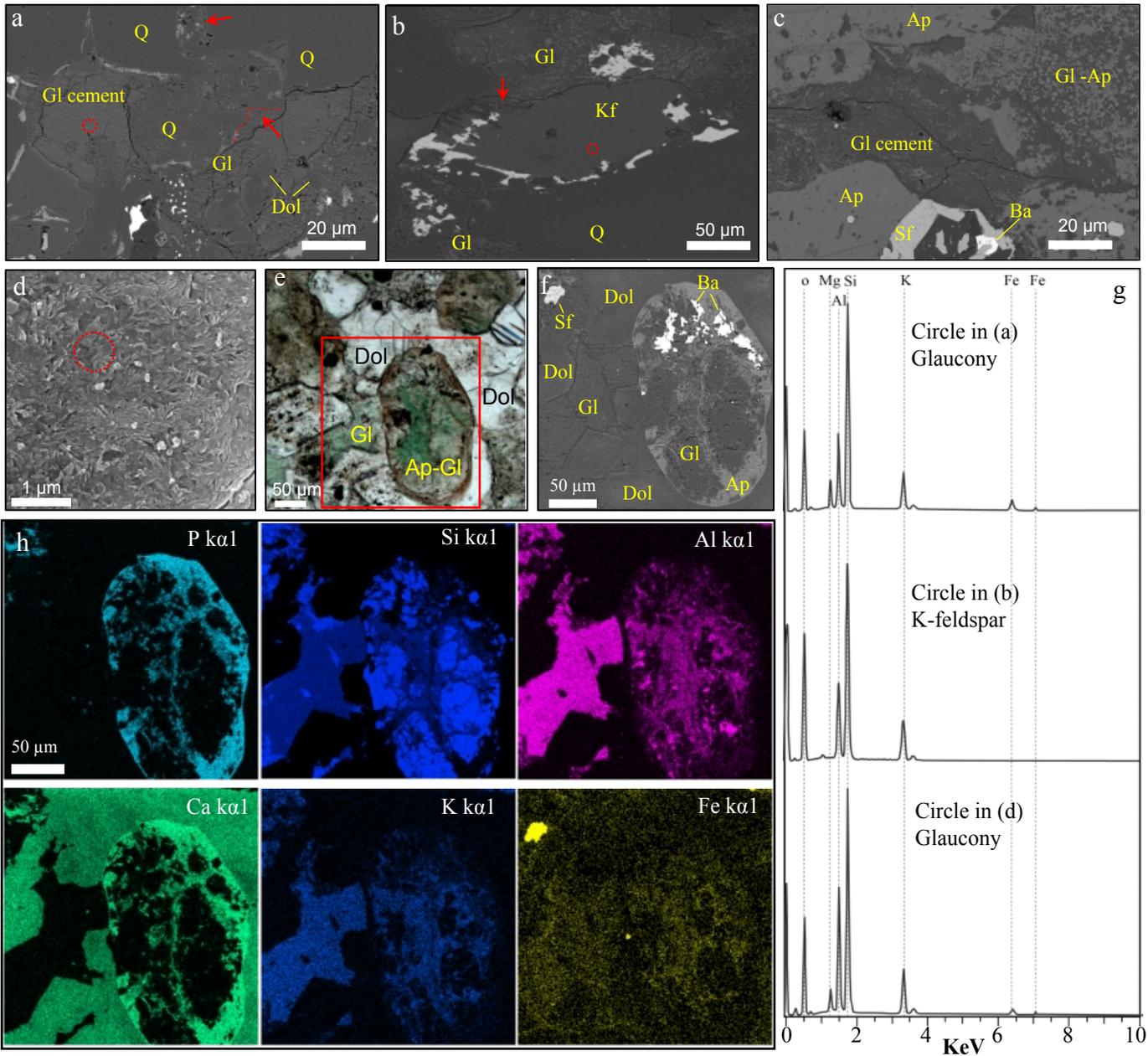


Fig. 3. SEM images (a-d, f), energy dispersive spectrometer (EDS) spectra (g) and elemental maps (h) of the Doushantuo glaucony and associated minerals. a) BSE image showing type 1 glaucony (Gl) occurring as cements as well as partly replacing detrital quartz grains (arrows). b) BSE image of a K-feldspar partly replaced by glaucony (arrow). c) BSE image showing type 2 glaucony, glauconitized apatite grain, sulfide and barite. d) High magnification secondary electron image of type 2 glaucony showing lamellar structure. e) Transmitted light image showing an apatite-glaucony grain and glaucony cement that were selected for EDS mapping. f) BSE image of the area outlined in (e). g) EDS spectra of the spots marked in (a), (b) and (d). h) EDS elemental maps of (f). Mineral abbreviations: Q, quartz; Ap, apatite; Gl, glaucony; Kf, K-feldspar; Sf, sulfide; Ba, barite.

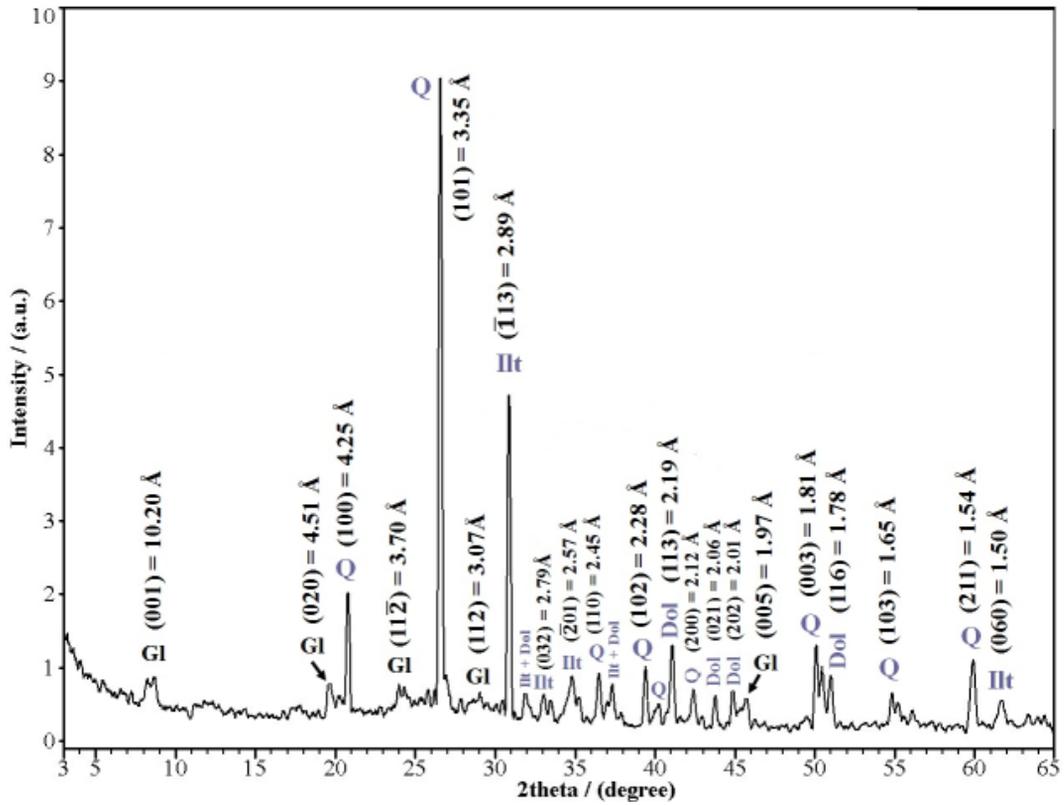


Fig. 4. XRD spectrum for a glauconitic sandstone (Zk1202-4). The diffraction peaks shown include those of glauconite (Gl), illite (Ill), quartz (Q), dolomite (Dol) and their mixtures.

