

Density Functional Theory Study of the Interaction of H₂O, CO₂ and CO with the ZrO₂ (111), Ni/ZrO₂ (111), YSZ (111) and Ni/YSZ (111) Surfaces

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This document contains the binding energies at the non-equivalent adsorption sites for H₂O, CO₂ and CO molecules on ZrO₂(111) (**Figure S1** and **Table S1**), Ni/ZrO₂(111) (**Figure S2** and **Table S2**), YSZ(111) (**Figure S3** and **Table S3**) and Ni/YSZ(111) (**Figure S4** and **Table S4**). We have not reported here the zero point energy. This document also contains the Zr_xO_yH_z clusters studied in Ref. 40 and discussed in our manuscript (**Figure S5**).

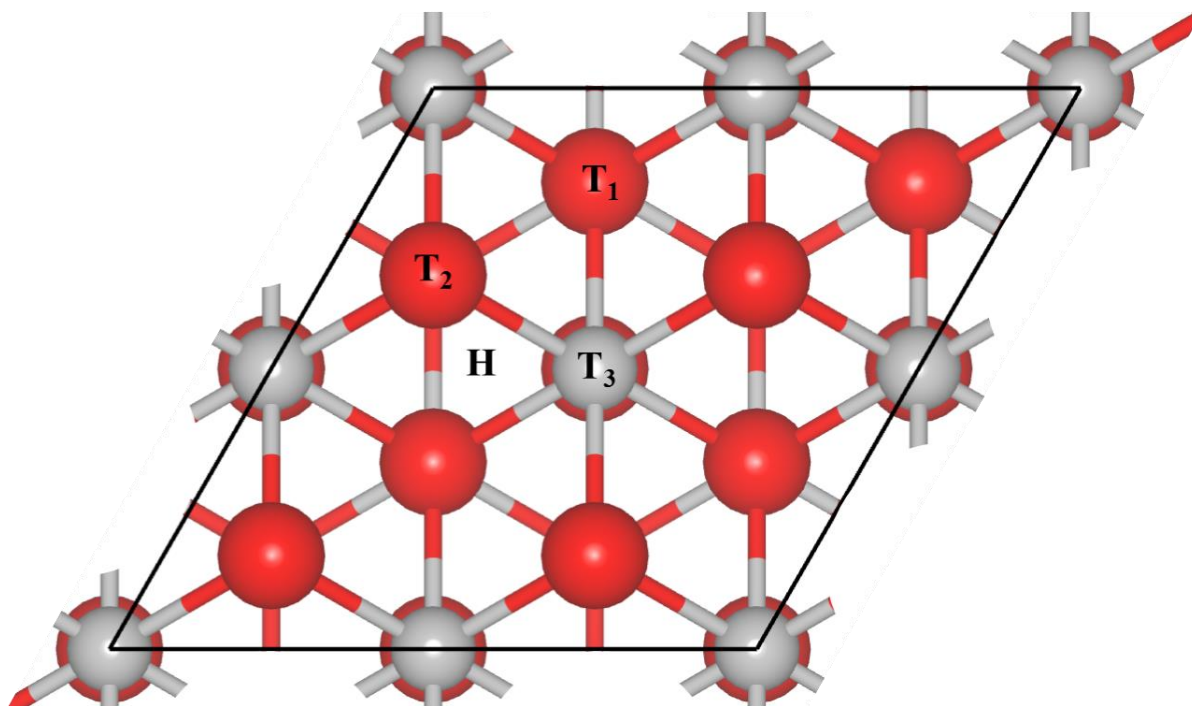


Figure S1 4 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on ZrO₂ (111) (top view). Colour key: red and grey spheres correspond to oxygen and Zr, respectively.

Table S1. Binding energies (in eV) of the 4 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on ZrO₂ (111).

