

1 **Thermo-economic analysis of composite district heating substation**
2 **with absorption heat pump**

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10 **ABSTRACT**

11 A novel composite district heating substation (CDHS) with absorption heat pump was proposed and
12 analyzed in this paper. The CDHS was composed of a water-LiBr absorption heat pump and two
13 plate heat exchanges, which could improve the utilization efficiency of geothermal water and the
14 primary supply water of the primary district heating network. The effect of geothermal water
15 temperature and mass flow rate, and primary supply water mass flow rate variation on system
16 performance were investigated and analyzed by case study. The objective was to maximize net profit
17 and minimize payback period by the cascade utilization of the geothermal water and primary supply
18 water. In addition, economic analysis and multi-objective optimization were conducted to find the
19 optimal mass flow rate of the primary supply water based on the TOPSIS decision making method.
20 Exergy loss analysis was applied under the optimal condition to discover which components had
21 the largest exergy loss. Results indicated that, the proposed system had a net profit of 16.22M\$ in
22 the life time and the minimum payback period was 2.2 years at the optimal primary supply water
23 mass flow rate of 46.16kg/s where the COP and exergy efficiency of the system were 1.85 and
24 59.81%, respectively.

¹ These authors contributed equally to this work.

25 **Keywords:** Geothermal water, Composite district heating system, Absorption heat pump,
 26 Economic analysis, Multi-objective optimization
 27

Nomenclature		t	temperature(°C)
A	area (m ²)	T	temperature (K)
a	solution circulation ratio	w	concentration of solution(%)
c	unit cost of exergy (\$/kJ)	\dot{Z}	capital cost rate(\$/s)
c_p	constant-pressure specific heat of water (kJ/kg/K)	Z	capital cost(\$)
\dot{C}	cost rate (\$/s)	Greek letters	
C_i	proximity index	φ	maintenance factor
COP	coefficient of performance	ψ	exergy efficiency
CRF	capital recovery factor	Subscripts	
$CDHS$	composite district heating system	abs	absorber
D	the mass flow rate of working fluid(kg/s)	dh	district heating
Ex	exergy rate (kW)	ds	driving source
EEV	electrical expansion valve	cond	condenser
F	correction factor	evap	evaporator
h	specific enthalpy (kJ/kg)	f	fuel
HP	heat price(\$/GJ)	ge	generator
i	annual interest rate(%)	gw	geothermal water
I	exergy loss rate(kW)	he	heat exchanger
K	overall heat transfer coefficient [W/(m ² ·K)]	i	ith element
\dot{L}	exergy loss rate (kW)	in	inlet
\dot{m}	mass flow rate (kg/s)	lm	logarithmic mean
n	system life time (yr)	out	outlet
N	annual operating hours (h)	p	product
NP	net profit (\$)	sol	solution
PP	payback period (year)	s	supply/strong
Q	heat rate (kW)	w	weak
s	specific entropy [kJ/(kg·K)]	ws	weak solution
S_i^+	distance from point i to positive ideal point	wf	working fluid
S_i^-	distance from point i to negative ideal point		

28 **1. Introduction**

29 Nowadays, coal, petroleum and other fossil energy have been widely applied in
30 urban district heating (DH), producing a lot of pollutants such as SO₂ and NO_x.
31 Although switching from coal to natural gas can reduce air pollution, there will be large
32 amounts of greenhouse. Therefore, air pollution and greenhouse gas problems caused
33 by heating are serious in countries with energy structure dominated by fossil fuels in
34 these countries like China [1]. With the rapid decline of fossil energy, the single
35 structure of energy will be greatly impacted by many factors in the future and must be
36 changed. Many researches for renewable energy have been conducted due to several
37 problems caused by fossil fuels, such as pollution, the greenhouse effect and acid rain.
38 Supplying energy via renewable energy is one of the most important methods for
39 environmental protection.

40 Geothermal energy is reliable, cheap, and environmental-friendly which is a
41 competitive alternative to substitute the conventional fossil fuels. Utilization of the
42 geothermal energy to district heating could be helpful to solve the energy and
43 environmental problems [2]. Due to the limitations of geothermal temperature and
44 equipment investment, most medium-low temperature geothermal source are used for
45 heating [3]. Thus, many researchers focus on the utilization of geothermal energy for
46 heating in different patterns. By 2015, there are 82 countries using geothermal energy
47 as heat source for heating [4]. Nian et al. presented a geothermal heating system with
48 abandoned oil wells, and built a heat transfer model. They examined the geothermal
49 production, room temperature and fluid production temperature by the model and
50 indicated that, an abandoned oil well with 3000m depth could be used for heating with
51 the area of 10000m² [5]. Jonas K et al investigated the exergy and economic
52 performance of a geothermal heat pump aided district heating system. The purpose was
53 to find the optimum solution of the system that the geothermal source was used in a
54 shell and tube type heat exchanger. They found that the system could heat for the
55 number of 13,766 residences and had an attractive investment for Simav region [6].

56 Miroslav V et al assessed a DH system using geothermal heat pump technology. The
57 environmental sustainability of geothermal heat pumps for district heating were
58 analyzed. The results showed that the geothermal heat pump had the advantages of
59 reducing the inlet primary energy by at least 30% with internal rate of return of up to
60 38% and payback period of 4.9 years [7].

61 Absorption heat pump (AHP) is one method for heating because it has large
62 potential to improve energy efficiency, save energy, protect environment and reduce
63 greenhouse gas. Sun et al presented a new configurations of district heating based on
64 natural gas and geothermal energy to reduce gas consumption and irreversible loss.
65 They analyzed the performance of thermodynamic and financial benefit and found that
66 the exergy efficiency could be improved by 12% and the natural gas consumption could
67 be reduced by 54% compared with conventional systems [8]. Lu et al designed a novel
68 gas-fired absorption heat pump that the sensible heat in high-grade and latent heat in
69 low-grade of flue gas were recovered. The pressure of intermediate evaporation and
70 absorption processes could be adjusted to enhance the adaptability in cold regions. The
71 simulated results showed that the energy saving potential of the system could be 39.6%
72 and payback period was 2.5 compared with conventional AHP and gas-fired boiler [9].
73 Wu et al selected 6 cities with typical climates and the performance of electric driven
74 ground source heat pump and the absorption ground source heat pump were compared
75 and analyzed. The research showed that the efficiency of primary energy utilization and
76 the balance between cold and heat of the absorption ground source heat pump were
77 significantly higher than that of the electric driven ground source heat pump. The
78 difference between them became more and more obvious with the operation of the unit
79 [10].