Algabri et al_Fig. 5

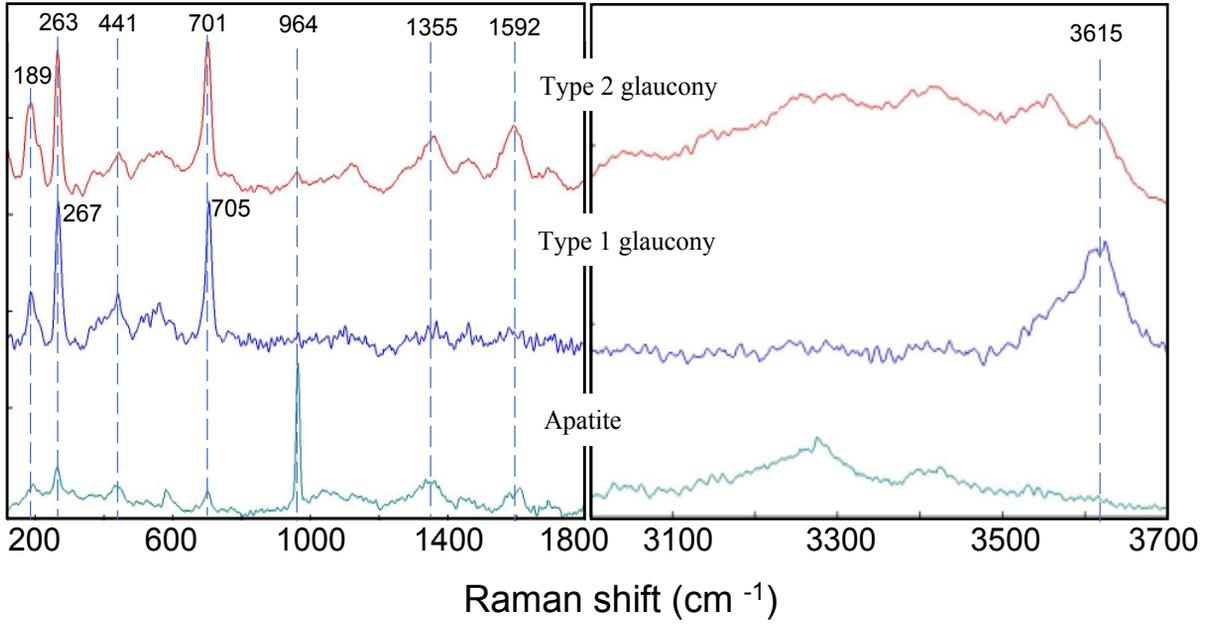


Fig. 5. Raman spectra of the Doushantuo glaucony and apatite.