Molecules	Initial Adsorption site	Initial Orientation	Binding energy (eV)	
H ₂ O	T ₁	Parallel	-1.14	
		Perpendicular	-1.14	
	T ₂	Parallel	-1.14	
		Perpendicular	-1.14	
	T ₃	Parallel	-1.14	
		Perpendicular	-1.14	
	H	Parallel	-1.14	
		Perpendicular	-1.14	
	CO ₂	T ₁	Parallel	-0.25
			Perpendicular	-0.25
T ₂		Parallel	-0.24	
		Perpendicular	-0.24	
T ₃		Parallel	-0.25	
		Perpendicular	-0.25	
H		Parallel	-0.25	
		Perpendicular	-0.25	
CO		T ₁	Parallel	-0.40
			Perpendicular	-0.10
	T ₂	Parallel	-0.40	
		Perpendicular	-0.40	
	T ₃	Parallel	-0.12	
		Perpendicular	-0.40	
	H	Parallel	-0.12	
		Perpendicular	-0.40	

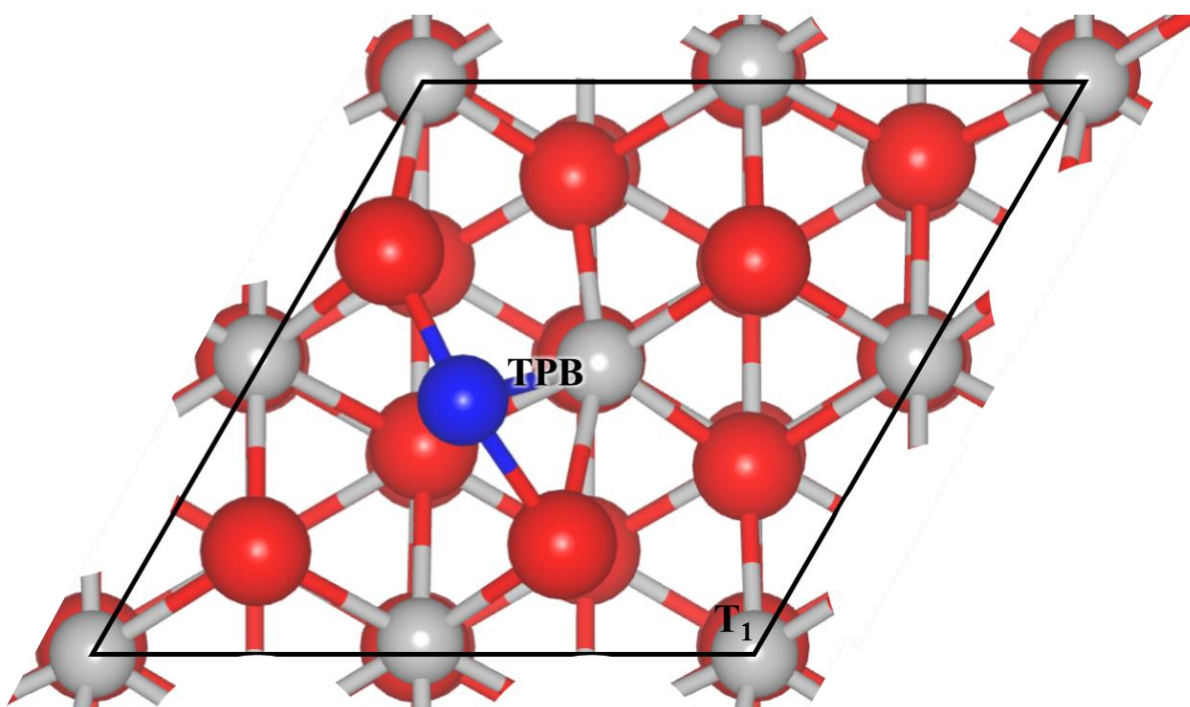


Figure S2. 2 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on Ni/ZrO₂ (111) (top view): at the triple phase boundary (TPB) and on top of the Zr (T₁) which is the most favourable adsorption sites for the three molecules on the clean ZrO₂(111) surface. Colour key: red, grey and blue spheres correspond to oxygen, Zr and Ni respectively.

Table S2. Binding energies (in eV) of the 2 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on Ni/ZrO₂ (111).

