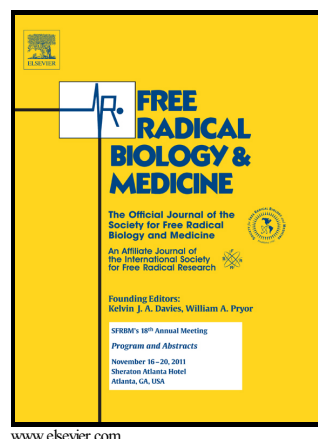


Author's Accepted Manuscript

Increasing extracellular H₂O₂ produces a bi-phasic response in intracellular H₂O₂, with peroxiredoxin hyperoxidation only triggered once the cellular H₂O₂-buffering capacity is overwhelmed

Lewis Elwood Tomalin, Alison Michelle Day, Zoe Elizabeth Underwood, Graham Robert Smith, Piero Dalle Pezze, Charalampos Rallis, Waseema Patel, Bryan Craig Dickinson, Jürg Bähler, Thomas Francis Brewer, Christopher Joh-Leung Chang, Daryl Pierson Shanley, Elizabeth Ann Veal



PII: S0891-5849(16)00094-0
DOI: <http://dx.doi.org/10.1016/j.freeradbiomed.2016.02.035>
Reference: FRB12775

To appear in: *Free Radical Biology and Medicine*

Received date: 14 August 2015
Revised date: 25 February 2016
Accepted date: 29 February 2016

Cite this article as: Lewis Elwood Tomalin, Alison Michelle Day, Zoe Elizabeth Underwood, Graham Robert Smith, Piero Dalle Pezze, Charalampos Rallis, Waseema Patel, Bryan Craig Dickinson, Jürg Bähler, Thomas Francis Brewer, Christopher Joh-Leung Chang, Daryl Pierson Shanley and Elizabeth Ann Veal, Increasing extracellular H₂O₂ produces a bi-phasic response in intracellular H₂O₂, with peroxiredoxin hyperoxidation only triggered once the cellular H₂O₂-buffering capacity is overwhelmed, *Free Radical Biology and Medicine*, <http://dx.doi.org/10.1016/j.freeradbiomed.2016.02.035>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting galley proof before it is published in its final citable form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Increasing extracellular H₂O₂ produces a bi-phasic response in intracellular H₂O₂, with peroxiredoxin hyperoxidation only triggered once the cellular H₂O₂-buffering capacity is overwhelmed

Lewis Elwood Tomalin¹, Alison Michelle Day¹, Zoe Elizabeth Underwood¹, Graham Robert Smith², Piero Dalle Pezze¹, Charalampos Rallis³, Waseema Patel¹, Bryan Craig Dickinson⁴, Jürg Bähler³, Thomas Francis Brewer⁵, Christopher Joh-Leung Chang⁵, Daryl Pierson Shanley^{1,*} and Elizabeth Ann Veal^{1,*}

¹Institute for Cell and Molecular Biosciences, Newcastle University, Framlington Place, Newcastle upon Tyne, NE2 4HH, UK

²Bioinformatics Support Unit, Newcastle University, Framlington Place, Newcastle upon Tyne, NE2 4HH, UK

³ University College London, Department of Genetics, Evolution & Environment and Institute of Healthy Ageing, Gower Street – Darwin Building, London WC1E 6BT, United Kingdom

⁴Department of Chemistry, The University of Chicago, Chicago, IL, 60637 USA

⁵Howard Hughes Medical Institute and Departments of Chemistry and Molecular and Cell Biology, University of California, Berkeley, Berkeley, CA 94720, USA

*Correspondence: e.a.veal@ncl.ac.uk and daryl.shanley@ncl.ac.uk

Peroxiredoxin hyperoxidation signals saturation of the thiol proteome

Abstract:

Reactive oxygen species, such as H₂O₂, can damage cells but also promote fundamental processes, including growth, differentiation and migration. The mechanisms allowing cells to differentially respond to toxic or signaling H₂O₂ levels are poorly defined. Here we reveal that increasing external H₂O₂ produces a bi-phasic response in intracellular

H₂O₂. Peroxiredoxins(Prx) are abundant peroxidases which protect against genome instability, ageing and cancer. We have developed a dynamic model simulating in vivo changes in Prx oxidation. Remarkably, we show that the thioredoxin peroxidase activity of Prx does not provide any significant protection against external rises in H₂O₂. Instead, our model and experimental data are consistent with low levels of extracellular H₂O₂ being efficiently buffered by other thioredoxin-dependent activities, including H₂O₂-reactive cysteines in the thiol-proteome. We show that when extracellular H₂O₂ levels overwhelm this buffering capacity, the consequent rise in intracellular H₂O₂ triggers hyperoxidation of Prx to thioredoxin-resistant, peroxidase-inactive form/s. Accordingly, Prx hyperoxidation signals that H₂O₂ defences are breached, diverting thioredoxin to repair damage.

INTRODUCTION

Reactive oxygen species (ROS) generated by the partial reduction of oxygen during aerobic metabolism, immune cell attack or following exposure to radiation can cause lethal levels of cell damage. Hence, there is a strong driving force to evolve and maintain ROS-protective mechanisms. Nevertheless, altered redox homeostasis and increased oxidative cell damage are associated with the development of many common diseases, including cancer, diabetes, cardiovascular and neurodegenerative diseases. However, there is increasing evidence that low levels of ROS can also have beneficial effects; acting as signalling molecules to regulate diverse biological processes (for a review see (1)). Hence, there is considerable interest in understanding how cells optimise ROS defences to provide adequate protection without compromising ROS-signalling functions. Here we have developed and used a mathematical model as a tool to understand what governs how cells respond to increases in hydrogen peroxide (H₂O₂).

Peroxiredoxins (Prx) are amongst the most prevalent enzymes involved in responses to H₂O₂. Prx are ubiquitous and highly expressed peroxidases which utilise reversibly oxidised cysteine residues to reduce peroxides (for a review see (2)) (Fig. 1A). Intriguingly, peroxiredoxins appear to have dual roles in cancer, acting as tumor suppressors but with

increased Prx expression also associated with poor prognosis tumors and increased metastasis. Prx have also been shown to promote longevity in yeast, worms, flies and mammals (for a review see (3)). There is therefore great interest in understanding how Prx influence cell responses to H_2O_2 .

The catalytic mechanism of the typical 2-Cys Prx subfamily involves the initial reaction of an active, peroxidatic cysteine (Cys_P) with peroxide to form a cysteine-sulfenic acid (SOH) (Fig. 1A). The sulfenylated peroxidatic cysteine then forms a disulfide with a second, resolving cysteine (Cys_R) in an adjacent Prx molecule. In eukaryotes, these Prx disulfides are reduced by the thioredoxin system. However, the sulfenylated peroxidatic cysteines of thioredoxin-coupled Prx are highly susceptible to further 'hyperoxidation' to thioredoxin-resistant sulfinic-derivatives, thus inactivating their thioredoxin peroxidase activity (4) (Fig. 1A). In contrast, bacterial 2-Cys Prx, such as the *E. coli* peroxiredoxin AhpC, are much less sensitive to hyperoxidation (5). Conserved YF and GG(L/V/I)G amino acid motifs found in all hyperoxidation-sensitive Prx are responsible for this sensitivity (5). The evolution of these conserved amino acid motifs suggests that Prx hyperoxidation confers a selective advantage in eukaryotes. Indeed, as described below, several possible functions have been proposed for hyperoxidation of Prx.

In eukaryotes, in which Prx are sensitive to inactivation, H_2O_2 is generated and utilised as a signalling molecule (2). Thus it has been proposed that the thioredoxin peroxidase activity of Prx might act as a barrier to this signalling, and that inactivation of Prx might be important to allow H_2O_2 to regulate target proteins (5). Although, Prx are not hyperoxidized in response to the low levels of H_2O_2 generated in response to growth factors (6), oscillations in the amount of hyperoxidized Prx have been associated with circadian rhythms across a wide range of species (for a review see (7)). Moreover, oscillations in the hyperoxidation of the mouse mitochondrial Prx, Prx3, have been shown to be important for circadian oscillations in p38 activation and adrenal steroid synthesis (8). Nevertheless, it remains unclear whether the hyperoxidation of Prx in any of these contexts serves to

increase the levels of H₂O₂ available for signalling. Moreover, work in the fission yeast *Schizosaccharomyces pombe* has shown that, rather than acting as a barrier, the thioredoxin peroxidase activity of the single *S. pombe* peroxiredoxin, Tpx1, is actually required for the H₂O₂-induced activation of the AP-1-like transcription factor Pap1 (9-11). We have shown that the role of the thioredoxin peroxidase activity of Tpx1 in H₂O₂-induced Pap1 activation is to competitively inhibit the reduction of the active, oxidized form of Pap1 by the thioredoxin-like protein, Tx11 (11). Accordingly, hyperoxidation of Tpx1 to a thioredoxin-resistant form prevents the H₂O₂-induced activation of Pap1 by increasing the availability of reduced Tx11 (11). Based on these studies, it has been proposed that hyperoxidation of Tpx1 to a thioredoxin-resistant form is important to increase the pool of reduced thioredoxin available to repair oxidatively damaged proteins and support the activity of other enzymes, such as the methionine sulfoxide reductase, Mxr1 (12). Consistent with this hypothesis, hyperoxidation of Tpx1 and the *C. elegans* peroxiredoxin PRDX-2 are important for cell survival under acute stress conditions (12, 13). Hyperoxidation of Prx has also been proposed to have other roles, for example, protecting cells against protein aggregation by promoting the ability of Prx to act as a chaperone (14). However, it is still unclear under what conditions extensive Prx hyperoxidation occurs *in vivo* and when and where this might be important.

Although all eukaryotic Prx are inherently sensitive to hyperoxidation, other factors affect this sensitivity. For example, *in vitro* studies revealed that human cytosolic Prx1 is much more sensitive to hyperoxidation than the mitochondrial Prx, Prx3 (15). This increased sensitivity reflects a 10-fold slower rate of Prx1 disulfide formation that increases the risk of further oxidation of the sulfenylated peroxidatic cysteine (16, 17). In addition to the intrinsic biochemical properties of the Prx itself, the *in vivo* sensitivity of Prx to hyperoxidation is also influenced by the local environment. For example, although *in vitro* the ER-localised Prx, Prx4, has a similar sensitivity to hyperoxidation to Prx1, a negligible proportion of Prx4 becomes hyperoxidized to its sulfinic form *in vivo*. This is due to the low abundance of

disulfide reductases in the ER which causes Prx4 disulfides to accumulate instead (18). As well as the availability of disulfide reductases, the extent to which Prx become hyperoxidized *in vivo* will also be influenced by other aspects of its local environment, such as local H₂O₂ concentration, compartmental volumes and competition with other peroxidases. Here we have developed a computational model incorporating both the biochemical properties and *in vivo* environment as a tool to investigate when Prx become hyperoxidized and the effect of hyperoxidation on cell responses to H₂O₂. Notably, our model predicts that cells contain a Prx-independent H₂O₂-removing activity that becomes saturated following exposure to specific concentrations of extracellular H₂O₂. This results in a bi-phasic response to increased levels of ectopic H₂O₂ with Prx hyperoxidation only occurring once this H₂O₂-removing activity is saturated and intracellular H₂O₂ levels start to rise more rapidly. Importantly, we have experimentally confirmed these predictions in both yeast and human cells. As we discuss, our study provides new mechanistic insight into how Prx hyperoxidation and regulation of thioredoxin activity allow cells to implement appropriate responses to the different levels of H₂O₂ encountered *in vivo*.

RESULTS

Quantitative analysis of how the *in vivo* oxidation of the Prx, Tpx1, changes in response to different concentrations of H₂O₂

To facilitate the generation of a mathematical model capable of simulating *in vivo* changes in Prx oxidation, we first obtained quantitative data for the *in vivo* oxidation of the single *S. pombe* peroxiredoxin Tpx1 following exposure of cells for 20s to a range of H₂O₂ concentrations (0 – 6mM) (Fig. 1 and Table S1). Western blot analysis of AMS-treated proteins using anti-Tpx1 antibodies detected multiple Tpx1-containing bands, as confirmed by the absence of these bands in $\Delta tpx1$ mutant cells (Fig. 1B). These included bands with mobilities of ~20kD and ~40kD consistent with representing reduced Tpx1 monomers

(Tpx1SH) and Tpx1 disulfide dimers respectively. Under normal growth conditions the reduced monomeric band (Tpx1SH) was prevalent (Fig. 1B, Table S1). However, following treatment with H_2O_2 , the levels of Tpx1SH decreased with the concomitant increased formation of disulfide dimers and higher molecular weight bands (>40kD), which likely include mixed disulfides with other proteins (Fig.1B and data not shown). The sensitivity of all of these bands to reduction by beta-mercaptoethanol to 20kD forms (Fig. S1A) indicating that each represents a different disulfide-bonded form of Tpx1.