Algabri et al_Fig. 6

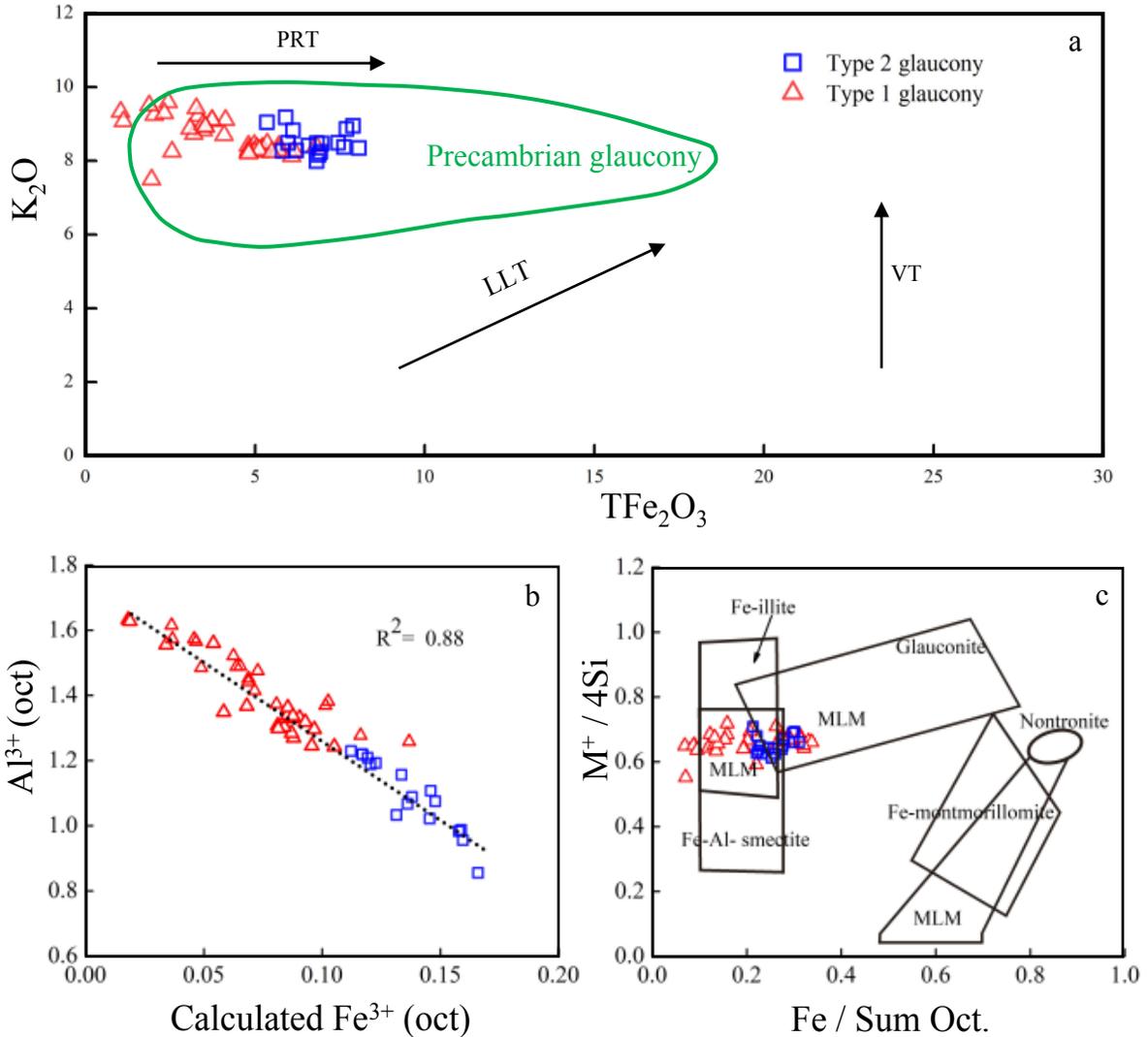


Fig. 6. Chemical composition of the two types of glaucony in the Doushantuo Formation. (a) Cross plot of K_2O vs. calculated TFe_2O_3 , showing constantly high K_2O but variable TFe_2O_3 contents. Also shown is the compositional field of typical Precambrian glaucony (Banerjee et al., 2016b; Tang et al., 2017). Arrows indicate evolutionary trends of glaucony predicted by three hypotheses: the ‘layer lattice theory’ (LLT), the ‘verdissement theory’ (VT), and the ‘pseudomorphic replacement theory’ (PRT). The K_2O - Fe_2O_3 relationship of most Phanerozoic glaucony closely follows the LLT and VT trends and are not presented here (cf. Banerjee et al., 2016b). (b) Diagram for Al^{3+} vs Fe^{3+} (calculated) in octahedral site for the Doushantuo Formation showing negative correlation. Note that the overall increase in Fe^{3+} from type 1 to type 2 glauconies at the expense of Al^{3+} . (c) Cross plot of $Fe/\text{Sum of octahedral charges}$ vs $(M^+/4Si)$ (M = interlayered cations, after Meunier and El Albani, 2007). Note that type 1 and 2 glauconies plots into the fields of glauconite, Fe-illite and Fe-Al smectite. MLM, Mixed layer minerals.

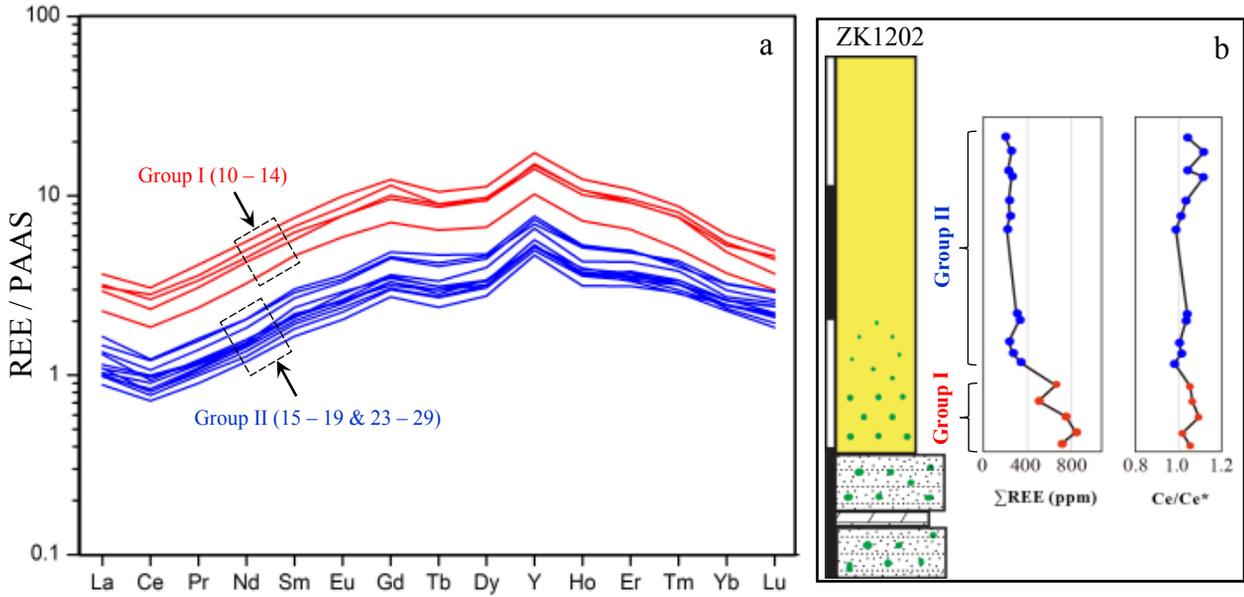


Fig. 7. Post-Archean Australian Shale (PAAS) -normalized REE + Y patterns (a) and stratigraphic variations of Σ REE and Ce/Ce* (b) for the Doushantuo phosphorites. Every spectrum in (a) represents the average composition of two to four spots on the same polished slab analyzed with LA-ICPMS. Note the roughly coincident decreases in Σ REE and glaucony abundance (shown by the green dots).