Molecules	Initial Adsorption site	Initial Orientation	Binding energy (eV)
H ₂ O	TPB	Parallel	-1.46
		Perpendicular	-1.40
	T ₁	Parallel	-0.78
CO ₂	TPB	Parallel	-1.37
		Perpendicular	-0.23
	T ₁	Parallel	-0.27
CO	TPB	Parallel	-0.28
		Perpendicular	-0.28
	T ₁	Perpendicular	-0.47

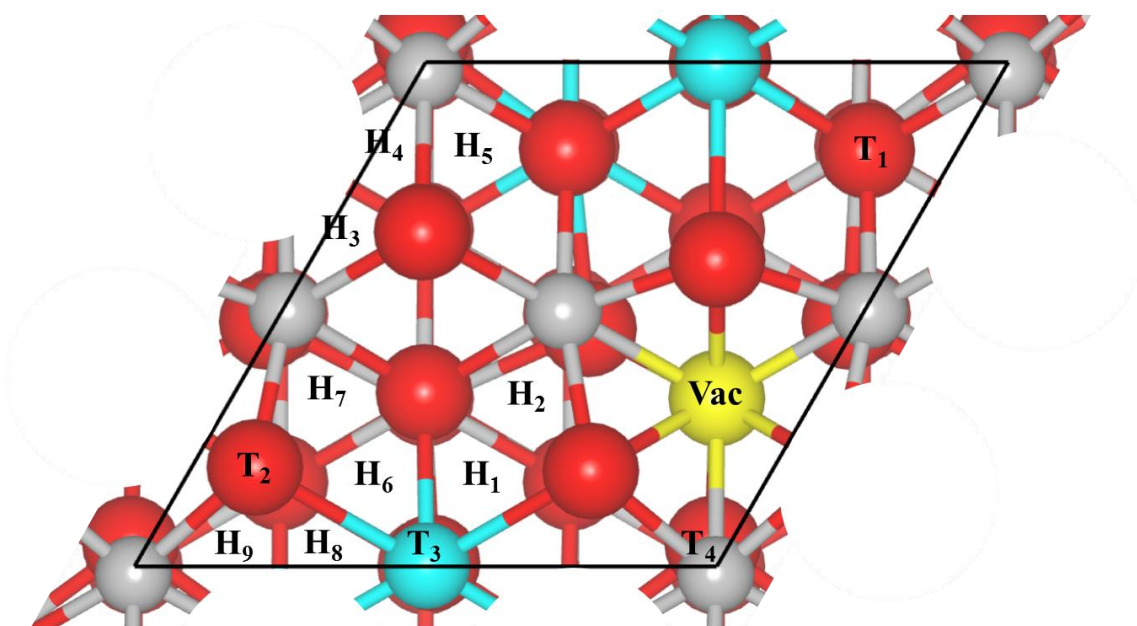


Figure S3. 5 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on YSZ (111) (top view). Colour key: red, grey and cyan spheres correspond to oxygen, Zr, and Y respectively. The oxygen vacancy is represented by the yellow sphere.

Table S3. Binding energies (in eV) of the 5 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on YSZ (111).

Molecules	Initial Adsorption site	Initial Orientation	Binding energy (eV)	Initial Adsorption site	Initial Orientation	Binding energy (eV)
H ₂ O	T ₁	Parallel	-0.85	H ₁	Parallel	-0.83
		Perpendicular	-0.85	H ₂	Parallel	-0.39
	T ₂	Parallel	-0.60	H ₃	Parallel	-0.50
		Perpendicular	-0.86	H ₄	Parallel	-0.54
	T ₃	Parallel	-0.88	H ₅	Parallel	-0.55
		Perpendicular	-0.87	H ₆	Parallel	-0.86
	T ₄	Parallel	-0.57	H ₇	Parallel	-0.85
		Perpendicular	-0.57	H ₈	Parallel	-0.85
	Vac	Parallel	-0.60	H ₉	Parallel	-0.59
		Perpendicular	-0.29			
CO ₂	T ₁	Parallel	-0.36	H ₁	Parallel	-0.31
		Perpendicular	-0.36	H ₂	Parallel	-0.35
	T ₂	Parallel	-0.27	H ₃	Parallel	-0.17
		Perpendicular	-0.23	H ₄	Parallel	-0.18
	T ₃	Parallel	-0.37	H ₅	Parallel	-0.18
		Perpendicular	-0.36	H ₆	Parallel	-0.36
	T ₄	Parallel	-0.36	H ₇	Parallel	-0.36
		Perpendicular	-0.36	H ₈	Parallel	-0.35
	Vac	Parallel	-0.28	H ₉	Parallel	-0.35
		Perpendicular	-0.28			
CO	T ₁	Parallel	-0.19	H ₁	Perpendicular	-0.36
		Perpendicular	-0.38	H ₂	Perpendicular	-0.25
	T ₂	Parallel	-0.27	H ₃	Perpendicular	-0.10
		Perpendicular	-0.27	H ₄	Perpendicular	-0.17
	T ₃	Parallel	-0.38	H ₅	Perpendicular	-0.11
		Perpendicular	-0.38	H ₆	Perpendicular	-0.37
	T ₄	Parallel	-0.27	H ₇	Perpendicular	-0.36
		Perpendicular	-0.27	H ₈	Perpendicular	-0.36
	Vac	Parallel	-0.27	H ₉	Perpendicular	-0.24
		Perpendicular	-0.23			

