#### Manuscript Draft

Manuscript Number: STOTEN-D-16-04655

Title: Pollutant sources in an arsenic-affected multilayer aquifer in the

Po Plain of Italy: implications for drinking water supply

Article Type: Research Paper

Keywords: Aquifer; Recharge; Cl/Br; As; Cl; Cremona.

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Abstract: We took an area of ~150 km2 around the town of Cremona to be an analogue for hydrogeological and hydrochemical conditions in arsenic-polluted and arsenic-free groundwaters across the Po Plain of northern Italy. We investigated anthropogenic influences on ground and surface water in the area using Cl/Br ratios,  $\delta 180/\delta 2H$  and other hydrochemical data from 32 groundwater wells, 9 surface waters, a sewage outfall and a rainwater sample.

The deep aquifer (160-260 m below ground level), which is tapped widely for public supply, is partly recharged by seepage from overlying aquifers (65-150 m below ground level) that are As-polluted (up to 144  $\mu$ g/L As), producing a trend of increasing As with time. This threatens drinking water quality across the Po Plain where natural As-pollution of groundwater in aquifers at intermediate depth (50-120 m below ground level) is a basin-wide problem.

Groundwater quality in deep aquifers appears free of anthropogenic influences. In contrast, shallow groundwater and surface water are strongly affected by such pollution, although in some areas, quality remains unaffected. Outfalls from sewage-treatment plants and black water from septic tanks firstly affect surface waters, which then locally infiltrate shallow aquifers under high channel stages. Wastewater permeating shallow aquifers carries with it NO3 and SO4 which suppress reduction of iron oxyhydroxides in the aquifer sediments and so suppress the natural release of As to groundwater.

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The Editor,
Science of the Total Environment

Pollutant sources in an arsenic-affected multilayer aquifer in the Po Plain of Italy: implications for drinking water supply

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Marco Rotiroti, John McArthur, Letizia Fumagalli, Gennaro A. Stefania, Elisa Sacchi and Tullia Bonomi

Dear Sir,

The authors of the script titled above submit it for publication in *Science of the Total Environment* as a research article. The study addresses the issue of human impacts on groundwater in the Po Valley of Italy, where groundwater is the principal source of water used for domestic supply. The paper exploits the developing use of Cl/Br as a tracer of anthropogenic impact, and shows that in some circumstances (that is, in our area) standard interpretations of Cl/Br do not work and need modification before correct conclusions can be drawn.

We find that groundwater pumped for municipal supply is threatened both by salinity and arsenic-pollution, both of which are drawn into the aquifer in response to pumping. Given that close to 10 million consumers in the Po Valley rely on groundwater for domestic supply, we feel a high-impact journal would be a suitable place to publish these results in order to help us alert appropriate national authorities to the problem they face. For these reasons, I think that this manuscript fits the Aims and Scope of *Science of the Total Environment* and thus is suitable for publication as a research article.

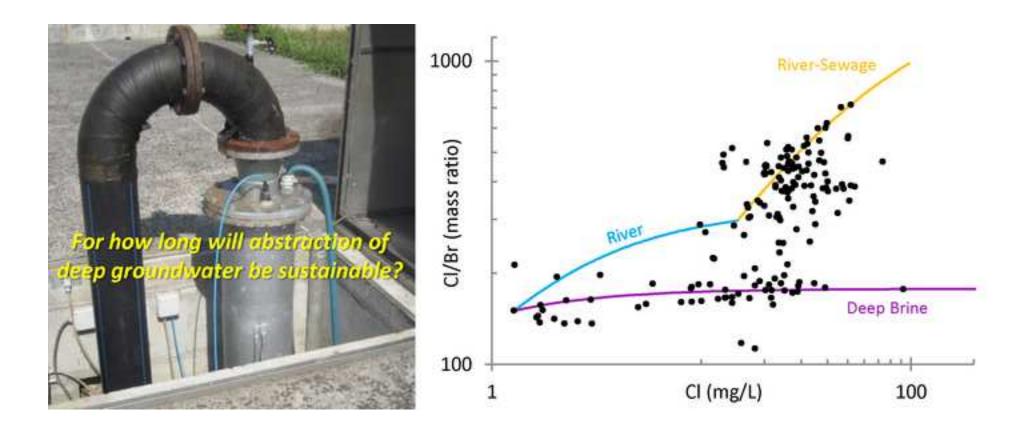
The manuscript has been prepared in accordance with the *Science of the Total Environment* guide for authors. It has not been previously published, in whole or in part, and that it is not under consideration by any other journal. All authors are aware of, and accept responsibility for, the manuscript.

All authors disclose any actual or potential conflict of interest including any financial, personal or other relationships with other people or organizations within three years of beginning the work submitted that might inappropriately influence, or be perceived as influencing, their work.

Sincerely,

Marco Rotiroti

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# \*Highlights (for review)

# Highlights

Groundwater in the Po Plain is impacted by geogenic As-pollution and human activities

These impacts are evaluated using Cl/Br ratios and stable isotopes as tracers

Shallow aquifers are recharged by surface channels impacted by sewage effluent

Deep aquifers are partly recharged by seepage from overlying aquifers enriched in As

This produces increasing As with time in the deep aquifer tapped for drinking supply

# Pollutant sources in an arsenic-affected multilayer aquifer in the Po Plain of Italy:

implications for drinking-water supply

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### **ABSTRACT**

We took an area of ~150 km² around the town of Cremona to be an analogue for hydrogeological and hydrochemical conditions in arsenic-polluted and arsenic-free groundwaters across the Po Plain of northern Italy. We investigated anthropogenic influences on ground and surface water in the area using Cl/Br ratios,  $\delta^{18}$ O/ $\delta^{2}$ H and other hydrochemical data from 32 groundwater wells, 9 surface waters, a sewage outfall and a rainwater sample.

The deep aquifer (160–260 m below ground level), which is tapped widely for public supply, is partly recharged by seepage from overlying aquifers (65–150 m below ground level) that are As-polluted (up to 144 µg/L As), producing a trend of increasing As with time. This threatens drinking water quality across the Po Plain where natural As-pollution of groundwater in aquifers at intermediate depth (50–120 m below ground level) is a basin-wide problem.

Groundwater quality in deep aquifers appears free of anthropogenic influences. In contrast, shallow groundwater and surface water are strongly affected by such pollution, although in some areas, quality remains unaffected. Outfalls from sewage-treatment plants and black water from septic tanks firstly affect surface waters, which then locally infiltrate shallow aquifers under high channel stages. Wastewater permeating shallow aquifers carries with it NO<sub>3</sub> and SO<sub>4</sub> which suppress reduction of iron oxyhydroxides in the aquifer sediments and so suppress the natural release of As to groundwater.

Keywords. Aquifer; Recharge; Cl/Br; As; Cl; Cremona.

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#### 1. Introduction

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Human settlements on alluvial plains commonly use groundwater for domestic consumption, industrial use, and irrigation. Natural and/or anthropogenic pollution in such settings may lower groundwater quality and so limit exploitation, as is the case in the Po Plain of northern Italy (Onorati et al., 2006). The Po Plain is Italy's largest alluvial basin and one of the larger in Europe, covering an area of 46,000 km<sup>2</sup>. The plain is crossed by the Po River, which flows 652 km from W to E collecting water from 141 total tributaries and hundreds of irrigation channels. These channels discharge into the Po River the excess of irrigation water which is originally diverted from other Alpine rivers that flow from N to S. The Po Plain is home to around 20 million inhabitants and is the most important economic area of Italy. As a consequence, human activities affect both the 75% of its area that is agricultural land and also the 10% that is urban development (Falcucci et al., 2007). Groundwater abstracted from the Po Basin is used for irrigation, domestic supply, and industrial processes (AdBPo, 2006). Irrigation in the Po Plain uses ~10 billion of m<sup>3</sup>/y taken from rivers (67%, and distributed to the field by channels), groundwater (22%) and springs (11%) (Zucaro, 2011). In order to avoid human impacts, groundwater for public supply was initially taken from depths > 150 m. In some areas, such deep abstraction encounters natural contamination by As (Carraro et al., 2015; Molinari et al., 2012; Rotiroti et al., 2014b). To avoid this As-pollution, abstractions were deepened 200 – 300 m below ground level (m bgl) where As concentrations are low, possibly because As is sequestered in neoformed sulphide minerals (Carraro et al., 2015; Rotiroti et al., 2014b). Groundwater from depths > 200 m is up to ~54,000 years old and its exploitation may constitute mining of resource that is replenished only slowly (Martinelli et al., 2014) by leakage from overlying aquifers that are As-polluted (Rotiroti et al., 2014b) and/or by upconing from underlying aquifers that are saline (Conti et al., 2000; Martinelli et al., 2014). Such migration is driven only by the hydraulic gradients caused by abstraction (Vassena et al., 2012). Deep abstraction may even induce land subsidence (Martinelli et al., 2014). Given the long-term threat to the sustainability of deep-groundwater posed by groundwater abstraction, we

Given the long-term threat to the sustainability of deep-groundwater posed by groundwater abstraction, we have set out primarily to (a) assess the source of recharge to deep aquifers tapped for domestic supply and (b) for deep groundwater in aquifers used for public supply, to test the potentially adverse effects on water quality in that

aquifer posed by influx of As-polluted groundwater and saline groundwater from adjacent. We also examine sources of recharge to shallow aquifers and evaluate, if any, their degree of anthropogenic contamination and develop a conceptual model of As release in the Po Plain.

#### 2. Materials and Methods

#### 2.1. Po Plain Aquifer Geology and Architecture

The alluvial systems of the Po Plain are underlain by Pliocene marine deposits and comprise Pleistocene sediments prograded from W to E and were then overlain by Holocene fluvial sediments (Garzanti et al., 2011; Marchetti, 2002). Alpine glaciations significantly increased the rate of glacio-fluvial aggradation and yielded gravel and sand units that are intercalated into units of silt/clay. The thickness of the silt/clay units increases from north (the Alpine foothills) to south (the Po River), reflecting the waning transport energy of glacial rivers (Ori, 1993). This geological setting is particularly evident in Lombardy Region (Figure 1), where monolithic aquifer of gravel and sand in the northern part of the plain (the higher plain) passes southwards into a multilayer system around the Po River (the lower plain) (Bonomi, 2009; Cavallin et al., 1983; Perego et al., 2014).

In the multilayer system of lower plain, the deeper aquifers have a sluggish circulation and so longer residence times for groundwater under natural conditions of flow (Martinelli et al., 2014). The long residence times, and confinement of the deeper aquifers, promote reducing conditions and the mobilization of As, Fe, Mn and NH<sub>4</sub> driven by degradation of organic matter buried in peat sediments (Carraro et al., 2013; Francani et al., 1994; Rotiroti et al., 2014b; Zavatti et al., 1995).

#### 2.2. Study Area

This work refers to a 150 km² area around the town of Cremona (lower Po Plain, N Italy; ~70,000 inhabitants). The details of aquifer architecture and aspects of groundwater quality, including As-pollution have been presented in Rotiroti et al. (2015a, 2015b, 2014a, 2014b) so only a summary is given here.

This multilayer aquifer comprises 5 aquifer units at differing depth ranges, where U means unconfined, S means semi-confined, and C means confined: U (0–25 m); S (30–50 m); C1 (65–85 m); C2 (100–150 m); C3 (160–260 m). Aquifers underlying C3, classified as Aquifer Group B (Carcano and Piccin, 2002), are not exploited in this

area since they are saline (Conti et al., 2000). Flow in aquifers U and S is from north to south owing to a strong topographic control. Flow direction in the deeper confined aquifers (C1, C2, C3) is from NW to SE, which is also the direction of regional groundwater flow (Vassena et al., 2012). The minimum hydraulic head is seen in C2 (Rotiroti et al., 2014b), resulting in convergent flow into aquifer C2 from both above and below. However, the presence of well-fields for public water supply tapping aquifer C3 locally induces a fall of hydraulic heads (up to 6 m) (Cambi et al., 2005) such as to induce a reversal of flow to C3.