Algabri et al_Fig. 8

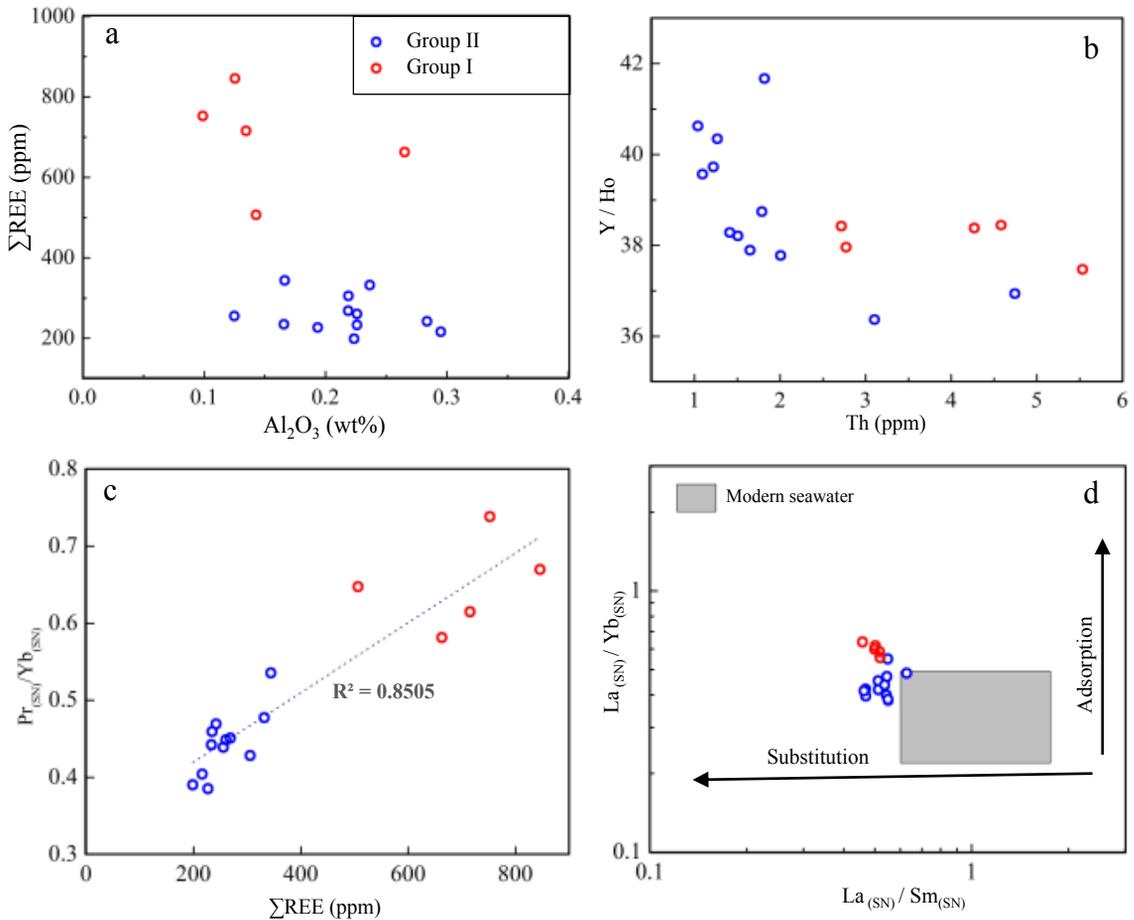


Fig. 8. Cross plots of Al_2O_3 , Th and various REE parameters. Compositional range of modern oceanic water in (d) is from Reynard et al. (1999).

Algabri et al_Fig. 9

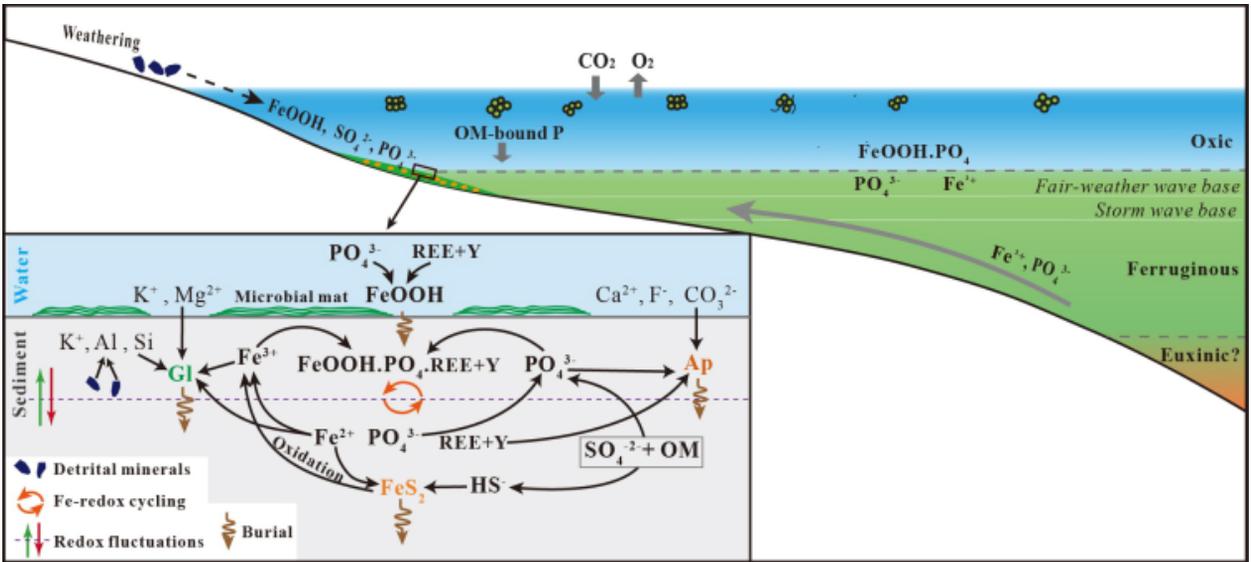


Fig. 9. Conceptual diagram of Fe redox cycling, glauconitization and phosphatisation processes for the lower Doushantuo Formation based on paleogeographic reconstructions and ocean chemistry models (Li et al., 2010; She et al., 2014; Muscente et al., 2015; Cui et al., 2016). In the peritidal environment near the redoxcline, Fe-redox processes at or below the sediment/water interface controlled the formation of glaucony (Gl) and apatite (Ap). Following deposition from the oxic seawater, Fe oxyhydroxides are dissolved in anoxic pore waters and Fe^{2+} and PO_4^{3-} are released. The reduced iron (Fe^{2+}) is refluxed back towards the sediment-water interface where it is incorporated into glaucony or again be oxidized and adsorb additional phosphate from oxygenated bottom waters, leading to the concentration of phosphate. Episodes of seawater oxygenation leads to redox variation in pore water and facilitate glaucony formation.

Table 1 Occurrence of glaucony and phosphorite in the Doushantuo Formation, South China

Depositional environment	Section	Locality	Glaucony abundance	Phosphate abundance	Reference
Shore*	Weng'an	Central Guizhou	Abundant	Abundant	This study; Zhou et al (2001); Zhu et al.(2007); Jiang et al (2011); Xiao et al (2014); Muscente et al (2015); Zhu and Jiang (2017) Chen et al. (2015); Zhang et al (2019)
	Kaiyang	Central Guizhou	Abundant	Abundant	
Inner shelf	Xiaofenghe	Western Hubei	None reported	Moderate	Zhu et al (2007); Xiao et al (2012); Muscente et al (2015); Cui et al (2016)
	Baokang	Western Hubei	None reported	Abundant	Zhou et al (2001); Xiao et al (2012); Muscente et al (2015); Cui et al (2016); She et al (2016)
	Zhangcunping	Western Hubei	None reported	Abundant	Zhu et al (2007); She et al (2013); Muscente et al (2015); Cui et al (2016)
Intra shelf	Jiulongwan	Western Hubei	None reported	Minor	McFadden et al (2008); Jiang et al (2011); Xiao et al (2012); Muscente et al (2015); Cui et al (2016)
Outer Shelf	Zhongling	Northwestern Hunan	None reported	Moderate	Zhu et al (2007); Li et al (2010); Jiang et al (2011); Cui et al (2015)
	Yangjiaping	Northwestern Hunan	None reported	Moderate	Zhu et al (2007); Jiang et al (2011); Cui et al (2015)
Slope	Siduping	Western Hunan	None reported	None reported	Jiang et al (2011)
	Taoying	Northeastern Guizhou	None reported	None reported	Jiang et al (2011)
	Wuhe	Southeastern Guizhou	None reported	None reported	Jiang et al (2011)
Basin	Longe	Southeastern Guizhou	None reported	None reported	Li et al (2010); Huang et al (2017)

* Sedimentary environment for Weng'an and Kaiyang have been alternatively interpreted as a shelf margin (Jiang et al., 2011).

Table 2 Quantitative EDS analysis results of oxide weight percentage and structural formula of glaucony in samples ZK1202-05 (type 1, spot No 1-18) and ZK1202-10 (type 2, spot No 19-35).