80 Both of geothermal energy and AHP contribute a lot for energy saving and
81 environment improvement. Therefore, many efforts have been conducted for theoretical
82 analysis and engineering application. Substantial studies about exergy and economic
83 analysis of AHP are available in open literatures. Luca et al proposed a reversible
84 absorption heat pump and internal combustion engine integration system which

85 employed a water-ammonia mixture. Energy analysis was conducted to evaluate the
86 economic viability and the second-law analysis was applied to compare the system
87 exergy efficiency with conventional systems [11]. Chen et al investigated a proposed
88 compression-absorption heat pump by heat-driven turbine. They built up the
89 mathematical models which included mass conversation, energy conversation and
90 exergy analysis [12]. It can be seen that energy analysis and exergy analysis have been
91 carried out about AHP. In addition, the optimization methods are effective to lower the
92 system cost. Cui et al proposed an innovative cascade absorption heat pump system for
93 recovering low-grade waste heat. They conducted the energy, exergy and economic
94 analyses of the system and optimized total annual cost and exergy destruction by multi-
95 objective optimization method. The results showed that multi-objective optimization
96 scheme was the most comprehensive and optimal scheme of the system [13].

97 There are numerous studies about the exergy analysis and multi-objective
98 optimization of geothermal water absorption heat pump. However, among district
99 heating technologies, conventional geothermal absorption heat pump has difficulty in
100 fully utilizing geothermal water and primary supply water has poor performance with
101 high return water temperature. For a conventional heating substation in China, return
102 water temperature(60°C-70°C) is too high which can be reduced to improve the energy
103 efficiency. For a conventional geothermal water, it is difficult to recharge water with
104 the temperature below 25°C. There is still a lack of knowledge of CDHS regarding the
105 thermo-economic analysis and multi-objective optimization. Therefore, this paper
106 presented a novel composite district heating substation (CDHS) integrated geothermal
107 water into district heating system which could achieve cascade utilization of geothermal
108 water and improve the utilization efficiency of primary supply water. Then a
109 mathematical model of CDHS in terms of energy, exergy, net profit and payback period
110 was developed to help experiment and engineering design. Further, the payback period
111 and net profit of CDHS was improved through the method of multi-objective
112 optimization. Multi-objective optimization of CDHS was carried out to find the optimal
113 operating condition. Finally, some recommendations were provided for the optimal

114 primary supply water mass flow rate. The main objectives include:

115 (1)Energy, exergy and economic performance of CDHS should be evaluated.

116 (2)The optimal mass flow rate of primary supply water of CDHS should be suggested.

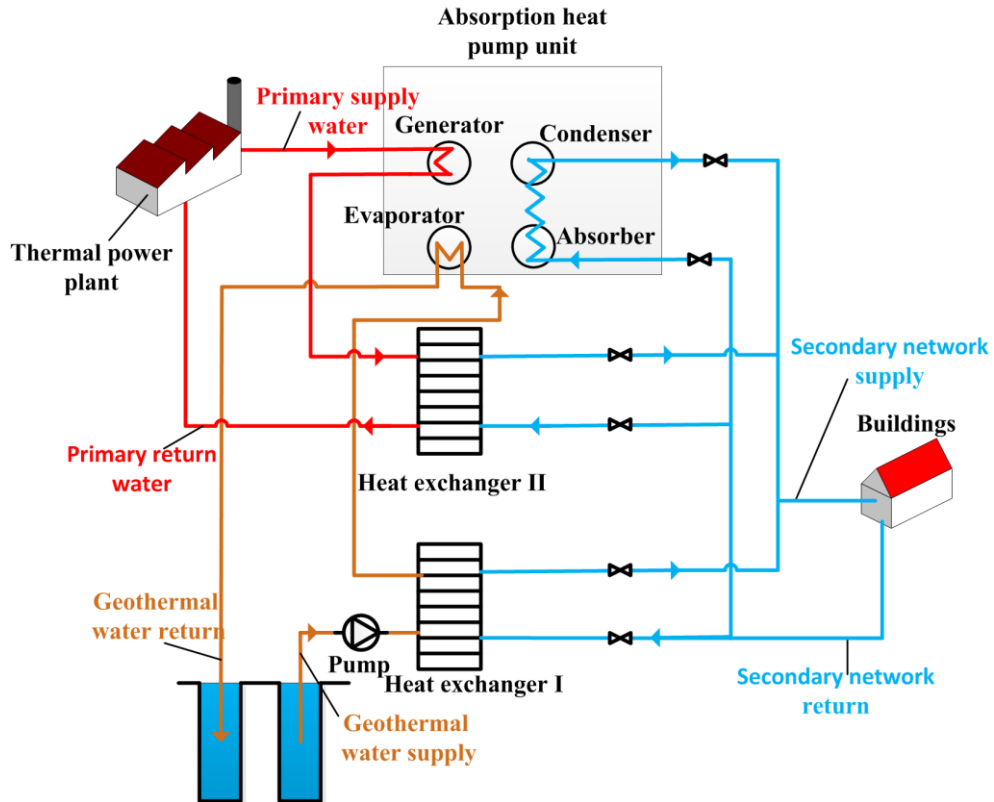
117 (3)The maximum net profit and the minimum payback period should be analyzed.

118 **2. System description**

119 The schematic diagram of the CDHS is shown in [Fig. 1](#). The system consists of a
120 water-LiBr absorption heat pump and two plate heat exchangers. There are three cycles
121 including driving source cycle, district heating cycle and the geothermal cycle
122 (illustrated by the red, blue and brown lines, respectively).