Although difficult to resolve, as expected, a magnified image revealed three distinct bands with mobilities consistent with Tpx1 disulfide homodimers (40kD) (Fig.1C). The differences in the mobility of these three Tpx1 dimer bands was consistent with each representing a different redox state, as observed previously for Tpx1 and human Prx (15, 16, 19) (Fig.1D). Indeed, in samples treated with NEM, which alkylates reduced cysteines producing a minimal increase in MW (0.0125kD), a single Tpx1-containing band was detected at ~40kD (Fig.1E). This confirms that the differences in the mobility of the 3 bands in AMS-treated samples reflect the different numbers of cysteine residues available to react with AMS, which increases the MW by 0.64kD per reduced cysteine. The single Tpx1 disulfide dimer band detected under normal conditions was significantly retarded by AMS. This is consistent with this band representing Tpx1-Tpx1 dimers containing a single disulfide bond between one peroxidatic (Cys_P) and one resolving cysteine (Cys_R) with the other cysteines reduced and AMS-reactive (Tpx1ox#1)(Fig.1D-E). In cells treated with 200 μ M H_2O_2 two additional Tpx1 disulfide bands were detected in the AMS-treated samples. The similar mobility of the lowest of these bands in NEM and AMS-treated samples suggests that it lacks any reduced cysteine thiols to react with NEM or AMS, consistent with it representing Tpx1 disulfide dimer containing two disulfide bonds (Tpx1ox#2) (Fig.1D-E). The third band detected after treatment with 200 μ M H_2O_2 (Tpx1ox:SOOH+AMS) had an intermediate mobility. This was consistent with the binding of a single AMS molecule to a single reduced cysteine thiol, as would be expected for Tpx1 disulfide dimers containing a single disulfide

bond and a hyperoxidized Cys_{SP} (Tpx1ox:SOOH) with the remaining Cys_R available to react with AMS (Fig.1D-E). This form has also been detected using anti-PrxSO₃ antibodies specific to the sulfenylated/sulfonylated peroxidatic cysteine (12). Thus we confirmed that the three distinct Tpx1 disulfide dimers we detected represented redox states previously described for *S. pombe* and human peroxiredoxins (15, 16, 19). In addition to the bands at 40kD, a band at 55kD was also detected (Fig. 1B, Fig. 2A-B, Fig. S1) that represents a Tpx1 dimer in a disulfide complex with Trx1 (Fig. S1B), a reaction intermediate in the reduction of Tpx1ox#2 by Trx1. Changes in the relative intensities of the various Tpx1 forms in cells treated with different H₂O₂ concentrations indicated that, as expected, Tpx1ox#1, Tpx1ox#2 and Tpx1ox:SOOH become the most prominent forms at the lowest, mid and high H₂O₂ concentrations respectively (Fig. 1 and Table S1).

Quantitative analysis of *in vivo* changes in the oxidation of Tpx1 with time following exposure to H₂O₂

To allow development of a dynamic model, we obtained kinetic data; determining how the relative abundance of different Tpx1 redox forms changed with time (≤ 600 sec) following treatment with 100 μ M or 200 μ M H₂O₂. This time-course was selected to represent the initial H₂O₂ response, before H₂O₂-induced increases in mRNA levels (20, 21) have had any significant effect on total Tpx1 protein levels, or the levels of other proteins which might impact on Tpx1 oxidation (9, 10, 22) (Fig. 2 and data not shown). Tpx1ox#1 and Tpx1ox#2 were detected following treatment with 100 μ M H₂O₂, but the relative levels of these forms did not change with time (Fig. 2A). Importantly, there was negligible formation of Tpx1ox:SOOH during the 100 μ M time course (Table S1), as was confirmed using antibodies specific to hyperoxidized Tpx1 (Fig. S2). In contrast, as expected (Fig. 1) Tpx1ox#1, Tpx1ox#2 and Tpx1ox:SOOH dimers were all detected in cells treated with 200 μ M H₂O₂ (Fig. 2B-C). However, the intensity of the Tpx1 dimer bands decreased over time with the concomitant formation of a Tpx1-containing monomeric band (~20kD). The increased mobility, compared with the AMS-reactive Tpx1-SH monomer band detected before addition of H₂O₂, suggests

this H₂O₂-induced band represents monomeric Tpx1 containing a single AMS-reactive cysteine thiol. This is consistent with the hyperoxidized Tpx1SOOH monomer previously observed following treatment of cells with higher H₂O₂ concentrations (9-12). The formation of Tpx1-SOOH monomer was confirmed using antibodies specific to hyperoxidized Tpx1 (Fig. 2C, S2). The 200 μ M time course data thus suggests that Tpx1ox:SOOH appears rapidly and is converted to Tpx1SOOH over time with the reaction essentially complete by 600s. The detection of Prxox:SOOH dimers prior to PrxSOOH monomers is consistent with previous studies of Prx hyperoxidation (15, 19). At 500 μ M and 1000 μ M H₂O₂ the monomeric Tpx1SOOH form was maximal by 60s, suggesting that increased H₂O₂ concentration results in an increased rate of formation of Tpx1SOOH (Fig. S2). Importantly, consistent with published studies (9, 10, 22), western blot analysis of beta-mercaptoethanol treated samples confirmed that differences in the intensity of different Tpx1-containing bands did not reflect changes in total Tpx1 levels (Fig. 2A-B). Thus quantitative analysis of images, obtained for multiple independent biological repeats of these experiments (Figs.1-2), was used to estimate the concentrations of each Tpx1 oxidation state at each time point and level of H₂O₂ (Table S1).

A kinetic model of Tpx1 oxidation was developed that can replicate the experimental data

Computational models to describe Tpx1 oxidation were then constructed using reaction networks and parameters that were selected based on published mechanisms, *in vitro* kinetic data, and our *in vivo* experimental data (Figs. 1-2, Tables S1, S2 and S3). We carried out qualitative and quantitative assessments of the ability of each model to describe the experimental data. This led to the selection of a final model, depicted in Fig. 3A, that used the parameter set in Table S3. This final model was able to simulate the removal of H₂O₂ from the extracellular space, indicating that this model could effectively represent the peroxide-removing activity of the cells (Fig. 3B). The final model also simulated the changes in the concentrations of the different Tpx1 redox states following 20s exposure to H₂O₂

concentrations between 0-1000 μM (Fig. 3C-G) or up to 600s following exposure to 100 or 200 μM H_2O_2 (Fig. 4). Importantly, the parameters that were predicted for this model were similar to published values (Table S3). For instance, the parameters $k_{\text{disulph_red1}}$, $k_{\text{disulph_red2}}$ were predicted to be 0.190 and 0.143 $\mu\text{M}^{-1}\text{s}^{-1}$ respectively (Table S3, Figs. S3-4), broadly consistent with experimentally determined values (23, 24). This indicated that the model accurately represents the rapid Trx1-mediated reduction of the disulfide bonds in Tpx1ox#1 and Tpx1ox#2. Interestingly, the rate constant for the reduction of Tpx1ox:SOOH, $k_{\text{disulph_red3}}$, was predicted, to be 0.029 $\mu\text{M}^{-1}\text{s}^{-1}$, 5 fold lower than either $k_{\text{disulph_red1}}$ or $k_{\text{disulph_red2}}$ (Table S3, Figs. S3-4). This suggests that Trx1 may be less efficient at reducing the disulfide bond in Tpx1ox:SOOH compared with the other Tpx1 disulfide dimers. Notably, the first order rate constant for Tpx1 disulfide formation (rate constant $k_{\text{disulph_form2}}$) estimated by our model as 3.44 s^{-1} was in a similar range to the rates of disulfide formation estimated for human Prx1 and Prx3 (2 s^{-1} and 20 s^{-1} respectively) from *in vitro* experimental investigations (16). This supports the use of our model and *in vitro* studies as complementary tools to understand the *in vivo* oxidation of yeast and human Prx.

Modelling suggests that oxidation of the peroxidatic cysteine thiol (Cys_P) in Tpx1ox#1 is not the major route for further oxidation of Tpx1

Having built and parameterised a dynamic model capable of simulating the *in vivo* oxidation of Tpx1 (Figs. 3-4), we tested which aspects of our model were important for simulation of the experimental data by comparing the ability of our final model and 2 alternative models to simulate the dynamics of Tpx1 oxidation. Although difficult to demonstrate experimentally, it seemed likely that the reaction of the free Cys_P in Tpx1ox#1 with H_2O_2 to form Tpx1ox:SOH would be involved in the formation of both the Tpx1ox#2 disulfide dimer and Tpx1ox:SOOH. However, unexpectedly, alternative Model A which included this additional reaction (Table S4) was less able to simulate the experimentally observed dynamics of Tpx1 oxidation (Fig. 5) and had an increased AIC compared with the final model (Table S5). Accordingly, this alternative model was rejected in favour of our final

model (Fig. 3A) in which the simultaneous oxidation of neighboring Cys_P-SH (disulph_{form1b}) is the major route of Tpx1ox:SOH formation. Interestingly, this suggests that the reduced Cys_P in Tpx1ox#1 may be less sensitive to H₂O₂ and that, *in vivo*, the reaction of neighboring peroxidatic cysteines with H₂O₂, prior to disulfide bond formation, may be prerequisite for the generation of hyperoxidized SOOH derivatives.

Although an alternative route for the formation of hyperoxidised Tpx1 would be via the oxidation of the sulfenylated peroxidatic cysteine in Tpx1-SOH, the inclusion of this reaction was not required to fit the experimental data, and actually rendered models less able to simulate the experimentally observed dynamics of Tpx1 oxidation (data not shown). Thus, we conclude that this route makes a negligible contribution to the pool of hyperoxidized Tpx1SOOH detected at these H₂O₂ concentrations. This is consistent with previous work (19) suggesting that the hyperoxidation of Tpx1ox:SOH precedes the formation of Tpx1SOOH disulfide dimers and is the dominant route for Tpx1 sulfinylation *in vivo*.

The removal of H₂O₂ by an additional process/es (H₂O₂_metab) was required to explain the dynamics of Tpx1 hyperoxidation

The final model (Fig.3A) accurately simulated our experimental observation that there was negligible formation of either of the hyperoxidized Tpx1 forms (Tpx1ox:SOOH & Tpx1-SOOH) at H₂O₂ concentrations below 200μM (Figs. 1-4 and S2). However, this particular feature of the experimental data was not captured by alternative Model B which did not contain the reaction 'H₂O₂_metab' (Table S4). Instead, alternative Model B predicted that Tpx1 would be hyperoxidized at all H₂O₂ concentrations, even below 200μM (Alt Model B Fig. 5C and E). This suggests that, although Tpx1 reacts with extremely low H₂O₂ concentrations to generate Tpx1-Tpx1 disulfides, other components of the cell's H₂O₂-buffering capacity normally inhibit hyperoxidation (hyp_ox in Fig. 3) *in vivo*. Together these results suggest that the sensitivity of peroxiredoxins to hyperoxidation is not just dependent

on the kinetics of Prx reaction with H_2O_2 but is also heavily influenced by the other peroxide-reactive molecules present in their *in vivo* environment.