Spot No	MgO	Al ₂ O ₃	SiO ₂	K ₂ O	CaO	TFe ₂ O ₃	Total	K ⁺	Na ⁺	Ca ²⁺	Fe ²⁺	Fe ³⁺	Mg ²⁺	Al ³⁺ (oct)	Si ⁴⁺	Al ³⁺ (tet)	R ³⁺ (oct)	R ³⁺ (tet)	Fe/Sum. Oct	M ²⁺ /4Si
1	4.0	22.3	52.6	8.2	0.0	4.8	91.9	0.72	0.00	0.00	0.10	0.14	0.40	1.38	3.59	0.41	1.53	0.41	0.17	0.64
2	5.4	22.0	61.3	8.3	0.0	6.2	103.1	0.64	0.00	0.00	0.12	0.16	0.49	1.28	3.71	0.29	1.44	0.29	0.20	0.59
3	4.1	21.8	54.1	8.2	0.0	4.8	92.9	0.70	0.00	0.00	0.10	0.14	0.41	1.37	3.64	0.36	1.51	0.36	0.17	0.64
4	4.1	26.0	56.6	9.2	0.0	3.3	99.2	0.74	0.00	0.00	0.07	0.09	0.38	1.49	3.56	0.44	1.58	0.44	0.11	0.66
5	4.0	28.9	57.1	9.6	0.0	2.5	102.0	0.75	0.00	0.00	0.05	0.07	0.37	1.57	3.49	0.51	1.63	0.51	0.08	0.65
6	3.7	29.5	56.0	9.5	0.0	1.9	100.5	0.75	0.00	0.00	0.04	0.05	0.34	1.62	3.46	0.54	1.67	0.54	0.06	0.65
7	3.8	28.1	55.3	9.3	0.0	2.3	98.9	0.75	0.00	0.00	0.05	0.06	0.36	1.57	3.48	0.52	1.64	0.52	0.08	0.65
8	3.8	26.3	54.3	8.9	0.0	3.1	96.3	0.73	0.00	0.00	0.06	0.09	0.37	1.52	3.52	0.48	1.61	0.48	0.11	0.64
9	3.3	24.3	64.2	7.5	0.0	2.0	101.2	0.57	0.00	0.00	0.04	0.05	0.29	1.57	3.86	0.14	1.62	0.14	0.06	0.55
10	4.6	24.9	56.4	9.0	0.0	3.5	98.4	0.73	0.00	0.00	0.07	0.10	0.44	1.44	3.58	0.42	1.54	0.42	0.12	0.65
11	5.2	25.4	58.9	9.1	0.0	3.8	102.3	0.71	0.00	0.00	0.07	0.10	0.47	1.41	3.59	0.41	1.51	0.41	0.12	0.63
12	3.6	24.9	58.8	9.4	0.0	3.3	100.0	0.75	0.00	0.00	0.06	0.09	0.33	1.49	3.66	0.34	1.58	0.34	0.11	0.68
13	4.7	22.9	59.2	9.1	0.0	4.1	100.0	0.72	0.00	0.00	0.08	0.11	0.44	1.37	3.69	0.31	1.49	0.31	0.14	0.67
14	3.8	25.2	58.2	9.1	0.0	3.7	100.0	0.72	0.00	0.00	0.07	0.10	0.35	1.48	3.63	0.37	1.58	0.37	0.12	0.66
15	4.0	24.7	58.4	8.9	0.5	3.5	99.5	0.71	0.00	0.03	0.07	0.10	0.37	1.45	3.64	0.36	1.55	0.36	0.12	0.68
16	3.2	21.2	64.8	8.2	0.0	2.6	100.0	0.64	0.00	0.00	0.05	0.07	0.29	1.49	3.96	0.04	1.55	0.04	0.09	0.64
17	4.0	22.3	55.5	8.4	1.6	6.8	93.8	0.69	0.00	0.11	0.14	0.19	0.38	1.26	3.57	0.43	1.45	0.43	0.23	0.71
18	4.0	24.7	58.4	8.9	0.5	3.5	100.1	0.71	0.00	0.03	0.07	0.10	0.37	1.45	3.64	0.36	1.55	0.36	0.12	0.68
Average	4.1	24.7	57.8	8.8	0.1	3.7	98.9	0.71	0.00	0.01	0.07	0.10	0.38	1.46	3.63	0.37	1.56	0.37	0.12	0.65
19	6.6	20.2	60.2	8.5	0.0	7.0	102.5	0.66	0.00	0.00	0.13	0.19	0.61	1.16	3.69	0.31	1.34	0.31	0.22	0.61
20	5.9	17.2	54.4	8.0	0.0	6.8	92.2	0.70	0.00	0.00	0.15	0.21	0.60	1.11	3.72	0.28	1.31	0.28	0.25	0.65
21	6.2	16.6	54.7	8.2	0.0	6.9	92.7	0.72	0.00	0.00	0.15	0.21	0.63	1.07	3.74	0.26	1.28	0.26	0.25	0.67
22	8.0	17.0	58.6	8.4	0.0	6.6	98.6	0.69	0.00	0.00	0.13	0.19	0.76	1.03	3.75	0.25	1.22	0.25	0.22	0.64
23	7.7	17.0	60.3	8.5	0.0	7.5	100.9	0.68	0.00	0.00	0.15	0.20	0.72	1.02	3.77	0.23	1.23	0.23	0.24	0.64
24	6.0	19.5	59.1	8.3	0.0	6.2	99.1	0.67	0.00	0.00	0.12	0.17	0.57	1.19	3.74	0.26	1.36	0.26	0.21	0.62
25	5.4	19.9	54.4	9.0	0.0	5.4	94.0	0.77	0.00	0.00	0.11	0.16	0.54	1.23	3.66	0.34	1.39	0.34	0.19	0.71
26	6.0	19.4	57.6	8.5	0.0	6.0	97.4	0.70	0.00	0.00	0.12	0.17	0.57	1.19	3.72	0.28	1.36	0.28	0.20	0.65