123 For the driving source cycle, primary supply water coming from the DH network
124 is used to drive the absorption heat pump and the heat is delivered to the generator. The
125 medium temperature primary supply water enters the heat exchanger II, to heat the
126 return district water, and the temperature is reduced. The secondary network consists of
127 three branches. In the first branch, the secondary return water passes through absorber
128 and condenser subsequently and absorbs heat. In the second branch, the return district
129 water enters the heat exchanger II and absorbs heat from primary supply water. The
130 secondary return water in the third branch is pumped into exchanger I and the
131 geothermal heat is delivered to it. For the geothermal cycle, the geothermal water(50-
132 70°C) goes through the heat exchanger I and releases heat to district heating water. Then
133 the medium temperature geothermal water (30-40°C) enters the evaporator of the AHP
134 and the temperature is further reduced, and finally it is discharged to the well.

135



136

137

Fig. 1 Schematic diagram of the CDHS

138

Fig. 2 shows the schematic of AHP which consists of five main sections:

139

evaporator, absorber, condenser, generator and solution heat exchanger. It includes two

140

cycles: solution cycle and working fluid cycle. In the solution cycle, the weak solution

141

from the absorber is pumped into solution heat exchanger. The pressure and

142

concentration remain constant but the temperature increases. Then it enters generator

143

and heated by the primary supply water. As the temperature of the solution rises, vapor

144

is produced and the solution becomes strong solution. The strong solution goes through

145

the solution heat exchanger and flows into the absorber. In the working fluid cycle, the

146

working fluid vapor generating in generator enters condenser and is cooled into liquid.

147

The liquid working fluid is throttled by the EEV and goes into evaporator. The working

148

fluid absorbs heat from the geothermal water in the evaporator and turns into low

149

pressure vapor, and then absorbed by the strong solution in absorber.

150

The suitable selection of working fluid and solution can reduce cycle

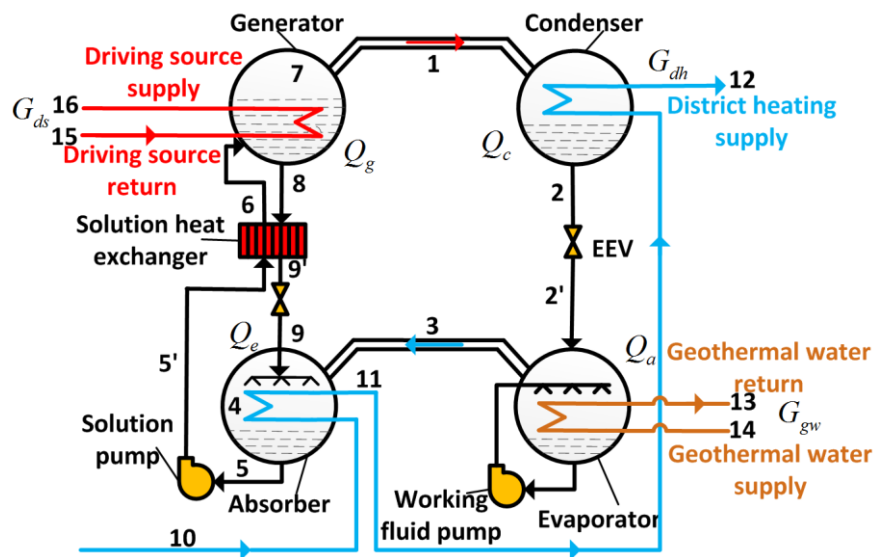
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thermodynamic inefficiencies and achieve higher energy conversion efficiency and

152 lower capital cost. The solution of LiBr + H₂O was used in absorption cycles around
 153 1930 which had strong hygroscopicity. Water is non-toxic, non-inflammable, in-
 154 explosive with large latent heat. LiBr has a high boiling point which is easily soluble in
 155 water with stability. Temperature difference of boiling point between H₂O and LiBr
 156 ensures the rapidly development of this combination. The efficiency of AHP cycle is
 157 higher than the cycle with NH₃, but the pressure was lower than that [14]. Therefore,
 158 LiBr + H₂O was selected as the working fluid and solution of the absorption heat pump.

159

160



161

162 Fig. 2 Schematic diagram of the water-LiBr AHP

163 3. Method

164 For simplicity, the following assumptions are made during the modeling process:

165 (1) The system is under state of thermal balance and stable operation, and heat
 166 exchange with the environment is neglected.

167 (2) The working fluid of evaporator and condenser outlet is saturated and reaches
 168 the thermal balance.

169 (3) The working fluid of absorber and generator outlet is saturated solution, and
 170 there is no insufficient absorption and insufficient occurrence.

171 (4) The losses of heat, pressure and flow resistance are ignored.

172 (5) The enthalpy of the working fluid remains constant before and after the
 173 throttling process.

174 (6) The works of the solution and solvent pumps are neglected.

175 (7) Logarithmic mean temperature difference is used in heat transfer calculation.

176 3.1. Energy analysis

177 In the generator and absorber, the energy equation can be formulated as:

$$\begin{cases}
 Q_{ge.1} = (G_{ws} - D_{wf})h_8 + D_{wf}h_1 - G_{ws}h_6 \\
 Q_{abs.1} = (G_{ws} - D_{wf})h_9 + D_{wf}h_3 - G_{ws}h_5 \\
 Q_{ge.1} = G_{hs}c_p(t_{15} - t_{16}) \\
 Q_{abs.2} = G_{hw}c_p(t_{11} - t_{10})
 \end{cases} \quad (1)$$

179 where G_{ws} is the mass flow rate of weak solution and D_{wf} is the mass flow rate of
 180 working fluid. The heat transfer rate of generator can also be expressed as:

$$Q_{ge.1} = D_{wf} [(a-1)h_8 + h_1 - ah_6] \quad (2)$$

182 where a is solution circulation ratio which can be calculated as:

$$a = \frac{w_s}{w_s - w_w} \quad (3)$$

184 The energy equation in evaporator and condenser can be defined as:

$$\begin{cases}
 Q_{cond,1} = D_{wf}(h_1 - h_2) \\
 Q_{evap,1} = D_{wf}(h_3 - h_2) \\
 Q_{cond,2} = G_{hw}c_p(t_{12} - t_{11}) \\
 Q_{evap,2} = G_{gw}c_p(t_{13} - t_{14})
 \end{cases} \quad (4)$$