In aquifer U, redox condition range from reducing to oxidising in response to local factors. In other aquifers, groundwater is anoxic and contains As, Fe, Mn and  $NH_4$  in concentrations that commonly exceed regulatory limits (10, 200, 50 and 500  $\mu$ g/L, respectively).

The Adda and Po Rivers flow across the area (Figure 1). In addition, multiple shallow irrigation channels both direct river water to fields for irrigation and act as drains at times of excess outflow when water tables are high *i.e.* from May to August. Two main collector channels, the Morbasco and Cerca channels (Figure 1) were sampled for this work. The Morbasco is ~32 km long and flows into the Po River. It receives treated sewage from the municipal sewage-treatment plant of Cremona as well as untreated discharges from unsewered sanitation. The Cerca Channel (~6 km long) flows into the Morbasco Channel and is the tail of the Naviglio Civico di Cremona Channel, a ~57 km long canal fed by Adda and Oglio Rivers.

The public water supply to Cremona comprises groundwater from two well-fields, one of 9 wells that lies 100 m west of Cremona (Figure 1, box B) and another of 10 wells that lies east 200 m of Cremona (Figure 1, box A). Both tap aquifer C3, which is the deepest aquifer. Before addition to the public supply, the groundwater is treated to reduce concentrations of As, Fe, Mn and NH<sub>4</sub> (Sorlini and Gialdini, 2014).

Rainfall in the study area is typically 750 mm per year (Bonomi et al., 2008), falling mostly in spring (April/May) and autumn (October/November) (Ginocchi et al., 2016). The study area is mostly agricultural, with prevailing maize cultivation (Bartoli et al., 2012). From May to August maize is grown under surface (border) irrigation. Irrigation water in Lombardy Region derives from rivers (96.8%), feeding an extensive channel network, groundwater (0.7%) and other sources as springs, lakes, etc. (2.5%) and counts a total volume of ~8 billion of m<sup>3</sup>/y (Zucaro and Corapi, 2009). This type of irrigation started as early as the 12<sup>th</sup> Century with the construction of the first important channels (Marchetti, 2002), and it has been practised on its present scale since the 50s/60s. Irrigation

constitutes an important source of recharge for shallow aquifers, together with precipitation, so much that their hydraulic heads have an increase generally from April to September (Facchi et al., 2004).

An important economic activity in the area that could have implications on water quality is livestock farming, in particular, piggery. Pig manure is often used as soil fertilizer, instead of synthetic compounds such as ammonium sulphate, so that this area was classified as nitrate-vulnerable zone (91/676/EEC).

#### 2.3. Water Quality Data

We collected 32 groundwaters from private and public supply wells, 9 surface waters (5 river and 4 channel), one sample of rainwater and a sample of outfall from Cremona's municipal wastewater treatment plant from July 2012 to October 2014 (Figure 1). In addition, the water supply company of Cremona provided unpublished historical data on water quality (data for 115 groundwaters from 37 wells over the period 2001-2011).

Samples were analysed for major ions, trace elements (As, B, Bi, Br, Cd, Cr, Co, Cu, Fe, I, Mn, Mo, Ni, Pb, Se, Sb, V, and U), dissolved organic carbon (DOC),  $\delta^{18}$ O and  $\delta^{2}$ H in water,  $\delta^{13}$ C in dissolved inorganic carbon (DIC) and  $\delta^{15}$ N in NH<sub>4</sub>. Water sampling and laboratory analyses were performed using standard methods, details are reported in Table S1 together with method quantitation limits and the full list of measured parameters.

#### 2.4. End-member Mixing Models

To identify the influence of salt, as a proxy for human influence on ground and surface waters, we use a mixing model of Cl/Br mass ratio against Cl concentration (Alcalá and Custodio, 2008; Davis et al., 2004, 1998; Hoque et al., 2014; Katz et al., 2011; McArthur et al., 2016, 2012; Panno et al., 2006; Vengosh and Pankratov, 1998). To investigate whether aquifers were being salinized by deep brine injection or flushed of brine by freshwater, we used the mixing-model approach of Ravenscroft and McArthur (2004), which uses B and Cl to identify ion-exchange of B through these processes. For these models, the end-members were as follows:

- (a) uncontaminated groundwater represented by our two most dilute groundwaters, Well 68 with 1.3 mg/L
   Cl, 19.1 μg/L B, and Cl/Br of 150, and Well 64, with 1.7 mg/L Cl, B 13.8 μg/L and Cl/Br of 151;
- 137 (b) river water with 14.8 mg/L Cl and Cl/Br of 297, the average composition of our 10 river samples;
- sewage effluent, with 98.2 mg/L Cl and Cl/Br of 981;
- 139 (d) road salt from the local highway company, which has 60.4% Cl and Cl/Br of 7545;

- domestic salt, with 59.7% Cl and Cl/Br of 3970, the average composition of 14 samples of table, cooking and dishwasher salts obtained from retail outlets in Cremona (Table S2);
- a deep brine of Monticelli (Boschetti et al., 2010), with 73,015 mg/L Cl, 17,673 μg/L B, and a Cl/Br of 152, and a deep brine of Cremona, end member brine with Cl/Br of 177 and a Cl of 19,500, the Cl/Br being derived from the slope of a linear regression of Br on Cl for aquifer C3 (Figure S1, *r* = 0.999).

#### 3. Results

#### 3.1. Water Quality

Measured water quality data are reported in Table S1. The compositions are similar to groundwater compositions from the area reported before (Rotiroti et al., 2014b). In particular, concentrations of As, Fe, Mn and NH<sub>4</sub> are high. Concentrations of B, Ba, Cd, Cr, Co, Cu, Mo, Ni, NO<sub>3</sub>, Pb, Se, Sb, SO<sub>4</sub>, V, and U are less than WHO and Italian regulatory limits. Depth profiles of conservative and redox-sensitive species measured in October 2014 are shown in Figure 2, and other relevant parameters in Figure S2.

The EC is higher in shallow aquifers (U and S; mean value of 947  $\mu$ S/cm) than in deep aquifers (C1, C2 and C3; mean of 516  $\mu$ S/cm) aquifers. Concentrations of Cl and B have similar profiles, with highest concentrations in aquifer U (up to 52 mg/L and 238  $\mu$ g/L, respectively), a downward decrease to a minimum in C2 with the underlying aquifer C3 having slightly higher concentrations (up to 28.8 mg/L and 87.6  $\mu$ g/L, respectively).

Nitrate is detectable only in aquifer U, with concentrations  $\leq$  42.8 mg/L. Concentrations of Fe and Mn are highest in reduced parts of aquifer U ( $\leq$  5.41 and  $\leq$  0.98 mg/L, respectively) and decrease downward to aquifer C3. Concentrations of SO<sub>4</sub> are highest in aquifer U ( $\leq$  171 mg/L) and a decrease downward, being at or below the detection limit (0.05 mg/L) in the deep aquifers. Concentrations of DOC have no clearly identifiable depth-trend and range between 2.8 and 7.6 mg/L, with one outlier of 14.1 mg/L in aquifer U (Well 4).

The profiles of Br and I comprise higher values in U (up to 132.8 and 13.1  $\mu$ g/L, respectively), a downward decrease and a peak in C3 (up to 166.1 and 25.6  $\mu$ g/L, respectively). Concentrations of As, NH<sub>4</sub> and PO<sub>4</sub> and H<sub>4</sub>SiO<sub>4</sub> increase downwards to S and C1 ( $\leq$  168  $\mu$ g/L, 2.0, 1.4, and 42 mg/L, respectively) but decrease downwards thereafter.

Duplicated sampling of groundwater for some wells after 27 months shows little difference in composition (Figure S3). Concentrations of As, Fe, Mn and NH<sub>4</sub> from July 2012 are slightly higher than those from October 2014 but not statistically difference at the 5% significance level (Mann-Whitney U test; p-value of 0.21, 0.67 and 1 for As, Fe and Mn, respectively). A small but significant difference in NH<sub>4</sub> (p-value of 0.03) might reflect the fact that samples collected in October 2014 were filtered through 0.2 μm whilst those collected in July 2012 were not.

For surface waters, Br and I apart, all concentrations were higher in channel waters than in river water (Table S1 and Figure 2). For Br and I, the converse was true, with Br being 50  $\mu$ g/L in rivers and 37  $\mu$ g/L in channels whilst I was 4.0 in rivers and 1.6 in channels.

No surface-water sample exceed the Italian regulatory limits for surface waters (D. Lgs. 152/06). River water quality results quite stable over the two monitoring dates (July and October 2014) whereas channel waters had higher mean concentrations for all measured species in October 2014 with respect to July 2014.

Rainfalls in Cremona during 3 days before the surface water sampling were 3.6 mm in July 2014 and 0.0 in October 2014; the daily average stage of Po River at Po3 location during the sampling was 28.56 m above sea level (a.s.l.) in July (22/07/14) and 28.26 m a.s.l. in October (07/10/2014) (ARPA Lombardia). On the basis of these data and visual observation in the field, it can be considered that channels had higher stages in July 2014 and lower stages in October 2014 whereas the Po River had comparable stages during the two sampling dates.

#### 3.2. Stable Isotopes

Values of  $\delta^2$ H vs  $\delta^{18}$ O (Figure 3a) plot on or close to the local meteoric water line (LMWL) of northern Italy (Longinelli and Selmo, 2003) and of Cremona (Francani et al., 1994) showing that evaporation has not affected our groundwaters during their time as recharge. The groundwaters cluster between the more enriched values for local precipitation and the more depleted values for the Po River, the latter reflecting the strong influence of water from the Western Alps (Marchina et al., 2015). Depth profiles of  $\delta^{18}$ O show little by way of trends (Figure 3b), with groundwaters from shallow and deep aquifers having similar values, except for 3 shallow wells (5, 11 and 26) that are more enriched. The  $\delta^{18}$ O values of shallow groundwaters fall between those of the Po River (-10.1 to -9.2; Marchina et al., 2016, 2015) and local precipitations (-7.73 to -5.95 %; Longinelli and Selmo, 2003). Values of  $\delta^{18}$ O for deep groundwaters fall within the range reported for deep aquifers of the Po Plain by Pilla et al. (2006) of -9.0 to -8.2 % and nearby sites at Lodi (-8.93 to -8.33 %; Guffanti et al., 2010), Pavia (~-8.7 to ~-8.5 %; Pilla, 1998),

Milano ( $\sim$  -9.8 to  $\sim$  -8.3 %; Avanzini et al., 1994), Piacenza (-9.42 to -7.38 %; Martinelli et al., 2014) and Parma (-9.86 to -7.05 %; Martinelli et al., 2014).

Measured  $\delta^{13}$ C values range from -9.3 to -14.9 ‰, an interval that reflects supersaturation with respect to calcite and dolomite (Rotiroti et al., 2015b). The values of  $\delta^{15}$ N measured in NH<sub>4</sub> range from 4.09 to 7.81 ‰. These data fall within the range of NH<sub>4</sub> produced by the mineralization of soil organic nitrogen and peat (from ~ +2 to ~ +8 ‰; Kendall, 1998), of sewage and septic tanks (~ +4‰; Gooddy et al., 2016) and of wastewaters outfalls (~ +4 to ~ +16 ‰; Hood et al., 2014), but are distinct from the NH<sub>4</sub> isotopic composition of synthetic fertilizers (-7.4 to +3.6‰; Vitòria et al., 2004b), of pig manure (+8 to +15‰; Vitòria et al., 2004a) and of landfill leachate (from ~ +7 to ~ +10 ‰; Gooddy et al., 2014).