Spot No	MgO	Al ₂ O ₃	SiO ₂	K ₂ O	CaO	TFe ₂ O ₃	Total	K ⁺	Na ⁺	Ca ²⁺	Fe ²⁺	Fe ³⁺	Mg ²⁺	Al ³⁺ (oct)	Si ⁴	Al ³⁺ (tet)	R ³⁺ (oct)	R ³⁺ (tet)	Fe/Sum. Oct	M ⁺ /4Si
27	5.8	20.9	57.8	9.2	0.0	5.9	99.6	0.74	0.00	0.00	0.12	0.16	0.55	1.22	3.66	0.34	1.38	0.34	0.20	0.68
28	7.0	17.9	58.2	8.2	0.0	6.9	98.1	0.67	0.00	0.00	0.14	0.19	0.67	1.09	3.73	0.27	1.28	0.27	0.23	0.62
29	7.4	18.0	58.2	8.5	0.0	6.8	98.9	0.69	0.00	0.00	0.14	0.19	0.71	1.07	3.71	0.29	1.26	0.29	0.23	0.64
30	7.7	15.8	57.9	9.0	0.0	7.9	98.3	0.74	0.00	0.00	0.16	0.22	0.75	0.96	3.75	0.25	1.18	0.25	0.27	0.69
31	7.4	16.3	55.7	8.4	0.0	7.6	95.4	0.71	0.00	0.00	0.16	0.22	0.74	0.99	3.71	0.29	1.21	0.29	0.27	0.66
32	6.4	21.6	58.0	8.8	0.0	6.1	101.0	0.70	0.00	0.00	0.12	0.17	0.60	1.21	3.62	0.38	1.38	0.38	0.20	0.64
33	7.4	16.3	56.8	8.9	0.0	7.7	97.0	0.74	0.00	0.00	0.16	0.22	0.72	0.98	3.72	0.28	1.21	0.28	0.26	0.69
34	9.0	14.0	57.3	8.3	0.0	8.1	96.7	0.70	0.00	0.00	0.17	0.23	0.88	0.85	3.77	0.23	1.09	0.23	0.28	0.66
35	5.9	20.2	57.1	8.3	0.0	5.8	97.3	0.68	0.00	0.00	0.12	0.16	0.57	1.22	3.68	0.32	1.38	0.32	0.20	0.63
Average	6.8	18.1	57.4	8.5	0.0	6.8	97.6	0.70	0.00	0.00	0.14	0.19	0.66	1.09	3.71	0.29	1.29	0.29	0.23	0.65

Note: R³⁺, trivalent cations in tetrahedral (tet.) and octahedral (oct.) sites of glauconite; M⁺, monovalent cations in interlayer sites of glauconite. An average value of Fe²⁺/ΣFe of typical Precambrian glaucony ([Banerjee et al., 2015](#); [Tang et al., 2017](#)) is used for the calculation of structural formula.

Table 3 EPMA analysis results of oxide weight percentage (wt%) and structural formula of glaucony in sample ZK1202-04 (type 1).

Spot No	SiO ₂	Al ₂ O ₃	TiO ₂	TFe ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	Total	Na ⁺	Ca ²⁺	K ⁺	Fe ²⁺	Fe ³⁺	Mg ²⁺	Al ³⁺ (oct)	Si ⁴⁺	Al ³⁺ (tet)	R ³⁺ (oct)	R ³⁺ (tet)	Fe/Sum. Oct	M ⁺ /4Si
1	54.0	26.3	0.1	2.0	3.5	0.1	0.1	9.2	95.4	0.01	0.01	0.77	0.04	0.06	0.35	1.56	3.53	0.47	1.61	0.47	0.23	0.69
2	55.0	22.3	0.1	3.5	5.3	0.2	0.1	8.8	95.2	0.01	0.01	0.74	0.07	0.10	0.51	1.35	3.62	0.38	1.45	0.38	0.35	0.68
3	52.4	26.9	0.1	3.2	3.0	0.1	0.1	8.7	94.6	0.02	0.02	0.73	0.07	0.09	0.30	1.56	3.46	0.54	1.65	0.54	0.24	0.67
4	55.7	22.6	0.1	4.1	4.5	0.3	0.1	8.7	96.0	0.01	0.01	0.72	0.08	0.12	0.43	1.37	3.63	0.37	1.48	0.37	0.33	0.68
5	55.4	21.1	0.1	4.8	4.8	0.3	0.1	8.4	94.8	0.01	0.01	0.71	0.10	0.14	0.47	1.31	3.66	0.34	1.45	0.34	0.37	0.66
6	55.8	20.4	0.1	5.2	4.9	0.2	0.1	8.4	95.1	0.01	0.01	0.70	0.11	0.15	0.48	1.27	3.68	0.32	1.42	0.32	0.39	0.66
7	56.4	20.1	0.1	5.7	4.9	0.3	0.1	8.4	95.9	0.01	0.01	0.70	0.12	0.16	0.48	1.25	3.70	0.30	1.41	0.30	0.40	0.66
8	56.5	20.8	0.1	5.2	4.8	0.3	0.0	8.3	96.1	0.00	0.00	0.69	0.11	0.15	0.47	1.28	3.69	0.31	1.43	0.31	0.38	0.64
9	55.4	21.2	0.1	5.0	4.7	0.2	0.0	8.4	95.1	0.00	0.00	0.71	0.10	0.14	0.47	1.30	3.65	0.35	1.45	0.35	0.38	0.66
10	55.6	21.0	0.1	4.8	4.8	0.3	0.1	8.3	95.0	0.01	0.01	0.70	0.10	0.14	0.48	1.30	3.67	0.33	1.44	0.33	0.38	0.65
11	55.7	18.5	0.0	6.1	4.0	0.4	0.1	8.1	92.9	0.01	0.01	0.70	0.13	0.18	0.40	1.24	3.77	0.23	1.42	0.23	0.39	0.68
12	56.3	20.7	0.0	5.4	3.6	0.3	0.0	8.5	94.8	0.00	0.00	0.71	0.11	0.16	0.36	1.33	3.72	0.28	1.49	0.28	0.34	0.67
13	56.4	19.8	0.0	5.7	3.8	0.3	0.1	8.3	94.4	0.01	0.01	0.70	0.12	0.17	0.38	1.30	3.75	0.25	1.46	0.25	0.36	0.67
14	52.9	26.1	0.0	1.1	2.7	0.1	0.0	9.3	92.3	0.00	0.00	0.80	0.02	0.03	0.28	1.63	3.56	0.44	1.66	0.44	0.17	0.72
15	55.6	26.2	0.0	1.1	2.8	0.1	0.0	9.1	95.0	0.00	0.00	0.75	0.02	0.03	0.27	1.63	3.62	0.38	1.66	0.38	0.17	0.68
16	55.7	19.9	0.1	5.1	3.6	0.2	0.1	8.3	93.0	0.01	0.01	0.72	0.11	0.15	0.36	1.33	3.75	0.25	1.48	0.25	0.33	0.68
17	55.9	20.2	0.0	5.0	3.6	0.2	0.1	8.2	93.2	0.01	0.01	0.70	0.10	0.15	0.36	1.34	3.75	0.25	1.49	0.25	0.33	0.67
18	57.4	21.2	0.0	5.1	3.6	0.2	0.0	8.2	95.8	0.01	0.01	0.68	0.10	0.15	0.35	1.36	3.74	0.26	1.51	0.26	0.32	0.65
19	55.8	20.2	0.1	5.1	3.5	0.3	0.0	8.3	93.4	0.00	0.00	0.71	0.11	0.15	0.35	1.34	3.74	0.26	1.49	0.26	0.33	0.67
20	56.7	20.0	0.1	5.5	3.7	0.3	0.0	8.2	94.5	0.01	0.01	0.69	0.11	0.16	0.36	1.32	3.76	0.24	1.48	0.24	0.34	0.66
Average	55.5	21.8	0.1	4.4	4.0	0.2	0.1	8.5	94.6	0.01	0.01	0.72	0.09	0.13	0.39	1.37	3.67	0.33	1.50	0.33	0.33	0.67

Table 4 Major element (wt%) and REE + Y concentrations (ppm) and calculated parameters for the ZK1202 phosphorites.