As predicted by our model, there is a two phase relationship between intracellular and extracellular H_2O_2 concentration

To further investigate the conditions which cause Prx to become hyperoxidized, we used our model to predict how increasing extracellular H_2O_2 concentrations ($[\text{H}_2\text{O}_2]_{\text{ex}}$) would affect the intracellular H_2O_2 concentration ($[\text{H}_2\text{O}_2]_{\text{int}}$) (Fig. 6A). Interestingly, the final model predicted that $[\text{H}_2\text{O}_2]_{\text{ex}} < 100 \mu\text{M}$ would cause little net change in $[\text{H}_2\text{O}_2]_{\text{int}}$ but that at $[\text{H}_2\text{O}_2]_{\text{ex}} > 150 \mu\text{M}$ the $[\text{H}_2\text{O}_2]_{\text{int}}$ increases linearly with increasing $[\text{H}_2\text{O}_2]_{\text{ex}}$ (Fig. 6A). This two phase relationship between intracellular and extracellular $[\text{H}_2\text{O}_2]$ was dependent on the reaction $\text{H}_2\text{O}_2_{\text{metab}}$ (Fig. 3A, 6A). This reaction, absent from alternative Model B, represents the cell's other peroxidase activities and is required for the model to accurately simulate the dynamics of Tpx1SOOH formation (Fig. 5C, 5E and 6B). Thus our model predicts that this peroxidase activity is saturated following treatment of wild-type cells with $150 \mu\text{M}$ H_2O_2 . To test this prediction we used the H_2O_2 sensitive fluorescent dye, PF3, to measure the rate at which $[\text{H}_2\text{O}_2]_{\text{int}}$ increases in cells following exposure to different $[\text{H}_2\text{O}_2]_{\text{ex}}$ (25). Although we detected a steady increase in the accumulation of intracellular H_2O_2 in cells exposed to low concentrations of H_2O_2 , as predicted by our model, the rate of intracellular H_2O_2 accumulation was much faster in cells exposed to extracellular H_2O_2 concentrations greater than $150 \mu\text{M}$ (Fig. 6C). Notably, as predicted by the model, and experimentally confirmed, significant formation of hyperoxidized Tpx1 (Tpx1ox:SOOH) only begins to occur following exposure of cells to similar $[\text{H}_2\text{O}_2]_{\text{ex}}$ to those saturating $\text{H}_2\text{O}_2_{\text{metab}}$ (Fig. 6B). Having established that our model was able to make accurate predictions in *S. pombe* we examined whether these findings held true in human cells.

To test whether there was also a biphasic effect of increasing $[\text{H}_2\text{O}_2]_{\text{ex}}$ on the rate at which intracellular H_2O_2 levels increase in human cells, we examined how exposure to

different extracellular concentrations of H_2O_2 affected the intracellular H_2O_2 concentration in human embryonic kidney (HEK293) cells. A lower level of extracellular H_2O_2 was required to breach the H_2O_2 -buffering capacity of HEK293 cells than *S. pombe*, possibly reflecting the increased H_2O_2 -permeability of human cells (Fig. 6C-D). Nevertheless, we observed a similar 2 phase relationship between extracellular and intracellular H_2O_2 concentrations; human cells were able to maintain a low intracellular H_2O_2 concentration following exposure to lower concentrations of H_2O_2 but once this buffering capacity was exceeded ($\geq 40\mu M$) the intracellular H_2O_2 concentration rose more rapidly (Fig. 6D). Furthermore, using anti-PrxSO₃ antibodies that recognise the hyperoxidized forms of all 4 human 2-Cys Prx (18, 26), we found that, similar to our findings in yeast (Fig. 6B), the *in vivo* hyperoxidation of human Prx only began to increase once the H_2O_2 -buffering capacity of HEK293 cells was overwhelmed and intracellular H_2O_2 levels started to increase more rapidly (Fig. 6E-F). Together these data reveal that there is a biphasic increase in intracellular H_2O_2 in response to increases in extracellular H_2O_2 in yeast and human cells, with cells able to buffer exposure to low levels of extracellular H_2O_2 more effectively. Moreover, these data are consistent with the model's prediction that saturation of this buffering capacity triggers the hyperoxidation of Prx.

Hyperoxidation of Prx detects the point at which other peroxide-removing processes (H_2O_2 _metab) become saturated

Accordingly, based on these modelling and experimental approaches, we propose that Prx hyperoxidation only occurs *in vivo* once the cell's peroxide-removing capacity becomes saturated and intracellular H_2O_2 levels start to increase more rapidly. This hypothesis, is consistent with *in vitro* work demonstrating that catalase is able to specifically inhibit the hyperoxidation of human Prx1 (16). However, it was possible that the increased accumulation of H_2O_2 observed in cells exposed to higher $[H_2O_2]_{ex}$ e.g. $>150\mu M$ H_2O_2 in *S. pombe* (Fig. 6C) was due to the coincident hyperoxidation of Prx to peroxidase-inactive forms, rather than the saturation of other cellular peroxidases, To test this possibility we began by examining the effect of increasing $[H_2O_2]_{ex}$ on intracellular H_2O_2 in $\Delta tpx1$ mutant *S.*

pombe. Notably, loss of Tpx1 had little effect on the H₂O₂-buffering capacity of cells (Fig. 7A-B). Furthermore, the peroxide-buffering capacity of cells expressing increased levels of either wild-type Tpx1 or a Tpx1 isoform that is 10x more resistant to hyperoxidation (12, 27), were also saturated following exposure to 150µM H₂O₂ (Fig. 7C-D). Together these data strongly suggest that the increase in the rate of H₂O₂ accumulation in cells exposed to [H₂O₂]_{ex}>150µM is due to the saturation of other cellular peroxide-buffering processes, rather than the inactivation of the thioredoxin peroxidase activity of Tpx1. Instead, this suggests that the hyperoxidation of Tpx1 *in vivo* actually detects the point at which the H₂O₂ buffering capacity of the cell is overcome and intracellular levels of H₂O₂ begin to rise more rapidly.

Thioredoxin is required to buffer low levels of extracellular H₂O₂

Having established that Tpx1 activity did not make an important contribution to the cell's ability to maintain low intracellular H₂O₂ levels following exposure to exogenous H₂O₂, we next investigated which other enzyme/s might be responsible for this H₂O₂-buffering capacity. Catalase has previously been demonstrated to protect Prx against hyperoxidation *in vitro* (16). However, although Ctt1 appears to limit the increase in intracellular H₂O₂ at higher concentrations of H₂O₂ (Fig. 7E), the similar 2 phase relationship between extracellular and intracellular H₂O₂ concentrations in $\Delta ctt1$ mutant cells suggests that catalase does not make an important contribution to the cell's ability to buffer low levels of H₂O₂ <150µM (Fig. 7E). It was possible that the Gpx/GSH system (28), or other subfamilies of Prx, contributed to the H₂O₂-buffering capacity of *S. pombe*, this seemed unlikely given the minimal effect that deletion of the genes encoding these enzymes (*gpx1*, *pmp20* or *dot5*) has on H₂O₂ resistance (10). Indeed, loss of Gpx1 had no effect upon the bi-phasic relationship between extracellular and intracellular H₂O₂ (7F).

Our previous studies have established that treatment with 0.2mM H₂O₂ causes the majority of Trx1 and Tx11 to become rapidly oxidized (11,12). As broad specificity oxidoreductases, thioredoxin family proteins are important cofactors for many enzymes and

for the reduction of other oxidized proteins that may also impact on intracellular H_2O_2 levels (12, 29, 30). Although, Tpx1 and Gpx1 were not important for buffering low levels of H_2O_2 (Fig. 7), it was still possible that the point at which the H_2O_2 -buffering capacity is breached might reflect the point at which thioredoxin reductase activity becomes limiting for the removal of H_2O_2 by other thioredoxin-dependent activities. Hence, to test whether thioredoxin-dependent processes are important for inhibiting increases in intracellular H_2O_2 in cells exposed to $\leq 150\mu\text{M}$ H_2O_2 , we examined how extracellular H_2O_2 treatment affected intracellular H_2O_2 concentration in mutant *S. pombe* lacking, or ectopically expressing additional thioredoxin and/or Trr1. Strikingly, the bi-phasic relationship between extracellular $[\text{H}_2\text{O}_2]$ and the rate of increase in intracellular $[\text{H}_2\text{O}_2]$ was lost in $\Delta\text{trx1}\Delta\text{txl1}$ cells in which both thioredoxin family proteins are absent (Fig. 8A). This linear relationship between intracellular and extracellular H_2O_2 concentrations up to $500\mu\text{M}$ H_2O_2 in $\Delta\text{trx1}\Delta\text{txl1}$ mutant cells indicated that the peroxide-removing activity that is normally saturated by low levels of H_2O_2 (H_2O_2 _metab) requires thioredoxin (Fig. 8A).

Accordingly, we tested whether the point at which the H_2O_2 -buffering capacity is breached might reflect the point at which thioredoxin reductase activity becomes limiting for the removal of H_2O_2 by thioredoxin-dependent processes. Importantly, overexpressing Trr1 prevented the rapid and sustained oxidation of Trx1 in cells treated with 0.2mM H_2O_2 , indicating that thioredoxin reductase activity, rather than NADPH levels, normally limits Trx1 reduction under these conditions (Fig.8B). However, overexpressing Trx1 and/or Trr1 had a negligible effect on the relationship between extracellular $[\text{H}_2\text{O}_2]$ and the rate of increase in intracellular $[\text{H}_2\text{O}_2]$ (Fig. 8C-E). This suggests that the saturation of the cell's H_2O_2 -buffering capacity is not due to the saturation of thioredoxin reductase activity.

Computational modelling and experimental studies have suggested that, although most protein thiols are relatively insensitive to oxidation, the reversible oxidation of H_2O_2 -sensitive protein cysteine-thiols (PSH) (the thiol proteome) might still make an important contribution to the peroxide-buffering capacity of human cells (31, 32). Indeed, it has been

estimated that the concentration of oxidant accessible protein thiols is around 13mM, similar to the total concentration of glutathione (32). Although most cellular thiols have a much lower reactivity with H_2O_2 than Prxs (33), the sheer abundance of these H_2O_2 -reactive thiols could mean that collectively they make a large contribution to the H_2O_2 -buffering capacity of the cell. Moreover, previous work has shown that, like Trx1 and Tx11, the thiol proteome is also maximally oxidised following exposure of *S. pombe* to 200 μM H_2O_2 (29). Hence, by causing constitutive, maximal oxidation of the thiol proteome (29), the absence of both cytosolic thioredoxin family proteins, Trx1 and Tx11, could potentially ablate the H_2O_2 -buffering ability of this thiol pool. To test whether, despite their lower reactivity, the intracellular concentration of H_2O_2 -oxidizable protein-thiols could be sufficient to make a significant contribution to the cell's capacity to buffer H_2O_2 , we constructed a simple computer model, assuming 13mM protein thiols with an average reactivity with H_2O_2 of $0.0005\mu\text{M}^{-1}\text{s}^{-1}$. This model was able to simulate the *in vivo* bi-phasic response in intracellular H_2O_2 , illustrating the concept that the oxidation of 13mM reduced protein-thiols (Table S6, S7), could prevent rises in intracellular H_2O_2 following treatment with up to 0.2mM H_2O_2 (Fig. 8F). This model also predicted that, provided the thioredoxin-reactivity of the resulting protein disulfides is low, the initial concentration of reactive thiols, will have a greater influence on the buffering capacity than the availability of thioredoxin (Trx1) (Fig. 8F, S5). Accordingly, this model simulates the negligible effect of overexpressing Trx1 and/or Trx11 on the relationship between extracellular $[\text{H}_2\text{O}_2]$ and the rate of increase in intracellular $[\text{H}_2\text{O}_2]$ (Fig. 8C-F). This is consistent with the idea that the initial oxidation of protein thiols, rather than other thioredoxin-dependent enzymatic processes, could be responsible for the observed buffering of low levels of external H_2O_2 (Fig. 7, 8C-E). Importantly, our model also recapitulated the effect of loss of Trx1 and Tx11 on intracellular H_2O_2 (Fig. 8A, 8F). Given that systems of redox-couples have previously been demonstrated to display 'apparent' Michaelis Menten kinetics (34,35), this suggests that the V_{max} of the H_2O_2 _metab, predicted in our model (Fig. 3, Table S3), could represent the apparent V_{max} for removal of H_2O_2 by oxidation of the thiol proteome. Clearly the representation of the diverse protein-thiol pool in our model is a gross oversimplification

and we note that the average reactivity we use requires a larger pool of protein thiols to react rapidly with H_2O_2 than might be expected from *in vitro* studies. Hence, although our experimental data indicate that thioredoxin is vital (Fig. 8A), it is possible that H_2O_2 -reactants, such as glutathione or methionine, not investigated here, also make important contributions to the cell's capacity to buffer the intracellular environment against rises in extracellular H_2O_2 . Nevertheless, our model and data are consistent with other studies that have suggested that the oxidation of the thiol proteome makes a major contribution to the *in vivo* removal of H_2O_2 (Fig. 8) (31,32). Importantly, here we show that the H_2O_2 -induced hyperoxidation of Prx only occurs once this peroxide-buffering capacity is saturated (Fig. 6, 8G). Based on these findings, we propose that hyperoxidation is a response to increased intracellular H_2O_2 , allowing downstream signalling, as well as protective chaperone functions of hyperoxidized peroxiredoxin (8,14). These findings are also consistent with our previous studies which indicated that peroxiredoxin hyperoxidation is important for thioredoxin-mediated repair and cell survival (Fig. 8G) (12).