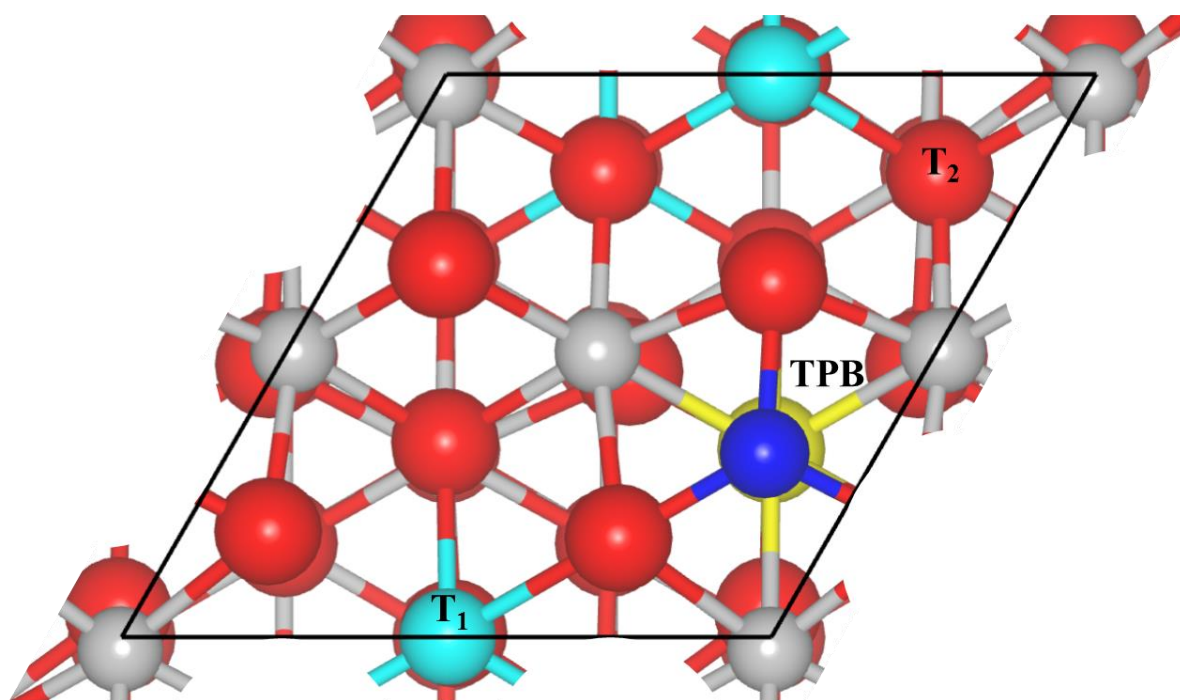


Figure S4. 2 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on Ni/YSZ (111) (top view): at the triple phase boundary (TPB) and on top of the Zr (T₁) and oxygen (T₂) which are the most favourable adsorption sites for H₂O and CO₂, and CO molecules on the clean YSZ (111) surface, respectively. Colour key: red, grey, cyan and blue spheres correspond to oxygen, Zr, Y and Ni respectively. The oxygen vacancy is represented by the yellow sphere.

Table S4. Binding energies (in eV) of the 2 non-equivalent initial adsorption sites for H₂O, CO₂ and CO molecules on Ni/YSZ (111).