186 In the solution heat exchanger, the energy equation can be expressed as:

187

$$Q_{sol,1} = (G_{ws} - D_{wf})(h_8 - h_9) \quad (5)$$

$$Q_{sol,2} = G_{ws}(h_6 - h_5)$$

188 The COP of the absorption heat pump is the ratio of the heat capacity to heat
 189 consumption and the flowchart is shown in Fig. 3:

190

$$COP = \frac{Q}{Q_{ge}} = \frac{Q_{abs} + Q_{cond}}{Q_{ge}} = \frac{Q_{ge} + Q_{evap}}{Q_{ge}} \quad (6)$$

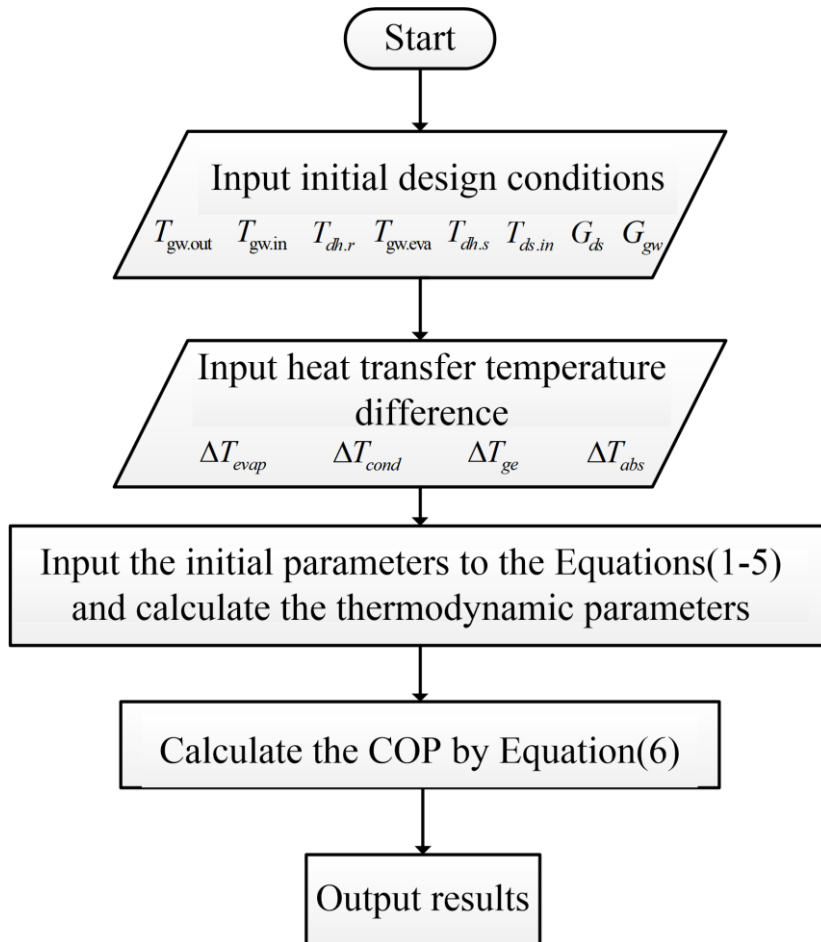
191 According to the Eqs(1-6), COP can be expressed as:

192

$$COP = 1 + \frac{Q_{evap}}{Q_{ge,1}} = 1 + \frac{(a-1)h_8 + h_1 - ah_6}{(h_3 - h_2)} \quad (7)$$

193

194



195

196

Fig. 3 Flowchart for thermodynamic calculation process.

197 **3.2. Exergy analysis**

198 Exergy rate is an effective energy which can be theoretically converted into work.
 199 It is made up of physical and chemical exergy when the kinetic and potential energies
 200 are ignored. The exergy rate at point i can be calculated as:

201

$$Ex_i = m[(h - h_0) - T_0(s - s_0)] \quad (8)$$

202 The exergy rate balance equation and the exergy efficiency for each component of
 203 the system are given in [Table 1](#).

204 The performance of the system can be evaluated by the second law of
 205 thermodynamics which is based on exergy loss and exergy efficiency. The exergy
 206 efficiency of the proposed CDHS can be defined as the ratio of the product exergy rate
 207 to the fuel exergy rate [\[15\]](#):

208

$$\psi = \frac{Ex_{p,i}}{Ex_{f,i}} \quad (9)$$

209 In this study, the product and fuel exergy rate can be defined as:

210

$$Ex_p = Ex_{dh,out} + Ex_{ds,out} \quad (10)$$

211

$$Ex_f = Ex_{ds,in} + Ex_{gw,in} - Ex_{gw,out} + Ex_{dh,in} \quad (11)$$

212 Accordingly, the exergy efficiency of the CDHS can be expressed as:

213

$$\psi = \frac{Ex_{dh,out} + Ex_{ds,out}}{Ex_{ds,in} + Ex_{gw,in} - Ex_{gw,out} + Ex_{dh,in}} \quad (12)$$

214

Table 1 Exergy rate balance equation and the exergy efficiency for each component

Component	Exergy balance equation	Exergy efficiency
Condenser	$Ex_1 + Ex_{11} = Ex_{12} + Ex_2 + I$	$\psi_{cond} = (Ex_{12} - Ex_{11}) / (Ex_1 - Ex_2)$
Evaporator	$Ex_{13} + Ex_2 = Ex_{14} + Ex_3 + I$	$\psi_{evap} = (Ex_3 - Ex_2) / (Ex_{13} - Ex_{14})$
Absorber	$Ex_9 + Ex_{10} + Ex_3 = Ex_{11} + Ex_5 + I$	$\psi_{abs} = (Ex_{11} - Ex_{10}) / (Ex_3 + Ex_9 - Ex_5)$
Generator	$Ex_{15} + Ex_6 = Ex_{16} + Ex_1 + Ex_8 + I$	$\psi_{ge} = (Ex_1 - Ex_6) / (Ex_{15} - Ex_{16} - Ex_8)$
Solution heat exchanger	$Ex_5 + Ex_8 = Ex_6 + Ex_9 + I$	$\psi_{sol} = (Ex_6 - Ex_5) / (Ex_8 - Ex_9)$
Solution pump	$Ex_5 + W_p = Ex_5 + I$	$\psi_{sp} = (Ex_5 - Ex_5) / W_p$

