- 1 High-temperature structural phase transitions in neighborite; a high-resolution neutron
- 2 powder diffraction investigation
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ABSTRACT

- 15 The nature of the apparently continuous structural phase transition at 1049 K in the
- 16 perovskite-structured, MgSiO₃ isomorph, neighborite (NaMgF₃), from the orthorhombic
- 17 (*Pbnm*) hettotype phase to the cubic ($Pm\bar{3}m$) aristotype structure, has been re-investigated
- using high resolution, time-of-flight neutron powder diffraction. Using data collected at 1 K
- 19 intervals close to the nominal phase transition temperature, the temperature-dependence of
- 20 the intensities of superlattice reflections at the M point $\left(\frac{2\pi}{a}\left[\frac{1}{2},\frac{1}{2},0\right]\right)$ and the R point (

21	$\frac{2\pi}{a}\left[\frac{1}{2},\frac{1}{2},\frac{1}{2}\right]$) of the pseudocubic Brillouin zone indicate the existence of a new
22	intermediate tetragonal phase in space group $P4/mbm$, with a narrow phase field extending
23	from ~1046.5 K to ~1048.5 K, at ambient pressure. Group theoretical analysis shows that the
24	structural transitions identified in this study, $Pbnm - P4/mbm$, and $P4/mbm - Pm\overline{3}m$, are
25	permitted to be second order. The observation of the tetragonal phase resolves the
26	longstanding issue of why the high temperature phase transition, previously identified as
27	$Pbnm - Pm\bar{3}m$, and which would be expected to be first order under Landau theory, is in fact
28	found to be continuous. Analysis of the pseudocubic shear strain shows it to vary with a
29	critical exponent of 0.5 implying that the phase transition from <i>Pbnm</i> to <i>P4/mbm</i> is tricritical
30	in character. The large librational modes that exist in the MgF_6 octahedron at high
31	temperature, and the use of Gaussian probability density functions to describe atomic
32	displacements, results in apparent bond shortening in the Mg – F distances, making mode
33	amplitude determination an unreliable method for determination of the critical exponent from
34	internal coordinates. Crystal structures are reported for the three phases of $NaMgF_3$ at 1033 K
35	(<i>Pbnm</i>), 1047 K (<i>P4/mbm</i>), and 1049 K (<i>Pm</i> 3 <i>m</i>).

Keywords: Neighborite; NaMgF₃; Phase transitions; Neutron diffraction; MgSiO₃ perovskite;
 MgSiO₃ post-perovskite

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39 Introduction

Knowledge of the physical properties of the perovskite and post-perovskite phases of MgSiO₃
is crucial to our understanding of the Earth's lower mantle. In particular, the D'' region on the
mantle side of the core-mantle-boundary, which forms the mantle's lower thermal boundary
layer, is associated with the transition in MgSiO₃ from the perovskite to post-perovskite

44 structure. For MgSiO₃, however, many of these properties are difficult to measure accurately 45 in the perovskite phase and most of them cannot be measured at the extreme pressures of stabilisation of post-perovskite (120 GPa). The best chance of constraining them is through a 46 47 combination of measurements on low-pressure analogue materials (with the same crystal 48 structure but a different chemical composition) and *ab initio* simulations of both the analogue 49 and natural systems. Experimentally, fluoride analogue phases can be considered to have 50 advantages over oxides as their high-pressure phases become stable at much lower pressures. 51 Thus, for example, ABF₃ perovskites have been used in investigations of the viscosity and 52 electrical conductivity of the mantle (Poirier et al., 1983; Watson et al., 1995; Li and Weidner, 2012). The seismic velocity changes in the D" region have been interpreted via 53 54 measurements of the equations of state of perovskite and post-perovskite NaMgF₃ (Hustoft et al., 2008) and very recently, studies of the perovskite to post-perovskite transition in single 55 crystals of NaNiF₃ have greatly improved understanding of the possible development of 56 texture and the resulting seismic anisotropy in D" (Dobson et al., 2011; Dobson et al., 2013). 57 58 The relatively low transition pressures found in the ABF₃ compounds also imply that their 59 ultrahigh pressure polymorphs, of relevance to super-Earths, might be investigated 60 experimentally as well as by computer simulations (Umemoto and Wentzcovitch, 2006; Grocholski et al., 2010). 61

NaMgF₃, found naturally as the mineral neighborite (Chao *et al.*, 1961), provides arguably the closest fluoride analogue to MgSiO₃ as the two compounds are isoelectronic, have similar ionic masses, and crystallise in both the orthorhombic (*Pbnm*) perovskite and post-perovskite structures (see e.g. Umemoto *et al.*, 2006). Preliminary high-temperature crystallographic investigations by Chao *et al.* (1961) indicated that neighborite undergoes at least one structural phase transition at a temperature of ~1173(25) K. A tetragonal, or pseudotetragonal phase was inferred to exist between 1033 K and 1173 K, however characterisation of this

