- Controls on cave drip water temperature and implications for speleothem-based
 paleoclimate reconstructions
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30 Abstract

While several studies explore cave climate and thermal regimes, little is known about the 31 32 controls on cave drip water temperature. Yet water temperature significantly influences 33 biogeochemical processes associated with cave drips. To identify the processes that control 34 the cave drip water temperature, we measured the temperatures at multiple locations along a 35 speleothem flow path and drip sources (stalactites) concurrently with the drip rates in 36 Cathedral Cave, Wellington, Australia. We monitored long-term drip water temperature, drip 37 rates, surface and cave climate and in-cave evaporation rates and conducted 3 infiltration 38 experiments with different flow, temperature and isotopic conditions. Our results show that the drip water temperature is controlled by multiple superimposed heat transport mechanisms 39 40 that act upon the infiltrating water in the epikarst, the water film after it enters the cave and 41 before it becomes a drip. The two main heat sources/sinks for drip water are the cave air and 42 the surrounding rock. The subsurface temperature is coupled to the surface temperature by 43 conduction through the soil and rock mass, but the cave climate is also coupled to the surface 44 climate by venting. On a regional scale drip temperatures are mainly driven by the annual 45 ground surface temperature signal but damped with depth and shifted in time compared to the 46 surface. On a local scale, the drip water temperature can differ significantly from cave air and 47 speleothem temperature due to the latent heat exchange of evaporation and localised water 48 film convection. The main controls are ground surface temperature, subsurface depth, air 49 density induced ventilation, distance from entry and drip rate. We present a conceptual model 50 that explains drip water temperature signals and provide signal driven guidance on best type 51 and location for speleothem sampling. We anticipate that our results will significantly 52 improve the understanding of temperature-dependent paleoclimate signals from speleothem 53 archives.

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54 **1. Introduction**

55 **1.1. Context and Aims**

56 Surprisingly little work has been done on what controls the temperature of cave drip water and yet this is of fundamental importance as it controls biogeochemical processes in caves. 57 58 For example, drip water temperature influences the growth rate of speleothems [Dreybrodt, 59 1981; Baker et al., 1998], fractionation of isotopes [Epstein et al., 1953], and deposition of biomarkers [Schouten et al., 2007]. In speleometeorology, latent heat exchange processes 60 61 such as condensation or evaporation alter the thermal energy content of drip water [De Freitas 62 and Schmekal, 2003] and can lead to cooling of speleothems [Cuthbert et al., 2014a]. Finally, in geomicrobiology, the habitat of cave microorganisms is strongly influenced by temperature 63 64 [Northup and Lavoie, 2010].

65 Cave drip water originates from precipitation or surface flow, which infiltrates the soil surface. It is well recognised that the dynamic temperatures at the earth's surface propagate 66 67 into the subsurface [Stallman, 1965; Baker and Ruschy, 1993]. Near-surface temperature 68 measurements can be used to quantify water flow [Rau et al., 2014], for example by 69 exploiting temperature-time variations [Taniguchi and Sharma, 1993; Bendjoudi et al., 2005] 70 or temperature depth profiles [Tabbagh et al., 1999; Cheviron et al., 2005]. Fluctuating 71 ground surface temperatures are damped with depth until a stable temperature is reached 72 [Taniguchi, 1993; Smerdon et al., 2003]. The dominant mechanism of subsurface heat 73 transfer beyond the soil zone is by conduction [Smerdon et al., 2003]. However, the influence 74 of rock, as opposed to air, temperature profiles on cave drip water temperature has not been investigated. 75

76 Water commonly flows over speleothem surfaces such as flowstones, stalactites and draperies 77 inside caves before arriving at the drip source (falling films) [i.e., Camporeale and Ridolfi, 78 2012]. During film flow a number of different heat and mass transfer mechanisms act simultaneously. While the engineering literature reports on simultaneous heat and mass 79 80 transfer during film flow [i.e., Yan and Soong, 1995], cave related sciences have not investigated the effects of film flow heat transport on the cave drip water temperature. Yet it 81 82 is well accepted that water films will exchange moisture and heat with the cave air [Atkinson 83 et al., 1983; Faimon et al., 2012].

Cave water is generally in contact with cave air for some time before forming drips. Cave climate must therefore be considered when investigating what controls cave drip water temperatures for caves that are open to the atmosphere. It has been shown that surface air 87 temperature anomalies can affect cave air temperature [Dominguez-Villar et al., 2013, 2014]. 88 A change in cave climate is associated with advective air flow by venting [De Freitas et al., 89 1982; De Freitas and Littlejohn, 1987]. Cave venting is caused by barometric pressure 90 changes, density differences between cave and surface air (chimney effect) [Conn, 1966; 91 Wigley, 1967; Oh and Kim, 2011] or through winds across the entrances (venturi effect) 92 [Kowalczk and Froelich, 2010]. Cave-atmosphere air exchange results in spatiotemporal 93 variability of otherwise stable cave air temperature [Smithson, 1991; Perrier et al., 2010]. In a 94 comprehensive investigation of cave air venting Faimon et al. [2012] determined the key 95 drivers of the microclimatic variability.

96 The cave climate also responds rapidly but predictably to changing atmospheric climate 97 conditions [Atkinson et al., 1983; De Freitas and Littlejohn, 1987]. Air flow can cause 98 significant loss of water due to evaporation from caves [McLean, 1971] with increasing 99 moisture loss for only small decreases in cave relative humidity below the saturation point 100 [Buecher, 1999]. Cuthbert et al. [2014a] reported significant cooling of speleothems, and drip 101 water, through in-cave evaporation.

102 Conversely, cave condensation and its change to the overall thermal energy balance were also 103 found to relate to cave air temperatures [De Freitas and Schmekal, 2003]. Condensation can increase the temperature of cave walls [Dreybrodt et al., 2005]. Further, considerable 104 105 speleothem dissolution can be caused by condensation through the formation of calcite 106 undersaturated drips [Rozemarijn et al., 1998]. Importantly, cave climate exerts significant 107 control on speleothem deposition through the temperature dependence of both kinetic and 108 equilibrium drip water geochemical processes [Spötl et al., 2005; Baldini et al., 2008]. 109 However, the in-cave climatic controls on cave drip water temperature have also yet to be 110 explored systematically.

111 When considering temperature as a control for water related cave processes and the 112 interpretation of temperature-dependent speleothem paleoclimate proxies, the cave air 113 temperature is generally used, since it is easily measured. Here, we illustrate that the true 114 cave drip water temperature can differ significantly from cave air temperature and we 115 identify the processes exerting control. Hence the aim of this paper is to identify and describe 116 the controls on cave drip water temperature. We systematically investigate the dominant 117 influences on cave thermal regimes and drip water temperature by analysing subsurface heat (and mass) transport through the karst and the atmospheric connection. Examples for the 118 119 different controls are presented using measurements of drip rate, speleothem and drip water

temperature as well as climate data monitored inside the cave and on the land surface. Using this data we demonstrate how a surface air temperature climate signal will be propagated to a cave, and how the resulting drip water temperatures may deviate from the mean annual air temperature.

124 **1.2. Description of the field site and prior work**

125 Data presented in this paper was acquired at Cathedral Cave in the Wellington Caves Reserve 126 (Latitude -32.622°, Longitude 148.940°) in New South Wales, Australia. Figure 1 shows the 127 location and horizontal dimensions of Cathedral Cave. Cathedral Cave is located in a 128 temperate semi-arid zone. The Caves Reserve is exposed to a significant seasonal variation in 129 the surface air temperature between approx. 0 to 45 °C, with a mean annual maximum 130 temperature of 24.3 °C. Long-term annual average rainfall in the area is episodic with approx. 131 617 mm/year, and the relative humidity varies between 6-98 % with a mean annual value of 132 68 % [BOM, 2014].

133 The cave system is located in the Molong Anticlinorial Zone and intersects a massive and 134 thinly bedded Devonian limestone [Osborne, 2007]. Cathedral Cave is one of the larger caves 135 featuring two nearby entrances and has a vertical depth of approx. 25 m. As a show cave it is 136 well-developed with infrastructure suitable for tourist groups. The cave is easily accessible 137 and offers an ideal opportunity to investigate subsurface karst processes, such as karst 138 hydrology, geochemistry and paleoclimate signals in speleothems. The cave has been subject 139 to long-term drip rate and drip water monitoring starting in 2009 and ongoing. Jex et al. 140 [2012] correlated spatially distributed drip records and found that they group into distinct 141 categories of differently behaving clusters indicative of the flow path features. Mariethoz et 142 al. [2012] identified chaos in drip rates and concluded that this contains information about 143 flow routing in fractured media. Rutlidge et al. [2014] found clear soil and limestone 144 signatures in the drip water through trace elements and organic matter analysis. Cuthbert et 145 al. [2014b] reported that cave drip water is only activated after long duration and high volume 146 rainfall, and that evaporation from the epikarst is an important control on drip water isotopic 147 composition.

148 **2.** Materials and methods

149 2.1. Surface irrigation

150 Owing to the temperate semi-arid climate at the Wellington Caves Reserve, rainfall events sufficient to overcome the soil moisture deficit and trigger cave dripping are erratic [Jex et 151 152 al., 2012; Mariethoz et al., 2012]. To induce dripping in the shallow cave so that controls on 153 cave drip water temperature could be investigated a total of 3 controlled surface irrigation 154 experiments were conducted over a two-year period (2013-2014). Geochemical results of the 155 first irrigation experiment were previously published in Rutlidge et al. [2014] and drip water 156 temperature data from the second irrigation experiment has been reported in Cuthbert et al. 157 [2014a].