DISCUSSION

The role of peroxiredoxins in cell responses to H_2O_2 has come under considerable scrutiny in recent years since the discovery that the thioredoxin peroxidase activity of Prx is sensitive to inactivation by H_2O_2 . Several functions have been proposed for the H_2O_2 -induced inactivation of this peroxidase activity. However, to assess if/when any or all of these functions are important, it is important to understand under which circumstances Prx become hyperoxidized (inactivated) *in vivo*. Here we have developed a mathematical model describing the kinetics of oxidation of the single *S. pombe* 2-Cys Prx in response to H_2O_2 as a tool to investigate the precise circumstances that cause Prx to become hyperoxidized *in vivo*. This model has made several unexpected predictions.

Firstly, our model suggested that there is a biphasic relationship between extracellular and intracellular H_2O_2 such that exposure to low levels of H_2O_2 produces only

small increases in the intracellular H_2O_2 concentration, whereas above a certain threshold the cell's peroxide-removing capacity becomes overwhelmed and intracellular H_2O_2 concentrations increase at a much faster rate. Importantly, this prediction was experimentally confirmed in both yeast and human cells.

Secondly, our model reveals that this peroxide-removing activity protects Tpx1 from hyperoxidation. This explains why hyperoxidized Tpx1 is only detected following exposure to extracellular concentrations of H_2O_2 above $100\mu\text{M}$. Notably, this is consistent with work in mammalian cells, which can effectively buffer extracellular H_2O_2 concentrations of $10\mu\text{M}$ that cause some hyperoxidation of Prx2 *in vitro* (16, 36) but which require higher extracellular H_2O_2 concentrations to increase intracellular H_2O_2 levels and also cause *in vivo* hyperoxidation of Prx (6, 36, 37). Indeed, not only do we confirm that the biphasic relationship between extracellular and intracellular H_2O_2 also holds true in HEK293 cells, but also that hyperoxidation of human Prx only occurs once the ability of these cells to buffer H_2O_2 becomes saturated (Fig. 6D-F).

Thirdly, our modelling suggests that the main route for formation of the hyperoxidized Prx involves reaction of 2 neighboring catalytic centres (Cys_P) with H_2O_2 prior to formation of Prx disulfides. Interestingly, this suggests that the reduced Cys_P in Tpx1ox#1 may be less sensitive to H_2O_2 than that in reduced Tpx1. This still needs to be experimentally validated, but could be explained by the propensity of disulfide formation to destabilize Prx decamers resulting in Prx dimers which are approximately 100 fold less reactive with H_2O_2 (38, 39). Intriguingly, this would provide an explanation as to why Prx hyperoxidation only occurs once the peroxide-buffering capacity of the cell is saturated, as the more rapid increase in intracellular H_2O_2 concentration would greatly increase the probability of neighboring Cys_P reacting with H_2O_2 prior to disulfide formation.

Fourthly, our modelling and experimental investigations together indicate that the peroxide-removing processes that are saturated in *S. pombe* are likely to include the

reversible oxidation of the thiol proteome. Notably, although the H_2O_2 -buffering capacity predicted by our model is eliminated in the absence of disulfide reductase activity, loss of Trx1 alone had little effect upon the ability of cells to buffer intracellular H_2O_2 (data not shown). Moreover, ectopically overexpressing Trx1 and Trx1 did not increase the intracellular H_2O_2 -buffering capacity (Fig. 8C-E). This suggests that the maximum rate of H_2O_2 -removal is independent of the availability of these disulfide reductases. Instead, this suggests that the limiting factor for H_2O_2 -buffering is the initial availability of reduced protein thiols, with the low affinity of these protein disulfides for Trx1/Txl1 limiting the rate at which they are regenerated. It has been estimated that the concentration of oxidant accessible protein thiols is around 13mM, similar to the total concentration of glutathione (32). Although it is possible that other activities, such as glutathione and methionine oxidation, also contribute, this is consistent with the saturation of this abundant pool of free protein-thiols making the most important contribution to the saturatable H_2O_2 buffering capacity revealed by our computer model.

In response to 200 μM H_2O_2 both thioredoxin family proteins (Trx1 and Txl1) are completely oxidised in the reduction of Tpx1-Tpx1 disulfides (11, 12). This Tpx1-dependent inhibition of Txl1, allows the sustained activation of the Pap1 transcription factor which promotes the expression of a host of oxidative stress defence enzymes (11). However, thioredoxin (Trx1) is also vital for the reduction of oxidized protein cysteine-thiols and the activity of methionine sulfoxide reductase enzymes. Indeed, the thiol proteome is maximally oxidised in cells where Trx1 and Txl1 are inhibited genetically, or as a result of Tpx1-dependent oxidation (29). Hence, it is logical that, once reactive protein thiols have fully reacted with H_2O_2 , it is important to target thioredoxin activity away from Prx disulfides, for which they have much greater affinity, towards reducing these oxidised cysteine and methionine residues. Hyperoxidized Prx cannot be reduced by thioredoxin. Accordingly, it has been proposed that, by converting Prx to a form that is no longer a thioredoxin substrate, Prx hyperoxidation enables thioredoxin to be targeted to other oxidised proteins

instead (12). Interestingly, our model predicts that Tpx1 disulfides are less efficiently reduced by thioredoxin if the non-bonded Cys_P is sulfinylated (Tpx1ox:SOOH). It is possible that, under conditions where thioredoxin reductase is limiting, this may also help redirect thioredoxin activity towards other substrates, for which it has a much lower affinity. Indeed, consistent with these findings, the hyperoxidation of Tpx1 is important to maintain thioredoxin activity, allowing the repair of oxidized proteins, and cell survival following exposure of cells to higher concentrations of H₂O₂ (12).

The high H₂O₂-scavenging activity of Prx, such as Tpx1, when recycling systems are provided in excess *in vitro*, supports previous reports suggesting that the thioredoxin peroxidase activity of Tpx1 is important for removing the low levels of endogenous H₂O₂ generated during normal aerobic growth and metabolism (16, 40). However, here we show that this thioredoxin peroxidase activity makes a negligible contribution to *S. pombe*'s capacity to buffer the internal environment against extracellular increases in H₂O₂ (Fig. 7A-D). Perhaps this is not surprising given that under these *in vivo* conditions thioredoxin reductase activity is limiting, preventing the efficient recycling of Tpx1 disulfides (Fig. 8B) (12). Indeed, our previous work, has suggested that, rather than its H₂O₂-detoxifying capacity, the important role of the thioredoxin peroxidase activity of Tpx1 in cells exposed to these levels of exogenous H₂O₂ is to promote the oxidation of Tx11 and hence H₂O₂-induced gene expression and oxidative stress resistance (11). Consistent with this, a model of Prx2 oxidation in human erythrocytes has also suggested that the abundance and peroxidase activity of the Prxs favours a signalling rather than a peroxide-detoxification role (41).

The hyperoxidation of 2-Cys Prx has been identified as a conserved feature of circadian rhythms in eukaryotes (42-44) (for a review see (7)). Our model suggests that hyperoxidation only occurs when the cell's capacity for H₂O₂-removal is breached, allowing the concerted reaction of 2 H₂O₂ molecules with adjacent peroxidatic cysteines. This raises the possibility that the hyperoxidized Prx detected in each of these organisms reflects a transient, daily increase in the intracellular H₂O₂ concentration above the cell's peroxide-

buffering capacity. Consistent with the possibility that a cyclic increase in ROS might be important for circadian rhythms, Nrf2, the transcription factor controlling the levels of peroxidase-removing enzymes in mammals, was recently shown to be regulated in a circadian pattern (45). If an increase in intracellular H_2O_2 is important for circadian control of cellular activities, then it is possible that loss of this regulation may contribute to the deleterious effects that can be associated with increased dietary antioxidants and constitutively activated stress defences.

The inactivation of 2-Cys Prx by hyperoxidation has been proposed to allow H_2O_2 to act as a signal (5). However, where the amount of hyperoxidized has been compared with the total Prx, only a small proportion of the total pool of 2-Cys Prx appears to be hyperoxidized under normal growth conditions/during circadian rhythms (6, 8, 46). Moreover, hyperoxidation of Prx is undetectable in response to the low H_2O_2 levels produced in response to growth factor activated NADPH oxidases (6, 37). As 2-Cys Prx are highly abundant, and only 1 of a repertoire of peroxidase enzymes, it has seemed unlikely that inactivation of a small proportion would significantly impact on intracellular H_2O_2 levels. Indeed, as predicted by our model, our experimental data suggests that the complete inactivation of Tpx1, either by deletion or hyperoxidation, has minimal effect on *S. pombe*'s ability to prevent the intracellular accumulation of H_2O_2 (Fig. 7). Instead, our model is consistent with other work suggesting that hyperoxidation of Prx may have other functions in signalling or protein homeostasis (12, 14, 16, 47, 48).

In summary, our study provides new insight into the underlying causes and function of Prx hyperoxidation. Moreover, the discovery that extracellular increases in H_2O_2 produce non-linear increases in intracellular levels, which are dependent upon the levels of thioredoxin activity, has important implications for the host of studies which have used a bolus of H_2O_2 either as a stress or signalling stimulus. Indeed, the model we have developed provides an important new tool to predict responses to altered redox conditions. For example, our model reveals how differences between the thioredoxin or Prx activity in

individual cells could precisely tailor the sensitivity/response of specific cells within a population to H₂O₂ signals and oxidative stress. Therefore, this has important implications for how dynamic redox changes initiate changes in cell function and behaviour during normal physiology and in disease.

MATERIALS AND METHODS

Cell culture conditions

The *S. pombe* strains (Table S8) and human kidney (HEK293) cells used in this study were maintained using standard media and growth conditions. For experiments, *S. pombe* were grown with agitation at 30°C in 50ml Edinburgh minimal media (EMM) supplemented with 0.48mM histidine, 0.56mM adenine, 0.67mM uracil, 1.91mM leucine. In experiments involving $\Delta\text{trx1}\Delta\text{txl1}$ cells, which are auxotrophic for cysteine, media was also supplemented with 0.52mM cysteine. HEK293 cells were grown in a humidified CO₂ incubator at 37°C in 24ml Dulbecco's Modified Eagles Medium (DMEM) supplemented with 10% (v/v) foetal calf serum (FCS), 100 units/ml penicillin, 100µg/ml streptomycin, 2mM L-glutamine and 1% non-essential amino acids (NES).

Quantitative measurement of Prx or FlagTrx1 oxidation

S. pombe: 50ml cultures of exponentially growing *S. pombe* (OD 0.4-0.5) were harvested before and after exposure to a range of H₂O₂ concentrations. At specific time points following addition of H₂O₂, 3ml of culture (2.4 – 4.0 x 10⁷ cells) was harvested by adding an equal volume of 20% Trichloroacetic acid (TCA). Protein extracts were prepared essentially as described previously (49) but without phosphatase treatment. Proteins were re-suspended and incubated in 100mM Tris-HCl pH 8.0, 1% SDS, 1mM EDTA, 1mg/ml PMSF containing 25mM AMS or 25mM NEM for 30min at 25°C then 5min at 37°C. *Human embryonic kidney cells*: — A 10cm plate of confluent HEK293 cells was washed three times with Dulbecco's

phosphate-buffered saline (DPBS) (Sigma), then incubated for 10 min at 37°C in DPBS supplemented, as indicated, with H₂O₂ (Sigma). Cells were washed three times in DPBS and then re-suspended in 500µl of lysis buffer (50mM Tris-HCl pH 7.5, 150mM NaCl, 0.5% NP40 (IGEPAL), 10mM imidazole, 2µg/ml pepstatin, 2µg/ml leupeptin, 100µg/ml PMSF and 1%(v/v) aprotinin). Western blotting: Insoluble material was pelleted by centrifugation (13,000 rpm, 3min). The concentration of the solubilised proteins (supernatant) was determined using a Pierce® BCA protein assay kit (Thermo Scientific). Protein samples were mixed with an equal volume of 2x SDS loading dye (625mM Tris-HCl, pH 6.7, 50%(v/v) glycerol, 10% sodium dodecyl sulfate (SDS), 0.5% Bromophenol Blue) and sample volumes equivalent to 2µg of protein were analysed by 15% SDS-PAGE and western blotting. Tpx1 was detected using anti-Tpx1 polyclonal antibodies (12) and hyperoxidized Prx were detected with monoclonal anti-peroxiredoxin-SO₃ antibodies (LabFrontiers) (26). FlagTrx1 was detected with monoclonal anti-Flag(M2-Sigma) antibodies. For *S. pombe* anti-tubulin antibodies and for HEK293 cells anti-actin (Sigma) antibodies were used to confirm that gels were evenly loaded. Primary antibodies were diluted 1 in 1000 in TBST (1mM Tris-HCl pH 8.0, 15mM NaCl, 0.01%(v/v) Tween 20). As appropriate, HRP-conjugated anti-rabbit or anti-mouse IgG secondary antibodies (Sigma) were used, followed by the fluorescent substrate ECL Plus to visualise antibody-labelled proteins (Thermo Scientific). Digital images of western blots were acquired with a Typhoon™ 9400 (GE Healthcare) and densitometry analysis performed using ImageQuantTL (Ver. 7).