#### 3.3. Cl/Br Ratio

On a cross-plot of Cl/Br against Cl (Figure 4) that includes both our data and historical, most groundwaters from the shallow aquifers (U and S) fall close the river-sewage mixing line or are more enriched in Cl. One sample (Well 3) is enriched in Br. Only one groundwater (Well 32) plots towards the unpolluted field marked by low Cl/Br and low Cl (McArthur et al., 2012 and reference therein).

Deep groundwaters show no anthropogenic influences. The deep aquifers (C1, C2 and C3) mostly plot along a mixing line between our dilute groundwater end-member and a deep brine end-member. At lower Cl concentrations, some addition of Br from organic decay may have pulled many of the more dilute groundwaters below brine-mixing lines. The Cl/Br value of 177 is consistent with the Cl/Br of 152 measured in the Monticelli brine (Boschetti et al., 2010), substantiating the hypothesis that salinity in aquifer C3 can be mainly governed by a mixing between dilute groundwater and deep brine. The sample from Well 50, and in a minor way that from Well 66, shows higher Cl/Br with respect to the others measured in deep aquifers.

River and channel data cluster separately, the former having lower Cl/Br. Po River waters (Po1, Po3 and Po4) are similar and show little difference between sampling dates (average Cl/Br of 327). Waters Ad1 and Po2 have similar Cl/Br (mean of 280 in July 2014 and 223 in October 2014). Water Po2 likely reflects the composition of the Adda rather than the Po as it was collected ~500 m downstream of the confluence of the Adda and Po Rivers before the rivers mixed fully. Channel samples plot close to the river-sewage mixing-line. Those collected in

October 2014 have an average Cl/Br of 451 that overlaps values for the shallow aquifers. Those collected in July 2014 have an average Cl/Br of 478 and plot slightly apart from shallow aquifers.

The sample of rainwater plots at low Cl but has a highish Cl/Br of 213. Similar high values were found for rain in south-eastern France at Gard (Cl/Br 201) and Hérault (Cl/Br 269) (Ladouche et al., 2009). The sample collected from Cremona's sewage-treatment plant has a Cl/Br of 981; this is slightly higher than values for sewage effluent from Israel (410 – 873; Vengosh and Pankratov, 1998) and the USA (300 – 600; Davis et al., 1998) but is within the range of septic-tank effluent in West Bengal, India (690 – 2530; McArthur et al., 2012).

#### 4. Discussion

### 4.1. Recharge to shallow aquifers

Groundwater quality of most shallow groundwaters are contaminated by human activity as shown by the high EC and Cl/Br values up to 1174 µS/cm and 624, respectively. Nevertheless, rare instances of uncontaminated groundwater occur *e.g.* Well 32, with Cl/Br of 197, a value close to that of 214 found in our spot rainwater sample. Such rare exceptions apart, the high Cl/Br of most shallow groundwater and channel waters arises from inputs of raw and treated sewage effluents with Cl/Br up to 981 (Figure 4). The Morbasco Channel, in particular, receives the outflow of 3 wastewater treatment plants and some untreated effluents from domestic unsewered sanitation.

Shallow groundwaters do not fall on the mixing line between effluent and uncontaminated groundwater, a fact that suggests direct contamination of groundwater by effluent is uncommon, rather, shallow groundwaters fall on a mixing line between channel waters and sewage effluent, a concordance that suggests effluent recharge aquifers indirectly as a component of channel water after dilution with natural channel flow. Our Cl/Br data therefore confirms the findings (Facchi et al., 2004; Pilla et al., 2006) that irrigation channels in the Po Plain mainly act as loosing streams and are an important source of recharge to shallow aquifers. Notwithstanding that, point-source contamination by untreated domestic sewage has been reported from parts of the Po Plain (Delconte et al., 2014; Sacchi et al., 2013).

The tendency of some shallow groundwater  $\delta^{18}O/\delta^2H$  values toward those of local precipitation (Figure 3) suggests that rainfalls are also an important source of recharge for shallow aquifers.

A minority of shallow groundwaters plot in a scatter array towards the low Cl/Br of groundwater from Well 3. These lower Cl/Br suggest inputs of Br from organic degradation may locally contribute Br preferentially to Cl and so lower Cl/Br. Groundwater from Well 3 contains 4.0 mg/L of Fe, which attests to the reducing capacity of aquifer U at this point and supports the suggestion that, locally, degradation of buried peat may be driving redox (Rotiroti et al., 2015b, 2014b). Such degradation would contribute additional Br to groundwater. Conformation of this suggestion derives from the TANGRAM<sup>©</sup> database (Bonomi et al., 2014) which shows that 4 m of the total 6-m screen-length of Well 3 straddles a peaty sand unit. Nevertheless, given the strong human impact on shallow groundwaters, we cannot say definitively that street runoff (Vengosh and Pankratov, 1998) and/or brominated flame retardants (Winid, 2015) are not the cause of the lower Cl/Br array in shallow groundwaters.

The generally higher Cl in shallow groundwaters than in channel water is probably not due to evaporation of channel water before and during infiltration because the  $\delta^{18}O$  and  $\delta^{2}H$  of shallow groundwaters show no evaporative trend (Figure 3). The data fall within a narrow range (–9.1 and –8.4 for  $\delta^{18}O$ ; –58.9 to –55.5 for  $\delta^{2}H$ ) and plot close to the LMWL for the area (Figure 3). The higher Cl concentrations in groundwaters compared to channel waters is therefore attributed to time-biased sampling of channel water that vary their Cl concentration through the year in response to changing inputs of sewage effluent and its dilution by natural flow.

Shallow groundwaters have higher Cl/Br than river waters (Figure 4), and a different  $\delta^{18}$ O/ $\delta^{2}$ H (Figure 3), proving that the study area's main rivers, the Po and the Adda, do not directly recharge shallow aquifers. This finding confirms similar conclusions of others (Marchina et al., 2016; Martinelli et al., 2014; Rotiroti et al., 2014b) that the Po River is mainly a gaining river fed by groundwater in the study area, rather than a losing river sourcing groundwater, despite widespread abstractions of groundwater for industrial and domestic use.

The high concentrations of  $SO_4$  found in shallow groundwater ( $\leq$  171 mg/L) must have an anthropogenic source, given the human impact on shallow groundwater revealed by Cl/Br values and  $SO_4$ /Cl mass ratios up to 7.2 and may derive from the use of manure fertilizer (Menció et al., 2016). Concentrations of  $NO_3$  are generally low, owing to the prevalence of reducing conditions, and so nitrate reduction, in much of the shallow aquifers (U and S) around Cremona, as is the case for most of the lower Po Plain (Balestrini et al., 2016; Sacchi et al., 2013). In rare oxic parts of the shallow aquifers nitrate persists at high concentrations (Well 5, 43 mg/L  $NO_3$ ) in association to high  $SO_4$ , further confirming its anthropogenic source.

#### 4.2. Recharge to deep wells

Deep groundwaters with  $\delta^{18}$ O in the range -9.0 to -8.2 ‰, a range similar to ours, have been traced to recharge areas in the high Po Plain and Alpine foothills (Pilla et al., 2006), with elevations of 150-250 m asl, so our deep groundwaters likely are also recharged at similar locations.

Concentrations of Cl in the deep aquifer C3, the most exploited for public supply, range up to 90 mg/L. With the exception of groundwater from Well 66, the fresher groundwater samples with Cl < 10 mg/L plot below the mixing line for brine with Cl/Br of 177 (Figure 4). This suggests excess Br from organic decay may be influencing these dilute samples. The suggestion is confirmed by a close co-variance of excess Br with both NH<sub>4</sub> (Figure S4a) and DOC (Figure S4b), all three of which are products of microbial fermentation of sedimentary organic matter, which is known to increase Br in groundwater at the expense of Cl: degradation of lignite and peat gave groundwaters in the Hula Valley of Israel values of Cl/Br as low as 4 (Nissenbaum and Magaritz, 1991). Degrading organic matter in soils of Western Australia were reported with Cl/Br between 6 and 10 (Gerritse and George, 1988), with similar low values in peats from Chile and Germany (Biester et al., 2012; Biester et al., 2006). A quantification of the relation between excess Br and DOC was achieved by Desbarats et al. (2014).

Deep groundwaters from aquifer C3 with Cl concentrations above 10 mg/L plot close to a mixing line with Cl/Br of 177 (Figure 4). Deep brines are common in the Po Plain, *e.g.* the Monticelli brine with a Cl/Br of 152 (Boschetti et al., 2010). Monticelli is located on the same buried thrust that underlies the study area of Cremona (Bonini, 2007). Buried thrust may channel uprising deep brines (Pilla et al., 2015) and impose a degree of uniformity to their composition. The higher Cl concentrations in our C3 groundwater therefore are assumed to come from minor mixing with such deep brines, which have Cl/Br considerably below the seawater value of 288.

This mixing appears to have been historical, and is now in reverse, as aquifer C3 is being flushed of brine. Six indicators attest to this flushing: (a) the high concentrations of B in groundwater from aquifer C3 ( $\leq$  87 µg/L) release by ion-exchange; (b) mass ratios of Na/Cl mostly above 2 and ranging up to 8.9, compared to 0.54 for seawater, showing the presence of excess Na over Cl; (c) a strong correlation between B and Na (Figure S5a); (d) a strong inverse correlation between Ca and Na (Figure S5a); (e) groundwater compositions that plot along the exchange line as defined by Re et al. (2013) (Figure S5b), and (f) compositions that plot in the freshwater–flushing field of a B vs Cl cross plot (Ravenscroft and McArthur, 2004) (Figure S5c).

Iodine concentrations in aquifer C3 range up to 26  $\mu$ g/L, giving Cl/I mass ratios between 890 and 2020 (Table S1; mean 1258). These values are well below the marine value of ~300 000 (Skinner and Berger, 2003). Such low Cl/I have been reported before for deep groundwater of the Po Plain (Boschetti et al., 2010; Conti et al., 2000) and must result from the input to groundwater of I from organic degradation.

The above discussion points out that at least part of groundwater abstracted from C3 seems to be replaced from underlying aquifers in Wells 67 and 110 and from overlaying aquifers in Well 66.

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#### 4.3. Temporal trends of groundwater composition

Our discussion above shows that aquifer C3 is being flushed of saline contamination but not at what rate. Development of aquifer C3 for most public supply in the area raises the question as to how sustainable the resource might be, how quickly groundwater composition is changing with time, and whether the quality is increasing or decreasing *i.e.* flushing is continuing now.