S.N.	MgO	Al ₂ O ₃	P ₂ O ₅	MnO	FeO	CaO	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Y	Th	ΣREE	Y/Ho	Ce/Ce*	Eu/Eu*	Pr _{(SS)/Yb_(SS)}
18-01	0.17	0.01	42.56	0.05	0.20	53.01	50.84	81.60	11.68	57.56	13.08	2.99	16.87	2.87	17.79	4.52	12.44	1.71	8.65	1.20	171.87	3.43	283.80	38.00	1.00	0.99	0.43
18-02	0.26	0.34	40.68	0.06	0.69	52.64	52.36	87.60	12.17	65.92	14.23	3.51	20.32	2.88	18.79	4.83	12.71	1.43	8.62	1.21	171.25	4.70	306.60	35.40	1.13	1.11	0.45
18-03	0.14	0.25	39.13	0.04	0.17	53.46	56.62	102.49	15.03	76.76	18.93	3.68	23.11	3.47	25.12	5.37	14.20	1.90	8.80	1.29	196.53	4.25	356.80	36.60	1.01	0.91	0.55
18-04	0.19	0.34	39.19	0.10	0.35	55.05	64.20	111.95	15.74	77.63	18.08	4.60	24.90	3.87	24.97	5.62	15.64	1.98	10.45	1.38	212.14	6.59	381.00	37.70	1.01	1.12	0.48
18 avg.	0.19	0.24	40.39	0.06	0.35	53.54	56.00	95.91	13.66	69.47	16.08	3.70	21.30	3.27	21.67	5.09	13.75	1.75	9.13	1.27	187.95	4.74	332.05	36.94	1.03	1.00	0.48
19-01*	0.34	2.31	28.33	0.09	1.07	46.88	42.97	80.14	11.56	59.12	15.21	3.70	18.44	2.71	16.50	3.89	11.48	1.43	7.69	0.97	157.39	2.62	275.80	40.40	1.02	1.16	0.48
19-02	0.14	0.39	36.24	0.04	0.21	54.86	52.35	88.96	12.60	66.11	15.37	3.79	20.37	3.04	19.92	4.97	13.46	1.51	8.86	1.24	192.35	2.40	312.50	38.70	1.07	1.12	0.45
19-03	0.11	0.15	37.42	0.04	0.66	54.94	46.62	68.33	9.42	49.60	11.19	2.61	17.12	2.73	18.98	4.77	13.51	1.57	9.01	1.15	187.06	1.38	256.60	39.20	1.10	0.97	0.33
19-04	0.06	0.12	37.88	0.03	0.12	55.25	55.42	97.39	14.56	71.52	18.30	4.29	25.00	3.63	23.74	5.68	15.64	1.81	9.35	1.39	217.39	1.58	347.70	38.30	0.95	1.07	0.50
19 avg.	0.10	0.22	37.18	0.03	0.33	55.02	51.46	84.89	12.19	62.41	14.95	3.57	20.83	3.14	20.88	5.14	14.20	1.63	9.08	1.26	198.93	1.79	305.62	38.74	1.04	1.10	0.43
23-01*	0.16	0.60	37.94	0.03	0.22	55.05	38.46	61.82	8.87	42.11	9.57	2.36	13.46	2.05	13.89	3.26	9.87	1.29	6.86	0.94	126.45	2.34	214.80	38.80	0.96	1.09	0.41
23-02	0.14	0.30	37.83	0.05	0.17	57.26	41.43	63.93	9.22	43.61	10.93	2.49	14.36	2.14	15.59	3.75	9.90	1.36	7.50	0.90	142.92	1.04	227.10	38.10	0.95	1.04	0.39
23-03	0.16	0.19	37.25	0.04	0.19	57.83	37.27	67.57	9.28	46.91	10.84	2.79	14.48	2.15	13.87	3.74	9.62	1.27	6.69	1.05	131.19	2.28	227.50	35.10	1.06	1.17	0.44
23-04	0.11	0.09	36.35	0.06	0.11	59.29	33.29	52.80	7.85	39.40	8.97	2.08	13.62	2.06	14.21	3.41	9.21	1.26	6.80	0.95	135.02	0.94	195.90	39.60	0.98	0.99	0.37
23 avg.	0.14	0.29	37.34	0.04	0.17	57.36	37.61	61.53	8.80	43.00	10.08	2.43	13.98	2.10	14.39	3.54	9.65	1.29	6.96	0.96	133.90	1.65	216.34	37.89	0.99	1.10	0.40
24-01	0.12	0.30	41.15	0.03	0.13	54.53	43.21	79.30	10.85	52.83	13.41	2.93	16.60	2.39	15.94	3.88	10.47	1.21	7.13	0.99	144.66	2.76	261.10	37.30	1.03	1.04	0.49
24-02	0.13	0.32	40.50	0.03	0.16	54.87	36.44	64.34	9.68	45.17	11.04	2.50	14.97	2.03	15.16	3.52	9.17	1.05	6.67	0.82	134.78	1.43	222.60	38.30	0.90	1.06	0.46
24-03*	0.21	0.69	37.26	0.04	0.51	52.20	40.48	68.58	9.29	43.60	10.60	2.48	13.00	2.00	13.44	3.34	8.93	1.19	6.53	0.98	126.17	2.47	224.40	37.80	1.00	1.08	0.45
24-04	0.14	0.22	38.79	0.03	0.11	55.37	38.17	72.18	9.82	50.82	12.16	2.91	14.86	2.18	16.73	3.71	10.10	1.37	6.84	1.01	140.02	1.83	242.90	37.80	1.10	1.13	0.46
24 avg.	0.13	0.28	40.15	0.03	0.13	54.92	39.27	71.94	10.12	49.61	12.20	2.78	15.48	2.20	15.94	3.70	9.92	1.21	6.88	0.94	139.82	2.01	242.18	37.78	1.01	1.10	0.47
25-01	0.14	0.34	30.57	0.04	0.10	53.26	35.38	60.03	8.83	48.57	11.34	2.72	12.94	2.06	14.37	3.51	9.23	1.01	7.08	0.86	131.98	1.07	217.90	37.60	1.08	1.13	0.40
25-02*	0.16	0.70	33.57	0.06	0.28	55.43	35.19	59.06	8.80	42.88	9.57	2.70	13.63	2.05	13.55	3.15	9.51	1.22	6.58	0.89	130.08	1.39	208.80	41.30	0.95	1.24	0.43

* These data are excluded from the calculation of average (avg.) values.