Hydrogen Peroxide colorimetric Quantitation in media

PeroXOquant Quantitative Peroxide Assay kit with aqueous compatible formulation (Thermo scientific) was used according to manufacturer's protocol. Briefly, *S. pombe* cells were grown in EMM media to OD₆₀₀ 0.5. H₂O₂ was added to the growing cultures at a final concentration of 50µM. 20µl of media taken at various time points as indicated, were mixed in a 96-well microplate with 200µl of working solution (freshly prepared according to manufacturer's instructions). The mix was incubated at room temperature for 20min. Absorbance at 560nm

was measured using a TECAN infinite M200 plate reader. The blank value (EMM without H₂O₂) was automatically subtracted from all sample measurements.

Measuring changes in intracellular H₂O₂ concentration

The H₂O₂ sensor, acetylated Peroxyfluorescein 3 (PF3) (25), was added to 10ml (2 x 10⁷) exponentially growing *S. pombe* cells (OD₅₉₅ 0.4 – 0.5) to a final concentration of 5μM. Cells were incubated in the dark for 20 minutes, washed once then re-suspended in an equal volume of PBS (137mM NaCl, 2.7mM KCl, 10mM Na₂HPO₄, and 1.8mM KH₂PO₄ pH 7.4). HEK293 cells were incubated for 20min in DPBS containing 5μM PF3, washed three times then re-suspended in DPBS to a final concentration of 2 x 10⁶cells/ml. Following re-suspension, 200μl aliquots (4x10⁶ *S. pombe* cells, 4x10⁵ HEK293 cells) of PF3-labelled cells, unlabelled cells or PBS controls were transferred to a 96well plate. Fluorescence measurements were made at 529nm following excitation at 495nm using a TECAN Infinite M200PRO plate reader and the average fluorescence of the unlabelled cells was deducted to calculate fluorescence due to PF3 (F). Measurements were taken 0 and 60 seconds before addition of H₂O₂ to determine the basal rate of reaction of PF3 with endogenously produced H₂O₂ ($\Delta F/\Delta t$)_{no_stress} (Eq. 1).

$$\frac{\Delta F}{\Delta t}_{no_stress} = \frac{F_{t60} - F_{t0}}{60} \quad \text{Eq. 1}$$

Fluorescence measurements were made 30, 60 and 120 seconds following addition of H₂O₂ (0 – 500μM) and used to calculate the rate of change of fluorescence ($\Delta F/\Delta t$)_{stress} (Eq. 2).

$$\frac{\Delta F}{\Delta t}_{stress} = \frac{F_{t120} - F_{t30}}{90} \quad \text{Eq. 2}$$

The rate of fluorescence change due to exogenous H₂O₂ ($\Delta F/\Delta t$) was then calculated using Eq.3 and plotted against H₂O₂ concentration using the graphics and statistics package R.

$$\frac{\Delta F}{\Delta t} = \frac{\Delta F}{\Delta t}_{stress} - \frac{\Delta F}{\Delta t}_{no_stress} \quad \text{Eq. 3}$$

The R linear model function (50) was used to calculate the derivative and 95% confidence interval for the change ($\Delta F/\Delta t$) when H_2O_2 is increased from 0 – 100 μM H_2O_2 (*S. pombe*) or increased from 0 – 20 μM (HEK293 cells). This gradient was extrapolated using the equation $y = mx + c$ and plotted on the same axis as the ($\Delta F/\Delta t$) data.

Computational Method: Selection of a suitable reaction network for the kinetic model of Tpx1 oxidation

To find a set of reactions able to describe the formation of the Tpx1 monomers and disulfide homodimers detected in our experiments, we built a series of alternative models which contained different sets of biochemically feasible reactions. Parameter estimation was performed for each preliminary model, including preliminary models A and B (Fig. 5, Table S4) (see below for details), and the model with parameters similar to published values that gave the lowest Akaike information criterion (AIC) parameter set (51) was selected for the final model (Fig. 3, Table S5).

Computational Methods: Rate laws and measured parameters

The model contained two compartments, an extracellular compartment with volume Vol_{ex} (l) representing the growth media around the cells and an intracellular compartment of volume Vol_{int} (l) representing the total volume of all of the cells. The volume of the intracellular compartment was estimated using Eq 4:

$$Vol_{int} = Cell_{vol} \times Cell_{Num} \quad (\text{Eq. 4})$$

where $Cell_{Num} = 4 \times 10^8$ (the number of cells in 50ml of an OD₅₉₅ 0.4 culture) and $Cell_{vol}$ is the mean volume of an *S. pombe* cell, measured for exponentially growing wild-type (972) cells(CASY®, Schärfe System) as 126 μm^3 . The rate of movement of H_2O_2 between these

compartments was modelled to move down its concentration gradient using the rate equations Eq. 5 (H_2O_2 _influx) & Eq. 6 (H_2O_2 _efflux).

$$v_{in} = k_{H_2O_2_perm} \times H_2O_2_{ex} \quad (\text{Eq. 5})$$

$$v_{eff} = k_{H_2O_2_perm} \times H_2O_2_{int} \quad (\text{Eq. 6})$$

Where v_{in} is the rate of influx and v_{eff} =rate of efflux, $[H_2O_2]_{ex}$ and $[H_2O_2]_{int}$ represent the extracellular and intracellular H_2O_2 concentration and $k_{H_2O_2_perm}$ is a constant representing all other factors that influence the rate of H_2O_2 movement between each compartment. The non-Tpx1 metabolism of H_2O_2 was modelled using Michaelis-Menten kinetics and all other reactions in the model were governed by mass action kinetics.

The parameters for the model were either derived from published work, measured experimentally or estimated from our experimental data using the parameter estimation algorithm implemented in COPASI 4.10 (55) (52). The initial concentrations of Tpx1 and Trx1 in the model were based on the global quantification of the *S. pombe* proteome (53). The copy number per cell for these two proteins was used to calculate the concentration of each protein in the intracellular compartment using:

$$M = \frac{cpc}{N_A \times CellVol} \quad (\text{Eq. 7})$$

Where M is the molar protein concentration, cpc = copy number per cell (53) and N_A =Avagadro constant $6.02 \times 10^{23} \text{mol}^{-1}$.

Rate constants for the oxidation of the peroxidatic cysteine (Cys_P-SH) and the hyperoxidation of the sulfenic acid intermediate (Cys_P-SOH) were taken from a recent study of human Prx1 and Prx3 (16). Michaelis-Menten parameters for the reduction of Trx1 by Trr1 were based on those experimentally determined for the orthologous *S. cerevisiae* enzymes (54). All other parameters were estimated from our experimental data using parameter estimation.

Computational Methods: Parameter estimation, time course simulation, identifiability analysis and data representation

Parameter estimation was performed in COPASI 4.13 (87). The data set used for the parameter estimation was calculated from the relative intensities of the Tpx1 monomer and disulfide homodimer bands detected in 2-5 independent biological replicates of Tpx1 oxidation experiments depicted in Fig.1A and Fig. 2A-B (Table S1), and PeroXOquant measurements for the removal of extracellular H₂O₂ (Fig. 3B, Table S1). Based on a broad range of experimental evidence (reviewed in (55)), a mock data set assuming a steady-state intracellular H₂O₂ concentration of 1nM in exponentially growing cells was also included in the parameter estimation.

Parameter estimation was performed 500 times from random initial parameter values using the Levenberg-Marquardt algorithm (56) for each model (Table S2). The parameter set used for the final model had an AIC of 77.6 and was found on 327 out of 500 estimations, each of these 327 parameter estimations converged on similar values as indicated by the frequency distributions for each parameter (Fig. S3). One-dimensional likelihood profiles for each parameter and 95% confidence intervals were calculated using a simple identifiability analysis (57). This analysis demonstrated that the estimated parameters were identifiable (Fig. S4) with acceptable 95% percent confidence regions calculated for each parameter (Table S3, Fig. S4). Time course simulation was performed in COPASI 4.10 (56) using the deterministic (LSODA) algorithm (52). All graphics and further analysis of the simulation and identifiability data were performed using R.

Abbreviations

Peroxiredoxins (Prx), Thioredoxin (Trx), Thioredoxin-like protein 1 (Tx11), Glutathione peroxidase 1 (Gpx1), 4-acetamido-4'-((iodoacetyl)amino)stilbene-2,2'-disulfonic acid (AMS), N-ethylmaleimide (NEM), Reactive oxygen species (ROS), Hydrogen peroxide (H₂O₂), peroxidatic cysteine (Cys_P), resolving cysteine (Cys_R), Protein disulfides (Pr-SS), Protein

thiols (Pr-SH), Dulbecco's Modified Eagles Medium (DMEM), Edinburgh minimal media (EMM), Trichloroacetic acid (TCA), Dulbecco's phosphate-buffered saline (DPBS), Horse radish peroxidase (HRP), Acetylated Peroxyfluorescein 3 (PF3), Akaike information criterion (AIC)

Acknowledgements

We are very grateful to Jonathon Brown for technical advice, Andy Karplus for helpful discussions and Brian Morgan, Janet Quinn and Simon Whitehall for discussions and comments on the manuscript. Thanks to Michelle Wray for technical assistance, Peter Banks for providing the catalase mutant strain and Mike Gray the HEK293 cells. LT, ZU, WP, AD and EV were funded by the BBSRC (BB/F015895/1, BB/F016980/1 and BB/F023065/1). Funding was also provided by a Wellcome Trust Senior Investigator Award (JB) #095598/Z/11/Z and NIH grant GM 79465 (C.J.C.). C.J.C. is an Investigator with the Howard Hughes Medical Institute. B.C.D. and T.F.B. were partially supported by a Chemical Biology Training Grant from the National Institutes of Health (T32 GM066698). The authors declare that they have no conflict of interest. Author Contributions: L.T. designed and performed experiments under the supervision of A.D. and E.V. L.T. developed and tested mathematical models under the supervision of D.S and E.V. with input from G.S and P.D.P. Z.U. contributed data to Figs. 7, 8 and S5. W.P. assisted L.T. with the generation of data in Fig. 6D-F with advice from B.D. and E.V. Experiment in Fig. 4A was carried out by C.R. under the supervision of J.B. B.D., T.B. and C.C. synthesised PF3 and advised on its use in experiments in Figs.6-8. L.T. and E.V. wrote the manuscript with input from all authors.

References

1. Holmstrom KM & Finkel T (2014) Cellular mechanisms and physiological consequences of redox-dependent signalling. *Nat Rev Mol Cell Biol* 15(6):411-421.