Our repeat-sampling interval of 27 months is too short to do other than place an upper limit on rate of change of groundwater composition. Nevertheless, we attempt to do so using the average concentration of As and Cl in C3 over the sampling dates available. These are reported in Figure 5. We use average concentration, rather than the time-series for each well because the available sampling dates are a few, and wells were not all sampled on the same dates. The average compositions suggest that no discernible trend is shown by Cl concentrations, but that As concentrations show a slight increase with time. The increase of As in C3 cannot result from upward leakage of Asrich groundwater in underlying aquifers (Conti et al., 2000) as that would be accompanied by Cl, which is not seen; indeed, the aquifer is freshening. The change in As concentration may result from continuing slow reduction of sedimentary iron oxyhydroxides with concomitant release of As to groundwater (Rotiroti et al., 2015b). Such an hypothesis is compatible with the near absence of SO<sub>4</sub> in groundwater in aquifer C3, reduction of which had it been present might have sequestered As from solution in neoformed pyrite. Anyhow, the As release via iron oxyhydroxides reduction coupled to organic matter degradation seems to have a minor contribution to As-pollution in C3 since here excess Br and As are uncorrelated (Figure S4c). Finally the increase in As concentrations might result from leakage downwards of As-polluted groundwater from overlying aquifers C1 and C2 in response to abstraction. This seems confirmed by the October 2014 data since Well 66, which has the highest As concentration in C3 (59.8 µg/L), is the sole from C3 that plots above the mixing line for brine.

#### 4.4. Strengthening the conceptual model for As release

In anoxic aquifers worldwide, natural pollution by arsenic arises from the reductive dissolution of sedimentary iron oxyhydroxides. This mechanism has been invoked to explain such pollution in the USA (Korte, 1991; Korte and Fernando, 1991; Matisoff et al., 1982), the Bengal Basin (Nickson et al., 1998, 2000 et seq.), the Red River Basin, Vietnam (Berg et al., 2001; Postma et al., 2007), alluvial aquifers in Pakistan (Husain et al., 2012; Nickson et al., 2005), the Po Plain of Italy (Carraro et al., 2013; Rotiroti et al., 2014b), the Mekong River Basin of Cambodia (Rowland et al., 2007 et seq.) and other worldwide locations (see Gulens et al., 1979; Ravenscroft et al., 2009 for reviews). The reduction of sedimentrary iron oxyhydroxides is driven by microbial metabolism of organic matter (Banfield and Nealson, 1997; Chapelle and Lovley, 1992; Chapelle, 2000; Lovley, 1997; Nealson, 1997). In the multilayer aquifers around Cremona, the reduction has been postulated to be driven by the degradation of peat incorporated into semi-permeable silty and clayey aquitards (Rotiroti et al., 2015b; Rotiroti et al., 2014b).

In groundwaters that show little anthropogenic influence (aquifers C1, C2, C3 and well 32 tapping aquifer S) concentrations of As, PO<sub>4</sub>, and NH<sub>4</sub> co-vary, albeit weakly (Figure S6). The covariance provides confirmation of the strong co-absorption onto iron-oxyhydroxides of As and PO<sub>4</sub> and their concomitant release on its reduction (Ravenscroft et al., 2001). Microbial metabolism of organic matter in anoxic waters not only reduces iron oxyhydroxides but also generates NH<sub>4</sub> from amino acids, so the positive correlation between As and NH<sub>4</sub> (Figure S6b) although weak, provides some confirmation that these process co-occur. Moreover, the measured values of  $\delta^{15}$ N, together with other indicators (Francani et al., 1994; Rotiroti et al., 2015b), suggest that NH<sub>4</sub> derives from natural organic matter. Therefore, the assumption that peaty aquitards are probably the main source of organic matter driving As release in this system seems reasonable.

Reduction of SO<sub>4</sub> can generate neoformed pyrite that sequesters As. Concentrations of As are highest in aquifers S and C1 but their concentrations of SO<sub>4</sub> differ markedly (Table S1). These observations suggest that the process of SO<sub>4</sub>-reduction, if it occurs at all in these aquifers, does not to reduce As concentrations to zero.

The release of As to shallow aquifers in Cremona is suppressed by anthropogenic inputs of NO<sub>3</sub> and possibly SO<sub>4</sub> from wastewater, as occurs in the Bengal basin. (McArthur et al., 2016, 2012). Reduction of iron oxyhydroxides is suppressed by NO<sub>3</sub> because NO<sub>3</sub> is reduced more easily than Fe<sup>3+</sup> by microbial processes. In addition, the presence of SO<sub>4</sub> allows co-precipitation of As in sulphides formed by SO<sub>4</sub>-reduction. A weak negative

correlation between Cl/Br, indicative of wastewater discharges high in NO<sub>3</sub> and SO<sub>4</sub> and concentrations of As (Figure S7) appears to confirm the suppressant effect of waste waters on As-pollution in the study area. Increasingly stringent regulations on the use of N-based fertiliser in the Po Plain (91/676/EEC), and so the designation of large areas of the Po Plain as nitrate-vulnerable zones (Sacchi et al., 2013) therefore has the potential to decrease NO<sub>3</sub> inputs to the shallow aquifers, thereby allowing the wider development of anoxia and so an increase in As-pollution.

#### 5. Conclusions

This study dealt with the assessment of the recharge to shallow aquifers and the source of water to deep wells used for drinking water supply in the As-affected multilayer aquifer in Cremona, considered as an analogue for the aquifers of the lower Po Plain. Our findings are that:

- shallow aquifers receive considerable recharge from irrigation/drainage channels that are polluted by sewage effluent, giving rise to high Cl/Br values in shallow groundwaters. Nevertheless, in some areas shallow aquifers are still free from such an influence;
- b) shallow groundwater is not much affected by As-pollution because of the high NO<sub>3</sub> and SO<sub>4</sub> concentrations in channel water due to sewage effluents. Reduction of NO<sub>3</sub> suppresses As-pollution by suppressing reduction of iron oxyhydroxides. Reduction of SO<sub>4</sub> generates pyrite that can sequester As from solution and so lower concentrations of pollutant As;
- c) deep groundwater abstractions from aquifers C2 and C3 supply much of the regions domestic supply, especially for the town of Cremona. The supplies are unaffected by anthropogenic pollution but have concentrations of Cl that are increased by historical mixing with deep brines. The deep aquifers are currently being flushed of this Cl-contaminated water by fresher water, as evidenced by strong ion-exchange loss of Ca from recharging waters, with concomitant increase in B and Na;
- d) the As-pollution of deep aquifers is likely;
- e) over a 27-month period, As-pollution in the deep aquifers may have increased slightly as a result of drawdown of As-rich groundwater from overlying aquifers. If confirmed, the increases pose some threat to the treatment plants currently dealing with purification of groundwater for public supply, since treatment may be based on an assumed constant composition of groundwater;

f) to assess this risk, routine monitoring for As on a monthly basis should be instituted. These findings related to the Cremona area could have implications on drinking supply management in the whole lower Po Plain (5-10 million inhabitants) since here natural As contamination in intermediate aquifers is a regional problem and many wells that serve drinking supply are tapped in deep aquifers underlying the peak of As concentration. Acknowledgments We thank Paolo Vicentini of Padania Acque S.p.A, Andrea Azzoni, Barbara Pisaroni, and Giuseppina Demicheli of Province of Cremona, Municipality of Cremona, AEM Cremona S.p.A and all private owners for letting us to sample their wells. We also thank Vittorio Barella and Enrico Allais of ISO4 (Torino, Italy) for performing isotope analysis. Funding: This work was supported by Italian Ministry of Education, University and Research though the PRIN-2008 fund [grant number 2008YP85ZH]. Appendix A. Supplementary material Additional tables (Tables S1-S2) and figures (Figures S1-S7) are the supplementary material related to this article. References 91/676/EEC. Council Directive 91/676/EEC of 12 December 1991 concerning the protection of waters against pollution caused by nitrates from agricultural sources. AdBPo. Caratteristiche del bacino del fiume Po e primo esame dell'impatto ambientale delle attività umane sulle risorse idriche "Characteristics of the Po river basin and first assessment of the environmental impact of human activities on water resources". Autorità di Bacino del Fiume Po, Parma, 2006. http://www.adbpo.it/download/bacino\_Po/AdbPo\_Caratteristiche-bacino-Po\_2006.pdf Alcalá FJ, Custodio E. Using the Cl/Br ratio as a tracer to identify the origin of salinity in aquifers in Spain and Portugal. J. Hydrol. 2008; 359: 189-207. http://dx.doi.org/10.1016/j.jhydrol.2008.06.028

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415	ARPA Lombardia. Hydrography Division Website,
416	http://ita.arpalombardia.it/ITA/servizi/richiesta_dati/idro_pluvio_termo.asp (accessed 13.08.16)
417	Avanzini R, Beretta GP, Francani V, Nespoli R. Indagine preliminare sull'uso sostenibile delle falde profonde nella
418	provincia di Milano "Preliminary investigation on the sustainable use of deep groundwater in the Milano
419	province". Milano: Consorzio Acqua Potabile, 1994.
420	Balestrini R, Sacchi E, Tidili D, Delconte CA, Buffagni A. Factors affecting agricultural nitrogen removal in
421	riparian strips: Examples from groundwater-dependent ecosystems of the Po Valley (Northern Italy).
422	Agric., Ecosyst. Environ. 2016; 221: 132-144. http://dx.doi.org/10.1016/j.agee.2016.01.034
423	Banfield JF, Nealson KH. Geomicrobiology: Interactions between Microbes and Minerals. Reviews in Mineralogy
424	and Geochemistry Vol 35. Washington, DC: Mineralogical Society of America, 1997.
425	Bartoli M, Racchetti E, Delconte CA, Sacchi E, Soana E, Laini A, et al. Nitrogen balance and fate in a heavily
426	impacted watershed (Oglio River, Northern Italy): in quest of the missing sources and sinks.
427	Biogeosciences 2012; 9: 361-373. http://dx.doi.org/10.5194/bg-9-361-2012
428	Berg M, Tran HC, Nguyen TC, Pham HV, Schertenleib R, Giger W. Arsenic Contamination of Groundwater and
429	Drinking Water in Vietnam: A Human Health Threat. Environ. Sci. Technol. 2001; 35: 2621-2626.
430	http://dx.doi.org/10.1021/es010027y
431	Biester H, Hermanns Y-M, Martinez Cortizas A. The influence of organic matter decay on the distribution of major
432	and trace elements in ombrotrophic mires – a case study from the Harz Mountains. Geochim. Cosmochim.
433	Acta 2012; 84: 126-136. http://dx.doi.org/10.1016/j.gca.2012.01.003
434	Biester H, Selimović D, Hemmerich S, Petri M. Halogens in pore water of peat bogs – the role of peat
435	decomposition and dissolved organic matter. Biogeosciences 2006; 3: 53-64. http://dx.doi.org/10.5194/bg-
436	3-53-2006
437	Bonini M. Interrelations of mud volcanism, fluid venting, and thrust-anticline folding: Examples from the external
438	northern Apennines (Emilia-Romagna, Italy). J. Geophys. ResSol. Ea. 2007; 112: B08413.
439	http://dx.doi.org/10.1029/2006jb004859
440	Bonomi T. Database development and 3D modeling of textural variations in heterogeneous, unconsolidated aquifer
441	media: Application to the Milan plain. Comput. Geosci. 2009; 35: 134-145.
442	http://dx.doi.org/10.1016/j.cageo.2007.09.006

443	Bonomi T, Canepa P, Del Rosso F, Rossetti A. Relazioni temporali pluridecennali di dati pluviometrici, idrologici e
444	piezometrici nella pianura lombarda tra Ticino e Oglio "Pluridecennial temporal relationship between
445	rainfall, hydrologic and piezometric data in the lombardy plain, between Ticino and Oglio". Giorn. Geol.
446	Appl. 2008; 9: 227-248.
447	Bonomi T, Fumagalli M, Rotiroti M, Bellani A, Cavallin A. The hydrogeological well database TANGRAM©: a
448	tool for data processing to support groundwater assessment. Acq. Sott. Ital. J. Groundw. 2014; 3: 35-45.
449	http://dx.doi.org/10.7343/as-072-14-0098
450	Boschetti T, Toscani L, Shouakar-Stash O, Iacumin P, Venturelli G, Mucchino C, et al. Salt Waters of the Northern
451	Apennine Foredeep Basin (Italy): Origin and Evolution. Aquat. Geochem. 2010; 17: 71-108.
452	http://dx.doi.org/10.1007/s10498-010-9107-y
453	Cambi C, Dragoni W, Passeri F, Valigi D. Contribution to the hydrogeological knowledge of the Cremona aquifer
454	system and the exploitation of new water resources. Ital. J. Eng. Geol. Envir. 2005; 1: 71-89
455	Carcano C, Piccin A editors. Geologia degli acquiferi padani della Regione Lombardia "Geology of Po Plain
456	aquifers in Lombardy region". Firenze: S.E.L.C.A., 2002.
457	Carraro A, Fabbri P, Giaretta A, Peruzzo L, Tateo F, Tellini F. Arsenic anomalies in shallow Venetian Plain
458	(Northeast Italy) groundwater. Environ. Earth. Sci. 2013; 70: 3067-3084. http://dx.doi.org/10.1007/s12665-
459	013-2367-2
460	Carraro A, Fabbri P, Giaretta A, Peruzzo L, Tateo F, Tellini F. Effects of redox conditions on the control of arsenic
461	mobility in shallow alluvial aquifers on the Venetian Plain (Italy). Sci. Total Environ. 2015; 532: 581-594.
462	http://dx.doi.org/10.1016/j.scitotenv.2015.06.003
463	Cavallin A, Francani V, Mazzarella S. Studio idrogeologico della pianura compresa fra Adda e Ticino
464	"Hydrogeological study of the plain between Adda and Ticino Rivers". Costruzioni 1983; 326/327: 1-39
465	Chapelle FH, Lovley DR. Competitive Exclusion of Sulfate Reduction by Fe(lll)-Reducing Bacteria: A Mechanism
466	for Producing Discrete Zones of High-Iron Ground Water. Ground Water 1992; 30: 29-36.
467	http://dx.doi.org/10.1111/j.1745-6584.1992.tb00808.x
468	Chapelle HF. The significance of microbial processes in hydrogeology and geochemistry. Hydrogeol. J. 2000; 8: 41
469	46. http://dx.doi.org/10.1007/pl00010973

4/0	Conti A, Sacchi E, Chiarle M, Martinelli G, Zuppi GM. Geochemistry of the formation waters in the Po plain
471	(Northern Italy): an overview. Appl. Geochem. 2000; 15: 51-65. http://dx.doi.org/10.1016/S0883-
472	2927(99)00016-5
473	D. Lgs. 152/06. Decreto Legislativo n. 152 del 3 aprile 2006 sulle norme in materia ambientale "Legislative Decree
474	on environmental regulations", 2006.
475	Davis SN, Fabryka-Martin JT, Wolfsberg LE. Variations of Bromide in Potable Ground Water in the United States.
476	Ground Water 2004; 42: 902-909. http://dx.doi.org/10.1111/j.1745-6584.2004.t01-8x
477	Davis SN, Whittemore DO, Fabryka-Martin J. Uses of Chloride/Bromide Ratios in Studies of Potable Water.
478	Ground Water 1998; 36: 338-350. http://dx.doi.org/10.1111/j.1745-6584.1998.tb01099.x
479	Delconte CA, Sacchi E, Racchetti E, Bartoli M, Mas-Pla J, Re V. Nitrogen inputs to a river course in a heavily
480	impacted watershed: A combined hydrochemical and isotopic evaluation (Oglio River Basin, N Italy). Sci.
481	Total Environ. 2014; 466–467: 924-938. http://dx.doi.org/10.1016/j.scitotenv.2013.07.092
482	Desbarats AJ, Koenig CEM, Pal T, Mukherjee PK, Beckie RD. Groundwater flow dynamics and arsenic source
483	characterization in an aquifer system of West Bengal, India. Water Resour. Res. 2014; 50: 4974-5002.
484	http://dx.doi.org/10.1002/2013wr014034
485	Facchi A, Ortuani B, Maggi D, Gandolfi C. Coupled SVAT-groundwater model for water resources simulation in
486	irrigated alluvial plains. Environ. Model. Software 2004; 19: 1053-1063.
487	http://dx.doi.org/10.1016/j.envsoft.2003.11.008
488	Falcucci A, Maiorano L, Boitani L. Changes in land-use/land-cover patterns in Italy and their implications for
489	biodiversity conservation. Landscape Ecol. 2007; 22: 617-631. http://dx.doi.org/10.1007/s10980-006-9056-
490	4
491	Francani V, Beretta GP, Bareggi A, Nobile A, Cremonini Bianchi M, Cattaneo F. Aspetti idrogeologici del
492	problema della presenza di azoto ammoniacale nelle acque sotterranee della provincia di Cremona
493	"Hydrogeological aspects of the occurrence of ammonium in groundwater in the province of Cremona".
494	Bologna: Pitagora Ed., 1994.
495	Garzanti E, Vezzoli G, Andò S. Paleogeographic and paleodrainage changes during Pleistocene glaciations (Po
496	Plain, Northern Italy). Earth-Sci. Rev. 2011; 105: 25-48. http://dx.doi.org/10.1016/j.earscirev.2010.11.004

497	Gerritse RG, George RJ. The role of soil organic matter in the geochemical cycling of chloride and bromide. J.
498	Hydrol. 1988; 101: 83-95. http://dx.doi.org/10.1016/0022-1694(88)90029-7
499	Ginocchi M, Crosta GF, Rotiroti M, Bonomi T. Analysis and prediction of groundwater level time series with
500	Autoregressive Linear Models. Rend. Online Soc. Geol. It. 2016; 39: 109-112.
501	http://dx.doi.org/10.3301/ROL.2016.59
502	Gooddy DC, Lapworth DJ, Bennett SA, Heaton THE, Williams PJ, Surridge BWJ. A multi-stable isotope
503	framework to understand eutrophication in aquatic ecosystems. Water Res. 2016; 88: 623-633.
504	http://dx.doi.org/10.1016/j.watres.2015.10.046
505	Gooddy DC, Macdonald DMJ, Lapworth DJ, Bennett SA, Griffiths KJ. Nitrogen sources, transport and processing
506	in peri-urban floodplains. Sci. Total Environ. 2014; 494–495: 28-38.
507	http://dx.doi.org/10.1016/j.scitotenv.2014.06.123
508	Guffanti S, Pilla G, Sacchi E, Ughini S. Characterization of the quality and origin of groundwater of Lodigiano
509	(Northern Italy) with hydrochemical and isotopic instruments. Ital. J. Eng. Geol. Envir. 2010; 1: 65-78
510	Gulens J, Champ DR, Jackson RE. Influence of redox environments on the mobility of arsenic in groundwater. In:
511	Jenne EA, editor. Chemical modeling in aqueous systems; speciation, sorption, solubility, and kinetics.
512	American Chemical Society Symposium Series Vol. 93, 1979, pp. 81–95.
513	Hood JLA, Taylor WD, Schiff SL. Examining the fate of WWTP effluent nitrogen using $\delta$ 15N–NH4 +, $\delta$ 15N–NC
514	$3$ – and $\delta$ 15N of submersed macrophytes. Aquat. Sci. 2014; 76: 243-258.
515	http://dx.doi.org/10.1007/s00027-013-0333-4
516	Hoque MA, McArthur JM, Sikdar PK, Ball JD, Molla TN. Tracing recharge to aquifers beneath an Asian megacity
517	with Cl/Br and stable isotopes: the example of Dhaka, Bangladesh. Hydrogeol. J. 2014; 22: 1549-1560.
518	http://dx.doi.org/10.1007/s10040-014-1155-8
519	Husain V, Nizam H, Arain GM. Arsenic and Fluoride Mobilization Mechanism in Groundwater of Indus Delta and
520	Thar Desert, Sindh, Pakistan. Int. J. Econ. Environ. Geol. 2012; 3: 15-23
521	Katz BG, Eberts SM, Kauffman LJ. Using Cl/Br ratios and other indicators to assess potential impacts on
522	groundwater quality from septic systems: A review and examples from principal aquifers in the United
523	States. J. Hydrol. 2011; 397: 151-166. http://dx.doi.org/10.1016/j.jhydrol.2010.11.017

524	Kendall C. Tracing Nitrogen Sources and Cycling in Catchments. In: Kendall C, McDonnell JJ, editors. Isotope
525	Tracers in Catchment Hydrology. Amsterdam: Elsevier Science B.V., 1998, pp. 519–576.
526	Korte N. Naturally occurring arsenic in groundwaters of the midwestern United States. Environ. Geol. Water Sci.
527	1991; 18: 137-141. http://dx.doi.org/10.1007/bf01704667
528	Korte NE, Fernando Q. A review of arsenic (III) in groundwater. Crit. Rev. Env. Contr. 1991; 21: 1-39.
529	http://dx.doi.org/10.1080/10643389109388408
530	Ladouche B, Luc A, Nathalie D. Chemical and isotopic investigation of rainwater in Southern France (1996–2002):
531	Potential use as input signal for karst functioning investigation. J. Hydrol. 2009; 367: 150-164.
532	http://dx.doi.org/10.1016/j.jhydrol.2009.01.012
533	Longinelli A, Selmo E. Isotopic composition of precipitation in Italy: a first overall map. J. Hydrol. 2003; 270: 75-
534	88. http://dx.doi.org/10.1016/S0022-1694(02)00281-0
535	Lovley DR. Microbial Fe(III) reduction in subsurface environments. FEMS Microbiol. Rev. 1997; 20: 305-313.
536	http://dx.doi.org/10.1111/j.1574-6976.1997.tb00316.x
537	Marchetti M. Environmental changes in the central Po Plain (northern Italy) due to fluvial modifications and
538	anthropogenic activities. Geomorphology 2002; 44: 361-373. http://dx.doi.org/10.1016/S0169-
539	555X(01)00183-0
540	Marchina C, Bianchini G, Knoeller K, Natali C, Pennisi M, Colombani N. Natural and anthropogenic variations in
541	the Po river waters (northern Italy): insights from a multi-isotope approach. Isotopes Environ. Health Stud.
542	2016: 1-24. http://dx.doi.org/10.1080/10256016.2016.1152965
543	Marchina C, Bianchini G, Natali C, Pennisi M, Colombani N, Tassinari R, et al. The Po river water from the Alps to
544	the Adriatic Sea (Italy): new insights from geochemical and isotopic (δ18Ο-δD) data. Environ. Sci. Pollut.
545	Res. 2015; 22: 5184-5203. http://dx.doi.org/10.1007/s11356-014-3750-6
546	Martinelli G, Chahoud A, Dadomo A, Fava A. Isotopic features of Emilia-Romagna region (North Italy)
547	groundwaters: Environmental and climatological implications. J. Hydrol. 2014; 519, Part B: 1928-1938.
548	http://dx.doi.org/10.1016/j.jhydrol.2014.09.077
549	Matisoff G, Khourey CJ, Hall JF, Varnes AW, Strain WH. The Nature and Source of Arsenic in Northeastern Ohio
550	Ground Watera. Ground Water 1982; 20: 446-456. http://dx.doi.org/10.1111/j.1745-6584.1982.tb02765.x

551	McArthur JM, Ghosal U, Sikdar PK, Ball JD. Arsenic in Groundwater: The Deep Late Pleistocene Aquifers of the
552	Western Bengal Basin. Environ. Sci. Technol. 2016; 50: 3469-3476.
553	http://dx.doi.org/10.1021/acs.est.5b02477
554	McArthur JM, Sikdar PK, Hoque MA, Ghosal U. Waste-water impacts on groundwater: Cl/Br ratios and
555	implications for arsenic pollution of groundwater in the Bengal Basin and Red River Basin, Vietnam. Sci.
556	Total Environ. 2012; 437: 390-402. http://dx.doi.org/10.1016/j.scitotenv.2012.07.068
557	Menció A, Mas-Pla J, Otero N, Regàs O, Boy-Roura M, Puig R, et al. Nitrate pollution of groundwater; all right,
558	but nothing else? Sci. Total Environ. 2016; 539: 241-251. http://dx.doi.org/10.1016/j.scitotenv.2015.08.151
559	Molinari A, Guadagnini L, Marcaccio M, Guadagnini A. Natural background levels and threshold values of
560	chemical species in three large-scale groundwater bodies in Northern Italy. Sci. Total Environ. 2012; 425:
561	9-19. http://dx.doi.org/10.1016/j.scitotenv.2012.03.015
562	Nealson KH. SEDIMENT BACTERIA: Who's There, What Are They Doing, and What's New? Annu. Rev. Earth
563	Pl. Sc. 1997; 25: 403-434. http://dx.doi.org/10.1146/annurev.earth.25.1.403
564	Nickson R, McArthur J, Burgess W, Ahmed KM, Ravenscroft P, Rahmann M. Arsenic poisoning of Bangladesh
565	groundwater. Nature 1998; 395: 338-338. http://dx.doi.org/10.1038/26387
566	Nickson RT, McArthur JM, Ravenscroft P, Burgess WG, Ahmed KM. Mechanism of arsenic release to
567	groundwater, Bangladesh and West Bengal. Appl. Geochem. 2000; 15: 403-413.
568	http://dx.doi.org/10.1016/S0883-2927(99)00086-4
569	Nickson RT, McArthur JM, Shrestha B, Kyaw-Myint TO, Lowry D. Arsenic and other drinking water quality issues,
570	Muzaffargarh District, Pakistan. Appl. Geochem. 2005; 20: 55-68.
571	http://dx.doi.org/10.1016/j.apgeochem.2004.06.004
572	Nissenbaum A, Magaritz M. Bromine-rich groundwater in the Hula Valley, Israel. Naturwissenschaften 1991; 78:
573	217-218. http://dx.doi.org/10.1007/bf01136083
574	Onorati G, Di Meo T, Bussettini M, Fabiani C, Farrace MG, Fava A, et al. Groundwater quality monitoring in Italy
575	for the implementation of the EU water framework directive. Phys. Chem. Earth, Pt. A/B/C 2006; 31:
576	1004-1014. http://dx.doi.org/10.1016/j.pce.2006.07.001
577	Ori GG. Continental depositional systems of the Quaternary of the Po Plain (northern Italy). Sediment. Geol. 1993;
578	83: 1-14. http://dx.doi.org/10.1016/S0037-0738(10)80001-6

570	
579	Panno SV, Hackley KC, Hwang HH, Greenberg SE, Krapac IG, Landsberger S, et al. Characterization and
580	Identification of Na-Cl Sources in Ground Water. Ground Water 2006; 44: 176-187.
581	http://dx.doi.org/10.1111/j.1745-6584.2005.00127.x
582	Perego R, Bonomi T, Fumagalli L, Benastini V, Aghib F, Rotiroti M, et al. 3D reconstruction of the multi-layer
583	aquifer in a Po Plain area. Rend. Online Soc. Geol. It. 2014; 30: 41-44.
584	http://dx.doi.org/10.3301/ROL.2014.09
585	Pilla G. Caratterizzazione idrochimica e geochimica isotopica delle falde idriche nel sottosuolo della città di Pavia
586	"Chemical and isotopic characterisation of groundwater from the city of Pavia". Atti Tic. Sc. Terra 1998;
587	40: 185-201.
588	Pilla G, Sacchi E, Zuppi G, Braga G, Ciancetti G. Hydrochemistry and isotope geochemistry as tools for
589	groundwater hydrodynamic investigation in multilayer aquifers: a case study from Lomellina, Po plain,
590	South-Western Lombardy, Italy. Hydrogeol. J. 2006; 14: 795-808. http://dx.doi.org/10.1007/s10040-005-
591	0465-2
592	Pilla G, Torrese P, Bersan M. The Uprising of Deep Saline Paleo-Waters into the Oltrepò Pavese Aquifer (Northern
593	Italy): Application of Hydro-Chemical and Shallow Geophysical Surveys. In: Lollino G, Arattano M,
594	Rinaldi M, Giustolisi O, Marechal J-C, Grant EG, editors. Engineering Geology for Society and Territory -
595	Volume 3: River Basins, Reservoir Sedimentation and Water Resources. Cham: Springer International
596	Publishing, 2015, pp. 393-397.
597	Postma D, Larsen F, Minh Hue NT, Duc MT, Viet PH, Nhan PQ, et al. Arsenic in groundwater of the Red River
598	floodplain, Vietnam: Controlling geochemical processes and reactive transport modeling. Geochim.
599	Cosmochim. Acta 2007; 71: 5054-5071. http://dx.doi.org/10.1016/j.gca.2007.08.020
600	Ravenscroft P, Brammer H, Richards K. Arsenic Pollution: A Global Synthesis. Chichester: Wiley-Blackwell, 2009
601	Ravenscroft P, McArthur JM. Mechanism of regional enrichment of groundwater by boron: the examples of
602	Bangladesh and Michigan, USA. Appl. Geochem. 2004; 19: 1413-1430.
603	http://dx.doi.org/10.1016/j.apgeochem.2003.10.014
604	Ravenscroft P, McArthur JM, Hoque BA. Geochemical and palaeohydrological controls on pollution of
605	groundwater by arsenic. In: Chappell WR, Abernathy CO, Calderon R, editors. Arsenic Exposure and
606	Health Effects IV. Oxford: Elsevier Science, 2001, pp. 53-78.

607	Re V, Sacchi E, Martin-Bordes JL, Aureli A, El Hamouti N, Bouchnan R, et al. Processes affecting groundwater
608	quality in arid zones: The case of the Bou-Areg coastal aquifer (North Morocco). Appl. Geochem. 2013;
609	34: 181-198. http://dx.doi.org/10.1016/j.apgeochem.2013.03.011
610	Rotiroti M, Di Mauro B, Fumagalli L, Bonomi T. COMPSEC, a new tool to derive natural background levels by the
611	component separation approach: application in two different hydrogeological contexts in northern Italy. J.
612	Geochem. Explor. 2015a; 158: 44-54. http://dx.doi.org/10.1016/j.gexplo.2015.06.017
613	Rotiroti M, Fumagalli M, Bonomi T. How to manage potential groundwater contaminations by As, Fe and Mn in
614	lower Po Plain: a proposal from the case study of Cremona. Acq. Sott. Ital. J. Groundw. 2014a; 3: 9-16.
615	http://dx.doi.org/10.7343/as-070-14-0096
616	Rotiroti M, Jakobsen R, Fumagalli L, Bonomi T. Arsenic release and attenuation in a multilayer aquifer in the Po
617	Plain (northern Italy): Reactive transport modeling. Appl. Geochem. 2015b; 63: 599-609.
618	http://dx.doi.org/10.1016/j.apgeochem.2015.07.001
619	Rotiroti M, Sacchi E, Fumagalli L, Bonomi T. Origin of arsenic in groundwater from the multi-layer aquifer in
620	Cremona (northern Italy). Environ. Sci. Technol. 2014b; 48: 5395–5403.
621	http://dx.doi.org/10.1021/es405805v
622	Rowland HAL, Pederick RL, Polya DA, Pancost RD, Van Dongen BE, Gault AG, et al. The control of organic
623	matter on microbially mediated iron reduction and arsenic release in shallow alluvial aquifers, Cambodia.
624	Geobiology 2007; 5: 281-292. http://dx.doi.org/10.1111/j.1472-4669.2007.00100.x
625	Sacchi E, Acutis M, Bartoli M, Brenna S, Delconte CA, Laini A, et al. Origin and fate of nitrates in groundwater
626	from the central Po plain: Insights from isotopic investigations. Appl. Geochem. 2013; 34: 164-180.
627	http://dx.doi.org/10.1016/j.apgeochem.2013.03.008
628	Skinner HCW, Berger AR. Geology and Health: Closing the Gap. New York: Oxford University Press, 2003.
629	Sorlini S, Gialdini F. Study on arsenic removal in the drinking water treatment plant of Cremona (Italy). J. Water
630	Supply Res. T. 2014; 63: 625-629. http://dx.doi.org/10.2166/aqua.2014.195
631	Vassena C, Rienzner M, Ponzini G, Giudici M, Gandolfi C, Durante C, et al. Modeling water resources of a highly
632	irrigated alluvial plain (Italy): calibrating soil and groundwater models. Hydrogeol. J. 2012; 20: 449-467.
633	http://dx.doi.org/10.1007/s10040-011-0822-2

634	Vengosh A, Pankratov I. Chloride/Bromide and Chloride/Fluoride Ratios of Domestic Sewage Effluents and
635	Associated Contaminated Ground Water. Ground Water 1998; 36: 815-824.
636	http://dx.doi.org/10.1111/j.1745-6584.1998.tb02200.x
637	Vitòria L, Grandia F, Soler A. Evolution of the chemical (NH4) and isotopic (δ15N–NH4) composition of pig
638	manure stored in an experimental pit. In: International Atomic Energy Agency, editor. Isotope Hydrology
639	and Integrated Water Resources Management. Conf. Symp. Papers, Vienna, 2004a, pp. 260-261.
640	Vitòria L, Otero N, Soler A, Canals À. Fertilizer Characterization: Isotopic Data (N, S, O, C, and Sr). Environ. Sci.
641	Technol. 2004b; 38: 3254-3262. http://dx.doi.org/10.1021/es0348187
642	Winid B. Bromine and water quality – Selected aspects and future perspectives. Appl. Geochem. 2015; 63: 413-435.
643	http://dx.doi.org/10.1016/j.apgeochem.2015.10.004
644	Zavatti A, Attramini D, Bonazzi A, Boraldi V, Malagò R, Martinelli G, et al. La presenza di Arsenico nelle acque
645	sotterranee della Pianura Padana: evidenze ambientali e ipotesi geochimiche "Occurrence of groundwater
646	arsenic in the Po Plain: environmental evidences and geochemical hypothesis". Quad. Geol. Appl. 1995;
647	S2: 2.301-2.326.
648	Zucaro R. Atlante nazionale dell'irrigazione "National atlas of irrigation". Roma: INEA, 2011.
649	Zucaro R, Corapi A. Rapporto sullo Stato dell'Irrigazione in Lombardia "Report on the Status of Irrigation in
650	Lombardy". Roma: INEA, 2009.
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Figure 1. Study area with rivers, channels and sampling points labelled with ID number; wells are classed for each

tapped aquifer; total 14 groundwater and 1 surface water samples were collected in July 2012, 9 surface water and 1

rain samples were collected in July 2014 and 28 groundwater, 9 surface water and 1 sewage samples were collected in

October 2014; Uox: zone with oxidising conditions in aquifer U; Ured: zone with reducing conditions in aquifer U;

Umix: zone with mixing conditions in aquifer U; S: aquifer S; C1: aquifer C1; C2: aquifer C2; C3 aquifer C3.

Figure 2. Groundwater and surface water concentrations of conservative and redox-sensitive species measured in 2014

over depth; symbol length corresponds to well screen interval; Uox: oxidized zone of aquifer U; Ured: reduced zone of

aquifer U; S: aquifer S; C1: aquifer C1; C2: aquifer C2; C3: aquifer C3.

Figure 3. (a) Plot of measured and reference  $\delta^2 H$  vs  $\delta^{18} O$  in groundwater, Po River water and precipitation; LMWL:

local meteoric water line by <sup>a</sup>Longinelli and Selmo (2003) and <sup>b</sup>Framcani et al. (1994); also plotted are weighted-means

of precipitation by <sup>c</sup>Longinelli and Selmo (2003) in 4 stations at Mantova, Milano, Parma and Piacenza, all near

Cremona and values of the Po River in Cremona by  $^{\rm d}$ Marchina et al. (2015, 2016). (b) Measured  $\delta^{18}$ O over depth;

symbol length corresponds to well screen interval; sampling points labelled with ID number are cited in the text; dashed

boxes represent reference range values (see Section 3.2 for range values). U: aquifer U; S: aquifer S; C1: aquifer C1;

C2: aquifer C2; C3: aquifer C3.

Figure 4. Plot of Cl/Br vs Cl for our data and legacy data, together with mixing lines between end-members given in

Section 2.4; sampling points labelled with ID number are cited in the text; percentages indicated on mixing lines

represent the fraction of the high-Cl end-member over the low-Cl end-member; Uox: oxidized zone of aquifer U; Ured:

reduced zone of aquifer U; S: aquifer S; C1: aquifer C1; C2: aquifer C2; C3: aquifer C3.

Figure 5. Average concentration of As and Cl in C3 over the sampling dates available (May 2006, October 2006,

November 2007, April 2010 and February 2011 from unpublished historical data provided by the water supply company

of Cremona and October 2014 from this study).

# Figure size

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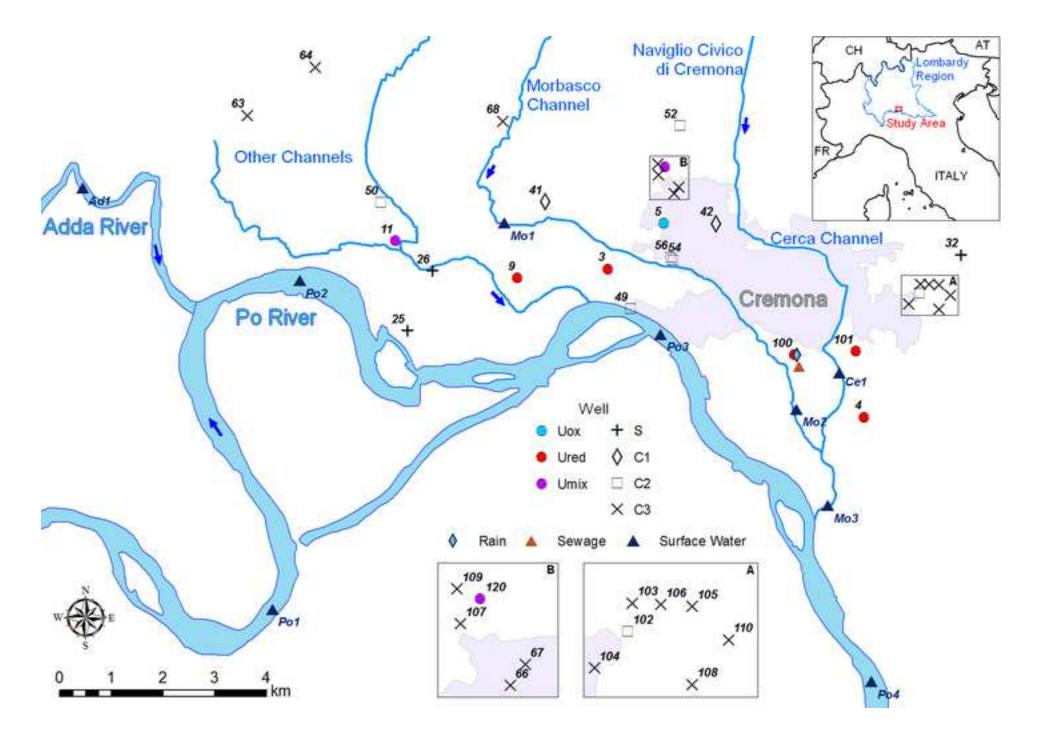
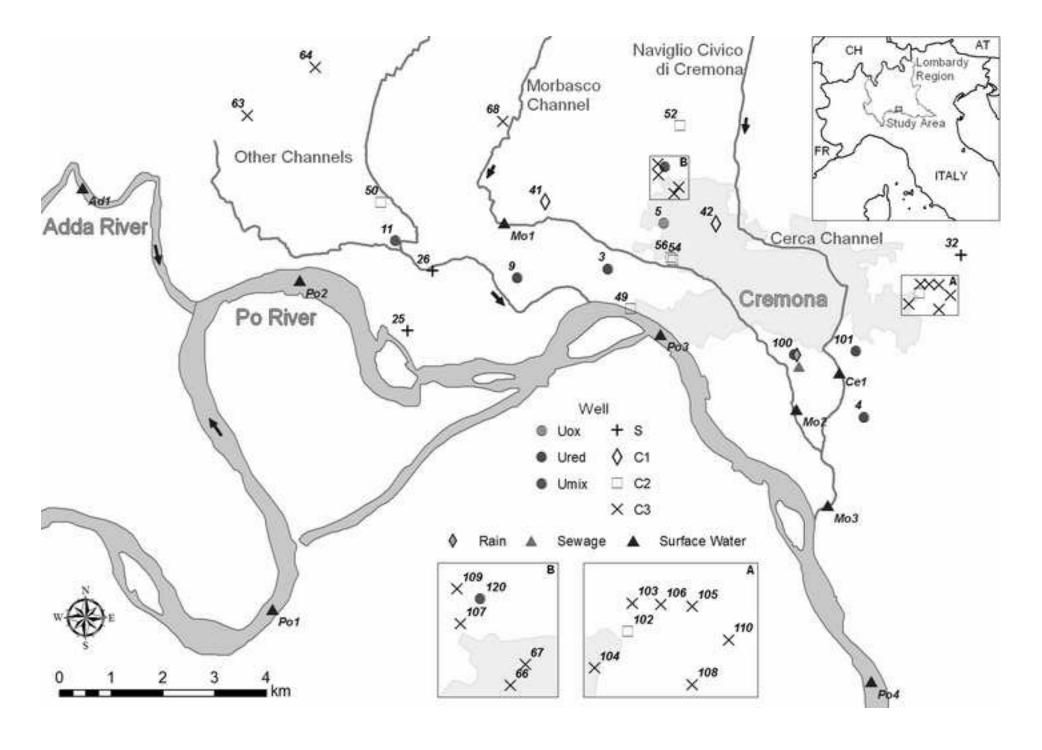
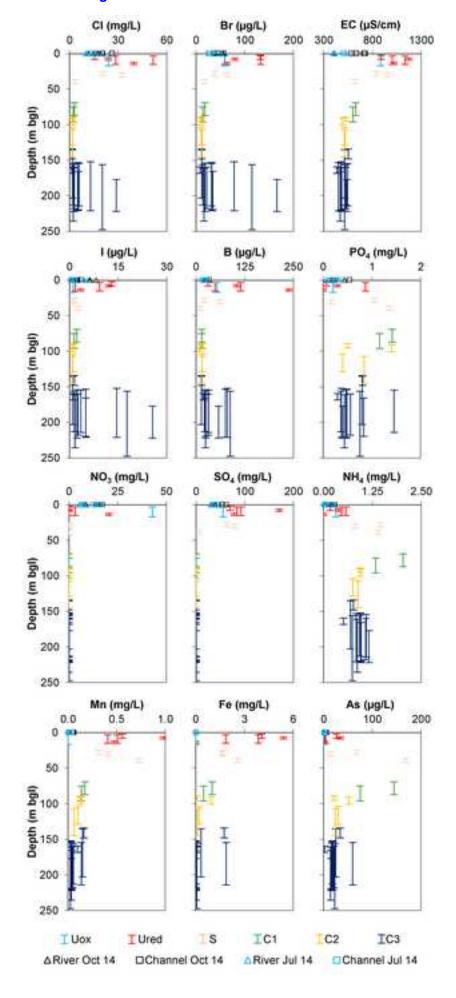
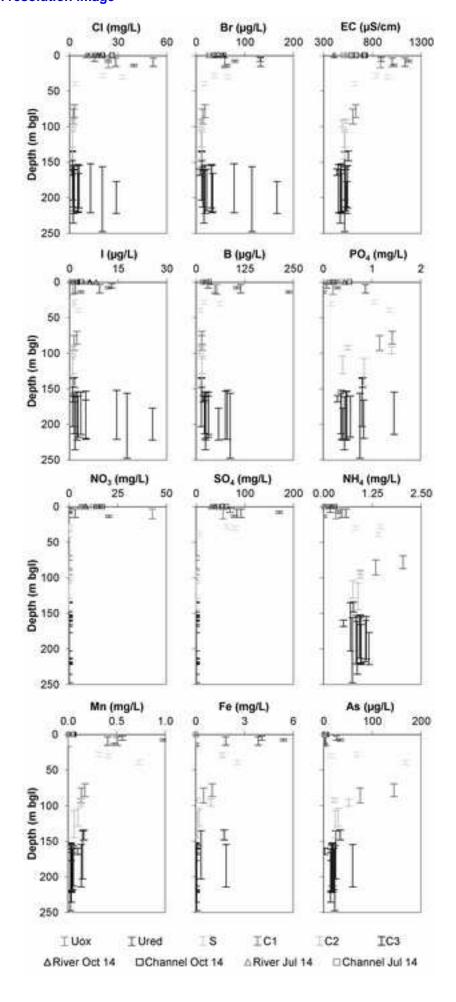


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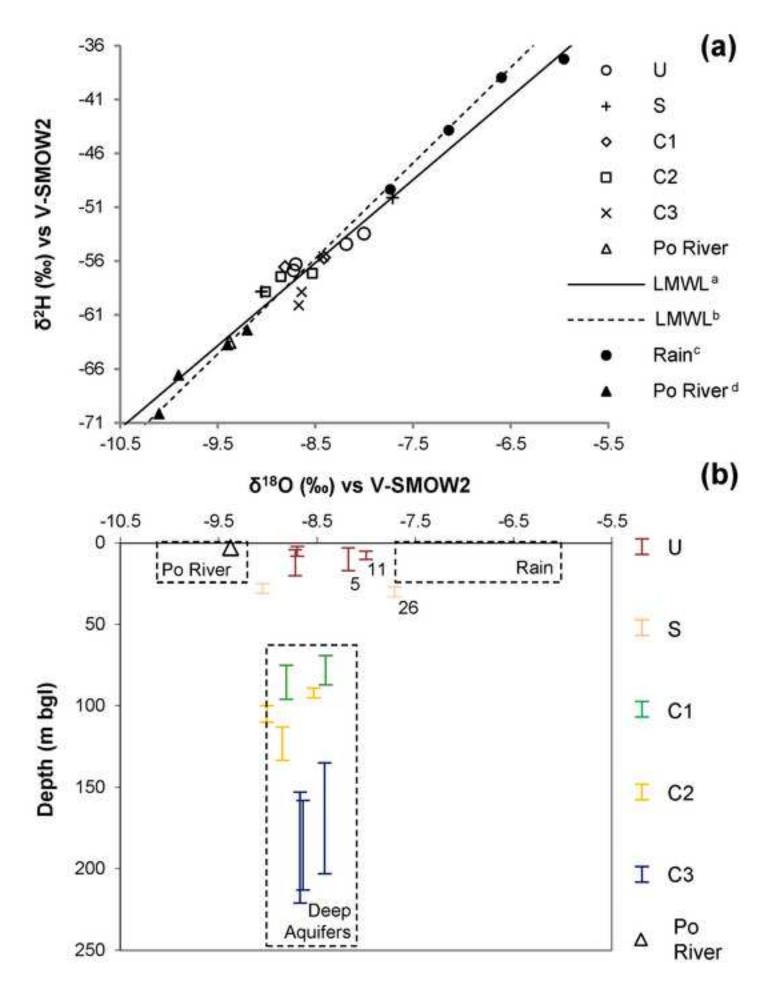


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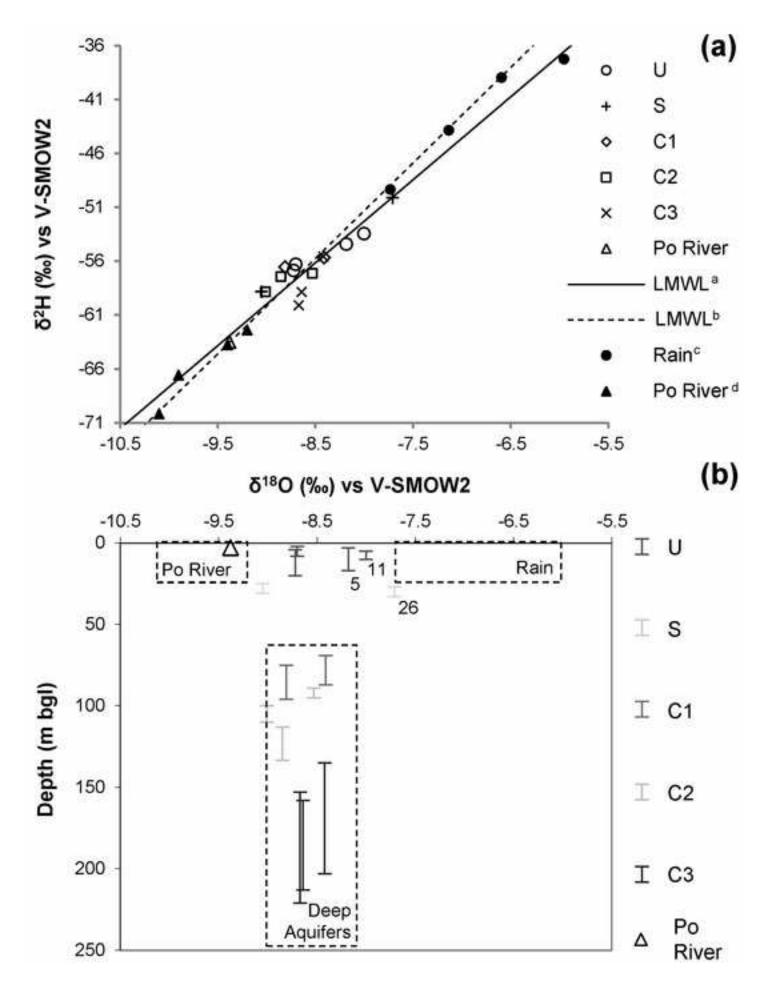


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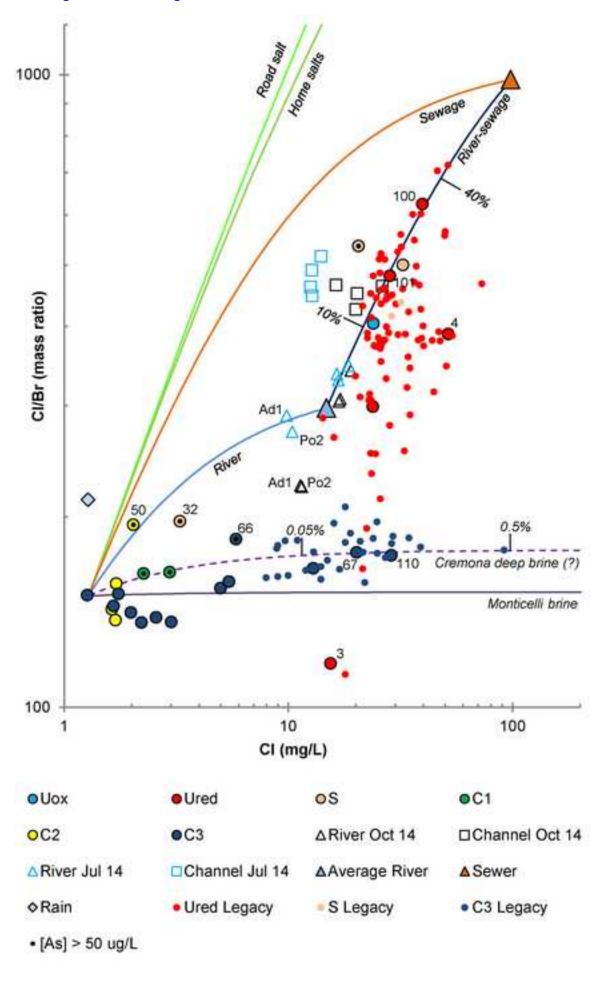


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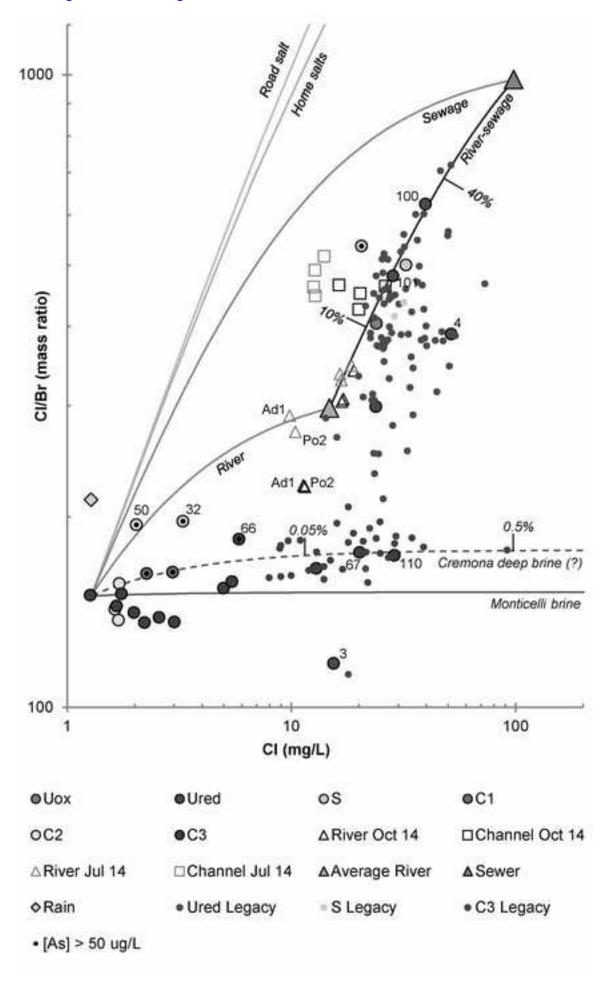


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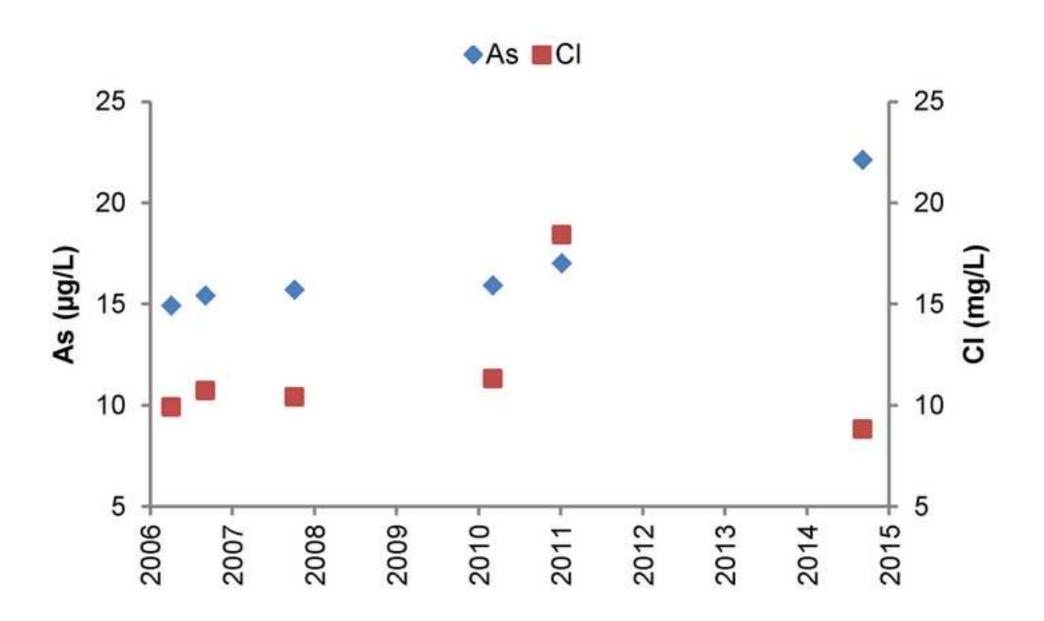
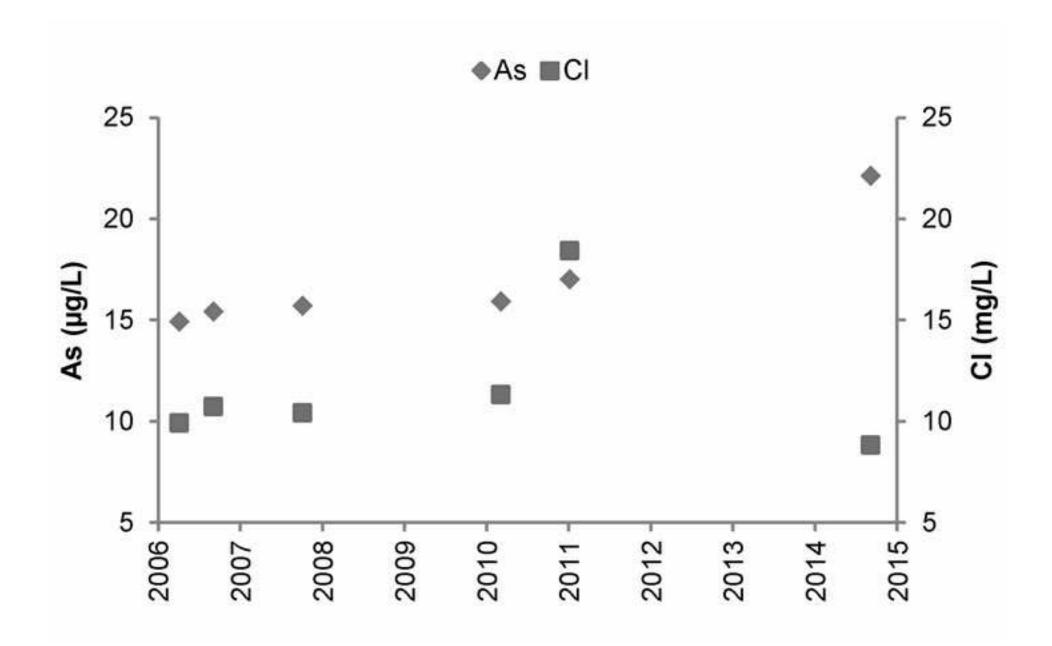


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