Table 4 continued

S.N.	MgO	Al ₂ O ₃	P ₂ O ₅	MnO	FeO	CaO	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Y	Th	ΣREE	Y/Ho	Ce/Ce*	Eu/Eu*	Pr _(SSY) /Yb _(SSY)
25-03	0.08	0.09	34.41	0.03	0.07	56.71	42.31	83.04	11.34	59.62	14.02	3.07	17.39	2.68	16.90	4.12	10.71	1.35	7.64	1.05	155.65	2.44	275.30	37.80	1.11	1.01	0.47
25-04	0.09	0.25	31.71	0.03	0.08	52.94	34.62	57.30	8.91	43.78	9.79	2.69	13.86	2.17	13.53	3.35	8.61	1.11	6.25	0.81	132.13	0.73	206.80	39.40	0.91	1.19	0.46
25 avg.	0.11	0.23	32.23	0.03	0.08	54.30	37.44	66.79	9.69	50.65	11.72	2.83	14.73	2.30	14.93	3.66	9.52	1.15	6.99	0.91	139.92	1.41	233.32	38.28	1.04	1.10	0.44
26-01*	0.28	0.92	34.02	0.08	0.91	55.47	35.50	59.62	7.86	38.89	9.94	2.15	11.82	1.87	13.28	2.80	8.86	1.14	6.49	0.84	115.41	4.98	201.00	41.20	1.08	1.00	0.39
26-02	0.20	0.22	36.84	0.05	0.35	56.95	38.68	62.37	8.77	44.55	9.93	2.33	13.89	2.08	13.68	3.36	8.99	1.24	6.36	0.98	129.62	1.80	217.20	38.60	1.04	1.04	0.44
26-03*	0.26	0.54	37.16	0.05	0.19	56.71	36.38	57.96	7.92	37.85	8.10	2.19	11.68	1.83	12.19	2.98	7.87	1.09	6.22	0.72	112.41	1.66	195.00	37.70	1.01	1.17	0.41
26-04	0.18	0.23	38.62	0.05	0.17	56.77	48.63	96.16	12.10	62.79	13.90	3.36	19.29	2.72	17.56	4.46	11.44	1.49	8.45	1.13	152.32	4.41	303.50	34.10	1.19	1.11	0.46
26 avg.	0.19	0.23	37.73	0.05	0.26	56.86	43.65	79.26	10.44	53.67	11.91	2.85	16.59	2.40	15.62	3.91	10.22	1.37	7.41	1.05	140.97	3.10	260.35	36.37	1.12	1.10	0.45
27-01	0.22	0.27	39.47	0.05	0.08	54.93	34.69	59.01	8.07	42.24	10.32	2.41	12.61	1.81	14.02	3.13	9.82	1.22	6.84	0.92	127.05	1.40	207.10	40.60	1.11	1.12	0.38
27-02	0.14	0.35	38.20	0.04	0.55	52.78	33.81	48.64	6.89	35.99	7.02	1.98	10.82	1.62	12.64	3.08	9.17	1.25	7.20	0.99	130.01	0.51	181.10	42.20	1.06	1.20	0.31
27-03	0.13	0.08	41.12	0.05	0.19	54.61	40.18	61.76	8.77	42.11	10.06	2.43	12.50	2.08	13.59	3.60	10.93	1.35	7.25	0.91	144.22	1.30	217.50	40.10	0.98	1.08	0.39
27-04	0.13	0.07	41.64	0.03	0.07	54.12	45.74	88.60	12.73	64.22	15.38	3.48	20.97	2.98	19.21	4.50	13.35	1.66	8.59	1.03	173.40	1.86	302.40	38.60	1.01	1.04	0.47
27 avg.	0.16	0.19	40.11	0.04	0.22	54.11	38.60	64.50	9.11	46.14	10.70	2.57	14.22	2.12	14.86	3.58	10.82	1.37	7.47	0.96	143.67	1.27	227.04	40.35	1.04	1.10	0.39
28-01	0.17	0.12	39.29	0.08	0.32	55.54	39.69	71.10	9.50	47.85	12.16	2.87	14.44	2.17	14.43	3.45	9.80	1.24	7.10	1.07	132.77	1.42	236.90	38.50	1.09	1.12	0.43
28-02*	0.31	2.36	36.08	0.13	0.51	50.24	39.45	74.99	9.53	48.80	12.31	2.73	14.85	2.14	14.20	3.42	9.40	1.28	6.98	0.87	131.98	1.42	241.00	38.60	1.16	1.06	0.44
28-03	0.14	0.17	42.80	0.07	0.32	53.13	49.50	102.79	13.11	66.12	17.38	4.18	21.23	3.15	20.14	4.66	12.14	1.51	8.02	1.06	167.15	2.00	325.00	35.90	1.14	1.14	0.52
28-04	0.09	0.08	44.25	0.04	0.44	52.24	35.04	59.35	7.98	41.29	10.17	2.39	12.82	1.94	13.05	3.10	8.46	1.16	6.92	1.00	124.55	1.10	204.70	40.20	1.11	1.09	0.37
28 avg.	0.13	0.12	42.12	0.06	0.36	53.64	41.41	77.75	10.20	51.75	13.24	3.15	16.17	2.42	15.87	3.73	10.13	1.30	7.35	1.04	141.49	1.51	255.51	38.21	1.11	1.10	0.44
29-01	0.12	0.19	42.83	0.02	0.16	52.60	37.97	66.15	9.09	45.94	11.01	2.63	14.19	2.18	14.43	3.26	9.54	1.30	7.03	0.85	133.33	1.08	225.60	40.90	1.06	1.09	0.41
29-02	0.07	0.14	41.56	0.03	0.14	52.66	33.25	53.84	7.60	38.23	8.94	2.15	12.05	1.95	13.29	3.29	9.08	1.26	6.89	0.93	133.38	0.85	192.80	40.50	1.03	1.05	0.35
29-03	0.14	0.31	38.28	0.10	0.27	50.77	25.15	36.17	5.31	24.86	5.67	1.24	8.77	1.21	8.12	2.24	7.09	0.87	5.28	0.66	97.42	0.58	132.60	43.50	0.92	0.97	0.32
29-04	0.10	0.26	36.11	0.03	0.11	51.83	38.56	73.04	9.84	52.32	10.91	2.81	16.00	2.06	15.88	3.71	10.12	1.24	6.62	0.76	139.41	1.65	243.90	37.60	1.14	1.20	0.47
29 avg.	0.11	0.22	39.70	0.04	0.17	51.96	33.73	57.30	7.96	40.34	9.13	2.21	12.75	1.85	12.93	3.12	8.96	1.17	6.45	0.80	125.89	1.04	198.70	40.63	1.04	1.10	0.39

* These data are excluded from the calculation of average (avg.) values.