215

216 3.3. Economic model

217 The economic performance of the proposed CDHS can be evaluated by the unit
 218 cost of exergy for the heat capacity which is expressed as:

219

$$c_{cost} = \frac{\dot{C}_{cost}}{\dot{Q}} \quad (13)$$

220 where \dot{C}_{cost} is the rate of cost of CDHS. It is the sum of heat source cost rate (\dot{C}_{hs})

221 and capital investment and maintenance cost rate (\dot{Z}_k), and the balance equation is:

222

$$\dot{C}_{cost} = \dot{C}_{hs} + \dot{Z}_k \quad (14)$$

223

$$\dot{C}_{hs} = c_{hs} \dot{Ex}_{hs,in} \quad (15)$$

224
$$Z_k = \frac{Z_k \times \varphi}{N \times 3600} CRF \quad (16)$$

225 where c_{hs} is the unit cost of exergy for the heat source. φ is the maintenance factor
 226 and N is the annual operating hours. Z_k is the capital investment cost of all the
 227 components and can be calculated by the equation as follows [16]:

228
$$Z_k = 1397 \times A_{He}^{0.89} \quad (17)$$

229 CRF is the capital recovery factor that present value can be converted into a
 230 stream of equal annual payments, and it can be described as follows:

231
$$CRF = \frac{i(1+i)^n}{(1+i)^n - 1} \quad (18)$$

232 where i is the annual interest rate and n is the system life time.

233 When calculating Z_k , the area of the equipment(A) must be taken into account
 234 and can be expressed as follows:

235
$$A = \frac{Q}{KF\Delta T_{lm}} \quad (19)$$

236 Where K and ΔT_{lm} are the overall heat transfer coefficient and logarithmic mean
 237 temperature difference in heat exchangers. F is the LMTD correction factor and can be
 238 determined by Fettaka et al [17].

239 **3.4. Multi-objective optimization**

240 In this system, the payback period (PP) and net profit (NP) are chosen as objective
 241 functions to analyze the thermo-economic performance of the CDHS. The NP is the
 242 system life time times the difference of the benefit and cost, and the PP equals to the
 243 cost of the system life time divided by the annual benefit. They can be defined as

244 follows:

$$245 \quad NP = (Q \times N \times HP - \dot{C}_{\text{cost}} \times 3600 \times N) \times n \quad (20)$$

$$246 \quad PP = \frac{\dot{C}_{\text{cost}} \times 3600 \times n}{Q \times HP} \quad (21)$$

247

248 where HP is the heat price.

249 In this system, it is expected to maximize the NP and minimize the PP, but these
250 two goals are conflict. Therefore, multi-objective optimization is applied to solve this
251 problem. A fuzzy non-dimensionalization method is used to analyze the data of NP and
252 PP, and the TOPSIS method is applied to find the optimal design point in the CDHS
253 [18-21]. The maximizing objective NP and minimizing objective PP can be defined as:

$$254 \quad NP_i^n = \frac{NP_i - NP_i^{\min}}{NP_i^{\max} - NP_i^{\min}} \quad (22)$$

$$255 \quad PP_i^n = \frac{PP_i^{\max} - PP_i}{PP_i^{\max} - PP_i^{\min}} \quad (23)$$

256 where i is the i th element.

257 TOPSIS method is a multi-objective decision making method. The principal of the
258 method is to define the ideal point and the negative ideal point of the decision making
259 problem which can find the optimal point in the feasible solution. The optimal point is
260 closest to the ideal point and furthest from the negative ideal point. The distance from
261 a point to a positive ideal point(S_i^+) and the distance to a negative ideal point(S_i^-) can
262 be expressed as:

$$263 \quad S_i^+ = \sqrt{(PP_i^n - PP_i^{\max})^2 + (NP_i^n - NP_i^{\max})^2} \quad (24)$$

264
$$S_i^- = \sqrt{(PP_i^n - PP_i^{\min})^2 + (NP_i^n - NP_i^{\min})^2} \quad (25)$$

265 The proximity index can be defined as:

266
$$C_i = \frac{S_i^-}{S_i^+ + S_i^-} \quad (26)$$

267 If PP_i and NP_i are the ideal solution, then the C_i equals to 1. If they are
 268 negative solution, the C_i is equal to 0. The closer C_i gets to 1, the closer the solution
 269 is to the ideal solution

270 4. Case study

271 In order to ensure the practicability of the system, a commercial center project
 272 located in Tianjin (China) is used for performance analysis. There are two geothermal
 273 wells. One has an average water production rate of 30kg/s at 60°C and the other is for
 274 recharging. The radiant floor heating is used for space heating in the commercial
 275 building, of which the secondary supply and return water temperature are 45°C and 35
 276 °C, respectively. Thermodynamic design condition of the system has been listed in
 277 [Table 2](#) and the parameters applied in thermo-economic analysis are shown in [Table 3](#).

278 Table 2 Thermodynamic design conditions of the case.

Parameters	Value
Total heating capacity(MW)	18
Temperature of the geothermal water outlet, $T_{gw.out}$ (°C)	60
Temperature of the geothermal water inlet, $T_{gw.in}$ (°C)	19
The temperature of geothermal water at the inlet of evaporator $T_{gw.eva}$ (°C)	37
Temperature of the district heating supply water, $T_{dh.s}$ (°C)	45
Temperature of the district heating return water, $T_{dh.r}$ (°C)	35

Temperature of the primary supply water, $T_{ds.in}$ (°C)	110
Mass flow rate of the primary supply water, G_{ds} (kg/s)	60
Mass flow rate of the geothermal water, G_{gw} (kg/s)	30
Temperature difference of evaporator, ΔT_{evap} (°C)	2
Temperature difference of condenser, ΔT_{cond} (°C)	5
Temperature difference of generator, ΔT_{ge} (°C)	3
Temperature difference of absorber, ΔT_{abs} (°C)	5
Temperature difference of solution heat exchanger, ΔT_{sol} (°C)	20
Overall heat transfer coefficient of the evaporator, K_{evap} [W/(m ² · K)]	2791
Overall heat transfer coefficient of the condenser, K_{cond} [W/(m ² · K)]	5234
Overall heat transfer coefficient of the absorber, K_{abs} [W/(m ² · K)]	1163
Overall heat transfer coefficient of the generator, K_{ge} [W/(m ² · K)]	1623
Overall heat transfer coefficient of the solution heat exchanger, K_{sol} [W/(m ² · K)]	465
Correction factor of LMTD, F	1
Ambient temperature, T_0 (°C)	25
Ambient pressure, p_0 (kPa)	101.3

279

280

Table 3 The parameters using in thermo-economic analysis[22].

Parameters	Value
Annual operational hours, N (h)	3600
Annual interest rate, i (%)	14
The system life time, n (year)	15
Heat price(gas-boiler), HP (\$/GJ)	6.7
Maintenance factor, φ	1.06
Unit cost of exergy for the geothermal source, c_{hx} (\$/GJ)	6.7

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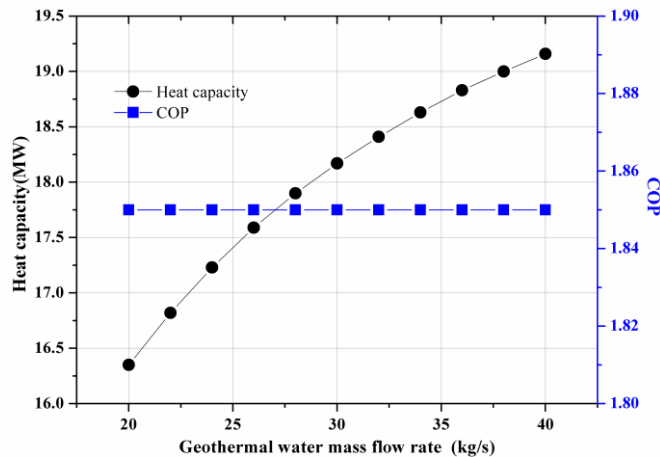
282 5. Results and discussion

283 5.1. Performance of the CDHS

284 The performance of CDHS at variable geothermal water mass flow rate(G_{gw}) and
285 temperature of geothermal water at the inlet of evaporator($T_{gw,eva}$) were conducted in
286 this paper.

287 Fig. 4 shows the variations of the heat capacity and COP with the geothermal
288 water mass flow rate. It can be seen that the heat capacity is almost linearly increased
289 with the increase of geothermal water mass flow rate. But the geothermal water mass
290 flow rate in the range of 20-40kg/s has no impact on the COP. As the geothermal water
291 mass flow rate increases from 20kg/s to 40kg/s, the heat capacity increases from
292 16.35MW to 19.16MW and the COP remains unchanged, respectively. This can be
293 explained by that the increase of geothermal water mass flow rate leads to the increase
294 of heat transfer in evaporator and exchanger I. Then more vapor generating in
295 evaporator is absorbed which will enhance the absorption capacity and result in the
296 increase of heat capacity. However, operating conditions of the system including
297 temperature, pressure, temperature difference, enthalpy and solution concentration
298 remain unchanged with the increase of geothermal water mass flow rate. According to
299 the Eq(7), COP remains unchanged.

300



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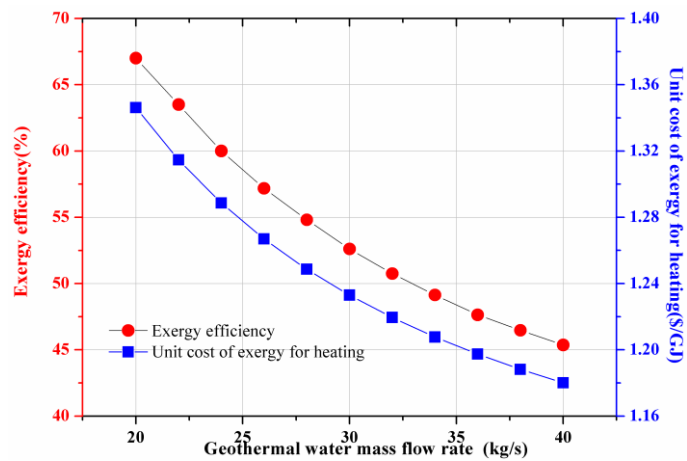
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Fig.4 Variation of heat capacity and COP with geothermal water mass flow rate

304

305 Fig. 5 displays the tendency of exergy efficiency and unit cost of exergy for heating
306 with geothermal water mass flow rate. What can be seen is that the exergy efficiency
307 and unit cost of exergy for heating decrease with the increase of the geothermal water
308 mass flow rate. According to the Eqs. (8-12), the exergy rate of driving source supply
309 and return water remain unchanged with the increase of geothermal water mass flow
310 rate. The exergy rate of district heating supply and return water, and the exergy rate of
311 geothermal supply and return water rise with the increase of geothermal water mass
312 flow rate. The decrease of exergy efficiency with the increase of geothermal water mass
313 flow rate can be explained by above. Additionally, with the increase of geothermal
314 water mass flow rate, there is an increase in the heat capacity and cost rate. But the
315 growth rate of heat capacity is greater than the cost rate which leads to the decrease of
316 unit cost of exergy for heating. The working fluid changes its phase from liquid to vapor
317 in evaporator.

318



319

320

321 Fig. 5 Variation of exergy efficiency and unit cost of exergy for heating with geothermal

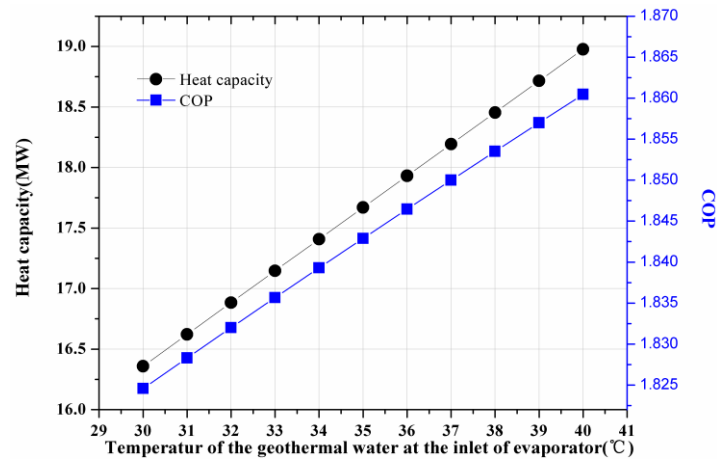
322

water mass flow rate

323

324 Figs. 6 and 7 show the effect of the temperature of geothermal water at the inlet of
 325 evaporator variation on the system performance. As the temperature of geothermal
 326 water at the inlet of evaporator increases from 30°C to 40°C, the heat capacity increases
 327 from 16.36MW to 18.98MW and the COP increases from 1.82 to 1.86. This is because
 328 the increase of temperature of geothermal water at the inlet of evaporator results in the
 329 increase of solution temperature. Then complete evaporation of working fluid takes
 330 place at higher temperature in evaporator and the ability of the solution in the absorber
 331 to absorb working fluid vapor is enhanced which causes the increase of heat transfer
 332 rate of evaporator. According to the Eq(7), COP is proportional to the heat transfer rate
 333 of evaporator. As is shown in the Fig. 7, the exergy efficiency and unit cost of exergy
 334 for heating decrease with the increase of temperature of geothermal water into the
 335 evaporator.

336

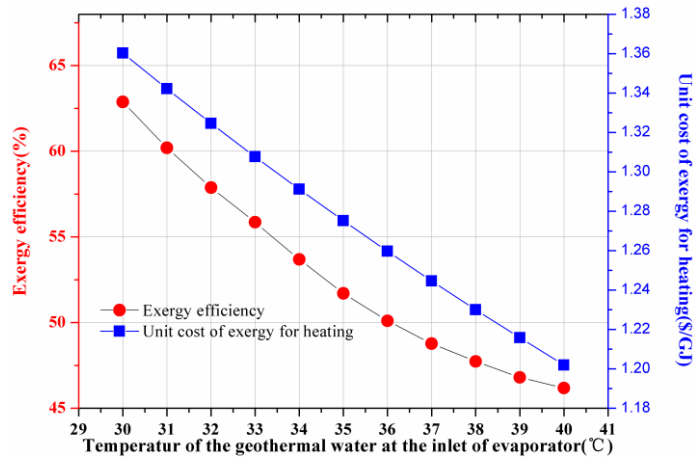


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339 Fig. 6 Variation of heat capacity and COP with the temperature of geothermal water at the
 340 inlet of evaporator

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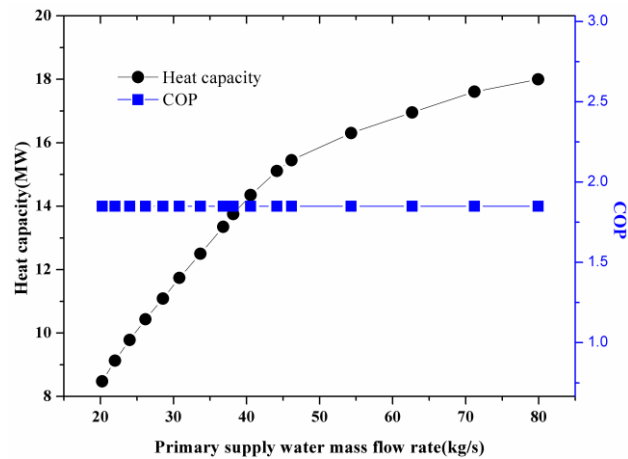
344 Fig. 7 Variation of exergy efficiency and unit cost of exergy for heating with the temperature
 345 of geothermal water at the inlet of evaporator

346

347 **5.2. Optimization analysis**

348 Fig. 8 shows the effect of the primary supply water mass flow rate variation on the
 349 system performance. It can be seen that the heat capacity indicates increase as the
 350 primary supply water mass flow rate increases. It is obvious that the system has a
 351 maximum heat capacity of 18MW and a constant COP of 1.85. Operating conditions of
 352 the system remain unchanged with the mass flow rate of the primary supply water.
 353 According to the Eq (7), COP remains unchanged. In order to find the optimal mass
 354 flow rate of the primary supply water, the multi-objective optimization for the CDHS
 355 was conducted.

356



357

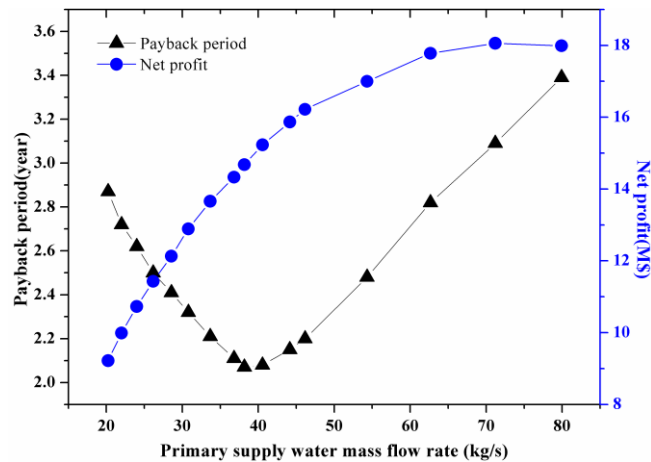
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359 Fig. 8 Variation of heat capacity and COP with the mass flow rate of the primary supply
360 water

361

362 Fig. 9 indicates that the net profit increases with the increase of mass flow rate of
363 the primary supply water. However, the payback period shows a trend of falling first
364 and then rising. As the mass flow rate of the primary supply water increases from
365 20.24kg/s to 79.92kg/s, the net profit increases from 9.22M\$ to 17.99M\$. The system
366 has a minimum payback period of 2.07 year at the mass flow rate of the primary supply
367 water of 38.17kg/s, where the net profit is 14.68\$. The maximum net profit is
368 19.99M\$ at the mass flow rate of the primary supply water is 79.92kg/s, where the
369 payback period is 3.39year.

370



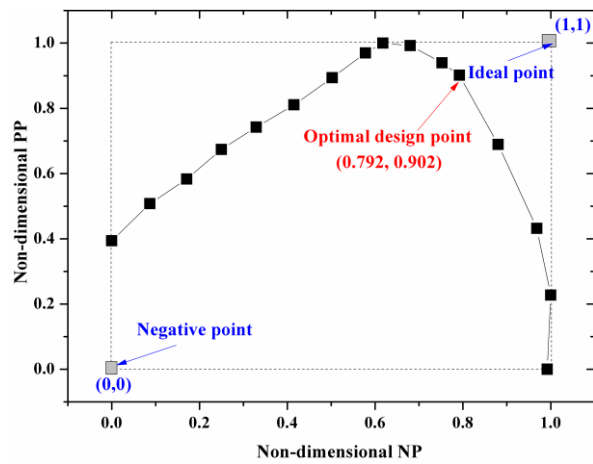
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373 Fig. 9 Variation of payback period and net profit with the mass flow rate of the primary
374 supply water

375 Fig. 10 shows the relation of non-dimensional net profit and payback period. The
376 TOPSOS decision making method is used to find the optimal design point. It can be
377 seen that the payback period and net profit at optimal point are 2.2year and 16.22M\$,
378 where the mass flow rate of the primary supply water is 46.16kg/s and COP is 1.85.
379 Table 4 exhibits the thermodynamic values of the system under the optimal condition.

380



381

382

383

Fig. 10 Non-dimensional payback period and net profit

384

385

Table 4 Thermodynamic properties of the system under the optimal condition.

Parameters	Value
Generator heat rate(kW)	2326.4
Evaporator heat rate(kW)	1987.7
Condenser heat rate(kW)	2122.6
Absorber heat rate(kW)	2192.2
Heat exchanger II heat rate(kW)	8797.4
Heat exchanger I heat rate(kW)	2539.8
Working fluid mass flow rate(kg/s)	0.856
Exergy efficiency	53%
COP	1.85
Total heating capacity(kW)	15652.2
Net profit(M\$)	16.22
Payback period(year)	2.2

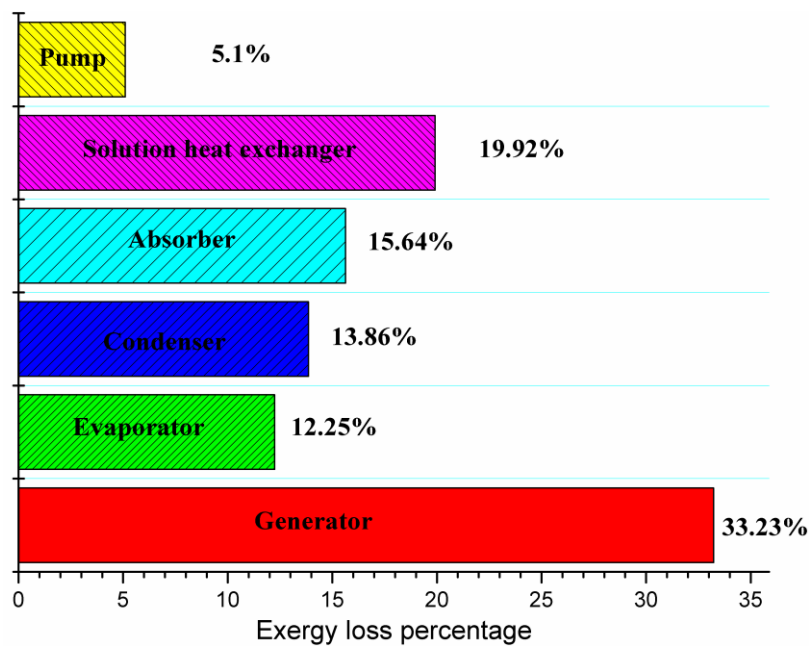
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388 **5.3. Exergy loss analysis**

389 The exergy destruction analysis of each component are carried out and shown in
390 Fig. 11 under the optimal condition where the volume of primary supply water is
391 46.16kg/s. It is necessary to analyze which component has the largest exergy loss. It
392 can be observed that the generator accounts for the biggest amount of the exergy loss
393 which is more than 33%, followed by the solution heat exchanger which is nearly 20%.
394 The evaporator, absorber and condenser have almost the same exergy loss, and the
395 exergy loss of the pump is minimal. This is basically due to that the temperature
396 difference between the primary supply water and working fluid in generator is bigger.
397 Therefore, it is generator that has the biggest potential to decrease the exergy loss of
398 the system and the exergy loss of the generator can be reduced by lowering the
399 temperature difference.

400



401

402

403 Fig. 11 Exergy destruction percentage of each component under the optimal condition

404

405 **6. Conclusion**

406 In this paper, the performance of a novel composite district heating substation
407 using absorption heat pump based on energy, exergy and economic analysis was carried
408 out. To determine the maximum value of net profit and the minimum value of the
409 payback period, the mass flow rate of the primary supply water were considered. The
410 main conclusions drawn from this paper were summarized as follows:

411 (1) The heat capacity of the system was mainly influenced by the geothermal water
412 mass flow rate and the temperature of geothermal water at the inlet of evaporator, but
413 they had no effect on the COP. The heat capacity was almost linearly increased with the
414 increase of geothermal water mass flow rate and the temperature of geothermal water
415 at the inlet of evaporator, but the exergy efficiency and unit cost of exergy for heating
416 were opposite.

417 (2) The optimal primary supply water mass flow rate of the proposed system,
418 based on the TOPSIS decision-making method, was 46.16kg/s for the system under the
419 design conditions.

420 (3) The system had a net profit of 16.22 M\$ in the life time and a small payback
421 period of about 2.2 years under the optimal condition. In addition, COP and exergy
422 efficiency of the system were 1.85 and 59.81%, respectively. The results demonstrated
423 the feasibility and economy of the system.

424 (4) The generator accounted for the largest share in the total exergy loss, which
425 was due to the large temperature difference between the primary supply water and
426 working fluid.

427

428 **Declarations of interest**

429 None

430 **Acknowledgements**

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