2. Rhee SG & Woo HA (2011) Multiple functions of peroxiredoxins: peroxidases, sensors and regulators of the intracellular messenger HO, and protein chaperones. *Antioxid Redox Signal* 15(3):781-794.
3. Nystrom T, Yang J, & Molin M (2012) Peroxiredoxins, gerontogenes linking aging to genome instability and cancer. *Genes Dev* 26(18):2001-2008.
4. Yang KS, *et al.* (2002) Inactivation of human peroxiredoxin I during catalysis as the result of the oxidation of the catalytic site cysteine to cysteine-sulfinic acid. *J Biol Chem* 277(41):38029-38036.
5. Wood ZA, Poole LB, & Karplus PA (2003) Peroxiredoxin evolution and the regulation of hydrogen peroxide signaling. *Science* 300(5619):650-653.
6. Woo HA, *et al.* (2010) Inactivation of peroxiredoxin I by phosphorylation allows localized H₂O₂ accumulation for cell signaling. *Cell* 140(4):517-528.
7. Stangherlin A & Reddy AB (2013) Regulation of circadian clocks by redox homeostasis. *J Biol Chem*:26505-26511.
8. Kil IS, *et al.* (2012) Feedback control of adrenal steroidogenesis via H₂O₂-dependent, reversible inactivation of peroxiredoxin III in mitochondria. *Mol Cell* 46(5):584-594.
9. Bozonet SM, *et al.* (2005) Oxidation of a eukaryotic 2-Cys peroxiredoxin is a molecular switch controlling the transcriptional response to increasing levels of hydrogen peroxide. *J Biol Chem* 280(24):23319-23327.
10. Vivancos AP, *et al.* (2005) A cysteine-sulfinic acid in peroxiredoxin regulates H₂O₂-sensing by the antioxidant Pap1 pathway. *Proc Natl Acad Sci U S A* 102(25):8875-8880.
11. Brown JD, *et al.* (2013) A peroxiredoxin promotes H₂O₂ signaling and oxidative stress resistance by oxidizing a thioredoxin family protein. *Cell Rep* 5(5):1425-1435.
12. Day AM, *et al.* (2012) Inactivation of a peroxiredoxin by hydrogen peroxide is critical for thioredoxin-mediated repair of oxidized proteins and cell survival. *Mol Cell* 45(3):398-408.

13. Olahova M, *et al.* (2008) A redox-sensitive peroxiredoxin that is important for longevity has tissue- and stress-specific roles in stress resistance. *Proc Natl Acad Sci U S A* 105(50):19839-19844.
14. Jang HH, *et al.* (2004) Two enzymes in one; two yeast peroxiredoxins display oxidative stress-dependent switching from a peroxidase to a molecular chaperone function. *Cell* 117(5):625-635.
15. Cox AG, *et al.* (2009) Mitochondrial peroxiredoxin 3 is more resilient to hyperoxidation than cytoplasmic peroxiredoxins. *Biochem J* 421(1):51-58.
16. Peskin AV, *et al.* (2013) Hyperoxidation of peroxiredoxins 2 and 3: rate constants for the reactions of the sulfenic acid of the peroxidatic cysteine. *J Biol Chem* 288(20):14170-14177.
17. Haynes AC, Qian J, Reisz JA, Furdul CM, & Lowther WT (2013) Molecular basis for the resistance of human mitochondrial 2-Cys peroxiredoxin 3 to hyperoxidation. *J Biol Chem* 288(41):29714-29723.
18. Cao Z, Subramaniam S, & Bulleid NJ (2014) Lack of an efficient endoplasmic reticulum-localized recycling system protects peroxiredoxin IV from hyperoxidation. *J Biol Chem* 289(9):5490-5498.
19. Jara M, *et al.* (2007) The peroxiredoxin Tpx1 is essential as a H₂O₂ scavenger during aerobic growth in fission yeast. *Mol Biol Cell* 18(6):2288-2295.
20. Chen D, *et al.* (2003) Global transcriptional responses of fission yeast to environmental stress. *Mol Biol Cell* 14(1):214-229.
21. Chen D, *et al.* (2008) Multiple pathways differentially regulate global oxidative stress responses in fission yeast. *Mol Biol Cell* 19(1):308-317.
22. Lackner DH, Schmidt MW, Wu S, Wolf DA, & Bahler J (2012) Regulation of transcriptome, translation, and proteome in response to environmental stress in fission yeast. *Genome Biol* 13(4):R25.

23. Chae HZ, Kim HJ, Kang SW, & Rhee SG (1999) Characterization of three isoforms of mammalian peroxiredoxin that reduce peroxides in the presence of thioredoxin. *Diabetes Res Clin Pract* 45(2-3):101-112.
24. Manta B, *et al.* (2009) The peroxidase and peroxynitrite reductase activity of human erythrocyte peroxiredoxin 2. *Arch Biochem Biophys* 484(2):146-154.
25. Dickinson BC, Huynh C, & Chang CJ (2010) A palette of fluorescent probes with varying emission colors for imaging hydrogen peroxide signaling in living cells. *J Am Chem Soc* 132(16):5906-5915.
26. Woo HA, *et al.* (2003) Reversible oxidation of the active site cysteine of peroxiredoxins to cysteine sulfinic acid. Immunoblot detection with antibodies specific for the hyperoxidized cysteine-containing sequence. *J Biol Chem* 278(48):47361-47364.
27. Koo KH, *et al.* (2002) Regulation of thioredoxin peroxidase activity by C-terminal truncation. *Arch Biochem Biophys* 397(2):312-318.
28. Makino N, Mise T, & Sagara J (2008) Kinetics of hydrogen peroxide elimination by astrocytes and C6 glioma cells analysis based on a mathematical model. *Biochim Biophys Acta* 1780(6):927-936.
29. Garcia-Santamarina S, *et al.* (2013) Is oxidized thioredoxin a major trigger for cysteine oxidation? Clues from a redox proteomics approach. *Antioxid Redox Signal* 18(13):1549-1556.
30. Garcia-Santamarina, S., Boronat, S., Ayte, J., and Hidalgo, E. (2013) Methionine sulphoxide reductases revisited: free methionine as a primary target of H₂O₂ stress in auxotrophic fission yeast. *Mol Microbiol* **90**, 1113-1124
31. Adimora NJ, Jones DP, & Kemp ML (2010) A model of redox kinetics implicates the thiol proteome in cellular hydrogen peroxide responses. *Antioxid Redox Signal* 13(6):731-743.
32. Hansen RE, Roth D, & Winther JR (2009) Quantifying the global cellular thiol-disulfide status. *Proc Natl Acad Sci U S A* 106(2):422-427.

33. Winterbourn CC (2008) Reconciling the chemistry and biology of reactive oxygen species. *Nat Chem Biol* 4(5):278-286.
34. Pillay, C. S., Hofmeyr, J. H., Olivier, B. G., Snoep, J. L., and Rohwer, J. M. (2009) Enzymes or redox couples? The kinetics of thioredoxin and glutaredoxin reactions in a systems biology context. *Biochem J* 417: 269-275
35. Pillay, C. S., Hofmeyr, J. H., and Rohwer, J. M. (2011) The logic of kinetic regulation in the thioredoxin system. *BMC Syst Biol* 5: 15
36. Miller EW, Dickinson BC, & Chang CJ (2010) Aquaporin-3 mediates hydrogen peroxide uptake to regulate downstream intracellular signaling. *Proc Natl Acad Sci U S A* 107(36):15681-15686.
37. Cuddihy SL, Winterbourn CC, & Hampton MB (2011) Assessment of redox changes to hydrogen peroxide-sensitive proteins during EGF signaling. *Antioxid Redox Signal* 15(1):167-174.
38. Wood ZA, Poole LB, Hantgan RR, & Karplus PA (2002) Dimers to doughnuts: redox-sensitive oligomerization of 2-cysteine peroxiredoxins. *Biochemistry* 41(17):5493-5504.
39. Parsonage D, *et al.* (2005) Analysis of the link between enzymatic activity and oligomeric state in AhpC, a bacterial peroxiredoxin. *Biochemistry* 44(31):10583-10592.
40. Paulo E, *et al.* (2014) A genetic approach to study H₂O₂ scavenging in fission yeast -distinct roles of peroxiredoxin and catalase. *Mol Microbiol* 92(2):246-257.
41. Benfeitas R, Selvaggio G, Antunes F, Coelho PM, & Salvador A (2014) Hydrogen peroxide metabolism and sensing in human erythrocytes: a validated kinetic model and reappraisal of the role of peroxiredoxin II. *Free Radic Biol Med* 74:35-49.
42. Edgar RS, *et al.* (2012) Peroxiredoxins are conserved markers of circadian rhythms. *Nature* 485(7399):459-464.

43. O'Neill JS, *et al.* (2011) Circadian rhythms persist without transcription in a eukaryote. *Nature* 469(7331):554-558.
44. O'Neill JS & Reddy AB (2011) Circadian clocks in human red blood cells. *Nature* 469(7331):498-503.
45. Pekovic-Vaughan V, *et al.* (2014) The circadian clock regulates rhythmic activation of the NRF2/glutathione-mediated antioxidant defense pathway to modulate pulmonary fibrosis. *Genes Dev* 28(6):548-560.
46. Cho CS, Yoon HJ, Kim JY, Woo HA, & Rhee SG (2014) Circadian rhythm of hyperoxidized peroxiredoxin II is determined by hemoglobin autoxidation and the 20S proteasome in red blood cells. *Proc Natl Acad Sci U S A* 111(33):12043-12048.
47. Phalen TJ, *et al.* (2006) Oxidation state governs structural transitions in peroxiredoxin II that correlate with cell cycle arrest and recovery. *J Cell Biol* 175(5):779-789.
48. Turner-Ivey B, *et al.* (2013) Role for Prdx1 as a specific sensor in redox-regulated senescence in breast cancer. *Oncogene* 32(45):5302-5314.
49. Delaunay A, Isnard AD, & Toledano MB (2000) H₂O₂ sensing through oxidation of the Yap1 transcription factor. *Embo J* 19(19):5157-5166.
50. Wilkinson G & Rogers CE (1973) Symbolic Description of Factorial Models for Analysis of Variance. *J Roy Stat Soc C-App* 22(3):392-399.
51. Akaike HJ ed (1973) *Information theory and an extension of the maximum likelihood principle* (Akademia Kiado, Budapest), pp 267-281.
52. Petzold L (1983) Automatic Selection of Methods for Solving Stiff and Nonstiff Systems of Ordinary Differential-Equations. *Siam J Sci Stat Comp* 4(1):136-148.
53. Marguerat S, *et al.* (2012) Quantitative analysis of fission yeast transcriptomes and proteomes in proliferating and quiescent cells. *Cell* 151(3):671-683.
54. Oliveira MA, *et al.* (2010) Insights into the specificity of thioredoxin reductase-thioredoxin interactions. A structural and functional investigation of the yeast thioredoxin system. *Biochemistry* 49(15):3317-3326.

55. Stone JR & Yang S (2006) Hydrogen peroxide: a signaling messenger. *Antioxid Redox Signal* 8(3-4):243-270.
56. Marquadt DW (1963) An algorithm for least squares estimation of nonlinear parameters. *Siam J Algebra Discr* 11:431-441.
57. Schaber J (2012) Easy parameter identifiability analysis with COPASI. *Biosystems* 110(3):183-185.

Figure legends

Figure 1. The Prx, Tpx1, undergoes oxidation to multiple redox states following exposure to different concentrations of H₂O₂.

[A] The thioredoxin peroxidase activity of Prx, such as Tpx1, involves the reversible oxidation of catalytic cysteines but is inactivated by hyperoxidation of the peroxidatic cysteine at high concentrations of H₂O₂. The catalytic breakdown of H₂O₂ by 2-Cys peroxiredoxins (Prx) involves the reaction of the peroxidatic cysteine (Cys_P) with H₂O₂. In the catalytic cycle the sulfenylated cysteine is stabilised by forming a disulfide with the resolving cysteine (Cys_R) in a neighboring Prx molecule. These Prx disulfides are reduced by the thioredoxin system using electrons from NADPH. In eukaryotes Prx disulfide formation is slow, rendering the sulfenylated Prx, Prx-SOH, susceptible to further oxidation to a sulfinylated, 'hyperoxidized' form (Prx-SOOH) that cannot be reduced by thioredoxin. This hyperoxidation is favoured at higher concentrations of H₂O₂ inactivating the thioredoxin peroxidase activity of the Prx. Western blot analysis (anti-Tpx1 antibodies) of **[C-D]** AMS-treated protein extracts from wild-type (972) and $\Delta tpx1$ mutant (VX00) cells treated, as indicated, for 20s with 0-6mM H₂O₂ reveals that Tpx1 undergoes oxidation to a number of redox states following exposure to H₂O₂. The absence of bands in $\Delta tpx1$ mutant (VX00) cells indicates that all the bands detected in wild-type cells represent Tpx1 or Tpx1-containing complexes. **[C]** A magnified image of the ~40kD region outlined by the dotted line in [B] shows that 3 different Tpx1-containing disulfide dimers (Tpx1 ox) are detected following

treatment with concentrations $\geq 200\mu\text{M}$. **[D]** Different Tpx1ox forms are depicted which were separated in **[E]** on the basis of the reduced mobility associated with modification of free cysteine thiols by AMS (0.64kD) compared with NEM (0.0125kD). **[E]** disulfide dimers (Tpx1ox) in duplicate samples extracted from wild-type or $\Delta tpx1$ mutant (VX00) cells before or following treatment for 20s with the indicated concentration of H_2O_2 then reacted with AMS or NEM. As in **[B]**, only low levels of Tpx1ox (disulfides) are detected in untreated cells or cells treated with $50\mu\text{M}$. See also Table S1 and Fig. S1.