69 phase was hampered by the limited real space resolution of the diffractometer employed. 70 More recently, Zhao et al. (1993a,b) re-investigated the structural behaviour of neighborite at high temperature using higher-resolution synchrotron powder diffraction; they found no 71 72 evidence for an intermediate phase, and a continuous transition to the aristotype cubic phase. This result is surprising as the $Pbnm - Pm\bar{3}m$ transition would be expected to be first order 73 under conventional Landau theory (Tolédano and Tolédano, 1987), rather than continuous 74 75 (unless, by accident, the critical temperatures of the two phonon modes of different symmetry 76 are exactly, or very close to, the same temperature). Critical behaviour was inferred from an 77 analysis of the magnitudes of the two independent order parameters (the in-phase, and anti-78 phase octahedral tilts; Glazer, 1972, 1975) with the conclusion that both modes condensed 79 simultaneously with an identical critical exponent of 0.25, i.e. continuous tricritical behaviour 80 in both order parameters. The crystallographic analysis performed, and the conclusions drawn from this study will be critiqued in detail in the Results and Discussion section of this 81 82 communication.

The precise characterisation of the crystal structures of polycrystalline, zone-boundary tilted, 83 84 perovskite-structured materials requires the ability both to determine the metric of the 85 pseudocubic subcell of the hettotype phase and also to observe key diagnostic superlattice 86 reflections at special points on the surface of the pseudocubic Brillouin zone (R, M, and X 87 points; Glazer, 1975). For the former, data must be collected at the highest possible real space 88 resolution, whereas for the latter, as the diagnostic superlattice reflections arise principally 89 from the displacements of the anions, neutron powder diffraction offers significant 90 advantages over X-ray diffraction, especially when the octahedral and the quasi-dodecahedral 91 sites contain heavy metal cations and the anion species are light. We present here a reinvestigation of the high-temperature phase transitions in the perovskite form of NaMgF₃. In 92 93 this study, we have exploited the intrinsic advantages of neutron diffraction, using a powder

94 diffractometer with a high real space resolution ($\Delta d/d = 4 \times 10^{-4}$), which, to first order, is 95 independent of the magnitude of the scattering vector.

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97 Experimental

The powder sample for neutron diffraction was synthesised by solid state reaction of a 98 99 stoichiometric mixture of NaF and MgF₂ (nominal purity 99.99%) at 750°C, for 15 h, in air (Street et al., 1997). The ground and sieved powder was placed in an 11 mm diameter, 100 101 cylindrical vanadium sample can and located in a vacuum furnace equipped with vanadium 102 heat shields and element. Time-of-flight neutron powder diffraction data were collected in the range 30 - 130 ms, using the high resolution backscattering detectors of the powder 103 104 diffractometer HRPD, at the ISIS spallation neutron source (Ibberson et al., 1992). Data were 105 collected at the following temperatures: at 823 K, then in 10 K steps to 993 K, at 999 K, then 106 in 10 K steps to 1029 K, in 2 K intervals to 1041 K, at 1044 K, then in 1 K steps to 1054 K. 107 All data collections were made for 30 µAh (approximately 50 minutes) with a 10 minute 108 thermal equilibration time once the set point temperature was achieved. Above 1054 K, data 109 were collected in 5 K intervals to 1224 K, for 6 µAh, with a 10 minute equilibration time. 110 Thermal stability was ± 0.1 K for all temperatures.

111 The raw data were focussed, normalised to the incident flux distribution using an upstream

beam monitor, and corrected for wavelength-dependent detector efficiency to produce a

113 dataset in the time-of-flight range 32 - 120 ms, equivalent to 0.64 - 2.4 Å in the

backscattering bank (maximum Q resolution 9.82 Å⁻¹), and suitable for profile refinement

using the GSAS suite of programs (Larson and Von Dreele, 1986). Instrumental calibrations

116 were made using SRM640b standard silicon powder to determine the time-of-flight (ms) to d-

spacing (Å) conversion, and a ceramic CeO₂ pellet to characterise the intrinsic line shape

from the methane moderator. Data were analysed using the Rietveld method, as implemented within GSAS, using a pseudo-Voigt convoluted with back-to-back exponentials for the line shape. An initial model for the 999 K refinement was derived from the results of Zhao *et al.* (1993a); once this refinement had converged, the result was used as a seed for the next temperature, with the whole process carried out iteratively to the highest temperature measured. Atomic displacement parameters were refined as isotropic for both cations, and as anisotropic for the anions.

125 Discussion of earlier high-temperature crystallographic results

The structural behaviour of NaMgF₃ has been investigated by X-ray diffraction as a function 126 of composition, with substitution of K for Na (Martin et al., 2005), as a function of pressure 127 128 at room temperature (Liu et al., 2005; Martin et al., 2006a,b), and at simultaneous high pressure and temperature (Chen et al., 2005). At atmospheric pressure, X-ray diffraction at 129 130 high temperature has been used to study both the orthorhombic and cubic phases, refining 131 both the average (Zhao et al., 1993a; Zhao et al., 1993b) and the local (Martin et al., 2007) 132 crystal structures, and neutron powder diffraction has been combined with molecular dynamics to investigate thermal motions in the high-temperature cubic phase (Street *et al.*, 133 134 1997). The structural crystallography of the orthorhombic phase has been studied at low temperatures using powder neutron diffraction, at medium resolution, by Mitchell et al. 135 (2007), and at significantly higher resolution by Knight (2014) who found negative linear 136 thermal expansion for the *b* axis (*Pbnm* setting) between 20 K - 90 K, and evidence for 137 quadratic coupling of the in-phase tilt to the anti-phase tilt. Of particular relevance to this 138 139 current communication are the results and conclusions drawn by Zhao et al. (1993a,b) in two papers published in Physics of the Earth and Planetary Interiors, which for convenience in 140 this article, we name PEPI I (Zhao et al., 1993a), and PEPI II (Zhao et al., 1993b). 141