Molecules	Initial Adsorption site	Initial Orientation	Binding energy (eV)
H ₂ O	TPB	Parallel	-0.34
		Perpendicular	-0.27
	T ₁	Parallel	-0.73
CO ₂	TPB	Parallel	-0.32
		Perpendicular	-0.32
	T ₁	Parallel	-0.31
CO	TPB	Parallel	-2.37
		Perpendicular	-1.83
	T ₂	Perpendicular	-0.36

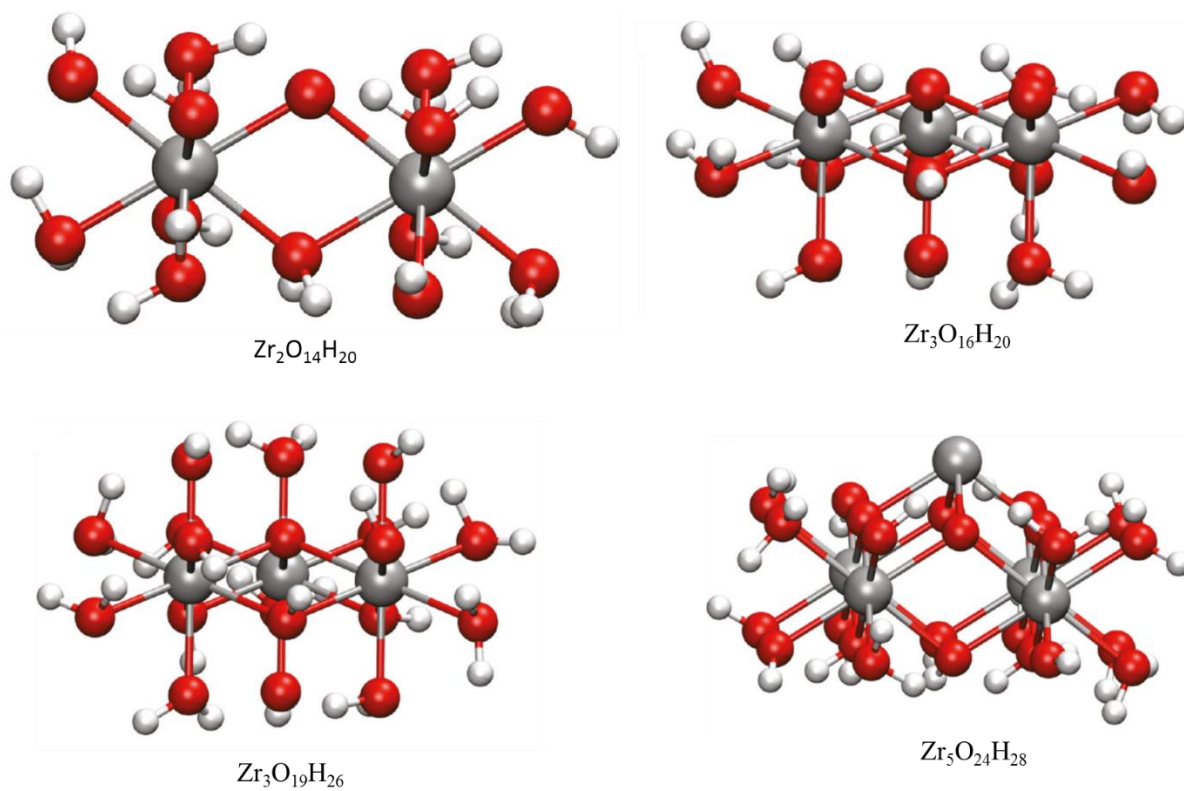


Figure S5 Representation of the $Zr_xO_yH_z$ clusters studied in Ref. 40 and discussed in our manuscript. Adapted with permission from V. Hornebecq, C. Kn, P. Boulet, B. Kuchta, P.L. Llewellyn, Adsorption of Carbon Dioxide on Mesoporous Zirconia: Microcalorimetric Measurements, Adsorption Isotherm Modeling, and Density Functional Theory Calculations, *J. Phys. Chem. C.* 115 (2011) 10097–10103. Copyright (2011) American Chemical Society.