During the surface irrigations a patch size of $\sim 24 \text{ m}^2$ (2013) and $\sim 50 \text{ m}^2$ (2014) above the 158 near-surface chamber of Cathedral Cave (see Figure 1) was hand hosed with town water from 159 160 a storage tank. Two summer and one winter irrigation campaigns were conducted. The dates 161 and specifics of each of the three surface irrigation experiments are summarised in Table 1. 162 Importantly, during the first irrigation experiment, the temperature of first and third 163 continuous surface application was set to approx. 0.3 °C using ice bags. Further, deuterium was added as a conservative tracer to the batch of water first applied (enrichment of ~6100 ‰ 164 165 VSMOW) during surface irrigation in 2013. Markowska et al. [submitted] provide a detailed 166 analysis of the deuterium tracer measured during the same experiments as well as long-term 167 monitoring of natural isotopic composition.

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2.2. Cave and surface monitoring

169 Different sites were selected for monitoring at increasing cave depths and distance from cave 170 entrance (Figure 1). To measure the drip water temperature we affixed automated miniature temperature loggers (DST micro T, StarOddi, Iceland) along known flow paths of water on 171 172 top of the speleothem (flowstone), with a logger mounted to the tip of the drip source 173 (stalactite, Figure 2B). The loggers were selected based on their small size, rapid temperature 174 response time (~20 s), resolution (0.01 °C) and accuracy (± 0.2 °C). These features make the loggers an ideal choice for monitoring drip water temperature. The cave air temperature was 175 176 also measured in close proximity to the drip source (Figure 2A). During the irrigation 177 experiment in January 2013 (southern hemisphere summer) the shallow soil temperature of 178 the irrigation patch was monitored at 2 locations with DST micro T loggers (Figure 1).

In the January 2014 irrigations, in addition to the StarOddi loggers, detailed temperature
 measurements were acquired with high accuracy (±0.002 °C) and resolution (0.0006 °C)

custom-build instrumentation. The sensors consisted of Platinum resistors (Pt1000, 1 k Ω at 0 °C) embedded in flat aluminium housing (25 x 6 x 1 mm – see Figure 2C) designed for fast thermal response. Figure 2 shows sensors deployed along a flow stone and stalactite near the entry (site A, location in Figure 1). More details about method and results from this deployment are reported in Cuthbert et al. [2014a]. Here, we use a subset of this data for a more detailed and comprehensive description of the heat transport processes that exert control on cave drip water temperatures.

188 The drip locations were also monitored continuously with automated drip counters 189 (Stalagmate, Driptych, UK). Further, climate monitoring stations consisting of relative 190 humidity and temperature sensors (HMP155A, Campbell Scientific, USA) were deployed at 191 2 different locations to record the cave air. Cave barometric pressure was also measured 192 using a pressure transducer (Levellogger, Solinst Inc., Canada). Water samples were 193 regularly collected from drip sources at site A with 20 ml glass McCartney bottles. The 194 samples were analysed using a Los Gatos® cavity ring down laser spectrometer with overall precision of $\pm 2.0\%$ δ^2 H. Evaporation pans (9.5 cm inner diameter) were deployed at site A 195 196 and C (Figure 1) for extended periods of time. Volumetric water loss was measured using a 197 digital pipette, precision scale and the pan size, and the evaporation rate was calculated from 198 the time of pan deployment.

199 Surface climate variables, i.e. air temperature, shallow soil temperature and moisture, relative 200 humidity and barometric pressure, were monitored by a climate station (Hill Climate Station, Wellington, data download available: http://groundwater.anu.edu.au/) located in close 201 202 proximity south-east of Cathedral Cave. Precipitation data was recorded by a rain gauge in 203 Wellington ~6.5 km away (Agrowplow, station 065034) [BOM, 2014]. The thickness of the 204 soil zone was found to vary from 0 to 0.5 m estimated by inserting a thin metal rod into the 205 soil across the irrigation patch. During the 2014 experiment volumetric soil moisture 206 integrated across the upper 10 cm was measured frequently at random spots across the 207 irrigation patch with a handheld meter (MPM160, ICT International, Australia).

- 208 **2.3. Data processing**
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2.3.1. Surface to subsurface heat conduction

The Earth's surface is exposed to time variable heat influx from solar radiation, which forms a significant energy source for subsurface propagation. The periodicity of insolation is controlled through the earth and solar cycles. Hence, surface air temperature contains distinct frequencies, i.e. daily, annual, decadal, centennial, millennial, as well as aperiodic environmental influences related to local weather and climate, i.e. high and low pressure systems, and oscillation indices. Cave temperatures have been related to ground surface and surface air temperatures by analysing heat propagation with depth through conduction assuming that thermal properties can be depth averaged [Smerdon et al., 2003, 2004].

Carslaw and Jaeger [1959] formulated a 1D differential heat conduction equation. The equation was solved with a harmonic temperature boundary at the top and a constant temperature boundary at infinite depth. This resembles the subsurface environment between surface and cave. Since the heat transport equation is of linear nature, the analytical solution is valid for any harmonic component of temperature variation with an individual frequency (e.g. daily or annual) that is part of the total temperature signal [Goto et al., 2005].

Here, we consider that thermal diffusivity for soil can vary due to differences in saturation [Ochsner et al., 2001], compared to low porosity bedrock which can be assumed to be constant over time. Consequently, it is useful to separate the subsurface into two layers: soil zone and epikarst zone. While several studies have used shallow multi-level soil temperature measurements to calculate near-surface infiltration [Smerdon et al., 2004; Bendjoudi et al., 2005; Cheviron et al., 2005] the propagation of thermal waves into rock above the groundwater table is predominantly controlled by thermal diffusion [Smerdon et al., 2003].

To calculate the dynamic subsurface rock temperature through two layers, an analytical
solution [Carslaw and Jaeger, 1959; Goto et al., 2005] is modified as

233 (1)
$$T_i(z,t) = T_0 + A_i \cdot \exp\left(-\sqrt{\frac{\pi}{P_i}}\left(\frac{d_s}{\sqrt{D_s}} + \frac{z - d_s}{\sqrt{D_r}}\right)\right) \cdot \cos\left(2\pi \frac{t}{P_i} - \sqrt{\frac{\pi}{P_i}}\left(\frac{d_s}{\sqrt{D_s}} + \frac{z - d_s}{\sqrt{D_r}}\right) - \theta_i\right)$$

for $z \ge d_s$. Here, *i* is a distinct harmonic temperature component with period P_i [d]. T_i is the temperature [°C] due to harmonic temperature component *i* as a function of depth *z* below subsurface [m] and *t* is time [d]; T_0 is the mean surface temperature [°C]; A_i is the amplitude [°C] of the harmonic signal *i*; θ_i is a phase offset [rad]; d_s is the thickness of the soil layer [m].

In Equation 1, *D* is the effective thermal diffusivity for the soil layer (subscript *s*) and the epikarst (subscript *r*). In general, the thermal diffusivity $[m^2/d]$ is defined as [Carslaw and Jaeger, 1959]

242 (2)
$$D = \frac{\kappa}{\rho c}$$

where κ is the bulk thermal conductivity [W/m/K] for variably saturated soil or solid rock [de Vries, 1963; Tarnawski, 2011; Horai, 1971; Clauser and Huenges, 1995]. Analogously, the bulk volumetric heat capacity ρc [MJ/m³/K] is reported for sediments and rock [Schön, 1996; Schärli and Rybach, 2001].

247 Equation 1 can be used to predict the subsurface temperature response to a particular 248 frequency component of interest extracted from the ground surface temperature data. For 249 example the i-th component could be daily, annual, centennial, millennial, or any other 250 significant component determined using a Fourier transform analysis of dominant 251 frequencies. In Equation 1 the exponential part accounts for temperature amplitude damping 252 and the cosine part for the shift in phase over depth. The phase offset θ is the time relative to 253 the maximum insolation (summer solstice on 21 December in the southern hemisphere) and 254 accounts for any difference between the conduction theory and realistic conditions.

255 In this paper we use 2 different layers, one representing the soil and one the limestone. We 256 measured the thermal conductivity and heat capacity of soil and limestone samples collected 257 at the Cathedral Cave field site (Figure 1) using a KD2 Pro thermal analyser (Decagon 258 Devices, US). To account for the variable water saturation of the soil (i.e. dry and saturated 259 end members), the soil parameters were measured after oven drying (105 °C, 6 hours) and 260 after saturating the soil sample with water. Further, a piece of limestone bedrock had holes 261 drilled for inserting the instrument needles, and a highly conductive paste was used to ensure 262 optimal thermal bridging between needle and limestone sample. The measured thermal 263 parameters are listed in Table 2.

264 Equations 1 & 2 were used to simulate the annual temperature variations (with P = 365.25 d) 265 at various depths of interest. Models were fitted to temperature observations by varying 266 parameters as outlined in Table 4 and minimising the normalised root mean square error 267 (NRMSE). For the surface air temperature the parameters of interest were mean annual 268 temperature (T_0) , amplitude (A) and phase offset from solstice (θ) while the depth was set 269 to zero (z=0). For the cave air and flowstone temperatures the parameters of interest were 270 mean annual temperature (T_0) , depth of limestone (z) and phase offset (θ) . Here, the 271 remaining parameters were set as follows: Amplitude A as determined from the surface air temperature fit, soil zone thickness d = 0.1 m, thermal diffusivities as measured on soil and a limestone sample (Table 2).

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2.3.2. Air density calculation

A well-known process of cave atmosphere air and moisture exchange is venting stimulated by the difference in density between atmospheric and cave air (the chimney effect) [Conn, 1966; Wigley, 1967; Oh and Kim, 2011]. The density of air can be calculated taking into account thermodynamic properties of dry air as well as water vapour [Giacomo, 1982]. It is expressed as

280 (3)
$$\rho_a = \frac{pM_a}{ZRT} \left[1 - x_v \left(1 - \frac{M_v}{M_a} \right) \right]$$

where *p* is the barometric pressure [Pa]; *Z* is the compressibility coefficient, under the conditions reported here of value 0.999611566 [-]; *R* is the universal gas constant, 8.31441 [J/K/mol] and *T* is the temperature [K]; M_a and M_v are the molar mass of dry air 0.0289635 [kg/mol] and water vapour 0.018015 [kg/mol]. The mole fraction of water vapour in moist air x_v is defined as

286 (4)
$$x_v = \frac{h}{p} exp \left(AT^2 + BT + C + DT^{-1} \right) f$$

where *h* is the relative humidity (0 < h < 1); *f* is an enhancement factor, under the conditions reported here of value 1.0038 [-]; the saturation vapour pressure coefficients are published as A=1.2811805 $\cdot 10^{-5}$ [K⁻²], B=-1.9509874 $\cdot 10^{-2}$ [K⁻¹], C=34.04926034, D=-6.3536311 $\cdot 10^{3}$ [K]. For above parameter values please refer to Giacomo [1982].

Equations 3 and 4 require measurement of the common variables that define the thermodynamic state of moist air: barometric pressure, air temperature and relative humidity (RH). To investigate cave venting, air densities were calculated from the surface and cave climate records for a 2-week period during both summer and winter in 2014.

295 **3. Results**

3.1. Surface and cave climate 296 297 Figure 3 shows surface air temperature and rainfall recorded at the surface above Cathedral 298 Cave over a 2-year period between 2012 and 2014. Cave air temperature measured near site 299 A is also shown. A climatic summary for the period between Jan 2013 and Dec 2014 is as 300 follows: The minimum and maximum surface air temperatures were -2.9 °C and 43.5 °C. 301 Typical for a temperate semi-arid climate, relative humidity varied between 5-98 %, with a median of ~63 %. For more than half of the year (233 d) the volumetric soil moisture content 302 303 was below the median annual value of 21 % because evapotranspiration generally exceeds 304 precipitation.

While ~312 days/year were without significant rain (< 1 mm/day), below average yearly total of 550 mm was recorded from episodic rainfall events occurring on 86 days/year. On 1 March 2013 a maximum daily rainfall of 74 mm was recorded. The amount of rain from this natural event was comparable to the manual application of water on the irrigation patch during the surface irrigation experiments (Table 1 and Figure 1).

At Site C, the air and speleothem temperature was very stable at 17.8 °C with only minor

fluctuations of ~0.1 °C between January-December 2014. Cave relative humidity (RH) was

measured at 10 min intervals during parts of the year between January and November 2014.

313 The RH, recorded at site A, fluctuated significantly with minimum, maximum and median

values of 59.3 %, 97.9 % and 88.6 %, respectively. At site C the RH showed very minimal

- fluctuations around a median value of 97.1 %, with minimum and maximum RH of 96.5 %
- and 97.8 %, respectively. Evaporation rates as measured at the different locations (Figure 1)

during summer and winter 2014 are shown in Table 3. There is a clearly decreasing trend in

evaporation rate with increasing distance from entrance in summer, with RH values

319 increasing as expected. Noteworthy, however, is the stable but below saturation RH level at

320 site C leading to some potential for evaporation from the deepest part of the cave throughout

321 the year.

322 **3.2.** Drip water temperatures during irrigation experiments

Figure 4 presents high resolution temperature measurements, drip counts and relative humidity measured during irrigation experiment 2 conducted in January 2014 (summer). While the majority of the measurements in Figure 4 were previously published by Cuthbert et al. [2014a], we use this dataset as a starting point and present new results that reveal a detailed analysis of the different controls on cave drip water temperature.

328 Before the first surface irrigation the soil moisture across the irrigation patch was between 4-329 24 % indicating a high soil moisture deficit. Approx. 3 h after the start of the water 330 application (~68 mm rainfall equivalent) the drip source responded with a rapid increase to 331 approx. 140 drips/min (Figure 4B). Before the second irrigation the soil moisture was much 332 higher with measurements ranging between 20 % and 37 %. After applying less water in the 333 second irrigation (equivalent of ~48 mm rain) the drip source responded much quicker (~1 h 334 after start) and showed significantly faster drip frequency (~180 drips/min) and longer drip 335 activity compared to the previous day (Figure 4B).

336 Before the onset of dripping, temperature measurements taken on the dry speleothem surface 337 along the expected drip water flow path (Fig. 2a) were relatively constant in time but with 338 decreasing temperature from cave ceiling to drip source (stalactite) revealing a downward 339 gradient approx. -0.8 °C/m (Figure 4A). Measured air temperatures reflect a spatial gradient 340 that was similar to the one measured on the rock surface. Relative humidity measurements 341 (RH) varied between a minimum of 79 % and a maximum of 91.5 %. A spot measurement 342 near the chamber ceiling revealed RH of up to 98 % after the flowstone had been wet at the 343 end of the irrigation experiment in Jan 2014.

Temperatures, measured after activation of cave dripping, exhibit a rapidly increasing temperature on all sensors, peaking at approx. ~0.3-0.8 °C above the original measurement coinciding with a peak at the maximum drip count (Figure 4A). This is followed by a slow temperature decrease as the drip rate decreases. At ~ 20 drips/min, the drip water temperatures measured by the lower sensors returned to the level measured before the onset of flow.

After a period of relatively stable measurements, the drip water temperature started to decrease, with lower sensors showing a more rapid and pronounced cooling of up to 1.5 °C below the cave air temperature which was measured in close proximity. The onset of observable evaporative cooling was at a RH of 90 %, and the increase in drip water coolingcoincided with a rapid drop of RH to 79 %.

After the second surface irrigation the same temperature increases were observed but with stronger magnitude and longer duration, despite the application of less water at the surface. However, evaporative cooling was less pronounced reflecting the higher levels of RH (85-90 %) during this event compared to the first event.

359 Figure 5 summarises temperature and deuterium data as well as drip counts measured during irrigation experiment 1 conducted in January 2013. Note that the experimental procedure and 360 361 measurement setup differed compared to experiment 2 described in the last section. Here, 362 drip water temperature was only measured at the drip source (same stalactite as above). 363 However, in addition shallow soil temperatures (~5 cm and ~10 cm below the surface) were 364 measured, but cave air RH was not. It is noteworthy that 4 individual irrigations were applied 365 (35-63 mm rainfall equivalent) and with the water during the first 3 applications cooled to ~ 0 °C, ~10 °C and ~0 °C, respectively. 366

367 Cave air temperature was relatively stable at approx. 17.5 °C (Figure 5A), while the daytime outside air temperature peaked at approx. 40 °C. During the time of experimentation the cave 368 369 air temperature shows slight increases during the times at which the surface air temperature 370 was at its lowest (night time). This excludes one occasion on 10 January 2013 where the cave 371 air and drip water temperatures both decreased coincident with the surface air temperature 372 falling below the average cave temperature (grey arrow in Figure 5B). Also noteworthy here 373 is the response of the soil temperatures to the cooled irrigation water, with both sensors 374 showing measurements as low as 5 °C and 14 °C which are clearly below the minimum 375 surface air temperature of 15 °C during that time (Figure 5A).

376 Drip water temperatures responded similarly to the surface irrigation during the January 2013 377 experiment (Figure 5D) compared to the experiment in 2014 (Figure 4B). Interestingly, the drip water temperature at the first drip activation with an average drip response of 80 378 379 drips/min shows a cooling event during which there was a significant temperature difference 380 of -2.5 °C between drip water and air temperature (Figure 5B). This was the response to an 381 irrigation application where the water was cooled to 10 °C, less than during the first irrigation 382 (Figure 5A). A similar sized evaporative cooling event can be seen again during the drip recession caused by the last surface irrigation where ~24 °C water was applied without the 383 384 addition of ice. A clear deuterium enrichment (deuterium breakthrough) was measured in drip water samples after the third surface application originating from the deuterium that wasadded to the first irrigation batch (Figure 5C).

Drip water temperature after the third surface irrigation during which water was cooled again to 0 °C showed a very small decrease before warming and tracking close to the cave air temperature (Figure 5B). As soon as the drip rate fell below ~30 drips/min another evaporative cooling event was observed. This time, however, it was overwhelmed by the last surface application of water which carried warm water as film flow along the speleothem surface.

393

3.3. Long term air, speleothem and drip water temperature records

394 Figure 6 shows the temperature data measured on the speleothem surface (dry or wet cave 395 over speleothem surfaces) at three different locations along the drip water flow path at site A 396 (see Figure 2) including the drip source (stalactite), air temperature and drip rate over a time 397 period of ~11 months. Figure 6 includes the response to surface irrigation experiment 3 (also 398 highlighted in Figure 3). The trend in all temperature data complies with a distinct annual 399 harmonic but with different amplitude and phase compared to surface air temperature. This 400 originates from subsurface conduction of the annual surface temperature wave, and we will 401 refer to this as the "background temperature".

402 Results from fitting surface air, cave air, speleothem and drip water temperature time-series 403 to Equation 1 with an annual periodicity are presented in Table 4 ordered by increasing total 404 depth. The best fitting annual harmonics are also plotted in Figures 3 and 6. Noteworthy here 405 is the characteristic amplitude damping and phase shifting with increasing total depth. While 406 the surface air temperature is offset from summer solstice by 20 days, there is a relatively 407 constant phase offset of ~11 to 12 days (compared to the surface air temperature) once the 408 annual temperature harmonic propagated through the subsurface. This indicates compliance 409 with the subsurface heat conduction theory (Equation 1). Further, total depths obtained from 410 the fitting procedure are in good agreement with the vertical cave dimensions estimated from 411 an in-cave survey (Figure 2A).

The two upper measurement points show relatively stable temperature over time, when considering faster than annual frequencies, but with occasional upward and downward spikes indicating fast advective film flow in summer and winter, respectively. However, the temperature measured in air and the tip of the stalactite (Figure 2) shows marked fluctuations with a daily frequency and varying amplitude of up to ~1 °C superimposed on irregular lower frequency variations and the background temperature. A number of drip events with varying
magnitude and with a maximum of ~25 drips/min were recorded (Figure 6). At this point a
question arises: What causes the faster than annual temperature fluctuations?

420

3.4. Examples of venting induced drip temperature changes

Figure 7 shows a detailed snapshot of cave flowstone, 2 stalactites, and cave air temperature
(A, D) as well as cave RH (B, E), and surface and cave air density calculated using Equations
3-4 (C, F) during summer and winter in the year 2014.

424 In summer (Figure 7A-C), a small drip event triggered an upwards temperature spike ~ 0.5 °C on the stalactite, followed by multiple cooling fluctuations with magnitude ~1.5 °C 425 426 coinciding with rapid decreases in cave air RH due to the venting events. A decrease in cave 427 air temperature, with some delay, as a result of evaporative cooling, is also evident from the 428 data. The cooling events are similar to those observable during the irrigation experiments 429 (Figures 4A and 5A) but seem to occur with a daily frequency over certain periods (Figure 6). 430 When comparing this with the surface and cave air densities it is clear that the regular RH 431 decreases correlate well with periods where the surface air is denser than the cave air (note 432 that dry air is denser than humid air of the same temperature) in the early mornings causing 433 frequent cave venting events. Interestingly, evaporative cooling spikes also occur higher up 434 the profile where the drip water flows as a film along the speleothem surface (Figure 6).

Figure 7D-F contains the 2 weeks of winter monitoring that also coincide with the third

436 surface irrigation experiment 3. In winter (Figure 7D-F) the drip source shows regular daily

437 temperature fluctuations of ~ 0.8 °C. Inspection of cave climate parameters reveals that the

438 cave air temperature fluctuates more and the RH less compared to summer (Figure 7E vs 7B).

439 Further, the outside air is almost continuously denser than the air in the shallow entrance area

440 (Figure 7F). Interestingly, the drip water temperature mainly reflected the pattern of the cave

441 air temperature while the drip rate (resulting from artificial surface irrigation during winter)

did not exceed ~25 drips in a 15 minute interval.

443 **4. Discussion**

Results presented in this paper allow, for the first time, a detailed identification of what controls the temperature of cave drip water. First we identify the controls and analyse how they affect drip water temperature, then we discuss their significance and implications in relation to interpreting speleothem records as paleoclimate archives.

448 **4.1. What mechanisms control the cave drip water temperature?**

449 Water movement to the drip source often occurs as film flow on cave deposits along variable 450 distances [Dreybrodt et al., 2005; Camporeale and Ridolfi, 2012; Baker et al., 2014]. The data 451 presented here demonstrates that cave drip water temperature is controlled by a number of 452 simultaneous heat transport mechanisms that act upon the water film. Heat transfer between 453 rock and water in karst conduits was analysed in detail by Covington et al. [2011], Covington 454 et al. [2012] and Luhmann et al. [2015]. Dreybrodt et al. [2005] have theoretically analysed 455 the heat and mass interactions involved in condensation corrosion involving water films. The 456 engineering literature has recognised the complexity of film flow heat and mass exchange 457 [i.e., Yan and Soong, 1995]. In relation to speleology our results are first in reporting and 458 analysing heat transport processes that control cave drip water temperature.

The variety of different mechanisms and associated variables complicates quantification of the individual processes. Here, we focus on a detailed description of temperature characteristics that can be measured after water enters the cave and flows along cave features before arriving at the drip source. Figure 8 conceptualises the controls on drip water temperature. The individual heat transport mechanisms are discussed with reference to examples presented in the results.

465 **Convective heat transport:**

466 <u>Heat convection due to subsurface water percolation</u> $(q_{f,surf})$:

467 During the first surface irrigation experiment the water was deliberately cooled (Table 1) to 468 test whether its thermal signature, transported by heat convection through the soil zone and 469 the epikarst stores, is detectable at the drip source. The pre-existing large soil moisture deficit 470 prior to surface irrigations was responsible for the first irrigation not producing any flow in 471 the cave (Figure 5A). Due to the hot weather and general heat conduction towards the 472 irrigated patch the cooled soil recovered to near normal temperatures between each of the 473 cooled irrigations. While the second application was cold enough (~10 °C) for the thermal signature to be seen in the soil zone the cooling anomaly observed at the drip source 474

475 (locations k1 and k2 in Figure 5B) did not originate from the cooled surface irrigation. The 476 main evidence for this conclusion is the lack of breakthrough of the deuterium enriched water 477 (~6100 ‰ VSMOW) from the first irrigation (Figure 5C). The breakthrough of deuterium 478 occurred after the third irrigation, indicating that the water travel time was significantly 479 longer that the time between individual irrigations and the corresponding drip response in the 480 cave. Markowska et al. [submitted] concluded that the water activating the drip came from 481 epikarst stores. This also means that convection of cold water from the surface to the cave 482 will take longer than the individual drip response time.

483 While the soil zone clearly responded to the three applications of cooled water at 2 separate 484 locations (Figure 5A), the only signature attributable to the ice water detected at the drip 485 source was a sharp short temperature fluctuation of only -0.8 °C on 10 Jan 2012 at 09:18 486 while the air temperature remained constant (blue arrow in Figure 5A). Importantly, this happened at a time during which fast film flow occurred over the flowstone, so this is not a 487 488 temperature signal attributable to evaporative cooling (which only is dominant at slower 489 flow). Interestingly, this short lasting cooling event was detected shortly after the start of the 490 third surface irrigation (~35 mm rainfall equivalent) with ice-cooled water (~0 °C) while the soil was still cooled from the previous event. We interpret this as heat convection due to 491 subsurface water percolation caused by fast preferential flow through the well wetted soil and 492 493 fracture flow in the epikarst below. Note that first breakthrough of deuteriated water from the 494 first surface irrigation was observed at the same time (Figure 5C and Markowska et al. 495 [submitted]).

496 The above discussion illustrates that drip water temperature can be affected by thermal 497 energy transported from the surface to the drip source through convection caused by 498 subsurface water percolation. However, the prerequisites are that soil moisture is at field 499 capacity, that preferential flow paths are still present and that the volume of water applied to 500 the surface is much larger than the likely event based rainfall (105 mm was the maximum 501 event based total between Oct 2011 and Dec 2014). In our case it took more than 133 mm 502 rainfall equivalent (3 irrigations) of cooled water to produce a brief and small temperature 503 anomaly. Furthermore, the experimental conditions were a worst case scenario in two other 504 ways: 1) the temperature difference between the cooled irrigation water and the soil of 20-25 505 °C was unrealistically large for natural conditions, and 2) the section of the Cathedral Cave 506 used in these experiments is very shallow with only about 1.7 m of soil and rock mass 507 between the cave ceiling and the surface.

508 We expect that <u>heat convection from subsurface water percolation</u> caused by preferential 509 flow through the soil and fracture flow through the epikarst can rarely cause drip water 510 temperature anomalies that are significant for paleoclimate reconstructions from speleothems 511 under realistic conditions. However, we acknowledge that this will depend on the thickness 512 of the soil and epikarst as well as the fracture network above the cave. More research is 513 needed to determine the conditions for which heat convection due to preferential or fracture 514 flow from the surface can cause temperature anomalies that are of significance for speleothem-based paleoclimate reconstructions at drip sources. 515

516 <u>Heat convection (q_f) due to film advection (v_f) along cave walls:</u>

517 The mechanism of convective heat transport due to film advection is clearly illustrated in the 518 drip water temperature response during surface irrigations 1 and 2 (see labelled areas in 519 Figures 4A and 5B). Since it is summer, warmer water flows in films along the speleothem surfaces (v_f) where the thermal signature from above is carried with the water film (q_f) 520 (Figures 4A and 5A). As a result of convective heat transport due to film advection the drip 521 522 water temperature was raised by ~1 °C, but only at the start of the irrigation response (fastest 523 drip rates on an event basis, here > 50 drips/min) and when a negative temperature-depth 524 gradient existed (i.e., summer).

525 Temperature sensors located in the upper part of the profile (location b and c in Figure 2A) 526 near the point at which water enters the cave detected a warmer water film compared to the 527 surrounding air (Figure 4A). This thermal disequilibrium indicates heat convection due to fast 528 preferential or fracture flow triggered by the surface irrigation [Cuthbert et al., 2014a]. 529 However, it is important to note that the thermal energy causing the warming anomalies does 530 not originate directly from the water applied to the surface. Instead, the anomalies originate 531 from conduction between water film and rock higher up along the profile (explanation further 532 below). Convective breakthrough between surface and cave only occurred under extreme circumstances, as pointed out in the previous subsection. The warming anomalies express a 533 534 temporary downward shift of the localised conductive depth profile, i.e. they represent the 535 temperature of the re-equilibration between the water film and the rock mass a short distance 536 above the point of measurement. Here, we hypothesize that the magnitude of the convective 537 signature is a function of the film advection rate (v_f proportional to the drip rate), the film 538 thickness (b) and the flow distance (L). Baker et al. [2014] measured the thickness of water 539 films on speleothems and found a dependency on the curvature and roughness of its surface. 540 Considering the number of unknowns and the fact that convective and conductive heat 541 transport are both contributing during film flow, it is highly challenging to predict the water 542 temperature as a function of distance.

543 As can be seen in the long-term drip water temperature record (Figure 6), film heat 544 convection is initiated at the onset of drip events. However, it is most pronounced at the times 545 with a large (exponential) temperature-depth gradient along the profile. This thermal gradient 546 is caused by the conduction of the annual temperature signal into the subsurface rock mass. 547 Consequently, the thermal effect of convection on drip water is a pronounced heating after 548 summer and cooling after winter solstice. Further, it is most muted around the equinoxes due 549 to a reversing temperature depth profile. Importantly, any convective influence on drip water 550 temperature caused by film advection along cave walls will be muted at depths beyond the reach of the annual harmonic signal (see discussion further below). 551

552 Exchange of moisture (m_{atm}) and thermal energy $(q_{f,atm})$ between surface and cave:

When caves are open to the atmosphere air is exchanged [Conn, 1966], with the "chimney effect" (caused by an unstable density difference) being a common cause of venting [i.e., De Freitas et al., 1982; Oh and Kim, 2011]. Here we observe that the surface air is frequently denser than the shallow cave air during summer (Figure 7C) and continuously during winter (Figure 7F) which causes Cathedral Cave to be a well vented cave. At this point the question arises: How deep do venting events propagate into the cave?

- 559 Cuthbert et al. [2014] have shown that the drip water temperatures at a continuous slow drip source located ~40 m into the cave (site B) was continuously ~0.6 °C cooler than the 560 561 surrounding speleothem and air temperature, at a depth where conduction of heat from the surface is muted and where RH values are stable at ~92 %. Further, evaporation rates 562 563 measured at different locations increasingly deeper in the cave show that the venting effect 564 must dampen with distance from entry, which is consistent with observations in other caves [Perrier et al., 2010; Faimon et al., 2012]. However, despite the fact that the high frequency 565 566 venting events do not directly show up at site C (Figure 1) a potential for evaporation does exist since the RH is ~97 %. Maintaining RH at less than saturation would not be possible 567 without air exchange and, thus, drier and denser surface air must continuously replace moist 568 569 and lighter air from deep within the cave.
- 570 Our findings are consistent with those from Buecher [1999] who reported a significant 571 moisture loss at an average cave RH of 99.4 % due to venting in Kartchner Caverns located

572 in semi-arid Arizona. While cave venting has previously been investigated [Smithson, 1991; 573 Tarhule-Lips and Ford, 1998; Spotl et al., 2005] and its effects on the moisture loss have been 574 analysed [McLean, 1971; Buecher, 1999; De Freitas and Schmekal, 2003], we emphasize that 575 potentially significant amounts of thermal energy in the form of latent heat continuously 576 leaves the cave in the form of water vapour. This raises the question whether ongoing 577 evaporation and associated cooling can significantly lower the overall temperatures of caves 578 as well as individual drips? This could be answered by quantifying the energy lost through 579 latent heat as a fraction of the total cave energy balance.

580 **Conductive heat transport:**

581 <u>Conduction of the surface temperature signal into the subsurface $(q_{c,atm})$:</u>

Conduction of surface air temperature signals into the subsurface is a well-accepted 582 583 phenomenon [Smerdon et al., 2003, 2004]. Table 4 shows that the depth propagation of the 584 annual harmonic through rock mass complies well with the theory (Equation 1). Dominguez-585 Villar et al. [2013] made use of cave thermal anomalies, measured in the cave air, to infer that 586 vegetation change at the surface influenced subsurface conduction. Further, the signature of 587 global warming was found in cave air temperature data at a depth of 37 m [Dominguez-Villar 588 et al., 2014]. We present 2 years of surface air temperature and cave air measurements, as 589 well as 1 year of speleothem, water film and drip water temperatures at different depths along 590 a flow profile. We illustrate that Equation 1 is able to predict the subsurface penetration of 591 the annual harmonic component by conduction from the ground surface temperature signal 592 considering multiple layers with different thermal properties. This should equally apply to 593 any other harmonic contained in the surface temperature signal as long as it is of sufficient 594 magnitude and duration not to be damped beyond detectability.

The data shown in Figures 3 and 6 demonstrate that the penetration of the annual temperature variation controls the drip water temperature at site A. The surface temperature signal generates the "background" temperature for drip water, but with exponentially damped amplitude and linearly shifted phase proportional with depth. Here, the differences in mean annual temperature can be explained with temperature changes that are slower than annual.

600 <u>Conduction between speleothem and water film</u> $(q_{c,rock})$:

The mechanism of heat conduction between speleothem and the water film, albeit "smeared"

by convection, is evident from the drip water temperatures measured during both irrigation

603 experiments (Figure 4A and Figure 5A). The first irrigation experiment (cooled water was 604 applied to the surface on three consecutive days, Figure 5) clearly illustrated that the pre-605 existing temperature-depth gradient (the subsurface temperature decreases exponentially with 606 depth in summer) warmed the infiltrating colder irrigation water by conduction to produce 607 the arrival of warm pulses on the speleothem at the onset of dripping (Figure 5B). The time it 608 took for the deuteriated water to arrive at the drip source (Figure 5C) indicates a relatively 609 long residence time of water in the epikarst stores (~48 hours), for relatively large volumes of 610 water applied and an extreme temperature difference between water and rock. This 611 demonstrates that any temperature disequilibrium between rock and water from location b 612 onwards (Figure 2) must have originated from the subsurface rock mass. During irrigation 613 experiment 2 similar increases in the water temperature were observed after dripping had 614 started. Therefore, the increase in drip temperature after flow started was caused by 615 conduction from the warmer speleothem to the water further upstream of the profile 616 (exponentially decreasing rock temperature with depth in summer), followed by convective 617 heat transfer due to film advection, and subsequent conduction from the warmer water film 618 back into the rock further downstream [Cuthbert et al., 2014a].

619 The fact that the relative magnitude of the warming anomaly remained the same for sensors 620 located further along the profile is evidence for conduction between water film and rock 621 (Figure 4A). The amount of thermal energy conducted depends on the time that the water 622 film is in contact with a particular area of speleothem, the film thickness (b) and the temperature difference. The contact time is determined by the velocity of the film flow (v_f) , 623 624 which is proportional to the drip rate measured. There is a slow temperature tailing of the 625 water film and drip temperature (Figure 4A) in all records along the flow stone (L). This is 626 caused by conduction of thermal energy from the warmer water film back into the cooler 627 speleothem when convection becomes less significant than conduction at decreasing film 628 advection (= drip rates).

The temperature sensors that were inserted 4 cm into the speleothem confirm that the thermal anomaly caused by the flowing water film is transferred into the speleothem. These sensors show a temperature damping and lag with distance into the speleothem that is characteristic of heat conduction (red lines in Figure 4A). Below a certain film advection rate (~20 drips/min in this case), convective warming ceases to dominate and is overwhelmed by evaporative cooling (the cross-over of lines e & f in Figure 4A) illustrating that there is a temporary thermal equilibrium (Figure 4A). Consequently, if the water film advection rate is sufficiently slow or the film is thin enough the drip water temperature is controlled by the
speleothem temperature but only in the absence of impacts from cave climate (i.e.
evaporative cooling).

639 <u>Conduction between air and water film or rock wall</u> $(q_{c,air})$:

640 The cave air shows a vertical temperature gradient that is similar to the subsurface rock 641 temperature gradient under stable conditions, i.e. no flow and no venting events (Figure 4A). 642 Thermal anomalies can propagate much quicker through air than rock or water because the 643 thermal diffusivity of air is approx. 22 and 146 times larger than that of the rock and water, 644 respectively (Table 2). However, the heat capacity of air is in excess of \sim 4,000 and \sim 2,500 645 times smaller than water and rock, respectively. This means that the energy contained in 646 thermal anomalies brought into the cave by air venting is effectively damped by the rock 647 [Perrier et al., 2010]. Nevertheless, an example of heat conduction between air and drip water 648 can be seen during irrigation experiment 1: A venting event transports cooler air from the 649 atmosphere into the cave temporarily lowering the temperature of the drip source by $\sim 1 \, ^{\circ}C$ 650 (grey arrow in Figure 5A). The drip had ceased to be active at the time however the 651 speleothem surface was still wet.

During winter the cave is continuously vented and the cave air temperature fluctuates periodically with varying amplitudes that depend on the surface climate (Figure 7E). This thermal signature is almost exactly replicated by the drip source temperature showing the mechanism of conduction between air and speleothem or air and water film (Figure 7D). The magnitude of temperature variation depends on the magnitude of airflow which is proportional to the air density difference [Faimon et al., 2012].

658 Latent heat and mass transport:

659 <u>Latent heat $(q_{l,air})$ and mass (m_{air}) exchange between the water film and cave air</u>:

660 Cuthbert et al. [2014a] previously demonstrated evaporative cooling of speleothem drip 661 water, by as much as -1.5 °C compared to the cave air temperature. We have shown in new 662 data presented here that this may be as high as -2.5 °C (Figure 5). This anomaly was not 663 caused by heat convection due to subsurface water percolation transporting the cooled 664 irrigation water via preferential or fracture flow between surface and cave, as deuterium 665 breakthrough had not yet occurred (Figure 5C). The cooling occurred because the previously 666 dry flowstone surface was wetted by the drip response to surface irrigation. In fact, at one 667 location (k2 in Figure 5B) cooling of the wet flowstone to below air temperature continued 668 after film flow had ceased. As the absence of dripping (and therefore film flow) rules out the 669 possibility of convective cooling from cooled irrigation water applied to the surface, the 670 cooling anomaly must be caused by evaporation.

In Figure 6 we present a new longer record of temperatures measured on 3 points along the speleothem surface including drip source (stalactite). It is obvious that frequent evaporative cooling events (Figure 7A) are directly coupled to venting events lowering the RH during summer (Figure 7C). Without venting the cave air RH would reach saturation over time and diminish the potential for evaporation. While Buecher [1999] found that cave evaporation rates are very sensitive to changes in RH, we observe that the vapour deficit also directly influences the magnitude of evaporative drip water cooling (Figures 4 and 7A-C).

678 From results presented here it is clear that air venting causes a complex thermodynamic 679 coupling of cave and surface climate that influences the cave drip water temperature. We 680 illustrate frequent and significant evaporative cooling and associated moisture exchange 681 between drip water and cave air caused by frequent exchange of humid cave air with dry 682 surface air. Dreybrodt et al. [2005] reported that cave walls can be warmed due to the release 683 of latent heat during condensation in caves located in a humid climate. While our results 684 show that in-cave evaporation can cause cooling, we anticipate that condensation could warm 685 cave drip water. We illustrate that, when venting is present, cave drip water temperature near 686 cave entrances can contain significant diurnal fluctuations or continuous cooling relative to 687 cave air whenever RH is below a certain threshold. However, for drip water temperature to be 688 affected by the cave climate it must be exposed to the cave air for some time before arriving 689 at the drip source, e.g. as a water film flowing over speleothem surfaces such as flowstones, 690 stalactites and draperies.

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4.2.1. Relationship between temperature at the surface and drip source

Drip water temperature is a key variable to be considered when the paleoclimate records are reconstructed from speleothem archives. Current methods allow for paleo-temperature reconstruction (i.e. from $\delta^{18}O$) with seasonal and even monthly resolution [i.e., Treble et al., 2007; Orland et al., 2009]. The spatial resolution of speleothem milling, and therefore the temporal resolution of climate proxies, is likely to increase in the future with the development of better technologies. While the surface temperature is typically the result of interest, many

4.2. Implications for speleothem-based paleoclimate reconstructions

geochemical proxies depend on the temperature of the water at the drip source. This necessitates a better understanding of processes affecting the temperature at the surface of the speleothem at the time of its formation. Past surface climate estimates can be influenced by assumptions about the conditions along the flow path between surface and drip source.

704 Our results demonstrate that, in the absence of cave venting and convective thermal 705 breakthrough from the surface, the drip water temperature is primarily a function of 706 subsurface heat conduction, i.e. infiltrating surface water is quickly equilibrated to the 707 subsurface temperature-depth profile. A universally applicable model to describe the 708 relationship between surface and drip water temperature in this case is the differential 709 equation for conductive heat transport [Carslaw and Jaeger, 1959]. It is important to note that 710 thermal modelling requires subsurface thermal parameters such as presented in Table 2. 711 However, these are in general reasonably well constrained and references to suitable 712 literature can be found in Rau et al. [2014]. While significant temperature anomalies due to 713 convective heat transport from the surface that could imprint on paleoclimate proxies can be 714 ruled out in our case, we note that this could be possible under different karst settings. 715 However, we expect the likeliness of such temperature anomalies to decreases with 716 increasing subsurface depth.

717 The presence of the annual temperature signal in our data (Figure 3) facilitated the use of an 718 analytical solution that is based on a harmonic temperature input at the surface (Equation 1). 719 While this solution is useful for estimating the subsurface temperature response to cyclic 720 drivers (e.g. annual, decadal, centennial or millennial), many paleoclimate events of interest 721 are based on non-cyclic changes in the surface temperature, e.g. rapid climate change 722 [Holmes et al., 2011]. Modelling the latter would require the selection of a suitable model to quantify the temperature evolution between surface and drip source. For example, the 723 724 analytical solution used by Domínguez-Villar et al. [2013] describes the subsurface 725 temperature as a function of depth and time based on a step change in surface temperature. 726 Drip source temperature signals can be predicted from arbitrary surface temperature-time 727 signals using a time convolution of this model. Vice versa, a deconvolution can unravel the 728 surface temperature from a speleothem-based paleoclimate proxy.

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4.2.2. Optimising the speleothem sampling location

Our measurements show that drip water temperature is controlled by a complex thermal
coupling between the subsurface rock background temperature driven by the ground surface
temperature and the cave climate driven by ventilation. This requires careful consideration

when deciding speleothem sampling locations. For example, the stalactite on which the drip temperature was measured (Figure 2A-B) was exposed to an annual temperature variation of ~5.21 °C as conducted from the surface but with a delay of ~2.6 months (80 days) compared to the surface temperature signal. This is a significant variation when the temperature dependency of speleothem growth is considered [Hendy, 1971; Casteel and Banner, 2014] and if seasonal surface temperature is to be reconstructed.

739 Figure 9 shows the propagation of selected frequency components with an average soil zone 740 thickness of 0.1 m and an underlying epikarst to a depth of 100 m as a generic example but 741 also resembles the Cathedral Cave setting. Calculations are based on the laboratory 742 measurements of thermal parameters. Envelopes for minimum and maximum thermal 743 diffusivity for soil and bedrock as reported in the literature (Table 2) were also determined 744 for transferability of the results, i.e. when different materials are present at different field sites. Figure 9 clearly illustrates the characteristic amplitude damping and phase shifting with 745 746 depth, inherent to the different harmonic signals. For example, it might be useful for a 747 researcher to maximise or minimise the annual temperature signal (which may determine the 748 presence of annual geochemical laminae useful for chronology building) compared to the 749 long-term paleoclimate signal. If a speleothem location was to be selected where the 750 maximum annual temperature variation should not be larger than 1 °C (0.5 °C amplitude) the 751 surface amplitude damping factor is ~0.059 (0.5 °C/8.51 °C). In the absence of venting and 752 convective heat transport through preferential or fracture flow, the desired variation is not 753 exceeded at total depths of greater than ~8.6 m (red dot in Figure 9A).

754 Another important consideration, when paleoclimate is to be inferred from speleothem 755 archives, is the phase shift. Again an example close to our case: A surface temperature signal with centennial period is shifted by ~7.82 years (94 months) at 15 m depth (red dot in Figure 756 757 9B). Hence, this should be taken into account either when an accurate resolution of temporal 758 (i.e. seasonal) climate patterns is desired or when climatic patterns are compared to other 759 sources of information. Table 5 exemplifies minimum and maximum expected damping 760 factors and signal shifts for distinct depths extracted from Figure 9. This lag is within 761 resolution of long-record dating [Cheng et al., 2009] and could explain previous lag times 762 between drip source related signals and surface events [Domínguez-Villar et al., 2009].

The above discussion illustrates that the speleothem sampling location will not only depend on the type of proxy (i.e., $\delta^{18}O$, $\delta^{13}C$, Δ_{47} , trace metals, organics) but also on what archived 765 harmonic signal resolution is desired. The increasing temporal resolution for drip source 766 temperature dependent proxies makes shallow sampling attractive to maximise the high 767 frequency temperature signal (i.e., seasonal to annual). However, near-entrance locations 768 require a good quantitative understanding on the influence from cave climate, such as 769 evaporation (or condensation) discussed below. Deep samples are better for long-term surface 770 dependent proxies as higher frequency temperature harmonics are essentially damped out. 771 Equations 1 and 2, as visualised in Figure 9 and Table 5, can serve as a guide for targeted 772 speleothem sampling.

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4.2.3. Cave venting and evaporation

774 As a further point of discussion we illustrate that cave venting, besides influencing pCO_2 775 [Spotl et al., 2005; Baldini et al., 2008; Kowalczk and Froelich, 2010], can alter cave drip 776 water temperature and consequently influence speleothem growth. In fact, Casteel and 777 Banner [2014] illustrate that seasonal temperature variations control calcite growth rates and 778 trace element ratios. We emphasise that significant and frequent in-cave evaporation and drip 779 water cooling is to be expected for near-entrance parts of caves that are located in present (or 780 past) low humidity environments. Figure 10 summarises the evaporative cooling potential at 781 Cathedral Cave. While there is a weak correlation between drip water cooling and RH the 782 data exhibits significant scattering which indicates that additional parameters affect the 783 cooling, e.g. flow path, drip rate and air circulation. We observed up to -1.8 °C at a RH of < 784 95 % for drip water that is exposed to the cave air. Unravelling the dependency of drip water 785 evaporative cooling on venting clearly requires further research.

While we illustrate that evaporative drip water cooling is caused by regular ingress of dry air during summer (Figure 7A-C), in-cave evaporation also occurs during winter as the outside air is permanently denser (Figure 7D-F). Our results prove that Cathedral Cave is well vented near the entrance despite the lack of discernible air movement. Results also indicate that moisture escapes from even the deepest parts of the cave (RH < 100 %, evaporation rate > 0 mm) but measurable influences on the drip water temperature were not detected.

It is well accepted that venting influences geochemical signatures [Spotl et al., 2005; Baldini et al., 2008]. We point out that evaporation leads to isotopic enrichment of drip water [Cuthbert et al., 2014b; Markowska et al., submitted], and that evaporative drip water cooling could significantly influence chemical/isotopic signatures in speleothems [Kim and O'Neil, 1997]. This may be a further complication in reconciling clumped isotope thermometry Δ_{47} based temperature proxies in speleothems with mean air temperature, as Δ_{47} will be affected by the temperature of the water film from which the carbonate is precipitated [Affek et al., 2008].

800 Our results are consistent with Perrier et al. [2010] in that ventilation related effects, such as 801 evaporation and associated cave rock and drip water temperature anomalies, are damped with 802 increasing distance from the entrance. However, the magnitude of venting will strongly 803 depend on the cave geomorphology [De Freitas et al., 1982]. In fact considerable air flow has 804 been reported within caves [Conn, 1966; McLean, 1971; Cigna and Forti, 1986], in particular 805 when multiple entries located at different vertical elevations are present [Faimon et al., 2012; 806 Gregoric et al., 2013]. Figure 10 presents the first quantification of the effects of evaporative 807 cooling of cave drip water. Our data is just from two drip sites in one cave, and further 808 empirical field data is needed to develop a predictive model of factors determining the extent 809 of evaporative cooling. However, the implications for speleothem temperature proxies are 810 clear – in ventilated caves, researchers should consider the possibility that the speleothem 811 proxy temperature is systematically cooler than the external mean air temperature.

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4.2.4. Considerations for the type of speleothem to be sampled

A question arises as a result of the above discussion: What type of speleothem should be 813 814 sampled to best constrain the drip water temperature? Site 1 has a stalagmite fed from a 815 flowstone with a relatively long path (~ 3 m) where the water is exposed to the cave 816 atmosphere via film flow. While we expect this type of speleothem would have a large 817 potential for thermal disequilibrium affecting temperature proxies, it could still be a good 818 source for soil or vegetation derived signals (i.e., pollen). A stalagmite fed by a regular 819 conical-shaped stalactite will have drip water flowing along the outside of the deposit. This 820 type of speleothem would be cooled during periods when the drip rate is slow and regular 821 [Cuthbert et al., 2014a] which may imprint on the geochemical proxy and make interpretation 822 difficult. We believe that the best stalagmite (likely a candlestick shape) for sampling is fed 823 by a soda-straw stalactite because the flow path to the drip is surrounded by (thin) calcite and 824 the water is therefore less exposed to the cave atmosphere and potential evaporative cooling. 825 However, confirming this requires further research.

826

4.2.5. Summary

The implications of our results for speleothem paleoclimate reconstruction can be summarised as follows: The location that the proxy-derived temperature signal is representative for (i.e., surface or drip source) and the processes that could influence the signal must be carefully considered. Depending on the requirements, Equation 1 offers a quantitative model to convolve or deconvolve the "background" temperature signal between surface and drip source onto which in-cave signals will be superimposed.

The damping of surface temperature variations in the soil/epikarst is a function of subsurface depth and frequency (Figure 9). If a surface temperature signal is required as a paleoclimate proxy (i.e., a decadal-scale temperature signal) a near-surface chamber, again with minimum venting and maximum relative humidity, should fulfil the conditions for sampling.

Figure 9 illustrates the importance of considering the subsurface depth when speleothems are sampled for the purpose of accurately unravelling the surface temperature signal from isotope proxies. For example, highest amplitudes for the surface temperature during glacial-interglacial climate transitions and for the variability over the last 10,000 years are 5 °C and 0.5-1 °C, respectively [Cheng et al., 2009]. A rough guide for selecting appropriate sampling depths where the desired signal can be resolved is given in Table 5.

We stress that, consistent with the results of Cuthbert et al. [2014a], frequent evaporative 845 • 846 cooling events are to be expected in caves that could have been ventilated or exposed to 847 evaporation (RH < 100 %). Evaporative cooling can lower the drip water temperature 848 compared to cave air/speleothem temperature. The best cave locations to minimise this 849 effect are those with a long-term RH of 100 % and no air flow. These criteria were set out in the 1960s to determine where to best sample speleothems for temperature records from 850 ¹⁸O [Hendy, 1971]. Here we show that, while the premise was correct, correction of the 851 temperature signal should be considered. The influence could be assessed by checking for 852 853 a difference in air and drip water temperature.

The best speleothems to sample and analyse to obtain paleoclimate records of surface air temperature changes are minimum diameter stalagmites that are supplied by soda-straw stalactites. While the speleothem-water contact is maximised over water-air contact, the drip rates for these specimens are likely to be slow and evaporation could still occur, and therefore caves of RH of 100% and no air flow would provide ideal sampling locations.

28

859 **5.** Conclusion

Cave drip water temperature is controlled by multiple heat transfer mechanisms acting 860 861 simultaneously during the movement of water through soil and bedrock and as film flow over 862 speleothem surfaces, i.e. conduction, convection and latent heat and mass exchange. The two 863 main heat sources/sinks are: 1) conduction of the dynamic surface temperature signal 864 vertically into the subsurface, 2) the cave atmosphere as is coupled to the surface atmosphere 865 by different venting mechanisms. The relative importance of each mechanism depends on the 866 thickness of the overburden, the distance of film flow between entering the cave and the 867 arriving at the drip source, and the advective velocity of the water film which is proportional 868 to the drip rate.

869 While cave air temperatures have been measured and analysed in detail, there is a general 870 lack of data and understanding relating to controls of cave drip water temperature. We 871 deployed multiple specialised high-resolution sensors along an in-cave flow path and drip 872 source to measure the evolution of the speleothem/water temperature. In-cave dripping was 873 induced through manual surface irrigation experiments with cooled water and deuterium as a 874 conservative tracer. In combination with measurements of drip rates, surface and cave 875 climate, in-cave evaporation rates and deuterium concentrations we identified and analysed, 876 for the first time, the heat transfer processes that exert control on the cave drip water 877 temperature between surface and drip source.

878 Temperature harmonics contained in the surface temperature signal propagate conductively 879 into the subsurface and undergo frequency dependent exponential amplitude damping and 880 linear phase shifting with subsurface depth. For example, we observed that there is a clear 881 exponential temperature-depth gradient induced by the annual surface temperature harmonic 882 which controls the drip water temperature ("background" temperature). Film flow along the 883 speleothem surface can convectively carry this signal down along the flow path causing 884 temperature anomalies that depend on the film advection rate (which is proportional to the 885 drip rate). However, this convective temperature anomaly is damped ("smeared") by 886 conduction back into the speleothem along the flow path depending on the temperature-depth 887 gradient at the time.

At the same time the water film is exposed to the cave air which can significantly change drip water temperature through convection/conduction or latent heat and mass exchange, with magnitudes that depend on the distance from the cave entrance. The influence on the water

29

temperature, however, depends on the film advection rate and the complex coupling between surface and cave climate through venting (i.e. air exchange induced by a density difference between surface and cave air). We observed regular evaporative drip water cooling events of -1.5 °C and up to -2.5 °C during summer when denser low-RH air enters the cave. Further, the drip water temperature can also fluctuate due to air-induced convection/conduction in winter when surface air is continuously denser (constant venting).

897 Drip water temperature is a key parameter controlling many biogeochemical in-cave 898 processes that must be quantified when the paleoclimate is reconstructed from speleothem-899 based archives. We advise how the drip water "background" temperature can be modelled 900 using simple analytical solutions of the differential heat conduction equation. We show how a 901 data supported conceptual model for cave drip water temperature can assist with constraining 902 a range of temperature sensitive biogeochemical speleothem processes. Further, we offer 903 guidance on the type and location of speleothems that are sampled for paleoclimate signals 904 with the intent to either maximise or minimise the drip water temperature signature. We 905 anticipate that our findings will lead to significant improvements in the understanding of 906 climate signals from speleothem based paleoclimate archives.

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- 1102

1103 **Figure captions:**

Figure 1: Survey map of Cathedral Cave located in the Wellington Caves Reserve in NSW,Australia. Instrumented sites are marked with red on the map.

1106

Figure 2: A) Schematic subsurface cross-section of Site A (Figure 1) showing the drip water flow path along a flow stone to the stalactite (drip site) and the sensors deployed to measure water film and drip temperature and cave air temperature as well as climate parameters (humidity and pressure). B) A StarOddi micro T temperature sensor measuring at the drip source. C) Example of high-precision aluminium temperature sensor mounted on flow stone along the flow path (Australian 1-dollar coin with 25 mm diameter for scale).

1113

1114 Figure 3: Data from two years of monitoring at the Cathedral Cave: Surface air temperature, 1115 daily precipitation, and cave air temperature (measured at Site A1 Figure 1). For air 1116 temperatures, best fit to Equation 1 is indicated by dashed black lines. Blue lines are the drip 1117 water temperatures measured at Site A. Vertical dark grey bars show the times at which 1118 surface irrigation experiments were conducted coinciding with intense data collection 1119 periods. The light grey background indicates the times at which longer-term cave flowstone 1120 and drip water temperature was measured. The blue lines are speleothem and drip water 1121 temperature measurements enlarged in Figure 6 and explained later.

1122

1123 Figure 4: Drip monitoring with high time-resolution at site A during summer 2014. A) 1124 Temperature measured along a drip water flow path (for locations see Fig. 2a) on top of the 1125 flowstone (blue), at ~40 mm depth into the flowstone (red) and in the air (green). Surface air 1126 temperature is also plotted (grey). B) Drip rate and relative humidity. A total of 2 irrigations 1127 were conducted (vertical black lines indicating equivalent rainfall) with 3400 L and 2400 L 1128 applied to the surface irrigation patch. Parts of this data were previously published in 1129 Cuthbert et al. [2014a] to demonstrate evaporative cooling of speleothems. Light grey shaded 1130 areas indicate periods dominated by evaporative cooling. Dark grey shaded areas depict 1131 periods dominated by film convection.

1132

1133 Figure 5: Drip monitoring with high time-resolution at site A during summer 2013. A total of 4 irrigations were conducted with rainfall equivalents of 35 mm and 63 mm. A) Temperature 1134 1135 measured at the tip of two neighbouring stalactites, and in the air (see Figure 2 for locations). 1136 Irrigations 1, 2 and 3 were cooled using bags with ice (irrigation water temperature is 1137 indicated next to the vertical black lines in a). B) Vertically enlarged temperature data from 1138 A. C) Deuterium measured in drip water samples during the irrigation experiment. Deuterium was added to the first irrigation (~6100 ‰ VSMOW). Min/max of the 2-year average from 1139 1140 various drip sources at site A [Markowska et al., submitted]. D) Drip rate of both stalactites. The grey arrow (A and B) depicts the time when the surface air temperature was lower than 1141 1142 the cave air temperature indicating cave venting. The blue arrow (B) shows the time at which 1143 the cooled surface irrigation caused a drip water temperature anomaly. Light grey shaded 1144 areas indicate periods of evaporative cooling. Dark grey shaded areas depict periods of film 1145 convection.

1146

Figure 6: Temperatures measured at Site A on the flowstone surface where film flow occurred during times at which the drip source is active. Locations of the records are marked according to Figure 2. Data framed by grey vertical bars are highlighted in Figure 7. The highlighted winter dataset coincides with the surface irrigation experiment 3 (see also Figure 3).

1152

Figure 7: Summer (A-C) and winter (D-F) snapshots of dry/wet speleothem and cave air temperature (A and D), cave climate (B and E), surface and cave air density (C and F). Note that the winter dataset (D-F) shows the response to surface irrigation experiment 3 (see Figure 3). Note that y-axes of subplot B, C, E and F have the same range for better signal comparison.

1158

Figure 8: Conceptual model of the different controls on cave drip water temperature between surface and drip source. Individual heat and mass transfer mechanisms are depicted by arrows and described as follows: $q_{c,atm}$ is conduction between surface and subsurface, $q_{f,surf}$ is convection between surface and subsurface, $q_{f,atm}$ is convection between surface and cave air, m_{atm} is moisture exchange between surface and cave air, $q_{c,rock}$ is conduction between speleothem and water film, $q_{c,air}$ is conduction between water film and air, $q_{l,air}$ is latent heat 1165 exchange between water film and air, m_{air} is moisture exchange between water film and air, 1166 q_f is convection of the water film, v_f is advection of the water film, L is the film flow 1167 distance between water entering the cave and drip source, b is the thickness of the water 1168 film.

1169

Figure 9: Depth penetration of surface temperature components based on Equations 1-2 and thermal parameters in Table 2 with selected frequencies (daily, annual, decadal, centennial and millennial): A) amplitude damping, B) phase shift. The grey bands enveloping the curves reflect the variability arising from min/max thermal parameters reported in the literature. The red dots illustrate practical examples given in the discussion.

1175

Figure 10: The evaporative cooling potential: Difference between cave air and drip water
temperature plotted against RH. Site A: ~2 months of summer data (Figure 6). Site B: Data
from the irrigation experiment 2 (Figure 3B in Cuthbert et al. [2014a]). Site C: ~4 months of

1179 measurements.

1180 Table captions:

Table 1: Detailed summary of the individual surface irrigations conducted at 3 different timesover a two year period between 2013 and 2014.

1183

Table 2: Summary of thermal parameters of water, air, soil and limestone: ¹Water and air properties can be found in NIST [2014]. ²Soil and limestone properties were measured in the laboratory using samples collected in the field. ³Ranges for soil thermal parameters and limestone bedrock are from Ochsner et al. [2001] and Vosteen and Schellschmidt [2003].

1188

1189 Table 3: Cave evaporation rates measured at different locations and opposing seasons.

1190

1191 Table 4: Summary of results obtained by analysing temperature data from different locations

1192 with Equation 1 using an annual signal period (P = 365.25 days), soil zone thickness d = 0.1

1193 m (except for surface air temperature), soil and limestone thermal diffusivity listed in Table

1194 2. Phase offset is relative to summer solstice. The fitting algorithm minimised the NRMSE by

1195 varying the bold parameters.

1196

1197 Table 5: Max/min damping factors (ratio between subsurface and surface amplitude) and

signal shifts for distinct depths and different harmonic signals extracted from Figure 9.

Date	Experiment / application	Water volume [litres]	Equiv. rain [mm]	Duration of irrigation [hours]	Equiv. rainfall intensity [mm/h]	Irrigation water temperature [°C]
8/01/2013	1/1	840	~35	1.75	~20	0.3
9/01/2013	1/2	1500	~63	1.75	~35	10.6
10/01/2013	1/3	840	~35	1.75	~20	0.3
11/01/2013	1/4	1500	~63	1.75	~35	24.2
14/01/2014	2/1	3400	~68	2.85	~24	~25
15/01/2014	2/2	2400	~48	3.00	~16	~25
22/07/2014	3/1	1460	~29	1.00	~29	~12
23/07/2014	3/2	745	~15	0.50	~30	~12
24/07/2014	3/3	1460	~29	1.00	~29	~12

Table 1: Detailed summary of the individual surface irrigations conducted at 3 different times over a two year period between 2013 and 2014.

	Thermal	Specific heat	Thermal	Min. thermal	Max. thermal
Material	conductivity	capacity	diffusivity	diffusivity	diffusivity
	[W/m/K]	$[MJ/m^3/K]$	$[m^2/d]$	$[m^2/d]$	$[m^2/d]$
Water @ 18 °C	0.595 ¹	4.180 ¹	0.01231	-	-
Air @ 18 °C	0.025 ¹	0.001 ¹	1.8014 ¹	-	-
Soil (dry)	0.545 ²	1.188 ²	0.0396 ²	-	-
Soil (saturated)	0.835 ²	2.939 ²	0.0245 ²	-	-
Soil	-	-	0.03 ³	0.01 ³	0.06 ³
Limestone	2.356 ²	2.518 ²	0.0808^2	0.06 ³	0.14 ³

Table 2: Summary of thermal parameters of water, air, soil and limestone: ¹Water and air properties can be found in NIST [2014]. ²Soil and limestone properties were measured in the laboratory using samples collected in the field. ³Ranges for soil thermal parameters and limestone bedrock are from Ochsner et al. [2001] and Vosteen and Schellschmidt [2003].

Evaporation rate [mm/year]							
Location	Summer (January 2014)	Winter (July 2014)					
Near entrance	440						
Site A	50	>56					
Site B	40						
Site C	13	4.8					

Table 3: Cave evaporation rates measured at different locations and opposing seasons.

Temperature	Mean	Amplitude	Phase	Phase	Total	NRMSE	Number
measurement				offset	depth		of data
location							points
Parameter [unit]	<i>T</i> ₀ [°C]	A [°C]	[d]	θ [d]	z [m]	[-]	[-]
Surface air	16.90	8.51	0	20.0	0	0.1827	52,376
Flowstone	17.18	5.03	30.7	31.9	1.55	0.2579	30.810
(b, Site A)							
Flowstone	16.62	4.11	42.4	31.8	2.16	0.2341	30,811
(c, Site A)							,
Stalactite	16.11	2.61	68.8	31.2	3.55	0.1653	30,814
(J, Site A)							
Cave air	16.32	2.38	74.1	31.4	3.83	0.3609	23,750
(Site A)							
Cave air	15.70	1.65	95.6	31.1	4.95	0.3400	29,338
(Site A1)							
Cave air	18.10	-	-	-	~25	-	17,959
(Site C)							

Table 4: Summary of results obtained by analysing temperature data from different locations with Equation 1 using an annual signal period (P = 365.25 days), soil zone thickness d = 0.1 m (except for surface air temperature), soil and limestone thermal diffusivity listed in Table 2. Phase offset is relative to summer solstice. The fitting algorithm minimised the NRMSE by varying the bold parameters.

Harmonic		da	daily		annual		decadal		centennial		millennial	
Dept h [m]		Min	Max	Min	Max	Min	Max	Min	Max	Min	Max	
0.1	Amp [-]	0.17	0.49	0.91	0.96	0.97	0.99	0.99	1.00	1.00	1.00	
	Phase [months]	0.0	0.0	0.1	0.1	0.3	0.4	0.9	1.3	2.8	4.0	
1	Amp [-]	0	0.01	0.65	0.77	0.87	0.92	0.96	0.97	0.99	1.00	
	Phase [months]	-	0	0.5	0.8	1.6	2.5	5.1	7.8	16.3	24.5	
10	Amp [-]	0	0	0.02	0.08	0.30	0.46	0.68	0.78	0.89	0.92	
	Phase [months]	-	-	4.8	7.3	15.1	23.0	47.8	72.8	151.0	230.3	
100	Amp [-]	0	0	0	0	0	0	0.02	0.08	0.30	0.46	
	Phase [months]	-	-	-	-	-	-	473	723	1498	2288	

 Table 5: Max/min damping factors (ratio between subsurface and surface amplitude) and signal shifts for distinct depths and different harmonic signals extracted from Figure 9.