142 PEPI I describes the evolution of the crystal structure of neighborite between 293 K and 1173 K using both laboratory and synchrotron source X-ray powder diffraction. The 143 crystallographic results from this paper were subsequently used to infer the critical behaviour 144 145 of the orthorhombic – cubic phase transition, as detailed in PEPI II. Despite the apparently comprehensive nature of this study, Knight (2014) has shown it to be substantially flawed, 146 containing inconsistent crystallographic results, poorly determined order parameter 147 148 magnitudes, incorrect application of Landau theory, and a mathematical error that leads to the wrong conclusion concerning the critical exponent associated with the phase transition. 149

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In PEPI I, laboratory source data were only collected to a O resolution of 5.23 Å⁻¹ which is 151 152 insufficient to derive precise structural parameters from a pseudosymmetric crystal structure (see Knight et al. (2004) for discussion of an example of an incorrect space group and 153 implausible crystal structure, at ambient temperature and pressure, for KCaF₃ perovskite 154 155 derived from data collected to a similar resolution). The derived structural parameters (bond 156 lengths and angles) shown in Table 3 of PEPI I are inconsistent with estimated standard deviations (esds) of the structural parameters listed in Table 2, and bond length corrections 157 158 for thermal motion are based on a model that assumes uncorrelated atomic displacements, which is inappropriate for a 3-dimensional, fully-connected, polyhedral network-structure. 159 160 The order parameters for zone-boundary tilted perovskite phases are proportional to the amplitudes of displacements that transform as the basis vectors of the irreducible 161 representations (irreps) R_4^+ (anti-phase tilt, Glazer, 1972), and M_3^+ (in-phase tilt, Glazer, 1972) 162 163 of the cubic Brillouin zone (Cowley, 1964). The expressions for the two tilts used in PEPI I are only approximations to these amplitudes, and are derived assuming that all octahedral 164 distortive modes are absent, which will not be the case. The correct method of analysis of the 165

166 atomic displacements is given by a full mode decomposition of the crystal structure (Perez-Mato *et al.*, 2010), which in the case of a perovskite crystal structure (ABC₃) in space group 167 Pnma, contains 2 mode amplitudes associated with the A site, and 5 associated with the 168 169 anions bonded to the octahedral B site (Cochran and Zia, 1968; Knight 2009, 2011, 2014). 170 In PEPI II, the magnitudes of the tilt angles, as determined by the methods detailed in PEPI I, were used in a simplified Landau free energy expression to determine the critical behaviour 171 associated with the orthorhombic - cubic phase transition. It was assumed, without 172 justification, that quartic terms could be ignored, leading to the expressions below 173

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$$\frac{\partial G}{\partial q_4} = 0 = A_1 (T - T_{c1}) q_4 + C_1 q_4^5 + \lambda q_4 q_2^2$$
$$\frac{\partial G}{\partial q_2} = 0 = A_2 (T - T_{c2}) q_2 + C_2 q_2^5 + \lambda q_4^2 q_2$$

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175 where q_2 is the magnitude of the in-phase tilt, q_4 is the magnitude of the anti-phase tilt, A_1 , A₂, C₁, C₂ are Landau coefficients, and λ is a coupling coefficient. These equations are 176 simultaneous, non-linear and coupled, and hence only have closed solutions for particular 177 178 values of the Landau and coupling coefficients. However, the solutions to q₂ and q₄ given in 179 PEPI II have been derived from an assumption that these equations are *independent*, and based on this mathematical error, they have been found to have identical critical exponents of 180 181 0.25. This conclusion that the in-phase and anti-phase tilts exhibit simultaneous tricritical 182 behaviour violates Landau theory for continuous phase transitions as the two order parameters have different symmetries (Tolédano and Tolédano, 1987), and hence the 183 184 conclusions for critical behaviour in neighborite drawn in PEPI II are formally invalid.