Figure 2. Changes in Tpx1 oxidation over time following treatment with 100 or 200 μM H_2O_2 .

Western blot analysis with **[A-B]** anti-Tpx1 or **[C]** anti-PrxSO₃ antibodies of AMS-treated protein extracts from wild-type (972) and $\Delta tpx1$ mutant (VX00) cells treated, as indicated with **[A]** $100\mu\text{M}$ **[B-C]** $200\mu\text{M}$ H_2O_2 for 0-600s shows how the oxidation of Tpx1 changes with time. **[B-C]** shows that Tpx1ox:SOOH disulfide formation at $200\mu\text{M}$ H_2O_2 precedes the formation of Tpx1-SOOH monomers. In **[B]** a section of the blot, outlined by the dotted line, is magnified and shown in the right hand panel to enable the additional Tpx1ox form present in cells treated with $200\mu\text{M}$ H_2O_2 (Tpx1ox:SOOH) to be seen more clearly. Western blot analysis of beta-mercaptoethanol(βME)-treated samples, run on a separate gel (lower panel in **[A]** and **[B]**), in which a single band represents all Tpx1 redox states (eliminating any influence that differences in mobility might have on transfer to the membrane) allowing total Tpx1 levels to be compared **[A]** confirms that differences between lanes reflect changes in Tpx1 oxidation rather than total Tpx1 levels. See also Table S1 and Fig. S2.

Figure 3. Diagram representing the mathematical model describing the *in vivo* oxidation of Tpx1 and qualitative analysis of the fit of the model to the experimental

data. [A] The model contains 9 different Tpx1 oxidation states which are interconverted by the indicated reactions. The rates of influx/efflux (H_2O_2 influx and H_2O_2 efflux) and removal of H_2O_2 (H_2O_2 metab) from the intracellular compartment were also included in the model. For rate laws see Table S2 and for parameters see Table S3. **[B-G]** Plots show simulated and experimentally determined concentrations of **[B]** extracellular H_2O_2 **[C]** reduced Tpx1; Tpx1SH **[D]** single Tpx1 disulfides; Tpx1ox#1 **[E]** double Tpx1 disulfides; Tpx1ox#2 **[F]** disulfide bonded hyperoxidized Tpx1; Tpx1ox:SOOH **[G]** hyperoxidized Tpx1 monomer; Tpx1SOOH in wild-type *S. pombe* following 20 second exposure to between 0 and 1000 μM H_2O_2 . Simulated data derived from the model were plotted against the experimental data used in the parameter estimation (Table S1). See also Table S2, S3, Figs. S3 and S4.

Figure 4. Qualitative analysis of the fit of the model to the experimental data for

between 0-600 s exposure to 100 or 200 μM H_2O_2 . Plots show simulated and experimentally determined concentrations of **[A-B]** reduced Tpx1; Tpx1SH **[C-D]** single Tpx1 disulfides; Tpx1ox#1 **[E-F]** double Tpx1 disulfides; Tpx1ox#2 **[G-H]** disulfide bonded hyperoxidized Tpx1; Tpx1oxSOOH **[I-J]** hyperoxidized Tpx1 monomer; Tpx1SOOH in wild-type *S. pombe* following 0-600s treatment with **[A, C, E, G and I]** 100 μM or **[B, D, F, H and J]** 200 μM H_2O_2 . Simulated data derived from the model were plotted against the experimental data used in the parameter estimation (Table S1).

Figure 5. Alternative models, Alt Model A and Alt Model B, were unable to simulate the experimentally determined changes in Tpx1 oxidation state following treatment of

wild-type *S. pombe* with H_2O_2 as effectively as the final model. All 3 models were constructed using the same reaction network (Fig.3A) and rate laws (Table S2) except that in alternative model A, Tpx1ox:SOH was also produced by the H_2O_2 -induced oxidation of Tpx1ox#1 in a reaction governed by the rate law $k_{cys_ox2} [Tpx1ox\#1][H_2O_2]_{int}$. Alternative model B was identical with the final model except that it lacked the peroxide-removing reaction, $H_2O_2_metab$. For parameters sets used in models see Tables S3 and S4. See also Table S5.

Figure 6. As predicted by the model, hyperoxidation of Prx only occurs when the H₂O₂-removing capacity of cells is saturated. [A-B] The effect of increasing extracellular H₂O₂ on **[A]** intracellular H₂O₂ concentration **[B]** hyperoxidation of Tpx1 was simulated using the final model with the rate constant $V_{\max_H2O2_metab}$ set as 59 μMs^{-1} (Final Model) or 0 μMs^{-1} (No H₂O₂_metab). In **[B]** the experimentally determined effects of increasing extracellular H₂O₂ concentrations on the hyperoxidation of Tpx1 are also shown. **[C-D]** Experimental measurements using a fluorescent H₂O₂-specific dye (PF3) (25) of the rate at which intracellular H₂O₂ increases ($\Delta F/\Delta t$) in **[C]** wild-type *S. pombe* (972) **[D]** human cells (HEK293) following exposure to increasing concentrations of H₂O₂. The results of three independent experiments are shown. The gradient ($m \pm \text{SE}$) and intercept ($c \pm \text{SE}$) were calculated using data points **[C]** 0 – 100 μM (*S. pombe*) **[D]** 0 – 20 μM (HEK293) and extrapolated to create the “Slope” and “Std Error of Slope” lines that are shown. **[E-F]** The increase in hyperoxidized Prx in HEK293 cells treated with increasing concentrations of H₂O₂ was determined experimentally by **[E]** western blotting with anti-PrxSO₃ antibodies of proteins extracted from HEK293 cells following 10min exposure to the indicated concentration of H₂O₂. This revealed an increase in beta-mercaptoethanol-resistant hyperoxidized Prx (indicated by the arrows) which were **[F]** quantified as total hyperoxidized Prx relative to a loading control (actin).

Figure 7. Tpx1, Gpx1 and catalase activity, do not influence the biphasic increase in intracellular H₂O₂ in response to increasing extracellular H₂O₂. The rate of intracellular H₂O₂ accumulation, measured using a fluorescent H₂O₂-specific dye (PF3) (25), following exposure of **[A]** wild type (NT4) **[B]** $\Delta tpx1$ (VX00) mutant **[C]** cells expressing higher levels of wild-type Tpx1; High Tpx1 WT (JR68) or **[D]** truncated, hyperoxidation-resistant Tpx1; Tpx1¹⁻¹⁸¹ (JR20) **[E]** $\Delta ctt1$ (LT3) **[F]** $\Delta gpx1$ (SB13) to a range of H₂O₂ concentrations. The results of three independent experiments are shown. The gradient ($m \pm \text{SE}$) and intercept ($c \pm \text{SE}$) for the data points 0 – 100 μM were calculated and extrapolated to create the lines “Slope 0-100 μM ” and “Std Error of Slope 0-100 μM ” (see methods).

Figure 8 Thioredoxin is required for the biphasic increase in intracellular H₂O₂ in response to increasing extracellular H₂O₂, consistent with the oxidation of protein-thiols limiting rises in intracellular H₂O₂ following exposure of *S. pombe* to low levels of extracellular H₂O₂ [A, C-E] The rate of intracellular H₂O₂ accumulation, using a fluorescent H₂O₂-specific dye (PF3) (25), following exposure of **[A]** Δ *trx1* Δ *txl1* (AD140) **[C-E]** wild type (NT4) cells containing **[C]** vector control (Rep1) **[D-E]** ectopically expressing additional Trr1 (Rep1Trr1) (Brown et al., 2013) and **[E]** Flag-tagged Trx1 (Rep2Trx1), to a range of H₂O₂ concentrations. **[B]** Western blot analysis of FlagTrx1 redox state in cells containing vector control or overexpressing Trr1 (Rep1Trr1) before and following exposure for the indicated times to 0.2mM H₂O₂. Image J quantification was used to determine the % FlagTrx1 oxidation. **[F]** The oxidation of protein thiols (Pr-SH) to protein disulfides (Pr-SS) with a low affinity for thioredoxin (Trx1/Txl1) is able to explain the H₂O₂-buffering capacity that is saturated following exposure to 150 μ M H₂O₂. In this model (Tables S6 and S7) increasing oxidoreductase activity, by doubling the concentrations of Trx1 or Trr1, produces a negligible effect on the H₂O₂-buffering capacity, whereas loss of both Trx1 and Txl1 ablates the H₂O₂-buffering capacity. **[G]** We propose that reduced protein thiols (Protein-SH), act like a sponge, contributing to a buffer that prevents increases in intracellular accumulation of H₂O₂ but that, following maximal oxidation of these cellular protein-thiols to disulfides (Protein-SS), intracellular H₂O₂ concentrations increase, causing peroxidatic cysteines in Prx to become hyperoxidized to sulfinic derivatives (SOOH) and diverting thioredoxin towards the reduction of other protein disulfides. See also Table S6-S7 and Fig. S5.

List of figures:

Figure 1

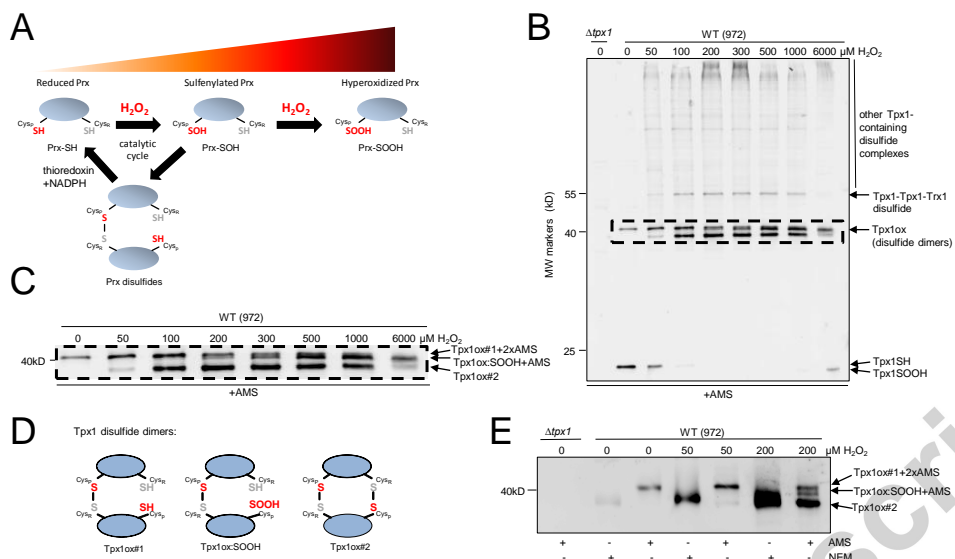


Figure 2

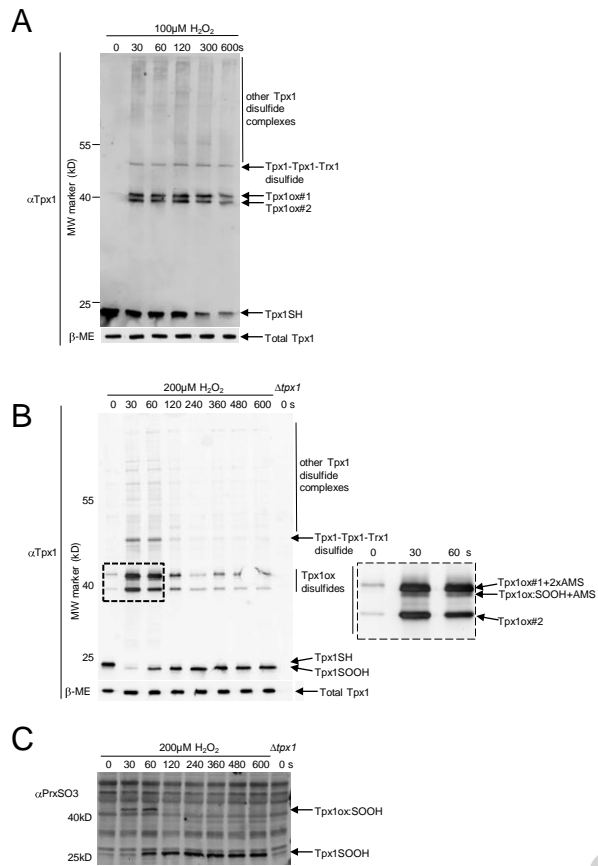


Figure 3

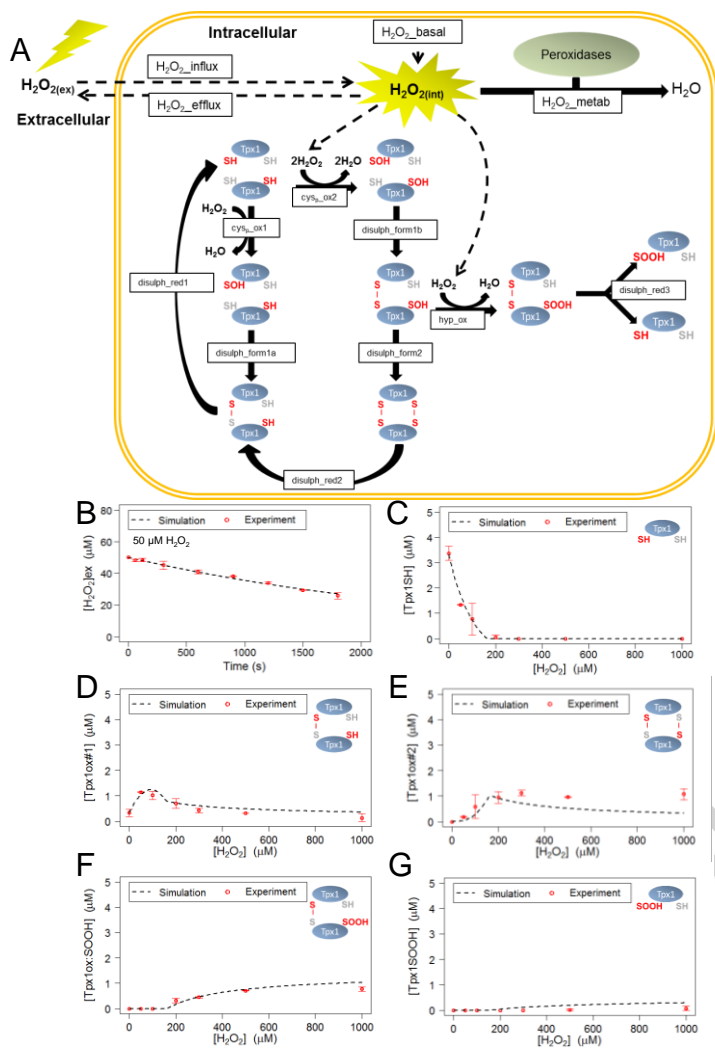


Figure 4

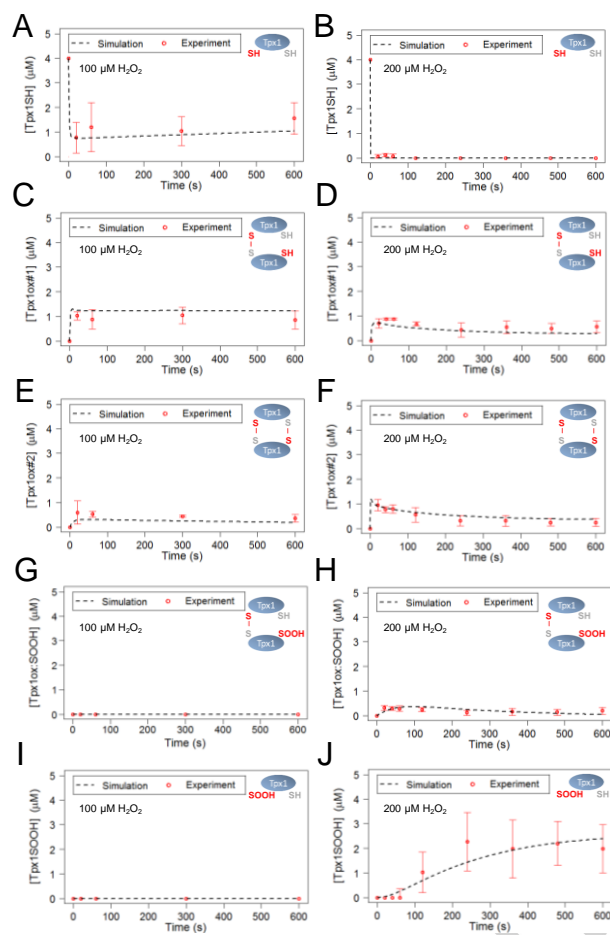


Figure 5

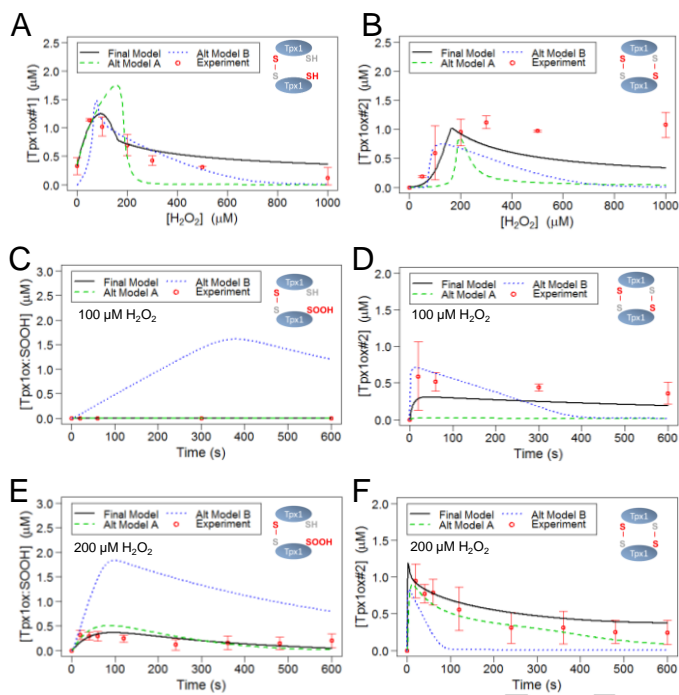


Figure 6

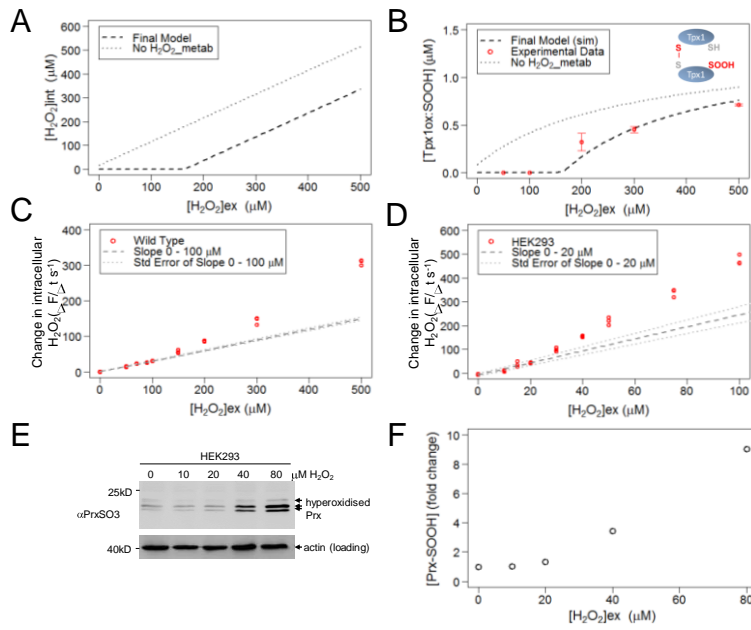


Figure 7

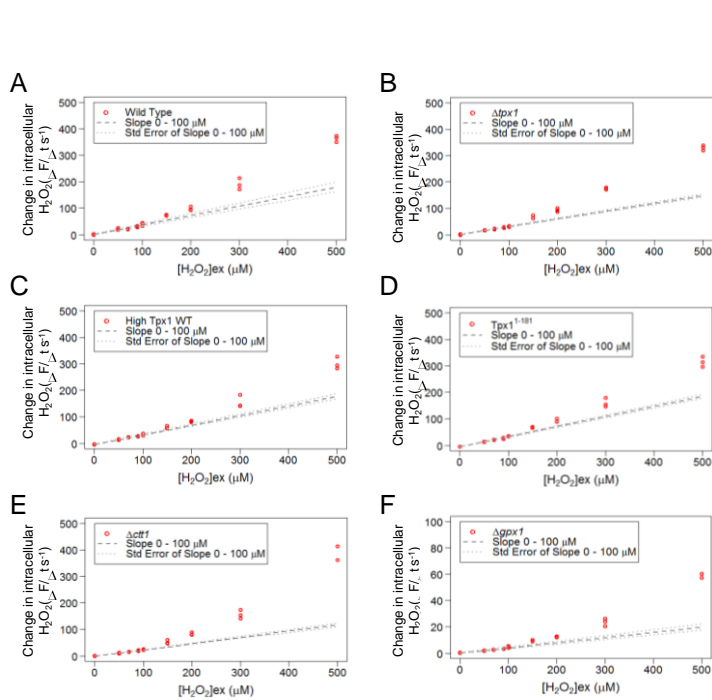
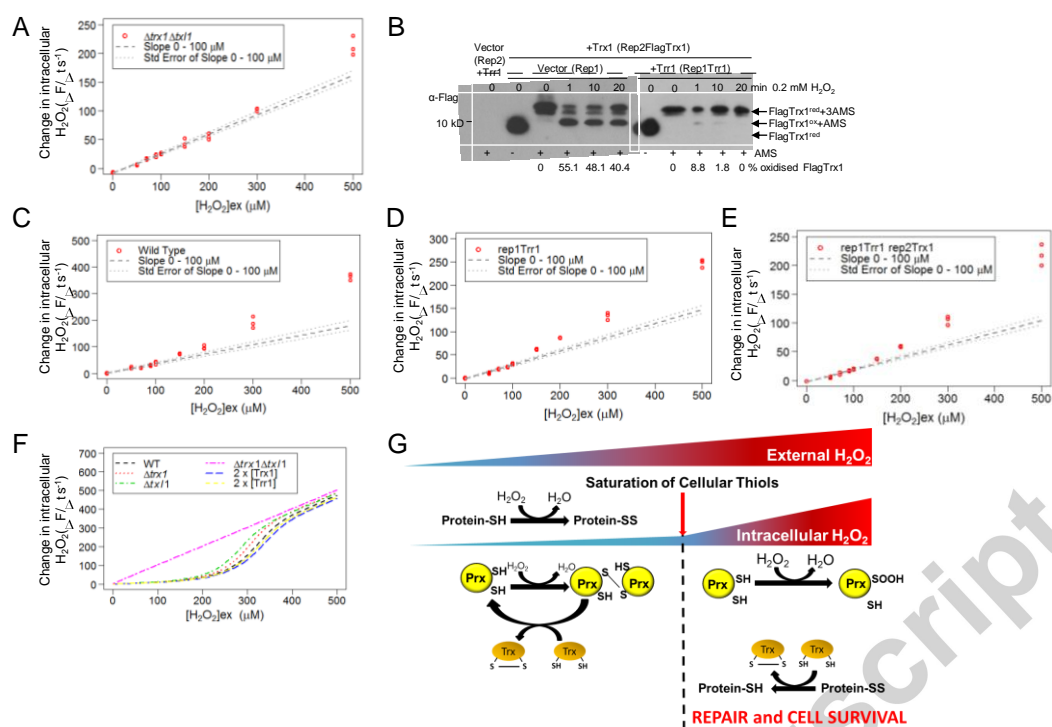


Figure 8



Highlights

- A mathematical model that simulates *in vivo* dynamic, H_2O_2 -induced, changes in Prx oxidation
- Increasing extracellular H_2O_2 produces a bi-phasic response in intracellular H_2O_2
- Thioredoxin and the thiol-proteome can efficiently buffer low levels of H_2O_2
- Prx hyperoxidation only occurs once this H_2O_2 buffering capacity is saturated

Graphical abstract:

