1 CALCIUM SILICATE PEROVSKITE'S SEISMIC VELOCITIES CAN EXPLAIN LLVPS IN EARTH'S

2 LOWER MANTLE

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11 Seismology records the presence of various heterogeneities throughout the lower mantle^{e.g.1,2}, however, the origins 12 of these signals, whether thermal or chemical, remain uncertain and therefore much of the information they hold 13 about the nature of the deep Earth is obscured. Accurate interpretation of observed velocities requires knowledge of the seismic properties of all of Earth's possible mineral components. Calcium silicate perovskite (hereafter "calcium 14 15 perovskite") is believed to be the third most abundant mineral throughout the lower mantle. Here we measure the 16 crystal structure, compressional and shear wave velocity of calcium perovskite samples, and provide direct 17 constraints for calcium perovskite's adiabatic bulk and shear moduli. We observe that titanium incorporation into 18 calcium perovskite stabilises the tetragonal structure to higher temperatures, and that the shear modulus of calcium 19 perovskite is significantly lower than is predicted by computations³⁻⁵ or thermodynamic datasets⁶. When combined 20 with literature data and extrapolated our results suggest subducted oceanic crust will be visible as low velocity 21 anomalies throughout the lower mantle. In particular we show that large low-velocity provinces (LLVPs) are 22 consistent with moderate enrichment of recycled oceanic crust, and mid-mantle discontinuities can be explained by a 23 tetragonal-cubic phase transition in Ti-bearing calcium perovskite.

The lower mantle is vast, extending from the seismic discontinuity observed at ~ 660 km depth to the core -mantle 24 boundary (CMB) at ~ 2890 km. Tomographic images demonstrate that despite a smooth variation of v_{ρ} , v_s and ρ in 25 26 1D velocity models the lower mantle is heterogeneous and regularly refertilised by subducting slabs^{7,8}. Sluggish 27 diffusive re-equilibration and incomplete mechanical mixing^{e.g. 9} means that large-scale patterns of mantle 28 convection may be directly observed via tomographic velocity anomalies and/or the distribution of seismic 29 scatterers. Identifying the causes of heterogeneities requires accurate mineralogical models of Earth's mantle to 30 facilitate comparisons between geophysical observations and predicted seismic velocities. However, a major uncertainty in many models^{e.g. 10,11} has been the influence of calcium silicate perovskite (capv, here corresponding to 31 32 $Ca[Si_Ti_{1-x_1}]O_3)$ on velocity, despite the widespread expectation that it is the lower mantle's third most abundant phase comprising 5-10 and 24-29 vol.% of peridotitic¹² and basaltic¹³ assemblages respectively. 33

Uncertainties stem from a sparsity of reliable measurements of capv's physical properties, which are technically
 challenging because CaSiO₃ is unrecoverable¹⁴, undergoing spontaneous amorphisation at room temperature during
 decompression. The widely used thermodynamic model of Stixrude et al.⁶ predicts that capv is significantly faster
 than PREM¹⁵, and therefore slow velocity anomalies are difficult to explain using recycled crustal material. Whilst

38 this is the widely adopted view, there is currently no consensus on the seismic properties of capy. Existing high-T39 calculations³⁻⁵ suggest that capv might be either slightly slower or much faster than PREM¹⁵ (Figure 1). In contrast, room temperature experiments^{16,17} have measured a shear velocity of capy that is at least 7% slower than the lowest 40 computational estimates; however there is difficulty extrapolating these to high-temperature conditions due to intervening 41 phase transformations of the capy structure^{3,18,19}. Very recently, high-temperature experimental velocity measurements 42 43 that are also slower than all computational values have been reported²⁰, although this study did not consider 44 extrapolations of capy's velocity throughout the deep mantle, leaving capy's contribution in generating lower mantle 45 signatures unresolved. Here we report synchrotron-based high PT experiments that simultaneously measure the 46 crystal structure and seismic velocities (v_p and v_s) of Ca[Si_xTi_[1-x]]O₃ compositions (x= 0.6 and 1) that bracket the 47 range of inclusions found in natural superdeep diamonds and are expected in lower mantle assemblages²¹. 48 Combining our new data with ab-initio calculations and literature data we directly address the influence of 49 crystallographic phase transitions on the velocity of capv and apply our results to provide a new understanding of

50 capv's geophysical signature throughout the lower mantle (Methods).

51 In-line with expectations from previous experiments^{e.g. 18}, in-situ diffraction confirms that capv is cubic at high temperature, and undergoes one (or more) structural distortions upon cooling to room temperature (Figure 2a). 52 53 Refinement and indexing of diffraction patterns reveals that endmember CaSiO₃ transforms on cooling from cubic 54 $(Pm\bar{3}m)$ at high temperatures into tetragonal (I4/mcm) perovskite between 380 – 420 K at ~ 12 GPa (Figure 2c). This 55 phase transition is identified by the non-linear splitting upon cooling (observed as broadening) of all diffraction peaks, except for those with indices hhh (i.e. 111 and 222), from the cubic aristotype unit cell ($a \sim 3.5$ Å, Figure 2b 56 57 and Extended Data Figure 1e). Additionally, weak superlattice reflections at d-spacings of ~ 2.11, 1.61, 1.07 and 0.98 58 Å (Extended Data Figure 2), which uniquely identify the *I*4/*mcm* structure, were observed below ~ 420 K. Titanium 59 incorporation (similarly to aluminium²³) increases the upper stability limit of tetragonal capv considerably, here by 60 nearly 800 K. We find that Ca[Si_{0.6}Ti_{0.4}]O₃ takes the $Fm\overline{3}m$ space group at high temperature (Extended Data Figure 61 1,3,4), possessing a double perovskite unit cell, with partial Si:Ti cation ordering that is apparently maintained throughout cooling. The cubic-tetragonal ($Fm\overline{3}m - I4/m$) transition in Ca[Si_{0.6}Ti_{0.4}]O₃ is observed at ~ 1200 K 62 63 (Extended Data Figure 1). Upon further cooling, a subsequent symmetry distortion, thought to be to $P2_1/c$, is 64 observed at ~ 700 K. These observations provide very strong evidence that capv follows the same structural 65 transitions on cooling as CaTiO₃, with the apparent reductions in symmetry from 14/mcm and Pbnm to their 14/m66 and $P2_1/c$ subgroups being a consequence of cation ordering:

$$Pm\overline{3}m \xrightarrow{\sim 420K} I4/mcm - CaSiO_3 (~ 12 GPa)$$

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$$Fm\overline{3}m \xrightarrow{\sim 1200K} I4/m \xrightarrow{\sim 700K} P2_1/c - Ca[Si_{0.6}Ti_{0.4}]O_3 (\sim 12 GPa)$$

$$Pm\overline{3}m \xrightarrow{\sim 1635K} I4/mcm \xrightarrow{\sim 1512K} Pbnm$$
 - $CaTiO_3 (0 GPa)^{-24}$

Acoustic velocities, determined simultaneously with synchrotron X-ray diffraction using pulse-echo ultrasonic interferometry, demonstrate that the observed phase transitions of capv are associated with significant elastic anomalies. CaSiO₃ and Ca(Si_{0.6}Ti_{0.4})O₃ samples undergo v_p and v_s reductions of 4-14 % and 8-20 % respectively due to 73 their cubic-tetragonal transitions (Figure 3). Continued cooling of Ca(Si_{0.6}Ti_{0.4})O₃ into its presumed monoclinic 74 structure sees the velocities increase near ambient temperature. The acoustic "shear-strengthening" with 75 temperature that we observe in tetragonal capv is also reported for polycrystalline BaTiO₃ samples²⁵. Such 76 behaviour is thought to result from a temperature-activated twin-domain wall process that also causes high 77 acoustic attenuation²⁶. Whilst we cannot rigorously measure acoustic attenuation, we do observe a diminution in 78 the intensity of reflected acoustic waves when samples are tetragonal. Experiments on endmember CaSiO₃ 79 demonstrate a modest reduction in v_p and v_s across the cubic – tetragonal transition at T < 450 K, however the 80 relatively small decrease observed is likely to continue at sub-ambient temperatures that could not be examined in 81 this study. We suggest, if cooled further, that CaSiO₃ likely undergoes similar magnitudes of velocity reduction to 82 those observed for Ca(Si_{0.6}Ti_{0.4})O₃. Absolute acoustic velocities measured for CaSiO₃ are lower than computational 83 predictions, but v_{ρ} and v_{s} in this study agree extremely well with previous experimental measurements made at room temperature^{16,17,20}. It is only with increasing temperature that our results diverge from previous experimental 84 data²⁰. The excellent room temperature agreement leads us to conclude that previous calculations must have 85 86 overestimated the velocities, specifically the shear modulus, of capv (as discussed below). We also observe the temperature dependences of velocities $\left(\frac{1dv}{vdT}\right)$ in cubic capv are 1.5 – 3 times larger than experimentally observed for 87 88 other mantle silicates^{e.g. 27}. However, the temperature dependence of the elastic moduli in our experiments (dK_s/dT and $dG/dT \sim -0.027 - -0.03$ GPa K⁻¹) match those observed for cubic [Ca,Sr]TiO₃ perovskites (-0.024 - -0.03 GPa K⁻¹) ²⁶ 89 90 and the bulk, but not the shear, modulus in previous experiments on CaSiO₃ ($dK_s/dT \sim -0.036$ and $dG/dT \sim -0.015$ 91 GPa K⁻¹)²⁰.

92 Using *ab initio* molecular dynamics, we have calculated the *PT* slope of the $14/mcm \rightarrow Pm\overline{3}m$ phase transition in CaSiO₃ perovskite (Methods, Extended Data Figure 5) in order to apply our results to Earth's deep mantle. The 93 calculated slope, ~ 15 K GPa⁻¹, is similar to results from previous calculations (~ 10 K GPa⁻¹)³ and experiments of the 94 $I4/mcm \rightarrow Pm\overline{3}m$ transition in SrTiO₃ (~ 18.5 K GPa⁻¹)²⁸. However, it is much larger than, but still within uncertainty of, 95 experimental estimates (≤ 2 K GPa⁻¹)^{18,23} for CaSiO₃. Assuming our *PT* slope is only shifted in temperature by Ti-96 97 incorporation, average MORB capv (~ $Ca[Si_{0.9}Ti_{0.1}]O_3$, ignoring other chemical components)¹³ should undergo a cubic \rightarrow tetragonal transformation at mid-mantle depths. In addition, capv subducted within slab assemblages, 98 99 particularly for Ti-rich capy compositions, may in fact retain the tetragonal structure throughout the entire mantle 100 at average temperatures (Extended Data Figure 5). Pure CaSiO₃, which is similar to the composition stable in 101 peridotitic and harzburgitic assemblages, is unlikely to become tetragonal in the ambient mantle, but could undergo 102 a cubic-tetragonal transition in cold slab assemblages reaching pressures greater than ~ 90 GPa without significant 103 super-adiabatic heating.

To evaluate our experimental results in the context of Earth's lower mantle we have fitted finite -strain equations of state (EoS) for cubic and tetragonal CaSiO₃ perovskite, using the thermodynamically self-consistent Mie-Debye-Grüneisen Birch-Murnaghan formalism²⁹ commonly adopted in mineralogical models⁶. The narrow pressure range of our experiments means additional constraints from literature data are required for extrapolations where velocities remain experimentally unconstrained. All available literature data judged to reliably constrain capv volume and/or acoustic velocities at high *PT* conditions (Supplementary Information Table 1) were collated and converted to a common pressure scale for joint inversion with new data from this study (see Methods for full fitting procedure). 111 We note that data from recent high-PT experiments²⁰ are not included, due to inconsistencies within this dataset (Methods). Recovered EoS parameters for tetragonal capv at 300 K are $V_0 = 46.10(6)$ Å³ per formula unit, $K_0 = 224(4)$ 112 GPa, $K_0' = 4$ (fixed), $G_0 = 107(6)$ GPa and $G_0' = 1.4(1)$ (Extended Data Figure 6a). The EoS for cubic capv, fitted using 113 high-temperature data lying above the phase transition calculated by *ab-initio* methods in this study (Figure 1, 114 115 Extended Data Figure 6b, Supplementary Information Table 2) has the parameters: $V_0 = 45.57(2)$ Å³, $K_0 = 248(3)$ GPa, 116 $K_0' = 3.6(1), G_0 = 107(1)$ GPa, $G_0' = 1.66$ (fixed, but manually varied by ± 0.22), $q_0 = 1.1(2), \gamma_0 = 1.67(4), \theta_0 = 771(90)$ K and $\eta_{s0} = 3.3$ (fixed to ~ 2 γ_0). Incorporation of our newly collected data, compared with only fitting literature data, 117 118 improves estimations of G_0 , γ_0 , q_0 and α , due to the high temperature resolution provided by this study, whilst the 119 dominant constraint on K_0 and K_0' come from the highest-pressure literature data. The narrow pressure range of our 120 velocity measurements mean they mainly constrain the shear modulus, but not its pressure derivative G_0' . However, literature values of G_0' from calculations and experiments^{4,5,20} are highly consistent and it can be fixed with some 121 122 confidence, although we have also varied it manually to assess its effect on extrapolations. Our approach relies on 123 high-precision data from four previous diffraction studies on capv at high PT conditions and results in a single 124 equation of state that explains all data without any significant outliers. This provides the best option to date to 125 investigate capy's velocity at deep mantle conditions and offers self-consistent EoS values without apparent 126 reduction in predictive capacity throughout the pressures and temperatures relevant to Earth's mantle.

127 Our results imply that cubic capy's compressional and shear velocity profiles are significantly slower than PREM¹⁵ (see Figure 1), whilst its bulk sound velocity is virtually indistinguishable from PREM¹⁵ (Extended Data Figure 7a). We 128 129 observe capy's velocities, especially v_s , to be much slower than those predicted from thermodynamic datasets⁶, by 130 previous high-temperature calculations³⁻⁵, or high *PT* experiments²⁰. In all cases the discrepancy is almost entirely due 131 to the much lower shear modulus observed in this study. Indeed, it is evident that the shear modulus of capv is one of the 132 most critical parameters for accurately modelling the velocity of basaltic assemblages throughout the lower mantle. A full comparison with results from previous studies is provided in the Methods. Whilst titanium-bearing capy has not been 133 134 included in finite-strain modelling, our experimental data demonstrates that titanium-incorporation will increase capv's velocity by < 1 km s⁻¹ for v_p and < 0.5 km s⁻¹ for v_s for Ca[Si_{0.6}Ti_{0.4}]O₃. Since MORB-derived capv has an 135 136 approximate composition of $Ca[Si_{0.9}Ti_{0.1}]O_3^{13}$, the effect of this titanium content on mantle velocities is expected to 137 be smaller.

138 Understanding the cause of the LLVPs remains one of the most prominent questions currently pursued by the deep 139 Earth research community³⁰. Identifying whether they are purely thermal anomalies or are thermo-chemical piles of 140 recycled or primordial material could have profound consequences for our understanding of mantle convection. In 141 order to address the question of LLVP composition we have incorporated our new cubic capy EoS in thermodynamic 142 models (Methods, Figure 4), extracting the acoustic velocities of recycled basalt and harzburgite assemblages, 143 relative to pyrolite, throughout lower mantle conditions (we are assuming pyrolite is representative of average 144 lower mantle composition). Along a 1500 K mantle adiabat, recycled MORB is predicted to have significantly slower 145 compressional and shear velocities than pyrolite. This result is in stark contrast with published thermodynamic 146 datasets, where predicted MORB assemblages are 1-2% faster than the average deep mantle⁶. Thus, despite 147 geochemical observations implying a long-lived reservoir of recycled crust in the deep Earth³¹ and the high density of MORB³² favouring its accumulation in the LLVPs^{e.g. 7}, this was previously considered incompatible with observed 148

149 slow velocities unless temperatures were extremely hot³³. However, our new results now imply LLVPs are well 150 explained by modest enrichments in recycled oceanic crust, without requiring excess temperature anomalies. 151 Compared to the bulk mantle composition, which is assumed to be a mixture of approximately 80:20 harzburgite:MORB, the v_s anomalies of -1.5 % and $v_s/v_p > 2$ observed in LLVPs³⁰ can be reproduced by a bimodal 152 mixture of MORB + harzburgite consisting 64 % MORB at ~ 100 GPa (v_p = -0.77 %) or 48 % MORB at ~ 125 GPa (v_p = -153 154 0.36 %) if capv is cubic. If the LLVPs are hotter, or if capv is tetragonal near the CMB, the proportion of basalt 155 required to explain the LLVPs would reduce. Our modelling further implies that the above-average velocities that 156 surround the LLVPs, lying beneath palaeo-subduction zones that are often considered to be slab-graveyards³⁰, could 157 potentially represent depleted assemblages. MORB-enriched LLVPs, surrounded by depleted material, also provide an explanation for the anti- or non-correlation of v_s and v_{ϕ} in D" if post-perovskite is stable^{see 30}. Taken at face value, 158 159 the predicted properties of CaSiO₃ suggest that pyrolite assemblages may be slightly slower than PREM¹⁵. However, 160 whilst we are confident that our work robustly demonstrates that subducted MORB assemblages are slow, the amount of capv in pyrolite is smaller and so further investigations on the effects of titanium and/or aluminium³² in 161 162 capv are required to determine whether or not the average velocity of the lower mantle remains compatible with a

163 pyrolitic bulk composition.

164 The properties of capv also provide an explanation for observed seismic reflectors throughout the mid-mantle². In 165 very cold slabs following a 1000 K adiabat, subducted basalts (if capv is cubic) are predicted to have very similar velocities to pyrolitic assemblages on a 1500 K adiabat and so may be seismically invisible. However, if stranded 166 167 fragments are thermally equilibrated with surrounding depleted materials, impedance contrasts with magnitudes of 168 up to $\sim \pm 2.8$ % will be created making them seismically visible as reflectors or as slow regions. Alternatively, if capy undergoes the cubic-tetragonal phase transition, this may also generate mid-mantle anomalies (Extended Data 169 170 Figure 5). Whilst constraining the depth and compositional dependence of this phase transition requires further studies, it is expected that cold downwelling capv is likely to experience the cubic to tetragonal transformation 171 somewhere beyond 1000 km depth. Stagnant or delaminated materials in the upper-lower mantle boundary region⁸ 172 173 may undergo the tetragonal-cubic transition during thermal equilibration, reducing MORB's shear velocity, which 174 may be the origin of observed mid-mantle reflectors².

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245 AUTHOR CONTRIBUTIONS

- A.R.T. designed, performed and analysed the experiments, gathered data from the literature and wrote the
- 247 manuscript. W.A.C. designed and developed the experimental procedure at ID06 of the ESRF. I.G.W. assisted with
- 248 interpretation and refinement of diffraction data. J.P.B., D.P.D, W.A.C and N.C.S. helped perform experiments over
- two sessions at the ESRF. J.M. performed the computational simulations. S.A.H. assisted with data analysis. All
- authors contributed to the scientific discussion and preparation of the manuscript.

251 AUTHOR INFORMATION

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254 DATA AVAILABILITY STATEMENT

Raw data were collected at the European Synchrotron Radiation Facility in Grenoble and are available from
http://doi.org/10.5285/6db95d87-365f-4018-abec-00e96e8fcf8d. Derived data from this study, which includes
source data from figure 2 & 3 and extended data figures 1 & 5, are provided in the supplementary information
tables.

259 FIGURE CAPTIONS

Figure 1: **Compressional and shear wave velocities of cubic CaSiO**₃ **perovskite from this and previous studies** (a) compressional wave velocity and (b) shear wave velocity of CaSiO₃ perovskite predicted in this, and previous, studies throughout the mantle. Individual experimental measurements are shown with symbols, coloured by temperature (white symbols are data collected at 300 K). All velocity curves are extracted along a 1500 K mantle adiabat. Thick coloured curve (and 95% confidence interval in grey) represents velocity of cubic capv based on finite-strain modelling (this study). Bold dashed line is the PREM velocity profile.

Figure 2: X-ray diffraction patterns demonstrating the cubic-tetragonal phase transition in CaSiO₃ perovskite. (a)
 Rietveld refined X-ray diffraction patterns collected at ~ 12 GPa and 1273 K (red, cubic) or 300 K (blue, tetragonal),

- with cubic CaSiO₃ peaks labelled by *hkl* and tick marks for other cell components, (b) normalised full-width-at-halfmaximum (fwhm) of selected diffraction peaks of CaSiO₃ perovskite as a function of temperature, (c) refined lattice
- 270 parameters of CaSiO₃ perovskite sample as a function of experimental temperature with 2σ uncertainties.
- Figure 3: Acoustic velocities of capv samples at high *PT* conditions. (a) Compressional, v_p , and (b) shear, v_s , velocities,
- with derived (c) shear, G, and (d) bulk, K_s, moduli of capv samples measured as a function of temperature at
- constant press load (~ 12 GPa), for Ca(Si_{0.6}Ti_{0.4})O₃ (blue circles) and CaSiO₃ samples (orange and green circles).
- 274 Uncertainties are all 2σ. Small blue circles are data with a low signal:noise ratio, due to low amplitude of the buffer
- 275 rod reflection, and have larger uncertainties. Experimental velocity measurements from previous studies are plotted
- as triangles (~ 12 GPa) and squares (orange 12±1 GPa and purple at 15±1 GPa). Dashed vertical lines indicate the
- temperatures of observed phase transitions for CaSiO₃ (orange) and Ca(Si_{0.6}Ti_{0.4})O₃ (blue). The blue temperature
- interval represents the extent of the 1st order I4/m to $P2_1/c$ transition in Ca(Si_{0.6}Ti_{0.4})O₃.

279 Figure 4: Modelled velocity profiles of lower mantle phase assemblages incorporating capv based on this study.

Models of (a) MORB, (b) harzburgite phase assemblages relative to pyrolite throughout the lower mantle. Red curves are calculated velocity profiles along a self-consistent 1500 K adiabat, blue curves are calculated along a 1000 K adiabat. Solid curves are compressional velocities, dashed are shear velocities, and dotted curves are bulk sound velocity profiles. The upper edge of each curve is the velocity when capv is cubic, whereas the lower bound is indicative of the velocity expected if capv forms a tetragonal structure.

285 Methods

286 STARTING MATERIALS

287 Starting materials of CaSiO₃ and Ca[Si_{0.6}Ti_{0.4}]O₃ compositions were initially prepared by grinding appropriate 288 quantities of high purity CaCO₃, SiO₂ and TiO₂ together in an agate mortar, prior to decarbonation and sintering into 289 crystalline wollastonite, in the case of CaSiO₃, or fusing to a glass, for Ca[Si_{0.6}Ti_{0.4}]O₃. All materials were analysed 290 with SEM EDS to check composition, and the CaSiO₃ was confirmed as pure wollastonite using X-ray powder 291 diffraction ($\lambda = 1.7904 \text{ Å}$). Ultrasonic experiments require fully-dense cylindrical samples with parallel polished 292 faces and an aspect ratio of the length/diameter < 2. Thus, the Ca[Si_{0.6}Ti_{0.4}]O_3 glass was hot-pressed into capv at \sim 293 14 GPa and 1400 °C using a 10/5 multi anvil assembly, and the recovered products manipulated into suitable forms 294 for subsequent acoustic experiments. Since CaSiO₃ perovskite is unrecoverable, undergoing a decomposition to 295 amorphous "glass" with significant fracturing at room temperature, it is not possible to prepare a similar sample of 296 the endmember perovskite composition. Instead samples of walstromite were prepared, which is the highest-297 pressure recoverable polymorph of CaSiO₃, by sintering at ~ 7 GPa and 1300 $^{\circ}$ C using a 14/8 multi-anvil assembly.

298 DIFFRACTION AND ULTRASONIC EXPERIMENTAL METHODS

299 X-ray diffraction and pulse-echo ultrasonic experiments were performed on beamline ID06-LVP of the ESRF 300 synchrotron, where modified 10/5 multi anvil assemblies were employed to allow simultaneous measurement of 301 sample diffraction, length and acoustic velocity. Samples were placed adjacent to a polished polycrystalline alumina buffer rod on one side that reached the cube to which the ultrasonic transducer was attached, and an MgO + NaCl 302 303 mixture serving as soft backing and (for the NaCl) as a pressure marker on the other. Thin Au foils were placed at 304 each end of the sample to allow sample length to be observed with radiographic imaging, and also between the 305 cube and buffer rod to assist with acoustic coupling. This acoustic column was encapsulated in a crushable MgO 306 sleeve, and subsequent Re or TiB₂+BN furnace, which was contained within ZrO₂ insulation and a Cr:MgO octahedral 307 pressure medium. A W-Re thermocouple was used, inserted from the opposite end of the cell assembly in an MgO 308 sleeve to monitor temperature adjacent to the sample throughout experiments (Extended Data Figure 8c).

Along the path of the x-ray beam, the normal ceramic materials were replaced by high transparency amorphous SiBCN(O) windows. Monochromatic synchrotron x-rays ($\lambda = 0.22542$ Å or 0.2296Å) were used to collect diffraction patterns from the sample and pressure markers throughout experiments using two different detectors available at ID06. In all experiments the standard Detection Technology X-scan 1D detector³⁴, which has a fixed 10 Hz integration rate was used to record diffraction patterns every 3.2 (continuous, with 32X rebin)-32 seconds (punctual, on increasing and decreasing pressure). Throughout one experiment on CaSiO₃ an additional Pixirad-8 detector was employed to assist with the identification of weak superlattice reflections from the sample. X-ray sample-detector geometry was calibrated using Si and/or LaB₆ NIST standards, and the collected diffraction patterns were suitably reduced and analysed using the Fit2d³⁵ software. Rietveld refinement of selected diffraction patterns was performed using the GSAS software package³⁶ (Supplementary Information Table 3 & 4). Experimental unit-cell volumes were also determined, and used for *PVT* EoS determination, by fitting the position and width of individual diffraction lines from the capv samples and pressure markers. In this case capv volumes are determined using the average volume calculated from the 200, 310, 321 and 222 diffraction peaks of the sample, whilst the volume of NaCl and Au were determined using their 220 and 310 peaks respectively (Supplementary Information Table 5).

- 323 Samples were compressed to target load and initially heated to T > 1000 K to remove stress in the sample that 324 might affect acoustic measurements. In the experiments on CaSiO₃, the starting material of walstromite was 325 converted into capv by annealing at constant load (initial pressure ~ 14 GPa) and 1200-1500 K for a period of 2-4 326 hours, until all signs of walstromite diffraction peaks were lost. Pressure throughout the experiments was determined from the unit-cell volumes of NaCl-B1 and/or Au using cross-calibrated high-temperature equations of 327 328 state³⁷, whilst sample length was measured using the standard imaging system installed on ID06-LVP. Sample 329 lengths determined with x-ray imaging were checked against those measured before and after (in the case of 330 Ca[Si_{0.6}Ti_{0.4}]O₃) experiments with a digital gauge (1 µm accuracy). Uncertainties in sample lengths are estimated 331 from images as \pm 5 pixels, corresponding to \pm 5.4 μ m (< 2% overall sample length), and it is this uncertainty that 332 produces the reported uncertainties in velocities (Supplementary Information Table 6 and Figure 3). Alongside ultrasonic measurements accompanied by diffraction and imaging, the crystallographic evolution of the samples 333 334 was specifically investigated using continuous diffraction collected during constant rate cooling ramps (25 - 50)335 Kmin⁻¹) for CaSiO₃ and Ca[Si_{0.6}Ti_{0.4}]O₃ samples without collecting ultrasonic data.
- Acoustic signals, always collected after sample annealing at high T, were transmitted into and received from the 336 337 sample assembly using a 10° Y-cut dual mode LiNbO₃ piezoelectric transducer that was fixed to the corner of the 338 "acoustic cube" opposite to the sample using Epo-tek 353ND epoxy. A signal generator (Tektronix AFG3101C or 339 Keysight 33622A) was used to create acoustic pulses composed of three consecutive periods of sine waves with 30-340 60 MHz frequency and $\pm 2.5 - 5$ V amplitude, which were passed to the transducer and oscilloscope. The resonant frequency of v_p and v_s from the transducer crystal were ~ 50 and 30 MHz respectively. Received echoes were 341 342 measured using the same oscilloscope (Tektronix DPO5140 or Keysight DSOS104A) to record the delay between 343 arrival times of compressional and shear wave signals at a rate of $2.5 - 5 \times 10^9$ samples/s. In later experiments a directional bridge (Keysight 86205A), preamplifier (LA020-OS) on the return signal and external trigger (trigger rate 344 345 2 kHz) were variously used, as the system was continuously developed throughout this study. Collection times of 346 individual acoustic spectra ranged from 5 - 300 s, with the various systems employed throughout this study. Two-347 way travel times of ultrasonic arrivals were converted into sample velocities using the "pulse-echo overlap method" 348 (for reflections from the near and far ends of the sample), which was implemented by maximising absolute values of 349 signal cross-correlation and sample lengths measured with x-ray imaging. Predicted reflection coefficients for both 350 interfaces (based on $R = (Z_2-Z_1)/(Z_1+Z_2)$, with $Z_i = \rho_i v_i$) are both negative (~ -0.025 and -0.25 for Al₂O₃ – capv and capv - NaCl+MgO respectively) at the PT conditions of the experiments, suggesting that no phase shift is expected in the 351 352 acoustic signals. The lack of observed phase shifts and measured pulse/echo amplitude ratios were found to be 353 consistent with these expectations, which provides assurance that the phase of acoustic arrivals has been correctly

354 identified. Measured velocities are reported in Supplementary Information Table 6. It should be noted that two 355 independent experiments were performed to measure the velocity of CaSiO₃ samples (which agree within 356 uncertainty for v_p) during separate visits to the ESRF, one employing a Re and the other a TiB₂:BN furnace. In the 357 second of these experiments on CaSiO₃, the shear wave signal from the sample was not observable above noise 358 levels and thus the shear-wave velocity was not determined. Additionally, the ultrasonic data reported for 359 Ca[Si_{0.6}Ti_{0.4}]O₃ above 975 K were collected on heating after annealing, whilst data below 975 K was subsequently 360 collected upon cooling the sample. This results in a small pressure difference between the two sets of 361 measurements and explains the discontinuity in Figure 3.

362 STRUCTURE DETERMINATION

363 Refined diffraction data from experiments on CaSiO₃ are consistent with a cubic crystal structure for capv at high 364 temperatures. All observed diffraction peaks have approximately constant widths at half maximum intensity, in line 365 with expectations from the diffractometer geometry, and all observed diffraction peaks could be attributed to 366 diffraction from the sample (in space group $Pm\overline{3}m$) or other cell components (MgO, Au, NaCl, TiB₂, Al₂O₃). Upon 367 cooling, the diffraction peaks from the sample undergo significant non-linear hkl-dependent broadening below ~ 420 K (Figure 2 and Extended Data Figure 1e). Similarly to previous studies^{18,23} we interpret this observation as the 368 369 result of a cubic-tetragonal transition in CaSiO₃. Close inspection of diffraction data from the Pixirad-8 detector, 370 which was employed during one experiment, reveals that weak superlattice peaks from CaSiO₃ (otherwise 371 unexplained by other cell components) appear in data collected at 373 and 300 K (Extended Data Figure 2). The strongest of these is observed at 2θ ~ 6.1° (indexed as 3/2 ½ ½ on the cubic sub-lattice) but additional peaks can 372 373 also be observed at 8.05° (3/2 3/2 ½), 12.1° (5/2 3/2 3/2) and 13.2° 20 (5/2 5/2 1/2) (*d*-spacings of ~ 2.11, 1.61, 1.07 374 and 0.98 Å). Assuming that CaSiO₃'s initial distortion upon cooling is to a tetragonal phase, there are three likely candidate structures, with space groups I4/mcm, P4/mbm and I4/mmm. The predicted superlattice peak positions 375 376 for the structure with I4/mcm symmetry exactly match the observed superlattice peak positions, explaining all four 377 observed peaks, whilst CaSiO₃ structures with space groups P4/mbm or I4/mmm cannot account for the 378 observations (Extended Data Figure 2) as these should both produce many additional superlattice peaks (e.g. those 379 indexed as ½ 0 3/2, 3/2 ½ 1 1/2 0 5/2, 5/2 1 ½ etc. on the basis of the cubic sub-cell) that are not observed. Given that there are no additional unexplained diffraction peaks, the knowledge that the most obvious superlattice peak 380 381 at ~ 6.1° 2 θ should indeed be the strongest in *I*4/*mcm* and that the same transition (*Pm* $\overline{3}m$ to *I*4/*mcm*) occurs in 382 $CaTiO_3^{24}$, we see no reason to doubt the structure of $CaSiO_3$ at room temperature and ~ 12 GPa is tetragonal with 383 space group 14/mcm.

384 The observed behaviour of $Ca[Si_{0.6}Ti_{0.4}]O_3$ is similar to $CaSiO_3$, but the intensities of the superlattice peaks are much 385 greater and the material is recoverable to ambient conditions. Diffraction patterns collected from the starting 386 material Ca[Si_{0.6}Ti_{0.4}]O₃ (measured alongside LaB₆ as a calibrant at 300 K, Extended Data Figure 3a) allows identification of superlattice peaks with odd and even Miller indices, which requires positive and negative 387 388 octahedral tilts³⁸. Thus, the highest possible symmetry of Ca[Si_{0.6}Ti_{0.4}]O₃ at ambient conditions is orthorhombic. 389 Refining the diffraction data in three likely spacegroups (Pbnm, Cmcm and P4₂/nmc) demonstrates that both Pbnm 390 and *Cmcm* can explain the patterns equally well, and so it is concluded ambient $Ca[Si_{0.6}Ti_{0.4}]O_3$ is orthorhombic 391 (actually monoclinic when B-cation ordered, see below). We choose to assume the space group is Pbnm, as this is

392 the known structure of $CaTiO_3^{24}$. Since no phase transition is observed during decompression, the room-393 temperature structure at high pressure is also assumed to be *Pbnm*, or its B-cation ordered equivalent $(P2_1/c)$. At 394 high-PT conditions Ca[Si_{0.6}Ti_{0.4}]O₃ is observed to undergo two structural phase transitions (Extended Data Figure 4). 395 At the highest temperatures, the diffraction pattern is well explained if the sample takes a cubic $Fm\overline{3}m$ structure. 396 With respect to the cubic form of $CaSiO_3$, this has a double unit-cell edge length and partial B-cation ordering (assumed to follow a 1:1 B-site scheme³⁹) of Si and Ti, a common ordering scheme in perovskites. On cooling, the 397 Ca[Si_{0.6}Ti_{0.4}]O₃ first distorts to a tetragonal structure marked by the appearance of the same set of superlattice 398 399 peaks that were observed for CaSiO₃. Since the cubic Ti-bearing capv is B-cation ordered, it is assumed the 400 tetragonal phase maintains this B-cation cation ordering, making the space group of the tetragonal phase I4/m. 401 Continued cooling sees distortion into the room temperature phase, which is assigned to be monoclinic $P2_1/c$ (the 402 B-cation ordered variant of Pbnm). Between the tetragonal and monoclinic structures there is a temperature 403 interval where the diffraction pattern cannot be indexed using a single structure model, and it is observed that this 404 temperature interval corresponds to very low acoustic velocities. This may be indicative of a first-order phase transition, which further implies a preference for Pbnm or $P2_1/c$ over Cmcm or C2/c, based on the analysis of 405 406 Glazer³⁸. Alternative explanations for this behaviour are a temperature interval of phase coexistence, an additional 407 perovskite structure or some other unexplained phenomena. Similar observations have been made for CaTiO₃⁴⁰, 408 where there is a small temperature interval between the I4/mcm and Pbnm structures where the behaviour is 409 attributed to an interval of phase coexistence caused by the kinetic energy barrier of the first-order phase 410 transition. A more detailed discussion of the crystallographic behaviour of capv samples lies beyond the scope of 411 the current study.

412 AB-INITIO CALCULATIONS AND THE SLOPE OF THE CAPV TETRAGONAL-CUBIC TRANSITION

413 Ab-initio calculations were performed to constrain the conditions of the tetragonal – cubic transition of capy 414 throughout the Earth's mantle. All simulations were carried out with the density-functional-theory (DFT) code 415 VASP⁴¹ using the projector-augmented-wave (PAW) method⁴² and the PBE formulation of generalised-gradientapproximation (GGA)⁴³. Molecular dynamics (MD) calculations used the Nosé thermostat and were run at the 416 417 gamma point with a cut-off of 600 eV and relaxed to within 10^{-5} eV and all forces to below 0.03 eV/Å. All 418 computational runs were at least 20 ps in duration, although all measured properties were observed to be fully 419 converged by 12 ps. Phonon calculations for calculating the force-constant matrix used an energy cut off of 850 eV, 420 4x4x4 K points and were relaxed to 10⁻⁸ eV with forces below 0.01 eV/Å. The finite difference method was used and 421 processing was done with the phonopy code⁴⁴. Ca atom semicore 3s and 3p states were treated as valence states. 422 All static and molecular dynamics runs were spin-polarised and CaSiO₃ was always simulated with an 80 atom 423 simulation box, based on a 2 x 2 x 4 assemblage of the perovskite cubic aristotype cell containing 5 atoms. For cubic 424 perovskite calculations the sub-cells from which the simulation box was constructed were fixed to have a geometry 425 of a=b=c, whereas for tetragonal CaSiO₃ the simulation geometry for was fixed such that $a=b\neq c$. Details of 426 simulation cell volumes are provided in Supplementary Information Table 7. In each case the final stress on the 427 crystal was correct to within 0.04 GPa for static calculations and to within 0.1 GPa (on average) for MD calculations. After the geometries were imposed at each PT condition the atoms were allowed to relax, within constraints 428 429 maintaining cubic or tetragonal structure, prior to simulations used to obtain free energies.

Free energy differences were calculated at 25, 75 and 125 GPa and at 0, 1000, 2000 and 3000 K. To calculate the
free energy of each state we used an approximation of the thermodynamic integration represented by eq. 1:

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$$F - F_0 \cong \langle U - U_0 \rangle_0 + \frac{1}{2k_b T} \langle [U - U_0 - \langle U - U_0 \rangle_0]^2 \rangle_0 \quad (eq. 1)$$

433 Where subscript 0 represents the reference state and other terms the state of interest. For a reference state we 434 used a harmonic oscillator with free energy defined by eq. 2:

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$$\boldsymbol{U}_{0} = \boldsymbol{U} + \sum_{i} \frac{p_{i}^{2}}{2m_{i}} + \frac{1}{2} \sum_{ij} \boldsymbol{u}_{i} \Phi_{ij} \boldsymbol{u}_{j} \qquad (eq. 2)$$

where u_x is a displacement vector and Φ_{ij} is the force constant matrix. As cubic CaSiO₃ is unstable at low 436 437 temperatures it has negative frequencies in its phonon spectrum and the free energy cannot be calculated directly from its force constant matrix. Thus, we applied a small correction to eliminate these imaginary frequencies and 438 439 allow free energy calculations. The correction matrix (Φ_{ii}^c) is formed by multiplying the onsite terms in the force constant matrix by kI (where I is the identity matrix and k is the smallest constant that eliminates all imaginary 440 frequencies). The correction procedure then subtracts the correction matrix from the true force constant matrix 441 442 $(\Phi_{ij} - \Phi_{ij}^c)$ to produce a modified force constant matrix Φ_{ij}^* which is used in eq. 2. Whilst this procedure cannot predict the correct absolute free energies, it should accurately calculate the free energy difference relative to a 443 444 reference state that has had the same correction procedure applied. This procedure is repeated for both cubic and tetragonal CaSiO₃ structures at each PT condition, and finally the free energy difference between these states is 445 446 determined using eq. 3:

$$\Delta G_{cub-tet} = \Delta G_{final-ref}^{cub} + G_{ref}^{cub} - (\Delta G_{final-ref}^{tet} + G_{ref}^{tet}) \qquad (eq. 3)$$

The final calculated free energy differences between cubic and tetragonal states are reported in Supplementary Information Table 7. All free energy differences are subject to uncertainties, which are assumed to be 1 meV/fu at 0K, and calculated to be < 10 meV/fu at high temperatures, from the statistical uncertainty in the simulation energies. These uncertainties were incorporated in the weighted least-squares regression used to determine the conditions and uncertainty of the tetragonal-cubic transformation at each pressure (25, 75 and 125 GPa) as plotted in Extended Data Figure 5.

454 CALCIUM PEROVSKITE EQUATION OF STATE

PVT-velocity data from experiments in this, and literature, studies were combined to allow fitting of an equation of state (EoS) for both tetragonal and cubic CaSiO₃ perovskite. The pressures of all reported literature data were converted to a single pressure-scale, which was used for pressure measurement in this study³⁷. This ensures consistency of the pressure-scale used throughout the dataset. The fitted EoS for tetragonal capv uses only room temperature data and all literature data collected on samples compressed in diamond anvil cells without use of a pressure-transmitting-medium were discarded (small symbols in Extended Data Figure 6a). The remaining PV data^{19,45}, including 1/20th of the data collected during room temperature decompression in this study prior to

462 amorphisation, were fitted to a second-order Birch Murnaghan EoS for V_0 and K_{T0} using the BurnMan software package¹¹ (Supplementary Information Table 2). Only 1/20th of the current data were used to ensure the final fitted 463 464 model was not overly biased to the data collected in this study. Subsequently, room temperature velocity measurements from this study were combined with literature data 17,20 to fit the shear modulus (G₀) and its pressure 465 derivative (G_0) within the SLB2005²⁹ formalism, as provided in BurnMan. As the equation of state is only calibrated 466 467 at room temperature, estimates of velocity reductions for tetragonal capv in Figure 4 are qualitatively based on the 468 magnitude of reductions observed in measurements from $Ca[Si_{0.6}Ti_{0.4}]O_3$, taken ~ 100 K below the phase transition, in this study, and are not calculated by using the tetragonal capy EoS. 469

470 The EoS for cubic capv (Supplementary Information Table 2) also fits data from this study and the literature⁴⁵⁻⁴⁸ using 471 the SLB2005²⁹ Mie-Grüneisen-Debye Birch-Murnaghan formalism implemented in BurnMan¹¹. Only data falling 472 above the calculated PT curve of the tetragonal-cubic phase transition (Extended Data Figure 5) were used to ensure than no data from tetragonal-structured capv were included (Extended Data Figure 6b). PVT data were first 473 used to fit V_0 , K_0 , K_0' and γ_0 . Subsequently, the complete PVT-velocity dataset was re-fitted for G_0 , γ_0 , q_0 and θ (V_0 , 474 K_{T0} , K_0' were unchanged) assuming $G_0' = 1.66$. G_0' was fixed to literature values^{4,5,20} due to the small pressure range 475 476 of velocity measurements in this study, a value consistent with literature scaling rules²⁹. η_{s0} was fixed at 3.3, based 477 on the scaling rules ($\eta_{s0}/\gamma_0 \simeq 2$) from Stixrude²⁹. Alternatively, η_{s0} can also be a fitted parameter (Supplementary 478 Information Table 2). However, since this second fit (with variable η_{s0}) results in slightly lower extrapolated velocities without significantly altering subsequent interpretation, fixing η_{s0} was viewed as a more conservative way 479 480 to evaluate the influence of capy. It is noted that the uncertainty bounds plotted in Figure 1 account for variation of 481 G_0 ' from 1.44 – 1.88 and capv remains slower than PREM and *ab initio* estimates throughout this entire range. We 482 recognise that, without using literature data, extrapolation throughout the mantle pressure range would be 483 completely unrealistic - and are beholden to accepting the reliability of literature data; however we note that four 484 previous studies on capy at high PT conditions, after conversion to a common pressure scale, can be combined and fitted to a single equation of state without any significant outliers. Readers are referred to Stixrude²⁹ for details of 485 486 the SLB2005 formalism.

487 THERMODYNAMIC MODELLING

488 The acoustic properties of MORB, peridotitic and harzburgitic assemblages have been calculated using the MMA-489 EoS software package⁴⁹. Simplified bulk compositions for MORB (NCFMAS) and pyrolite/harzburgite (CFMAS) from 490 the software's library were employed as typical of these assemblages throughout the lower mantle. Equilibrium 491 phase assemblages were calculated across a 0.5 GPa by 25 K grid throughout the mantle for each system, and the elastic properties of each assemblage extracted along self-consistent adiabatic temperature profiles beginning at 492 493 1000 K (representing slabs) and 1500 K (average mantle), which are plotted relative to one-another in Figure 4. The 494 latter temperature profile is very similar to the geotherm of Brown and Shankland ⁵⁰. Thermoelastic data from 495 Stixrude⁶ was used for all phases except for the MgSiO₃ bridgmanite endmember (which used updated properties 496 from Zhang⁵¹) and calcium perovskite which is defined in this study. We note that this database provides a 497 somewhat simplified view of lower mantle materials, as it does not include the effects of iron spin-transitions in ferropericlase or bridgmanite^{e.g. 52,53} or the ferroelastic phase transitions of stishovite^{e.g. 54}. We also highlight that the 498

modelling in this study inherently assumes that the database from Stixrude et al.⁶ accurately describes the elastic
 properties of all other lower mantle phases.

501 COMPARISON WITH PREVIOUS STUDIES

502 As noted in the main text, the acoustic velocities of CaSiO₃ observed in this study are observed to be significantly 503 slower than predicted in computational studies³⁻⁵ and mineralogical databases⁶. Additionally, they are also observed to be slower than those found in previous high PT experiments²⁰. Whilst we cannot fully explain the reasons for all 504 505 disagreements, we discuss some observations that may partially explain the mismatches. Database elastic properties⁶ predict the fastest capv velocities plotted in Figure 1, and it is these that the Earth Science community 506 507 currently uses when interpreting seismic observations. Results from the two ab initio molecular dynamics (AIMD) computational studies^{4,5} and experiments²⁰ all predict that capv should be equal to (v_p) or slower (v_s) than PREM. If 508 509 any of these results were adopted in mineralogical databases, slow velocity anomalies in the lower mantle could be 510 interpreted as they are in this study, an indicator of MORB enrichment, although not to the extent implied here. We 511 note that the other pseudo-high-temperature calculations³ do not provide enough information in the paper to 512 calculate acoustic properties at elevated temperatures, since the temperature effect on density is unquantified in 513 the original publication.

514 Comparing our work in detail, first with previous experimental results, it is observed that the room-temperature velocities 515 measured in this study are in excellent agreement with those of Gréaux et al.²⁰, Kudo et al.¹⁷ and extrapolated estimates 516 from Sinelnikov et al.¹⁶. Room-temperature velocities measured by Li et al.²² are somewhat faster, but given the lack of 517 details provided in that paper, which is a technical review, they are not considered further. It is observed that our reported 518 velocities disagree from previous experimental data only at high temperature²⁰, appearing to diverge as the reported 519 temperature increases. Given the similarities in methodology, it is most likely that temperature uncertainties are 520 responsible for the differences. Based on published details, we believe Gréaux et al. ²⁰ employed samples of 0.93-1.3 mm length (Figure 2 and Extended Data Figure 3 of Gréaux et al. 20) and 2mm diameter, with the thermocouple inserted radially 521 522 (through the furnace) adjacent to the far end of the pressure marker that in their experiments is initially ~ 1 mm in length. 523 This arrangement is substantially larger than the samples used in this study, which were 0.4-0.6 mm in length and 1.5 mm 524 diameter, with the thermocouple inserted axially to the end of the pressure marker that had a maximum length of 0.5 mm. 525 Thus, the maximum distance between the thermocouple and far end of the sample in our study is 1.1 mm, likely 0.75 mm at high pressure, approximately half of the equivalent distance in Gréaux et al.²⁰ (likely \ge 1.5 mm at pressure). Additionally, 526 527 by inserting the thermocouple axially in the cell at high-pressure, whereas it is unclear whether or not this would be the 528 case with a radial thermocouple, which could also have been affected by contacting the metal furnace. Given that, at 12-22 529 GPa, the sample column is likely to be 4-5 mm in length, the differences in geometry might have a very large influence on 530 the temperature conditions experienced by samples. Thermal modelling using finite element code⁵⁵ suggests that the thermal gradient across samples at a measured temperature of 1200 °C in our setup should be < 50-60 °C, with the 531 532 measured temperature likely to be lower than peak conditions. In contrast, assuming the thermocouple is centred (as 533 drawn by Gréaux et al.²⁰) and measuring 1200 °C, the range of temperatures experienced by a sample of 1 mm length x 2 534 mm diameter could very conceivably be 250 - 300 °C lower than that measured by the thermocouple. This implies that the 535 apparent effect of temperature on velocities should be smaller using the geometry of Gréaux et al.²⁰, since portions of 536 samples would be colder than believed. This is consistent with the observed differences in velocities between Gréaux et

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that the high-temperature velocities reported by Gréaux et al. ²⁰ might be less reliable is demonstrated by comparing the 538 independent estimates of bulk sound velocity expected for capv from pressure-temperature-volume systematics and 539 acoustic measurements $(V_{\phi} = \sqrt{V_P^2 - \frac{4}{3}V_S^2} = \sqrt{K_S/\rho}$, where K_s is a function of K_{0T} , K', γ and α). We observe that the bulk 540 sound velocity extracted from ultrasonic measurements in Gréaux et al.²⁰ are inconsistent with velocity extracted via a PVT 541 EoS using their diffraction data (Extended Data Figure 9). Ultrasonic V $_{\phi}$ values from Gréaux et al.²⁰ are offset to slower 542 543 values and have a much larger reduction at high temperature than those predicted via an EoS fitted using density from their or compiled literature data⁴⁵⁻⁴⁸. In contrast, bulk sound velocities from data in this study are consistent with 544 literature PVT EoS fitting. This inconsistency suggests velocities reported in Gréaux et al.²⁰ might be affected by large 545 546 temperature gradients.

al.²⁰ and the present study, where the offset in reported velocities increases at higher temperatures. Additional evidence

547 Considering calculated properties of capy, we observe that the database values³ best reproduce the adiabatic bulk 548 moduli of capv (compared with that from the global experimental dataset, Extended Data Figure 7b), whilst the two 549 AIMD studies^{4,5} predict a larger pressure effect on K_s than is observed in the experimental data. Other calculations 550 employing mean-field and Landau theory³ suggest that shear softening should be associated with the cubic-tetragonal transition, whilst AIMD approaches do not include this behaviour^{4,5}. However, it has been proposed⁴ that the choice of 551 552 cubic unit cell employed by Stixrude et al³ prevented rotations of the SiO₆ octahedra, leading to an anomalously large 553 shear modulus and explaining the high velocities. The AIMD results of Kawai and Tsuchiya⁴ should be preferred to those 554 from Li et al⁵, as the latter may not have fully converged and insufficiently sampled the Brillouin zone⁴ to accurately 555 predict crystal structure. Despite differences, all three computational studies predict a larger shear modulus than required 556 by experimental data (from both this and previous studies 16,17). It is possible that this discrepancy results from the strong 557 anharmonicity of capv, implying that extremely expensive calculations may be required to accurately describe capv's 558 elasticity using computational methods. Indeed, common first-principles methods inaccurately predict the elasticity or 559 phonon temperature dependence of other anharmonic cubic perovskites (SrTiO₃, BaTiO₃ and PbTiO₃)⁵⁶⁻⁵⁸. Pseudopotentials 560 that under/overestimate the cubic lattice parameter consistently over/underestimate the shear modulus in the opposite sense⁵⁶⁻⁵⁸. The Local-density-approximation (LDA), used by Kawai and Tsuchiya⁴, has been observed to overestimate the 561 shear modulus (c_{44}) of SrTiO₃, BaTiO₃ and PbTiO₃ by 8-18 % for ~ 1% underestimate of unit cell volume⁵⁶. Since we observe 562 563 a similar mismatch between the volume of cubic CaSiO₃ at adiabatic conditions based on our fit to experimental data and 564 the results of Kawai and Tsuchiya⁴, which are ~ 1 % too small, we expect that capv's velocities predicted by Kawai and Tsuchiya⁴ will be somewhat overestimated. However, it is unlikely this effect can explain the entirety of the disagreement 565 566 between previous calculations and our experimental results. A second contributor to the mismatch could be the presence 567 of crystallographic preferred orientation (CPO) within experimental samples, especially if alignment of an acoustically slow 568 direction coincided with the ultrasonic path. However, since refinement of X-ray diffraction patterns did not require CPO in 569 the cubic CaSiO₃ field to fit the data, this seems unlikely. Additionally, the way crystal symmetry is stipulated and the lack of 570 grain boundaries/defects in calculations may frustrate some phonon modes, further explaining the offset from 571 experimental values. Finally, we re-iterate that the finite-strain model we report in this paper is subject to very large 572 extrapolation from the experimental PT conditions (~ 12 GPa, 300 – 1500K) to those of the mantle (< 130 GPa, 1500 – 3000 573 K) and we acknowledge additional experiments are now required to investigate in better detail the changes of capv velocity 574 at more extreme conditions.

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Thomson et al. *(revision 1)*

631 Extended Data Figure 1: Lattice and diffraction peak parameters for CaSiO₃ and Ca[Si_{0.6}Ti_{0.4}]O₃ perovskite. Refined lattice

- parameters and pseudo-cubic unit cell volumes from (a) and (c) Ca[Si_{0.6}Ti_{0.4}]O₃ and (b) and (d) CaSiO₃ plotted as a
 function of experimental temperature. (e) Full-width-at-half-maximum (fwhm)of diffraction peaks of the CaSiO₃
- 634 perovskite sample, normalised to the fwhm at high temperature, measured at 100 °C intervals in a separate
- 635 experiment to that in Figure 2.

636 Extended Data Figure 2: X-ray diffraction patterns from CaSiO₃ perovskite. Stacked diffraction patterns of CaSiO₃ 637 perovskite at ~ 25, 100 and 200 °C. (a) full patterns, (b) and (c) are limited in 2θ range to allow indication of weak 638 superlattice peaks. Diffraction patterns are coloured by temperature conditions (see legend). The positions of the 639 diffraction peaks from the capv sample, MgO, NaCl and Au are indicated by markers – other small peaks are from 640 boron epoxy and/or furnace components. Cubic capv peaks are labelled with indices, hkl, in bold. The diffraction 641 patterns reveal the appearance of small superlattice reflections at T = 373 and 300 K at ~ 6.1°, 8.05°, 12.1° and 642 13.2° 2 θ (NB there is believed to be an additional superlattice reflection obscured at 10.5 ° 2 θ) labelled with *hkl* 643 indexed on the tetragonal (I4/mcm) unit cell and marked with gold stars.

Extended Data Figure 3: **Refined X-ray diffraction patterns from Ca**[Si_{0.6}**Ti**_{0.4}]**O**₃ **perovskite.** top to bottom – Rietveld refinements of Ca(Si_{0.6}Ti_{0.4})O₃ samples from (a) in $P2_1/c$ with LaB₆ calibrant and ambient pressure and 300 K. (b) in the tetragonal *I*4/*m* structure (with other cell components) at 890 K and high pressure (~ 12 GPa) and (c) in $Fm\overline{3}m$ *at* 1336 K and high pressure (12 GPa).

Extended Data Figure 4: X-ray diffraction patterns from Ca[Si_{0.6}Ti_{0.4}]O₃ perovskite. (a) Complete diffraction pattern of the Ca(Si_{0.6}Ti_{0.4})O₃ sample as a function of temperature at ~ 12 GPa, with diffraction intensity indicted by colour scaling. (b)-(h) magnified panels focussing on the temperature evolution of the 311, 222, 400, 422, 440, 620 and 444 diffraction peaks (indexed using a cubic lattice with *a* ~ 7.3 Å), demonstrating the change in thermal expansivity between cubic and tetragonal/monoclinic structures, and allowing visual identification of the observed phase transitions.

Extended Data Figure 5: Phase diagram of calcium perovskite throughout the mantle from *ab-initio* simulations and experiments. Cubic-tetragonal transition extrapolated throughout the mantle based on *ab initio* (solid circles) and experimental (triangles) constraints from this study. Vertical error bars (1 σ) and the grey envelope (80% confidence interval) represent the uncertainty in computational results from this study. A 1500 K mantle adiabat and cold slab temperature profile are plotted as red curves, with dashed red arrows indicating the warming occurring during slab stagnation at 700-1000 km depth. Results from previous experimental and computational studies are plotted as open symbols and grey curves respectively.

Extended Data Figure 6: **Equations of state for CaSiO₃ perovskite.** (a) *PV* EoS for tetragonal CaSiO₃ at 300 K, fitted to data from this study only (purple line) and combined with data from previous studies (thick black curve). Only data with large symbols, those that used pressure transmitting media, have been included in fitting the EoS. All small symbols are from experiments that did not employ a pressure-transmitting medium so have been excluded as volumes are expected to be affected by residual sample stress. Additionally, data from Wang et al. (1996)⁴⁷ were excluded as they used energy

18

dispersive diffraction in the large volume press, which can be subject to larger uncertainties in volume. Computational equations of state for tetragonal capv are plotted as dashed curves for comparison. ^{3,17,19,45,47,59-64} b) *PVT* EoS for cubic CaSiO₃ perovskite at 298 K and along a 1600 K adiabat fitted to data from this and previous studies. Small, partially transparent symbols, are literature data that were not included in the fitting, either due to falling below the calculated slope of the cubic-tetragonal transition (Methods) or due to concerns about data accuracy. The inset histogram shows the, approximately normal, distribution for the residuals for the fitted data compared with the best fit model, demonstrating the lack of outliers.

Extended Data Figure 7: Bulk sound velocity and bulk modulus of CaSiO₃ perovskite. (a) Bulk sound velocity of capv
predicted from the EoS in this study along a 1600 K mantle adiabat and at 300 K, compared with results from
previous computational studies on a 1600 K adiabat, a fit to previously published PVT diffraction data, and PREM.
(b) The adiabatic bulk modulus of CaSiO₃ perovskite calculated at 300 K and along a 1600 K mantle adiabat using the
finite strain model from this study, compared with thermodynamic results in Stixrude et al. ⁶ and previous high
temperature computational studies^{4,5}.

Extended Data Figure 8: Exemplar radiographic image, ultrasonic data and schematic of the experimental cell design. (a)
Example of synchrotron radiographic image used to measure sample length from Ca(Si_{0.6}Ti_{0.4})O₃ sample, (b) ultrasonic
signals from "runa" experiment of CaSiO₃ sample and (c) cross section of the experimental assembly (to scale) used in
ultrasonic experiments throughout this study.

Extended Data Figure 9: **Comparison of bulk sound velocity from diffraction and ultrasonic measurements.** Comparison of bulk sound velocity calculated from *PVT* equations of state, fitted to literature diffraction data (solid curves) and only that reported by Gréaux et al. (dashed curves), compared with bulk sound velocities calculated from ultrasonic measurements via $V_{bulk} = (V_p^2 - 4/3V_s^2)^{0.5}$ for data from Gréaux (squares) and this study (circles). All curves and symbols are coloured for temperature.

688

