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187 **Results**

188 In the present study, the normalised neutron powder diffraction patterns were examined for the presence or absence of key superlattice reflections at the R, M, and X points of the 189 pseudocubic Brillouin zone. By 999 K, all X point reflections were too weak for reliable 190 191 characterisation and the analysis presented here is, therefore, based on the stronger R and M point superlattice peaks. It should be noted, however, that, since the X point is only a 192 reciprocal lattice point in the orthorhombic phase by virtue of the simultaneous presence of 193 194 the R and M points, its absence does not affect the conclusions drawn. 195 The temperature dependences of the two strongest groups of diagnostic superlattice 196 reflections in the temperature interval 1031 K – 1050 K, indexed in the pseudocubic setting as 3/2 1/2 1 (M point, $d \sim 2.1$ Å), and 3/2 3/2 1/2 (R point, $d \sim 1.8$ Å), are shown in Figure 1. 197 The disappearance of the R point reflection, which is associated with the anti-phase 198 octahedral tilt, and cation and anion displacements that transform as the irreps R_4^+ and R_5^+ 199 respectively (Cochran and Zia, 1968; Knight, 2009, 2011), can be seen to have occurred 200 between 1046 K and 1047 K. At 1047 K, the M point superlattice reflection, which is 201 associated with the in-phase tilt and anion displacements that transform as the irreps M_3^+ and 202 M⁺₂ respectively (Cochran and Zia, 1968; Knight, 2009, 2011), still has appreciable intensity, 203 204 but can be seen from the Figure to vanish at a slightly higher temperature, in the interval 1048 K – 1049 K. Analysis of the full diffraction pattern measured at 1047 K shows the absence of 205 all reflections associated with the R and X points of the pseudocubic Brillouin zone, whereas 206 the diffraction pattern collected at 1049 K is consistent with the aristotype structure, and 207 exhibits no superlattice reflections at all. From these results it is clear that the apparently 208 continuous phase transition from Pbnm to $Pm\overline{3}m$ in fact proceeds by an intermediate phase 209 210 that exists in the very small range of temperature between ~1046.5 K and ~1048.5 K. Of the

211 four space groups consistent with the perovskite structure and in-phase tilting alone, *i.e.* exhibiting superlattice reflections only at the M point of the pseudocubic Brillouin zone 212 (Howard and Stokes, 1998), structural phase transitions from *Pbnm* to space groups *Immm*, 213 I4/mmm and $Im\bar{3}$ are first order, with a second-order transition being permitted only to 214 P4/mbm (Stokes and Hatch, 1988). Structural phase transitions to the aristotype phase are 215 first order for the space groups Immm and I4/mmm as they violate the Landau condition, but 216 permitted to be continuous for space groups $Im\bar{3}$ and P4/mbm (Stokes and Hatch, 1988). 217 From these results, together with the observation of a lambda anomaly in the heat capacity 218 219 which excludes the possibility of a first-order transition (Torpor at al., 1997), we conclude 220 that the intermediate phase must be tetragonal, in space group P4/mbm. To test this 221 conclusion, the data collected at 1047 K have been fitted using the Rietveld method. In this refinement, the starting value for the x coordinate of the fluorine anion on the Wyckoff site 222 4g was derived from the magnitude of the in-phase tilt determined from the crystal structure 223 of the orthorhombic phase at 1044 K (Knight, 2009, 2011). Convergence from this starting 224 225 point was rapid, with Figure 2 showing the excellent fit to the data, indicating the correctness 226 of the model; on the same figure, we make comparisons with data analysed in the orthorhombic phase at 1033 K, and the cubic phase at 1049 K. Fractional coordinates, atomic 227 228 displacement parameters, unit cell dimensions, and agreement factors for the three phases are reported in Table 1. 229

In Figure 3 we show the temperature evolution of the pseudocubic subcell in the orthorhombic phase, and the cubic lattice parameter above the second phase transition temperature. The pseudocubic subcell, subscripted *p*, is related to the orthorhombic unit cell via the transformation $-1/2 \ 1/2 \ 0 \ / \ 1/2 \ 1/2 \ 0 \ / \ 0 \ 0 \ 1/2$, and can be best described using the monoclinic metric $a_p = b_p \neq c_p$, $\alpha_p = \beta_p = 90^\circ$, $\gamma_p \neq 90^\circ$ since this description, with $\gamma_p \neq 90^\circ$, most clearly reveals that the 4-fold rotation in the cubic phase is reduced to a 2-fold rotation 236 in the orthorhombic phase (Knight, 2011). The lattice parameters are continuous through the phase transition, and hence there is no volume discontinuity at the phase transition 237 temperature, although there is a clear change in gradient of the linear thermal expansion 238 239 coefficient. Critical exponents for structural phase transitions are generally determined from the temperature-dependence of superlattice reflections measured on single crystals. However, 240 in the case of powder diffraction data where these weak reflections are more difficult to 241 242 measure, the temperature variation of the spontaneous strain and internal coordinates is normally the method of choice (Carpenter, 2007). The cosine of the pseudocubic inter-axial 243 244 angle γ_p is essentially identical to the spontaneous strain component e_6 (e_4 in the Cartesian basis of Carpenter, 2007) which varies as the square of the order parameter q4, which in turn 245 is related to the magnitude of the anti-phase tilt (Carpenter, 2007). The fit to the γ_p data in 246 Figure 3 is based on simple critical behaviour $(90-\gamma_n) = A(1-T/T_c)^{\beta}$ where A is a constant, T_c 247 248 is the critical temperature, and β is a critical exponent. Within error, $\beta = 0.5$, and hence the critical exponent for the order parameter of the Pbnm to P4/mbm phase transition is 0.25, and 249 250 the transition is tricritical in nature. Confirmation of the critical exponent can usually be achieved by determining the temperature-dependence of the order parameters directly from 251 252 the fractional coordinates and the unit-cell parameters. In the case of perovskite-structured 253 materials, as we have pointed out earlier, the order parameters should be determined through 254 mode decomposition. Unfortunately, in the case of neighborite, the MgF₆ octahedron is subject to large amplitude librations at temperatures close to the first, and second phase 255 256 transitions (Street et al., 1997). As a result, since the refinement of the atomic displacement parameters are based on Gaussian probability density functions, the true location of the 257 atomic positions, as determined from Bragg scattering, is severely compromised, with an 258 apparent shortening of the Mg-F bonds being observed (Zhao et al., 1993a, b). As a 259 consequence of this, mode decomposition cannot be used to derive a reliable confirmatory 260

261 critical exponent. However, at temperatures below 440 K, where librational motion can be safely ignored, Knight (2014) has indeed found that the order parameter q₄ varies with critical 262 exponent 0.25, and furthermore, that the order parameter q_2 is quadratically coupled to q_4 . At 263 264 low temperatures, the behaviour of neighborite is therefore dominated by the anti-phase tilt and the thermal evolution is such that only the R point behaves in a critical manner. Whether 265 this behaviour extends to the Pbnm - P4/mbm phase transition will require a detailed single 266 crystal study of the temperature-dependence of the superlattice reflection intensities; 267 however, the temperature-dependence of e_6 , which depends on q_4^2 in the numerator 268 269 (McKnight et al., 2009), strongly suggests that this is so. The unusual development of the other spontaneous strains at the gamma point around 700 K that has been discussed by 270 271 Carpenter *et al.* (1998) may arise from changes in the coupling between q_2 and q_4 which are 272 necessary for the evolution of the tetragonal phase, in which only q_2 is the active order 273 parameter.

274 In the case of the first structural phase transition, the narrowness of the tetragonal phase field 275 precludes further analysis of the data presented here and also explains the fact that the 276 tetragonal phase of NaMgF₃ has not been reported previously. In each of the two X-ray diffraction experiments reported by Zhao et al. (1993a), data were collected isothermally but 277 in steps that were too large (~20 K in the vicinity of the transitions) to resolve the two phase 278 279 transitions, whereas, in the more recent study of the local structure of NaMgF₃ by Martin et al. (2007), data were collected as the sample temperature was increased continuously, so the 280 281 diffraction pattern recorded at each datum was integrated over a 16 K range.

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285 Implications for terrestrial mineralogy and geophysics

286 It might be argued that the subtleties of the ferroelastic phase transformations in perovskites are of limited relevance to our understanding of the Earth, as it is now generally accepted that 287 288 the major perovskite phase, $MgSiO_3$, will remain orthorhombic throughout the lower mantle from its formation until the transition to the post-perovskite structure that occurs close to the 289 core-mantle boundary (Murakami et al., 2004; Oganov and Ono, 2004). This statement, 290 291 however, is not true for $CaSiO_3$ perovskite, thought to be the third most abundant lower 292 mantle mineral, and which also exists as a major component in subducted mid-ocean ridge basalt. Although CaSiO₃ may, or may not, be cubic under lower mantle conditions (see e.g. 293 294 Noguchi et al., 2013) it is definitely known to exhibit lower symmetry under conditions which many workers have considered to be geologically relevant. There is, however, 295 296 currently no consensus as to the number of high-temperature phase transitions (one or two), 297 the transition temperatures, or the space groups of the lower-symmetry phases of this mineral. At 40 GPa, for example, the transitions have been reported in a range from 550 K (e.g. 298 299 Komabayashi et al., 2007) to greater than 4000 K (Li et al., 2006). Space groups suggested 300 by computer simulation studies include I4/mcm and Imma (Adams and Oganov, 2006) or Pbnm and a tetragonal phase (Li et al., 2006; probably I4/mcm or P4/mbm from the 301 discussion in their paper). X-ray diffraction experiments by Uchida et al. (2009) gave data 302 303 that they considered consistent with *Cmcm* or possibly *Pbnm*, whereas other workers have 304 been more cautious and reported only that their diffraction patterns indicated a tetragonal crystal system (Komabayashi et al., 2007). Understanding ferroelasticity in these systems is 305 306 crucial to our understanding of the Earth. Not only will the phase transitions themselves cause potential seismic anomalies (see e.g. Komabayashi et al., 2007, for discussion of this 307 308 point with relevance to CaSiO₃), but also, even if a transition to a higher-symmetry structure 309 does not occur, ferroelastic twinning induced by seismic waves may cause a large seismic

- 310 attenuation, as determined in a recent study by Li and Weidner (2012) once again using
- 311 NaMgF₃ as an analogue for the silicate perovskites.

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318 **References**

319	Adams DJ, Oganov AR (2006) Ab initio molecular dynamics study of CaSiO ₃ perovskite at

320 P-T conditions of Earth's lower mantle. Phys Rev B 73:184106

- 321 Carpenter MA (2007) Elastic anomalies accompanying phase transitions in (Ca,Sr)TiO₃
- 322 perovskites: Part I. Landau theory and a calibration for SrTiO₃. Am Mineral 92:309–327
- 323 Carpenter MA, Salje EKH, Graeme-Barber A (1998) Spontaneous strain as a determinant of
- 324 thermodynamic properties for phase transitions in minerals. Eur J Mineral 10:619-691
- 325 Chen J, Liu H, Martin CD, Parise JB, Weidner DJ (2005) Crystal chemistry of NaMgF₃
- perovskite at high pressure and temperature. Am Mineral 90:1534-1539
- 327 Chao ECT, Evans HT, Skinner BJ, Milton C (1961) Neighborite, NaMgF₃, a new mineral
- from the Green River Formation, South Ouray, Utah. Am Mineral 46: 379-393
- Cochran W, Zia A (1968) Structure and dynamics of perovskite-type crystals. Phys Status
 Solidi 25:273–283
- Cowley RA (1964) Lattice dynamics and phase transitions of strontium titanate. Phys Rev
 134:A981–A997
- 333 Dobson DP, Hunt SA, Lindsay-Scott A, Wood IG (2011) Towards better analogues for
- 334 MgSiO₃ post-perovskite: NaCoF₃ and NaNiF₃, two new recoverable fluoride post-
- 335 perovskites. Phys Earth Planet Inter 189:171-175
- 336 Dobson DP, Miyajima N, Nestola F, Alvaro M, Casati N, Liebske C, Wood IG, Walker AM
- 337 (2013) Strong inheritance of texture between perovskite and post-perovskite in the D" layer.
- 338 Nature Geoscience 6:575-578

- Glazer AM (1972) The classification of tilted octahedra in perovskites. Acta Crystallogr B
 28:3384–3392
- Glazer AM (1975) Simple ways of determining perovskite structures. Acta Crystallogr A
 342 31:756–762
- 343 Grocholski B, Shim S-H, Prakapenka V.B. (2010) Stability of the MgSiO₃ analog NaMgF₃
- and its implication for mantle structure in super-Earths. Geophys Res Lett 37:L14204
- 345 Howard CJ, Stokes HT (1998) Group-theoretical analysis of octahedral tilting in perovskites.
- 346Acta Crystallogr B 54:782–789
- 347 Hustoft J, Catalli K, Shim S-H, Kubo A, Prakapenka VB, Kunz M (2008) Equation of state of
- 348 NaMgF₃ postperovskite: Implications for the seismic velocity changes in the D" region.
- 349 Geophys Res Lett 35:L10309
- 350 Ibberson R M, David WIF, Knight K S (1992) Report RAL-92-031. Rutherford Appleton
- 351 Laboratory, Didcot, Oxfordshire, England.
- 352 Knight KS (2009) Parameterization of the crystal structures of centrosymmetric zone-
- boundary-tilted perovskites: An analysis in terms of symmetry-adapted basis-vectors of the
- 354 cubic aristotype phase. Can Mineral 47:381–400
- 355 Knight KS (2011) Centrosymmetric perovskite crystal structures with space group Pbnm:
- 356 Crystallographic parameterization of KCaF₃ between 100 and 400 K in terms of the
- amplitudes of symmetry-adapted basis-vectors of the cubic aristotype phase. Can Mineral:
- 358 49:793-808
- 359 Knight KS (2014) A high resolution neutron diffraction study of the crystal structure of
- neighborite (NaMgF₃) between 9 K and 440 K. Am Mineral 99:824-838

- 361 Knight KS, Darlington CNW, Wood IG (2004) The crystal structure of KCaF₃ at 4.2 and 300
- 362 K: A re-evaluation using high-resolution powder neutron diffraction. Powder Diffr 20:7–13
- 363 Komabayashi T, Hirose K, Sata N, Ohishi Y, Dubrovinsky LS (2007) Phase transition in
- 364 CaSiO₃ perovskite. Earth Planet Sci Lett 260:564–569
- 365 Larson AC, Von Dreele RB (1986) GSAS, general structure analysis system. Los Alamos
- 366 National Laboratory Report No. LAUR 86-748
- Li L, Weidner DJ, Brodholt J, Alfè D, Price GD, Caracas R, Wentzcovitch R (2006) Phase
- 368 stability of CaSiO₃ perovskite at high pressure and temperature: insights from ab initio
- 369 molecular dynamics. Phys Earth Planet Inter 155:260–268
- 370 Li L, Weidner DJ (2012) Anelasticity and transient creep in NaMgF₃ perovskite at high
- 371 pressure. Phys Earth Planet Inter 194-195:98–106
- Liu H-Z, Chen J, Hu J, Martin CD, Weidner DJ, Häusermann D, Mao H-K (2005) Octahedral
- tilting evolution and phase transition in orthorhombic NaMgF₃ perovskite under pressure.
- 374 Geophys Res Lett 32:L04304
- 375 Martin CD, Chaudhuri S, Grey CP, Parise JB (2005) Effect of A-site cation radius on
- ordering of BX₆ octahedra in (K,Na)MgF₃ perovskite. Am Mineral 90:1522-1533
- 377 Martin CD, Chupas PJ, Chapman KW, Parise JB (2007) Local versus average structure: a
- 378 study of neighborite (NaMgF₃) utilizing the pair distribution function method for structure
- determination. J Appl Crystallogr 40:441-448
- 380 Martin CD, Crichton WA, Liu HZ, Prakapenka V, Chen JH, Parise JB (2006a) Phase
- 381 transitions and compressibility of NaMgF₃ (Neighborite) in perovskite and post-perovskite-
- related structures. Geophys Res Lett 33:L11305

- 383 Martin CD, Crichton WA, Liu H, Prakapenka V, Chen J, Parise JB (2006b) Rietveld structure
- 384 refinement of perovskite and post-perovskite phases of NaMgF₃ (Neighborite) at high
- 385 pressures. Am Mineral 91: 1703–1706
- 386 McKnight REA, Howard CJ, Carpenter MA (2009) Elastic anomalies with transformation
- 387 sequencies in perovskites: I. Strontium zirconate, SrZrO3. J Phys-Condens Mat 21:015901.
- 388 Mitchell RH, Alexander M, Cranswick LMD, Swainson IP (2007) A powder neutron
- diffraction study of the fluoroperovskite NaMgF₃ (neighborite) from 300 K to 3.6 K. Phys
- 390 Chem Minerals 34:507–712
- 391 Murakami M, Hirose K, Kawamora K, Sata N, Ohishi Y (2004) Post perovskite phase
- transition in MgSiO₃. Science 304:855–858
- 393 Noguchi N, Komabayashi T, Hirose K, Ohishi Y (2013) High-temperature compression
- 394 experiments of CaSiO₃ perovskite to lowermost mantle conditions and its thermal equation of
- 395 state. Phys Chem Minerals 40:81-91
- 396 Oganov AR, Ono S (2004) Theoretical and experimental evidence for a post-perovskite phase
- 397 of MgSiO₃ in Earth's D'' layer. Nature 430:445-448
- 398 Perez-Mato JM, Orobengoa D, Aroyo MI (2010) Mode crystallography of distorted
- 399 structures. Acta Crystallogr A, 66:558-590
- 400 Poirier JP, Peyronneau J, Gesland JY, Brebec G (1983) Viscosity and conductivity of the
- 401 lower mantle: an experimental study on a MgSiO₃ perovskite analogue, KZnF₃. Phys Earth
- 402 Planet Inter 32:273-287
- 403 Stokes HT, Hatch DM (1988) Isotropy subgroups of the 230 crystallographic space groups.
- 404 World Scientific Press, Singapore, 603 pp

- Street JN, Wood IG, Knight KS, Price GD (1997) The influence of thermal vibrations on the
 average structure of cubic NaMgF₃ perovskite: a combined molecular dynamics and neutron
 diffraction study. J Phys Condens Matter 9:L647–L655
- 408 Tolédano J-C, Tolédano P (1987) The Landau theory of phase transitions. World Scientific

409 Press, Singapore, 451 p

- 410 Torpor L, Navrotsky A, Zhao Y, Weidner DJ (1997) Thermochemistry of fluoride
- 411 perovskites: Heat capacity, enthalpy of formation, and phase transition of NaMgF₃. J Solid
 412 State Chem 132:131–138
- 413 Uchida T, Wang Y, Nishiyama N, Funakoshi K, Kaneko H, Nozawa A, Von Dreele RB,
- 414 Rivers ML, Sutton SR, Yamada A, Kunimoto T, Irifune T, Inoue T, Li B (2009) Non-cubic
- symmetry of CaSiO₃ perovskite up to 18 GPa and 1600 K. Phys Earth Planet Inter: 282:268274
- 417 Umemoto K, Wentzcovitch RM (2006) Potential ultrahigh pressure polymorphs of ABX₃
 418 compounds. Phys Rev B 74:224105
- 419 Umemoto K, Wentzcovitch RM, Weidner D J, Parise JB (2006) NaMgF₃: A low pressure
- 420 analog of MgSiO₃. Geophys Res Lett, 33:L15304
- 421 Watson GW, Wall A, Parker SC (1995) A molecular dynamics simulation of the effect of
- 422 high pressure on fast-ion conduction in a MgSiO₃-perovskite analogue; KCaF₃. Phys Earth
- 423 Planet Inter 89:137-144
- 424 Zhao Y, Weidner DJ, Parise JB, Cox DE (1993a) Thermal expansion and structural distortion
- $425 \qquad of perovskite data for NaMgF_3 perovskite. Part I. Phys Earth Planet Inter 76:1–16$

- 426 Zhao Y, Weidner DJ, Parise JB, Cox DE (1993b) Critical phenomena and phase transition of
- 427 perovskite data for NaMgF₃ perovskite. Part II. Phys Earth Planet Inter 76: 17–34

428 Figure Captions

429 Figure 1

430	The temperature-dependence of representative pseudocubic superlattice reflections at the M
431	and R points of the pseudocubic Brillouin zone of NaMgF ₃ . The R point reflection $3/2 \ 3/2 \ 1/2$
432	$(d \sim 1.8\text{\AA}, \text{ orthorhombic reflections: } 1 \ 2 \ 3 + 2 \ 1 \ 3 + 3 \ 0 \ 1)$ can be seen to vanish at a
433	temperature just greater than 1046 K, whereas the M point reflection $3/2 1/2 1(d \sim 2.1 \text{\AA})$,
434	orthorhombic reflections: $1 2 2 + 2 1 2$, tetragonal reflection $2 1 1$) passes through this
435	temperature and disappears between 1048 K and 1049 K. Consideration of group theoretical
436	results and analysis of diffraction data shows the phase field composed of the M point
437	superlattice reflections alone (~1046.5 K – 1048.5 K) corresponds to a new tetragonal
438	polymorph of neighborite.
439	
440	Figure 2
441	Rietveld fits to the three polymorphs of neighborite in the orthorhombic (Pbnm), tetragonal
442	$(P4/mbm)$ and cubic $(Pm\bar{3}m)$ phases.
443	

444 Figure 3

The temperature variation of the cubic and pseudocubic lattice parameters in the vicinity of the orthorhombic – tetragonal, tetragonal – cubic phase transitions. The full line on the figure shows that the pseudocubic shear angle varies with reduced temperature to the power 1/2, and hence the critical exponent for the tetragonal – orthorhombic phase transition is 1/4, i.e. tricritical in nature. The pseudocubic lattice parameters a_p , b_p and c_p are defined in terms of 450 those of the orthorhombic unit cell, *a*, *b* and *c* by: $a_p = b_p = \frac{\sqrt{a^2 + b^2}}{2}$, $c_p = c/2$ and

451
$$\cos(\gamma_p) = \frac{b^2 - a^2}{a^2 + b^2} \cong \frac{2\sqrt{2}a_0(b - a)}{4a_0^2} = \frac{(b - a)}{\sqrt{2}a_0} = e_6$$

452 where a_0 is the cell parameter of the cubic phase extrapolated into the tetragonal and

453 orthorhombic phase fields.

Structural parameters		Pbnm	P4/mbm	Pm3m
		1033 K	1047 K	1049 K
Lattice Parameters	а	5.53822(3)	5.55376(4)	3.92695(1)
(Å)				
	b	5.55272(3)	a	a
	С	7.85132(4)	3.92846(5)	а
Unit cell volume		241.445(2)	121.170(1)	60.557(1)
(Å ³)				
Na	x	-0.0034(11)		
	у	0.5098(25)		
	$100u_{iso}(\text{\AA}^2)$	7.45(10)	8.071(78)	8.107(75)
Mg	$100u_{iso}(\text{\AA}^2)$	1.86(4)	2.560(48)	2.399(33)
F1	x	0.0439(6)		
	у	-0.0048(14)		
	100u ₁₁ (Å ²)	8.29(25)	11.46(56)	12.94(6)
	100u ₂₂ (Å ²)	7.99(34)	u ₁₁	u ₁₁
	100u ₃₃ (Å ²)	1.05(16)	0.84(26)	2.34(7)
	100u ₁₂ (Å ²)	0.24(33)		
	$100u_{eq}(\text{\AA}^2)$	7(1)	7.9(3)	9.4(3)
F2	x	0.2836(4)	0.2612(5)	
	у	0.2169(4)		
	Z	-0.0221(4)		
	100u ₁₁ (Å ²)	4.33(12)	8.52(35)	
	100u ₂₂ (Å ²)	4.31(13)	u ₁₁	

454 Table 1. Crystallographic data for the three phases of $NaMgF_3$.

100u ₃₃ (Å ²)	10.05(21)	13.18(52)
100u ₁₂ (Å ²)	-2.75(12)	-5.45(42)
$100u_{13}(Å^2)$	0.51(21)	
100u ₂₃ (Å ²)	-0.45(21)	
$100u_{eq}(\text{\AA}^2)$	6.2(3)	10.1(7)

Pbnm. Na, F1: 4*c* x, y, 1/4; Mg: 4*a* 0, 0, 0; F2 8*d* x, y, z (Rp = 0.049, Rwp = 0.054, $\chi^2 = 1.6$ 455 for 30 variables) 456

- *P4/mbm*. Na: 2*c* 0, 1/2, 1/2; Mg: 2*a* 0, 0, 0; F1: 2*b* 0, 0, 1/2; F2: 4*g x*, 1/2-*x*, 0 (Rp = 0.043, Rwp = 0.049, χ^2 = 1.3 for 19 variables) 457
- 458
- $Pm\bar{3}m$. Na: 1b 1/2, 1/2, 1/2; Mg: 1a 0, 0, 0; F1: 3d, 0, 0, 1/2 (Rp = 0.043, Rwp = 0.049, $\chi^2 =$ 459
- 1.3 for 11 variables) 460



461

462 Figure 1.









466

467 Figure 3.